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# (12) United States Patent

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# (54) SUBSTITUTED PYRIDO[1,2-A]PYRAZINES AND SUBSTITUTED PYRIDO[1,2-A][1,4]DIAZEPINES FOR THE TREATMENT OF (INTER ALIA) ALZHEIMER'S DISEASE

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#### (57) ABSTRACT

The present invention is concerned with novel substituted 3,4-dihydro-2H-pyrido[1,2- a]pyrazine-1,6-dione derivatives of Formula (I) wherein  $R^1$ ,  $R^2$ ,  $R^3$ ,  $R^4$ ,  $R^5$ , Z and X have the meaning defined in the claims. The compounds according to the present invention are useful as gamma secretase modulators. The invention further relates to processes for preparing such novel compounds, pharmaceutical compositions comprising said compounds as an active ingredient as well as the use of said compounds as a medicament.

#### 14 Claims, No Drawings

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# SUBSTITUTED PYRIDO[1,2-A]PYRAZINES AND SUBSTITUTED PYRIDO[1,2-A][1,4]DIAZEPINES FOR THE TREATMENT OF (INTER ALIA) ALZHEIMER'S DISEASE

# CROSS REFERENCE TO RELATED APPLICATIONS

This application is the National Stage of International <sup>10</sup> Application No. PCT/IB2013/054014, filed May 16, 2013, which claims the benefit of European Patent Application No. 12168186.0, filed May 16, 2012; European Patent Application No. 12195875.5, filed Dec. 6, 2012; European Patent Application No. 12197010.7, filed Dec. 13, 2012; and European Patent Application No. 13159178.6, filed Mar. 14, 2013, the entire disclosures of each of which are incorporated herein by reference.

#### FIELD OF THE INVENTION

The present invention is concerned with novel substituted 3,4-dihydro-2H-pyrido[1,2-  $\alpha$ ]pyrazine-1,6-dione derivatives useful as gamma secretase modulators. The invention further relates to processes for preparing such novel compounds, pharmaceutical compositions comprising said compounds as an active ingredient as well as the use of said compounds as a medicament.

#### BACKGROUND OF THE INVENTION

Alzheimer's Disease (AD) is a progressive neurodegenerative disorder marked by loss of memory, cognition, and behavioral stability. AD afflicts 6-10% of the population over age 65 and up to 50% over age 85. It is the leading cause of 35 dementia and the third leading cause of death after cardiovascular disease and cancer. There is currently no effective treatment for AD. The total net cost related to AD in the U.S. exceeds \$100 billion annually.

AD does not have a simple etiology, however, it has been 40 associated with certain risk factors including (1) age, (2) family history and (3) head trauma; other factors include environmental toxins and low levels of education. Specific neuropathological lesions in the limbic and cerebral cortices include intracellular neurofibrillary tangles consisting of 45 hyperphosphorylated tau protein and the extracellular deposition of fibrillar aggregates of amyloid beta peptides (amyloid plaques). The major components of amyloid plaques are the amyloid beta (A-beta, Abeta or A $\beta$ ) peptides of various lengths. A variant thereof, which is the A $\beta$ 1-42-peptide (Abeta-42), is believed to be the major causative agent for amyloid formation. Another variant is the A $\beta$ 1-40-peptide (Abeta-40). A $\beta$  is the proteolytic product of a precursor protein, beta amyloid precursor protein (beta-APP or APP).

Familial, early onset autosomal dominant forms of AD 55 have been linked to missense mutations in the  $\beta$ -amyloid precursor protein ( $\beta$ -APP or APP) and in the presenilin proteins 1 and 2. In some patients, late onset forms of AD have been correlated with a specific allele of the apolipoprotein E (ApoE) gene, and, more recently, the finding of a mutation in 60 alpha2-macroglobulin, which may be linked to at least 30% of the AD population. Despite this heterogeneity, all forms of AD exhibit similar pathological findings. Genetic analysis has provided the best clues for a logical therapeutic approach to AD. All mutations found to date, affect the quantitative or qualitative production of the amyloidogenic peptides known as Abeta-peptides (A $\beta$ ), specifically A $\beta$ 42, and have given

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strong support to the "amyloid cascade hypothesis" of AD (Tanzi and Bertram, 2005, Cell 120, 545). The likely link between  $A\beta$  peptide generation and AD pathology emphasizes the need for a better understanding of the mechanisms of  $A\beta$  production and strongly warrants a therapeutic approach at modulating  $A\beta$  levels.

The release of Aβ peptides is modulated by at least two proteolytic activities referred to as β- and γ-secretase cleavage at the N-terminus (Met-Asp bond) and the C-terminus (residues 37-42) of the Aβ peptide, respectively. In the secretory pathway, there is evidence that β-secretase cleaves first, leading to the secretion of s-APPβ (sβ) and the retention of a 11 kDa membrane-bound carboxy terminal fragment (CTF). The latter is believed to give rise to Aβ peptides following cleavage by γ-secretase. The amount of the longer isoform, Aβ42, is selectively increased in patients carrying certain mutations in the region of a particular gene coding in a particular protein (presenilin), and these mutations have been correlated with early-onset familial AD. Therefore, Aβ42 is believed by many researchers to be the main culprit of the pathogenesis of AD.

It has now become clear that the  $\gamma$ -secretase activity cannot be ascribed to a single protein, but is in fact associated with an assembly of different proteins.

The gamma (γ)-secretase activity resides within a multiprotein complex containing at least four components: the presenilin (PS) heterodimer, nicastrin, aph-1 and pen-2. The PS heterodimer consists of the amino- and carboxyterminal PS fragments generated by endoproteolysis of the precursor protein. The two aspartates of the catalytic site are at the interface of this heterodimer. It has recently been suggested that nicastrin serves as a gamma-secretase-substrate receptor. The functions of the other members of gamma-secretase are unknown, but they are all required for activity (Steiner, 2004. 55 Curr. Alzheimer Research 1(3): 175-181).

Thus, although the molecular mechanism of the second cleavage-step has remained elusive until now, the  $\gamma$ -secretase-complex has become one of the prime targets in the search for compounds for the treatment of AD.

Various strategies have been proposed for targeting  $\gamma$ -secretase in AD, ranging from targeting the catalytic site directly, developing substrate-specific inhibitors and modulators of  $\gamma$ -secretase activity (Marjaux et al., 2004. Drug Discovery Today: Therapeutic Strategies, Volume 1, 1-6). Accordingly, a variety of compounds were described that have secretases as targets (Lamer, 2004. Secretases as therapeutics targets in AD: patents 2000-2004. Expert Opin. Ther. Patents 14, 1403-1420).

Indeed, this finding was supported by biochemical studies in which an effect of certain Non-Steroidal Anti-Inflammatory Drugs (NSAIDs) on  $\gamma$ -secretase was shown (US 2002/0128319; Eriksen (2003) J. Clin. Invest. 112, 440). Potential limitations for the use of NSAIDs to prevent or treat AD are their inhibition activity of cyclooxygenase (COX) enzymes, which can lead to unwanted side effects, and their low CNS penetration (Peretto et al., 2005, J. Med. Chem. 48, 5705-5720). More recently the NSAID R-flurbiprofen, an enantiomer lacking Cox-inhibitory activity and related gastric toxicity, has failed in large phase III trial since the drug did not improve thinking ability or the ability of patients to carry out daily activities significantly more than those patients on placebo.

WO-2010/100606 discloses phenyl imidazoles and phenyl triazoles for use as gamma-secretase modulators.

US20090062529 relates to polycyclic compounds effective as the rapeutic or prophylactic agents for a disease caused by  $A\beta$ .

WO-2010/070008 is concerned with novel substituted bicyclic imidazole derivatives useful as  $\gamma$ -secretase modulators

WO-2010/089292 is concerned with novel substituted bicyclic heterocyclic compounds useful as  $\gamma$ -secretase modulators.

WO-2011/006903 is concerned with novel substituted triazole and imidazole derivatives useful as  $\gamma$ -secretase modulators.

WO-2012/131539 relates to novel bicyclic pyridinones useful as brain-penetrable  $\gamma$ -secretase modulators.

There is a strong need for novel compounds which modulate  $\gamma$ -secretase activity thereby opening new avenues for the treatment of AD. It is an object of the present invention to overcome or ameliorate at least one of the disadvantages of the prior art, or to provide a useful alternative. The compounds of the present invention or part of the compounds of the present invention may have improved metabolic stability properties, improved central brain availability, improved solubilities, or reduced CYP inhibition compared with the compounds disclosed in the prior art. It is accordingly an object of the present invention to provide such novel compounds.

# SUMMARY OF THE INVENTION

It has been found that the compounds of the present invention are useful as  $\gamma\text{-secretase}$  modulators. The compounds according to the invention and the pharmaceutically acceptable compositions thereof, may be useful in the treatment or prevention of AD.

The present invention concerns novel compounds of Formula (I):

$$R^1$$
 $L$ 
 $R^2$ 
 $Z$ 
 $N$ 
 $R^3$ 
 $N$ 
 $X$ 
 $R^4$ 

tautomers and stereoisomeric forms thereof, wherein  $R^1$  is hydrogen,  $C_{1-4}$ alkyl or Ar;

provided however that R<sup>1</sup> is C<sub>1-4</sub>alkyl or Ar when L is a covalent bond;

 $R^2$  is hydrogen, phenyl, cyclo $C_{3\_7}$ alkyl, tetrahydro-2H-pyranyl, tetrahydro-2H-thiopyranyl, piperidinyl, or  $C_{1\_4}$ alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, hydroxyl,  $C_{1\_4}$ alkyloxy and  $NR^7R^8;$ 

Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2-ethanediyl is optionally substituted with one or two C<sub>1-4</sub>alkyl substituents;

L is a covalent bond, — $(CH_2)_{m-n}$ —Y— $(CH_2)_n$ —, 1,2-cyclopropanediyl,

 ${
m C_{1-6}}$ alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  ${
m C_{1-4}}$ alkyloxy ${
m C_{1-6}}$ alkanediyl wherein two geminal hydrogen atoms are replaced by  ${
m C_{2-6}}$ alkanediyl;

m represents 3, 4, 5, 6 or 7; n represents 1, 2 or 3; 4

Y is O or NH;

Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>,

C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and

C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl;

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C<sub>1-4</sub>alkyl optionally substituted with one or more halo substituents;

 $R^3$  is hydrogen, halo or  $C_{1-4}$ alkyl;

R<sup>4</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

R<sup>5</sup> is hydrogen or C<sub>1-4</sub>alkyl;

X is CR<sup>6</sup> or N;

(I)

 $^{25}$  R<sup>6</sup> is hydrogen or C<sub>1-4</sub>alkyl;

R<sup>7</sup> is hydrogen or C<sub>1-4</sub>alkyl;

R<sup>8</sup> is hydrogen or C<sub>1-4</sub>alkyl;

R° is a ring system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3-7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or C<sub>1-4</sub>alkyl optionally substituted with one or more halo atoms;

and the pharmaceutically acceptable addition salts, and the solvates thereof.

The present invention also concerns methods for the preparation of compounds of Formula (I) and pharmaceutical compositions comprising them.

The present compounds were found to modulate the  $\gamma$ -secretase activity in vitro and in vivo, and therefore may be useful in the treatment or prevention of AD, traumatic brain injury (TBI), dementia pugilistica, mild cognitive impairment (MCI), senility, dementia, dementia with Lewy bodies, cerebral amyloid angiopathy, multi-infarct dementia, Down's syndrome, dementia associated with Parkinson's disease and dementia associated with beta-amyloid; preferably AD and other disorders with Beta-amyloid pathology (e.g. glaucoma).

In view of the aforementioned pharmacology of the compounds of Formula (I), it follows that they may be suitable for use as a medicament.

More especially the compounds may be suitable in the treatment or prevention of AD, cerebral amyloid angiopathy, multi-infarct dementia, dementia pugilistica and Down syndrome.

The present invention also concerns the use of compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof, and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for the manufacture of a medicament for the modulation of  $\gamma$ -secretase activity.

The present invention will now be further described. In the following passages, different aspects of the invention are defined in more detail. Each aspect so defined may be combined with any other aspect or aspects unless clearly indicated to the contrary. In particular, any feature indicated as being

preferred or advantageous may be combined with any other feature or features indicated as being preferred or advantageous.

# DETAILED DESCRIPTION

When describing the compounds of the invention, the terms used are to be construed in accordance with the following definitions, unless a context dictates otherwise.

Whenever a radical or group is defined as "optionally sub- 10 stituted" in the present invention, it is meant that said radical or group is unsubstituted or is substituted. For instance, when Ar is defined as "Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-ben-15 zodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>, C<sub>1-4</sub>alkyl optionally substituted with one or more 20 substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and C<sub>1.4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and  $cycloC_{3-7}$ alkyl;" it is meant that:

"Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl; or

Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl; wherein said ring system is substituted with 35 one or more substituents each independently selected from the group consisting of halo,  $Ar^2$ ,  $R^0$ ,  $C_{1-4}$ alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo,  $C_{1-4}$ alkyloxy and cyclo $C_{3-7}$  alkyl, and  $C_{1-4}$  alkyloxy optionally substituted with 40 one or more substituents each independently selected from the group consisting of halo and cyclo $C_{3-7}$  alkyl".

Whenever the term "substituted" is used in the present invention, it is meant, unless otherwise is indicated or is clear from the context, to indicate that one or more hydrogens, in 45 R particular from 1 to 4 hydrogens, preferably from 1 to 3 hydrogens, more preferably 1 hydrogen, on the atom or radical indicated in the expression using "substituted" are replaced with a selection from the indicated group, provided that the normal valency is not exceeded, and that the substitution results in a chemically stable compound, i.e. a compound that is sufficiently robust to survive isolation to a useful degree of purity from a reaction mixture, and formulation into a therapeutic agent.

The term "halo" as a group or part of a group is generic for 55 fluoro, chloro, bromo, iodo unless otherwise is indicated or is clear from the context.

The term " $C_{1.4}$ alkyl" as a group or part of a group refers to a hydrocarbyl radical of Formula  $C_nH_{2n+1}$  wherein n is a number ranging from 1 to 4.  $C_{1.4}$ alkyl groups comprise from 60 1 to 4 carbon atoms, preferably from 1 to 3 carbon atoms, more preferably 1 to 2 carbon atoms.  $C_{1.4}$ alkyl groups may be linear or branched and may be substituted as indicated herein. When a subscript is used herein following a carbon atom, the subscript refers to the number of carbon atoms that the named 65 group may contain.  $C_{1.4}$ alkyl includes all linear, or branched alkyl groups with between 1 and 4 carbon atoms, and thus

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includes such as for example methyl, ethyl, n-propyl, i-propyl, 2-methyl-ethyl, butyl and its isomers (e.g. n-butyl, isobutyl and tert-butyl), and the like.

The term " $C_{1..4}$ alkyloxy" as a group or part of a group refers to a radical having the Formula  $OR^b$  wherein  $R^b$  is  $C_{1..4}$ alkyl. Non-limiting examples of suitable  $C_{1..4}$ alkyloxy include methyloxy (also methoxy), ethyloxy (also ethoxy), propyloxy, isopropyloxy, butyloxy, isobutyloxy, sec-butyloxy and tert-butyloxy.

The term "cycloC $_{3-7}$ alkyl" alone or in combination, refers to a cyclic saturated hydrocarbon radical having from 3 to 7 carbon atoms. Non-limiting examples of suitable cycloC $_{3-7}$ alkyl include cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and cycloheptyl.

The term " $C_{1-6}$ alkanediyl" as a group or part of a group defines bivalent straight or branched chained saturated hydrocarbon radicals having from 1 to 6 carbon atoms such as, for example, methylene or methanediyl, ethan-1,2-diyl, ethan-1, 1-diyl or ethylidene, propan-1,3-diyl, propan-1,2-diyl, butan-1,4-diyl, pentan-1,5-diyl, pentan-1,6-diyl, 2-methylbutan-1,4-diyl, 3-methylpentan-1,5-diyl and the like.

The term " $C_{2-6}$ alkanediyl" as a group or part of a group defines bivalent straight or branched chained saturated hydrocarbon radicals having from 2 to 6 carbon atoms such as, for example, ethan-1,2-diyl, ethan-1,1-diyl or ethylidene, propan-1,3-diyl, propan-1,2-diyl, butan-1,4-diyl, pentan-1,5-diyl, pentan-1,1-diyl, hexan-1,6-diyl, 2-methylbutan-1,4-diyl, 3-methylpentan-1,5-diyl and the like; in particular " $C_{2-6}$ alkanediyl" as a group or part of a group defines ethan-1,2-diyl.

Similarly, the term " $C_{1-3}$ alkanediyl" as a group or part of a group defines bivalent straight or branched chained saturated hydrocarbon radicals having from 1 to 3 carbon atoms.

Whenever variable 1' represents  $-(CH_2)_{m-n}$  Y  $-(CH_2)_n$ , it is intended that  $-(CH_2)_{m-n}$  is attached to 'R¹' and  $-(CH_2)_n$  is attached via the nitrogen atom to the remainder of the molecule. This is illustrated by formula (I'):

$$\mathbb{R}^{1 \longrightarrow (CH_{2})_{m - n}} Y \xrightarrow{(CH_{2})_{n}} \mathbb{R}^{2} \xrightarrow{\mathbb{R}^{3}} \mathbb{R}^{3}$$

$$\mathbb{R}^{2} \longrightarrow \mathbb{R}^{3} \longrightarrow \mathbb{R}^{4}.$$

The chemical names of the compounds of the present invention were generated according to the nomenclature rules agreed upon by the Chemical Abstracts Service, using Advanced Chemical Development, Inc., nomenclature software (ACD/Labs Release 12.00 Product version 12.01; Build 33104, 27 May 2009). In case of tautomeric forms, the name of the depicted tautomeric form was generated. It should be clear that the other non-depicted tautomeric form is also included within the scope of the present invention.

Hereinbefore and hereinafter, the term "compound of formula (I)" is meant to include the tautomers and stereoisomeric forms thereof, and the pharmaceutically acceptable addition salts, and the solvates thereof.

The terms "stereoisomers", "stereoisomeric forms" or "stereochemically isomeric forms" hereinbefore or hereinafter are used interchangeably.

The invention includes all stereoisomers of the compound of Formula (I) and tautomers thereof, either as a pure stereoisomer or as a mixture of two or more stereoisomers. Enantiomers are stereoisomers that are non-superimposable mirror images of each other. A 1:1 mixture of a pair of enantiomers is a racemate or racemic mixture. Diastereomers (or diastereoisomers) are stereoisomers that are not enantiomers, i.e. they are not related as mirror images. If a compound contains a double bond, the substituents may be in the E or the Z configuration. Substituents on bivalent cyclic (partially) saturated radicals may have either the cis- or trans-configuration; for example if a compound contains a disubstituted cycloalkyl group, the substituents may be in the cis or trans configuration. Therefore, the invention includes enantiomers, diastereomers, racemates, E isomers, Z isomers, cis isomers, trans isomers and mixtures thereof, whenever chemically possible.

The absolute configuration is specified according to the Cahn-Ingold-Prelog system. The configuration at an asymmetric atom is specified by either R or S. Resolved compounds whose absolute configuration is not known can be designated by (+) or (–) depending on the direction in which they rotate plane polarized light.

When a specific stereoisomer is identified, this means that said stereoisomer is substantially free, i.e. associated with less than 50%, preferably less than 20%, more preferably less than 10%, even more preferably less than 5%, in particular less than 2% and most preferably less than 1%, of the other isomers. Thus, when a compound of formula (I) is for instance specified as (R), this means that the compound is substantially free of the (S) isomer; when a compound of formula (I) is for instance specified as E, this means that the compound is substantially free of the Z isomer; when a compound of formula (I) is for instance specified as cis, this means that the compound is substantially free of the trans isomer.

For indications of stereochemistry in compounds of formula (I) wherein L is CH(CH<sub>3</sub>) or CH(CH<sub>2</sub>CH<sub>3</sub>), the following numbering has been used to indicate the stereocenters of the diastereomers:

$$R^1$$
 $R^2$ 
 $R^3$ 
 $R^3$ 
 $R^3$ 
 $R^4$ 

 $R^x$  is methyl or ethyl

Some of the compounds according to formula (I) may also exist in their tautomeric form. Such forms although not explicitly indicated in the above formula are intended to be 55 included within the scope of the present invention.

For therapeutic use, salts of the compounds of Formula (I) are those wherein the counterion is pharmaceutically acceptable. However, salts of acids and bases which are non-pharmaceutically acceptable may also find use, for example, in the 60 preparation or purification of a pharmaceutically acceptable compound. All salts, whether pharmaceutically acceptable or not are included within the ambit of the present invention.

The pharmaceutically acceptable acid and base addition salts as mentioned hereinabove or hereinafter are meant to 65 comprise the therapeutically active non-toxic acid and base addition salt forms which the compounds of Formula (I) are

able to form. The pharmaceutically acceptable acid addition salts can conveniently be obtained by treating the base form with such appropriate acid. Appropriate acids comprise, for example, inorganic acids such as hydrohalic acids, e.g. hydrochloric or hydrobromic acid, sulfuric, nitric, phosphoric and the like acids; or organic acids such as, for example, acetic, propanoic, hydroxyacetic, lactic, pyruvic, oxalic (i.e. ethanedioic), malonic, succinic (i.e. butanedioic acid), maleic, fumaric, malic, tartaric, citric, methanesulfonic, ethanesulfonic, benzenesulfonic, p-toluenesulfonic, cyclamic, salicylic, p-aminosalicylic, pamoic and the like acids. Conversely said salt forms can be converted by treatment with an appropriate base into the free base form.

The compounds of Formula (I) containing an acidic proton may also be converted into their non-toxic metal or amine addition salt forms by treatment with appropriate organic and inorganic bases. Appropriate base salt forms comprise, for example, the ammonium salts, the alkali and earth alkaline metal salts, e.g. the lithium, sodium, potassium, magnesium, calcium salts and the like, salts with organic bases, e.g. primary, secondary and tertiary aliphatic and aromatic amines such as methylamine, ethylamine, propylamine, isopropylamine, the four butylamine isomers, dimethylamine, diethylamine, diethanolamine, dipropylamine, diisopropylamine, di-n-butylamine, pyrrolidine, piperidine, morpholine, trimethylamine, triethylamine, tripropylamine, quinuclidine, pyridine, quinoline and isoquinoline; the benzathine, N-methyl-D-glucamine, hydrabamine salts, and salts with amino acids such as, for example, arginine, lysine and the like. Conversely the salt form can be converted by treatment with acid into the free acid form.

The term solvate comprises the hydrates and solvent addition forms which the compounds of Formula (I) are able to form, as well as the salts thereof. Examples of such forms are e.g. hydrates, alcoholates and the like.

The compounds of Formula (I) as prepared in the processes described below may be synthesized in the form of mixtures of enantiomers, in particular racemic mixtures of enantiomers, that can be separated from one another following art-known resolution procedures. An manner of separating the enantiomeric forms of the compounds of Formula (I) involves liquid chromatography using a chiral stationary phase. Said pure stereochemically isomeric forms may also be derived from the corresponding pure stereochemically isomeric forms of the appropriate starting materials, provided that the reaction occurs stereospecifically. Preferably if a specific stereoisomer is desired, said compound would be synthesized by stereospecific methods of preparation. These methods will advantageously employ enantiomerically pure starting materials.

In the framework of this application, a compound according to the invention is inherently intended to comprise all isotopic combinations of its chemical elements. In the framework of this application, a chemical element, in particular when mentioned in relation to a compound according to Formula (I), comprises all isotopes and isotopic mixtures of this element. For example, when hydrogen is mentioned, it is understood to refer to <sup>1</sup>H, <sup>2</sup>H, <sup>3</sup>H and mixtures thereof.

A compound according to the invention therefore inherently comprises a compound with one or more isotopes of one or more element, and mixtures thereof, including a radioactive compound, also called radiolabelled compound, wherein one or more non-radioactive atoms has been replaced by one of its radioactive isotopes. By the term "radiolabelled compound" is meant any compound according to Formula (I) which contains at least one radioactive atom. For example, a compound can be labelled with positron or with gamma emit-

ting radioactive isotopes. For radioligand-binding techniques, the <sup>3</sup>H-atom or the <sup>125</sup>I-atom is the atom of choice to be inserted. For imaging, the most commonly used positron emitting (PET) radioactive isotopes are <sup>11</sup>C, <sup>18</sup>F, <sup>15</sup>O and <sup>13</sup>N, all of which are accelerator produced and have half-lives of 5 20, 100, 2 and 10 minutes (min) respectively. Since the halflives of these radioactive isotopes are so short, it is only feasible to use them at institutions which have an accelerator on site for their production, thus limiting their use. The most widely used of these are  $^{18}\rm{F},~^{99m}\rm{Tc},~^{201}\rm{Tl}$  and  $^{123}\rm{I}.$  The handling of these radioactive isotopes, their production, isolation and incorporation in a molecule are known to the skilled person.

In particular, the radioactive atom is selected from the  $_{15}$   $C_{1-4}$ alkyloxy optionally substituted with one or more subgroup of hydrogen, carbon, nitrogen, sulfur, oxygen and halogen. In particular, the radioactive isotope is selected from the group of <sup>3</sup>H, <sup>11</sup>C, <sup>18</sup>F, <sup>122</sup>I, <sup>123</sup>I, <sup>125</sup>I, <sup>131</sup>I, <sup>75</sup>Br, <sup>76</sup>Br, <sup>77</sup>Br and

As used in the specification and the appended claims, the 20 singular forms "a", "an," and "the" also include plural referents unless the context clearly dictates otherwise. For example, "a compound" means 1 compound or more than 1 compound.

It should be understood that the term "compounds of For- 25 mula (I)" or "a compound of Formula (I)" as used in the specification, also covers the tautomers and stereoisomeric forms thereof, and the pharmaceutically acceptable addition salts, and the solvates thereof.

The terms described above and others used in the specifi-  $^{30}$ cation are well understood to those in the art.

Preferred features of the compounds of this invention are now set forth.

In an embodiment, the present invention concerns novel 35 and the pharmaceutically acceptable addition salts, and the compounds of Formula (I):

$$R^{1}$$
 $L$ 
 $R^{2}$ 
 $Z$ 
 $N$ 
 $R^{3}$ 
 $R^{3}$ 
 $R^{4}$ 

tautomers and stereoisomeric forms thereof, wherein R<sup>1</sup> is hydrogen, C<sub>1-4</sub>alkyl or Ar;

provided however that  $R^1$  is  $C_{1-4}$ alkyl or Ar when L is a covalent bond;

R<sup>2</sup> is hydrogen, phenyl, cycloC<sub>3,7</sub>alkyl, tetrahydro-2H-pyranyl, tetrahydro-2H-thiopyranyl, piperidinyl, or C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each 55 independently selected from the group consisting of hydroxyl,  $C_{1-4}$ alkyloxy and NR<sup>7</sup>R<sup>8</sup>;

Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2ethanediyl is optionally substituted with one or two C<sub>1-4</sub>alkyl substituents;

L is a covalent bond,  $-(CH_2)_{m-n}$  -Y  $-(CH_2)_n$  -, 1,2-cyclopropanediyl,

C<sub>1-6</sub>alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and C<sub>1-4</sub>alkyloxyC<sub>1-4</sub>alkyl, or C<sub>1-6</sub>alkanediyl wherein two geminal hydrogen atoms are replaced by C<sub>2-6</sub>alkanediyl;

m represents 3, 4, 5, 6 or 7; n represents 1, 2 or 3;

Y is O or NH;

Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>,

C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and

stituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl;

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C<sub>1-4</sub>alkyl optionally substituted with one or more halo substituents;

 $R^3$  is hydrogen, halo or  $C_{1-4}$ alkyl;

R<sup>4</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

 $R^5$  is hydrogen or  $C_{1-4}$ alkyl;

X is CR<sup>6</sup> or N;

(I)

R<sup>6</sup> is hydrogen or C<sub>1-4</sub>alkyl;

R<sup>7</sup> is hydrogen or C<sub>1-4</sub>alkyl;

R<sup>8</sup> is hydrogen or C<sub>1-4</sub>alkyl;

R<sup>0</sup> is a ring system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3-7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or  $C_{1-4}$ alkyl optionally substituted with one or more halo atoms;

solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein

40  $R^1$  is hydrogen,  $C_{1-4}$ alkyl or Ar;

provided however that  $R^1$  is  $C_{1-4}$ alkyl or Ar when L is a covalent bond;

R<sup>2</sup> is hydrogen, phenyl, cycloC<sub>3-7</sub>alkyl, tetrahydro-2H-pyranyl, tetrahydro-2H-thiopyranyl, piperidinyl, or C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, hydroxyl,  $C_{1-4}$ alkyloxy and NR<sup>7</sup>R<sup>8</sup>;

Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2ethanediyl is optionally substituted with one or two C<sub>1-4</sub>alkyl substituents;

L is a covalent bond, 1,2-cyclopropanediyl,

C<sub>1-6</sub>alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and C<sub>1-4</sub>alkyloxyC<sub>1-4</sub>alkyl, or C<sub>1-6</sub>alkanediyl wherein two geminal hydrogen atoms are replaced by  $C_{2-6}$ alkanediyl;

Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>,

C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and

C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl;

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C1-4alkyl optionally substituted with one or more halo substituents;

 $R^3$  is hydrogen, halo or  $C_{1-4}$ alkyl;

R<sup>4</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

R<sup>5</sup> is hydrogen or C<sub>1-4</sub>alkyl;

X is CR<sup>6</sup> or N;

R<sup>6</sup> is hydrogen or C<sub>1-4</sub>alkyl;

 $R^7$  is hydrogen or  $C_{1-4}$ alkyl;

 $R^8$  is hydrogen or  $C_{1-4}$ alkyl;

R<sup>0</sup> is a ring system selected from the group consisting of 15 piperidinyl, morpholinyl, cycloC3-7alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or C<sub>1-4</sub>alkyl optionally substituted with one or more halo atoms:

and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein

 $R^1$  is hydrogen,  $C_{1-4}$ alkyl or Ar;

provided however that  $R^1$  is  $C_{1-4}$ alkyl or Ar when L is a covalent bond:

R<sup>2</sup> is hydrogen, phenyl, cycloC<sub>3-7</sub>alkyl, tetrahydro-2H-pyranyl, tetrahydro-2H-thiopyranyl, piperidinyl, or C<sub>1.4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, hydroxyl, C<sub>1-4</sub>alkyloxy and NR<sup>7</sup>R<sup>8</sup>;

Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2ethanediyl is optionally substituted with one or two 35 C<sub>1-4</sub>alkyl substituents;

L is a covalent bond, 1,2-cyclopropanediyl, C<sub>1-6</sub>alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  $C_{1-4}$ alkyloxy $C_{1-4}$ alkyl, or

C<sub>1-6</sub>alkanediyl wherein two geminal hydrogen atoms are replaced by  $C_{2-6}$ alkanediyl;

in particular L is  $C_{1-6}$ alkanediyl;

Ar is indolyl optionally substituted with one or more substituents each independently selected from the group consisting 45 m represents 3, 4, 5, 6 or 7; of halo, Ar<sup>2</sup>, R<sup>0</sup>,

C<sub>1,4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and

C<sub>1-4</sub>alkyloxy optionally substituted with one or more sub- 50 stituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl;

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C<sub>1-4</sub>alkyl optionally substituted with 55 one or more halo substituents;

R<sup>3</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

R<sup>4</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

 $R^5$  is hydrogen or  $C_{1-4}$ alkyl;

X is CR<sup>6</sup> or N;

 $R^6$  is hydrogen or  $C_{1-4}$ alkyl;

 $R^7$  is hydrogen or  $C_{1-4}$ alkyl;

R<sup>8</sup> is hydrogen or C<sub>1-4</sub>alkyl;

R<sup>0</sup> is a ring system selected from the group consisting of piperidinyl, morpholinyl,

cycloC<sub>3-7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more 12

substituents each independently selected from the group consisting of halo or C1.4alkyl optionally substituted with one or more halo atoms;

and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein

 $R^1$  is Ar;

10 R<sup>2</sup> is hydrogen, or methyl; in particular methyl;

Z is methylene;

L is a covalent bond or  $C_{1-6}$ alkanediyl;

Ar is indolyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar2,

C<sub>1,4</sub>alkyl optionally substituted with one or more halo substituents, and

C<sub>1.4</sub>alkyloxy optionally substituted with one or more halo substituents each independently selected from the group consisting of halo:

Ar<sup>2</sup> is phenyl;

R<sup>3</sup> is hydrogen;

R<sup>4</sup> is methyl;

R<sup>5</sup> is hydrogen;

25 X is CH;

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and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein

 $R^{1}$  is hydrogen,  $C_{1-4}$ alkyl or Ar;

provided however that R' is C<sub>1-4</sub>alkyl or Ar when L is a covalent bond;

R<sup>2</sup> is methyl;

Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2ethanediyl is optionally substituted with one or two C<sub>1-4</sub>alkyl substituents;

L is a covalent bond,  $-(CH_2)_{m-n}$  -Y  $-(CH_2)_n$  -, 1,2-cyclopropanediyl,

 $C_{1\text{--}6}$ alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and C<sub>1-4</sub>alkyloxyC<sub>1-4</sub>alkyl,

or C<sub>1-6</sub>alkanediyl wherein two geminal hydrogen atoms are replaced by C<sub>2-6</sub>alkanediyl;

n represents 1, 2 or 3; in particular n represents 1 or 2; Y is O or NH:

Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar2, R0,

C<sub>1,4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and

C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC3-7alkyl;

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C<sub>1-4</sub>alkyl optionally substituted with one or more halo substituents;

 $R^3$  is hydrogen, halo or  $C_{1-4}$ alkyl;

 $R^4$  is hydrogen, halo or  $C_{1-4}$ alkyl;

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R^{5} is hydrogen or C_{1-4}alkyl;
X is CR^{6} or N;
R^{6} is hydrogen or C_{1-4}alkyl;
R^{7} is hydrogen or C_{1-4}alkyl;
R^{8} is hydrogen or C_{1-4}alkyl;
R^{0} is a ring system selected from the group
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R<sup>0</sup> is a ring system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3-7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or C<sub>1-4</sub>alkyl 10 optionally substituted with one or more halo atoms;

and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric 15 forms thereof, wherein

 $R^1$  is hydrogen,  $C_{1-4}$ alkyl or Ar;

provided however that  $R^1$  is  $C_{1-4}$ alkyl or Ar when L is a covalent bond;

R<sup>2</sup> is methyl:

Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2-ethanediyl is optionally substituted with one or two C<sub>1,4</sub>alkyl substituents;

L is ethylidene;

m represents 3, 4, 5, 6 or 7;

n represents 1, 2 or 3; in particular n represents 1 or 2; Y is O or NH:

Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>,

 $\rm C_{1-4}$ alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo,  $\rm C_{1-4}$ alkyloxy and cyclo $\rm C_{3-7}$ alkyl, and

C<sub>1.4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group 40 consisting of halo and cycloC<sub>3.7</sub>alkyl;

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C<sub>1-4</sub>alkyl optionally substituted with one or more halo substituents;

 $R^3$  is hydrogen, halo or  $C_{1-4}$ alkyl;

 $R^4$  is hydrogen, halo or  $C_{1-4}^{1-4}$  alkyl;

 $R^5$  is hydrogen or  $C_{1-4}$ alkyl;

X is CR<sup>6</sup> or N;

 $R^6$  is hydrogen or  $C_{1-4}$ alkyl;

 $R^7$  is hydrogen or  $C_{1-4}$ alkyl;

 $R^8$  is hydrogen or  $C_{1-4}$ alkyl;

R<sup>o</sup> is a ring system selected from the group consisting of piperidinyl, morpholinyl,

cycloC<sub>3.7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said 55 ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or C<sub>1.4</sub>alkyl optionally substituted with one or more halo atoms;

and the pharmaceutically acceptable addition salts, and the 60 solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein

 $R^1$  is hydrogen,  $C_{1-4}$ alkyl or Ar;

provided however that  $R^1$  is  $C_{1-4}$ alkyl or Ar when L is a covalent bond;

R<sup>2</sup> is hydrogen, phenyl, cycloC<sub>3-7</sub>alkyl, tetrahydro-2H-pyranyl, tetrahydro-2H-thiopyranyl, piperidinyl, or C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, hydroxyl, C<sub>1-4</sub>alkyloxy and NR<sup>7</sup>R<sup>8</sup>;

Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2-ethanediyl is optionally substituted with one or two C<sub>1,4</sub>alkyl substituents;

L is a covalent bond,  $-(CH_2)_{m-n}-Y-(CH_2)_n-$ , 1,2-cyclopropanediyl,

 $C_{1-6}$ alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  $C_{1-4}$ alkyloxy $C_{1-4}$ alkyl,

or  $C_{1-6}$ alkanediyl wherein two geminal hydrogen atoms are replaced by  $C_{2-6}$ alkanediyl;

m represents 3, 4, 5, 6 or 7;

n represents 1 or 2;

Y is O or NH;

25

20 Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>,

C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and

C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl;

 ${
m Ar}^2$  is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and  ${
m C}_{1.4}$ alkyl optionally substituted with one or more halo substituents;

R<sup>3</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

R<sup>4</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

 $R^5$  is hydrogen or  $C_{1-4}$ alkyl;

X is CR<sup>6</sup> or N;

 $R_{\underline{\phantom{A}}}^{6}$  is hydrogen or  $C_{1-4}$ alkyl;

 $R^7$  is hydrogen or  $C_{1-4}$ alkyl;

 $R^8$  is hydrogen or  $C_{1-4}$ alkyl;

45 R<sup>0</sup> is a ring system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3.7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or C<sub>1.4</sub>alkyl optionally substituted with one or more halo atoms; and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein

 $R^1$  is hydrogen,  $C_{1-4}$ alkyl or Ar;

provided however that R' is C<sub>1-4</sub>alkyl or Ar when L is a covalent bond;

R<sup>2</sup> is hydrogen;

Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2-ethanediyl is optionally substituted with one or two C<sub>1,4</sub>alkyl substituents;

L is a covalent bond, —(CH<sub>2</sub>)<sub>m-n</sub>—Y—(CH<sub>2</sub>)<sub>n</sub>—, 1,2-cyclo-propanediyl,

C<sub>1-6</sub>alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and C<sub>1-4</sub>alkyloxyC<sub>1-4</sub>alkyl,

or  $C_{1\text{--}6}$  alkanediyl wherein two geminal hydrogen atoms are replaced by  $C_{2\text{--}6}$  alkanediyl;

m represents 3, 4, 5, 6 or 7;

n represents 1 or 2;

Y is O or NH:

Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>,

C<sub>1.4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1.4</sub>alkyloxy and cycloC<sub>3.7</sub>alkyl, and

C<sub>1.4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3.7</sub>alkyl;

 ${
m Ar}^2$  is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and  ${
m C}_{1.4}$ alkyl optionally substituted with one or more halo substituents;

R<sup>3</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

R<sup>4</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

R<sup>5</sup> is hydrogen or C<sub>1-4</sub>alkyl;

X is CR<sup>6</sup> or N;

 $R^6$  is hydrogen or  $C_{1-4}$ alkyl;

 $R^7$  is hydrogen or  $C_{1-4}$ alkyl;

 $R^8$  is hydrogen or  $C_{1-4}$ alkyl;

R<sup>o</sup> is a ring system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3-7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or C<sub>1-4</sub>alkyl optionally substituted with one or more halo atoms; and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein

 $R^1$  is hydrogen,  $C_{1-4}$ alkyl or Ar;

provided however that R' is  $C_{1-4}$ alkyl or Ar when L is a 45 covalent bond;

 $R^2$  is hydrogen, phenyl, cyclo $C_{3\_7}$ alkyl, tetrahydro-2H-pyranyl, or  $C_{1\_4}$ alkyl optionally substituted with one or more substituents each independently selected from the group consisting of hydroxyl,  $C_{1\_4}$ alkyloxy and  $NR^7R^8$ ;

Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2-ethanediyl is optionally substituted with one or two C<sub>1.4</sub>alkyl substituents;

L is a covalent bond,  $-(CH_2)_{m-n}$ — $Y-(CH_2)_n$ —, 1,2-cyclopropanediyl,  $C_{1-6}$ alkanediyl optionally substituted with 55 one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  $C_{1-4}$ alkyloxy $C_{1-4}$ 4alkyl, or  $C_{1-6}$ alkanediyl wherein two geminal hydrogen atoms are replaced by  $C_{2-6}$ alkanediyl;

m represents 3;

n represents 2;

Y is O;

Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

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wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>,

C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and

C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl;

O Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C<sub>1-4</sub>alkyl optionally substituted with one or more halo substituents;

R<sup>3</sup> is hydrogen or halo;

R<sup>4</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

R<sup>5</sup> is hydrogen or C<sub>1-4</sub>alkyl;

X is CR<sup>6</sup> or N;

R<sup>6</sup> is hydrogen or C<sub>1</sub> ₄alkyl;

 $R^7$  is hydrogen or  $C_{1-4}$ alkyl;

 $R^8$  is  $C_{1-4}$ alkyl;

R<sup>o</sup> is a ring system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3-7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more C<sub>1-4</sub>alkyl groups optionally substituted with one or more halo atoms;

and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein

R<sup>1</sup> is hydrogen, tert-butyl or Ar;

provided however that R<sup>1</sup> is tert-butyl or Ar when L is a covalent bond:

R<sup>2</sup> is hydrogen, phenyl, cyclopropyl, tetrahydro-2H-pyranyl, or C<sub>1.4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of hydroxyl, C<sub>1.4</sub>alkyloxy and NR<sup>7</sup>R<sup>8</sup>;

40 Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2ethanediyl is optionally substituted with one or two methyl substituents;

L is a covalent bond,  $-(CH_2)_{m-n}-Y-(CH_2)_n-$ , 1,2-cyclopropanediyl,

 $\rm C_{1-3}$  alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  $\rm C_{1-4}$  alkyloxyC $_{1-4}$  alkylor or  $\rm C_{1-3}$  alkanediyl wherein two geminal hydrogen atoms are replaced by  $\rm C_{2-6}$  alkanediyl;

m represents 3;

n represents 2;

Y is O:

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Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of Br, Cl, F, Ar<sup>2</sup>, R<sup>0</sup>,

C<sub>1.4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of F, C<sub>1.4</sub>alkyloxy and cyclopropyl, and

C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of F and cyclopropyl;

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of F, Cl and C<sub>1-4</sub>alkyl optionally substituted with one or more F substituents;

R<sup>3</sup> is hydrogen or Cl;

R<sup>4</sup> is hydrogen, Cl, methyl or ethyl;

R<sup>5</sup> is hydrogen or C<sub>1-4</sub>alkyl;

X is CR<sup>6</sup> or N;

R<sup>6</sup> is hydrogen or methyl;

R<sup>7</sup> is hydrogen or methyl;

R<sup>8</sup> is methyl;

 $R^{\rm o}$  is a ring system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3.7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more C<sub>1.4</sub>alkyl groups optionally substituted with one or more halo atoms;

and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention concerns novel 20 compounds of Formula (I), wherein one or more of the following restrictions apply:

(a) R<sup>1</sup> is Ar;

- (b) R² is phenyl, cycloC<sub>3-7</sub>alkyl, tetrahydro-2H-pyranyl, tetrahydro-2H-thiopyranyl, piperidinyl, or C<sub>1-4</sub>alkyl option-25 ally substituted with one or more substituents each independently selected from the group consisting of hydroxyl, C<sub>1-4</sub>alkyloxy and NR<sup>7</sup>R<sup>8</sup>;
- (c) Z is methylene or 1,2-ethanediyl;
- (d) L is a  $(CH_2)_{m-n}$ —Y— $(CH_2)_n$ —,  $C_{1-6}$  alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  $C_{1-4}$  alkyloxy $C_{1-4}$  alkyl;
- (e) m represents 3, 4, 5 or 6;
- (f) n represents 1;
- (g) Y is O;
- (h) Ar is a ring system selected from the group consisting of phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>,
  - $C_{1-4}$ alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo,  $C_{1-4}$ alkyloxy and cyclo $C_{3-7}$ alkyl, and
  - C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group 45 consisting of halo and cycloC<sub>3-7</sub>alkyl;
- (i) Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C<sub>1.4</sub>alkyl optionally substituted with one or more halo substituents;
- (j) R<sup>3</sup> is hydrogen;
- (k)  $R^4$  is  $C_{1-4}$ alkyl;
- (1) R<sup>5</sup> is hydrogen;
- (m) X is  $CR^6$ ;
- (n) R<sup>6</sup> is hydrogen;
- (o)  $R^7$  is  $C_{1-4}$ alkyl;
- (p)  $R^8$  is  $C_{1-4}$ alkyl;
- (q) R<sup>0</sup> is a ring system selected from the group consisting of piperidinyl optionally substituted with one or more substituents each independently selected from the group consisting of halo or C<sub>1-4</sub>alkyl optionally substituted with one or more halo atoms.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein R<sup>0</sup> is a ring 65 system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3-7</sub>alkyl, pyrazolyl and pyrrolidinyl;

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wherein said ring system is optionally substituted with one or more  $C_{1\rightarrow 4}$ alkyl groups optionally substituted with one or more halo atoms.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein  $R^8$  is  $C_{1-4}$ alkyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein  $\rm R^2$  is phenyl, cycloC $_{\rm 3-7}$ alkyl, tetrahydro-2H-pyranyl, tetrahydro-2H-thiopyranyl, piperidinyl, or  $\rm C_{1-4}$ alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, hydroxyl,  $\rm C_{1-4}$ alkyloxy and NR $^7\rm R^8$ .

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein  $R^2$  is  $C_{1.4}$ alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, hydroxyl,  $C_{1.4}$ alkyloxy and  $NR^7R^8$ .

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein  ${\rm R}^2$  is  ${\rm C}_{1.4}$  alkyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein R<sup>2</sup> is methyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein R<sup>2</sup> is H.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein R³ is hydrogen or halo.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein Z is methylene or 1,2-ethanediyl, wherein methylene is optionally substituted with one or two  $C_{1-4}$ alkyl substituents.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein m represents 3, 4, 5 or 6; in particular 3, 4 or 5; more in particular 3 or 4; even more in particular 3.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein n represents 1 or 2; in particular 2.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein  $R^1$  is Ar.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein m represents 3 or 4 and wherein n represents 1 or 2; more in particular wherein m represents 3 and wherein n represents 2.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein Y represents O

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein L is a covalent bond, 1,2-cyclopropanediyl,

C<sub>1-6</sub>alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2Hpyranyl, phenyl and  $C_{1-4}$ alkyloxy $C_{1-4}$ alkyl,

or C<sub>1-6</sub>alkanediyl wherein two geminal hydrogen atoms are replaced by C<sub>2-6</sub>alkanediyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein Ar is indolyl optionally substituted as defined in any of the other embodiments, and wherein L is a covalent bond, 1,2-cyclopropanediyl,

C<sub>1-6</sub>alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2Hpyranyl, phenyl and C<sub>1-4</sub>alkyloxyC<sub>1-4</sub>alkyl,

or C<sub>1-6</sub>alkanediyl wherein two geminal hydrogen atoms are replaced by C<sub>2-6</sub>alkanediyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein L is a covalent bond or  $C_{1-6}$ alkanediyl;

 $R^1$  is Ar;

Ar is indolyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>,

C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and

C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group  $_{30}$  R<sup>2</sup> is hydrogen or C<sub>1-4</sub>alkyl; consisting of halo and cycloC<sub>3-7</sub>alkyl.

An interesting group of compounds relates to those compounds of formula (I) wherein the position of R<sup>3</sup> is fixed as shown in (I-x)

tautomers and stereoisomeric forms thereof,

wherein all the substituents have the same meaning as defined in any of the embodiments hereinbefore,

solvates thereof.

An interesting group of compounds relates to those compounds of formula (I) wherein the position of R<sup>3</sup> is fixed as shown in (I-x)

tautomers and stereoisomeric forms thereof. wherein all the substituents have the same meaning as defined

in any of the embodiments hereinbefore or hereinafter, and the pharmaceutically acceptable addition salts and the solvates thereof.

An interesting group of compounds relates to those compounds of formula (I) wherein the position of R<sup>3</sup> is fixed as shown in (I-x1)

$$R^{1}$$
 $R^{2}$ 
 $R^{2}$ 
 $R^{3}$ 
 $R^{4}$ 
 $R^{4}$ 

tautomers and stereoisomeric forms thereof,

wherein all the substituents have the same meaning as defined in any of the embodiments hereinbefore or hereinafter, and the pharmaceutically acceptable addition salts and the solvates thereof.

In an embodiment, the present invention concerns novel compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein:

 $R^1$  is Ar;

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Z is methylene;

L is a  $C_{1-6}$ alkanediyl, in particular methylene;

Ar is a ring system selected from the group consisting of phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>, C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl;

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C<sub>1-4</sub>alkyl optionally substituted with one or more halo substituents;

R<sup>3</sup> is hydrogen;

 $R^4$  is  $C_{1-4}$ alkyl;

R<sup>5</sup> is hydrogen;

X is CH;

and the pharmaceutically acceptable addition salts and the 50 Ro is a ring system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3-7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or C<sub>1-4</sub>alkyl optionally substituted with one or more halo atoms; in particular R<sup>o</sup> is piperidinyl;

and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention concerns novel 60 compounds of Formula (I), tautomers and stereoisomeric forms thereof, wherein:

 $R^1$  is Ar;

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(I-x)

 $R^2$  is  $C_{1-4}$ alkyl; in particular methyl

Z is methylene;

65 L is a C<sub>1-6</sub>alkanediyl, in particular methylene;

Ar is phenyl substituted with one or more substituents each independently selected from the group consisting of

C<sub>1-4</sub>alkyl optionally substituted with one, two or three substituents each independently selected from the group consisting of halo;

R<sup>3</sup> is hydrogen;

 $R^4$  is  $C_{1-4}$ alkyl; in particular methyl;

R<sup>5</sup> is hydrogen;

X is CH;

and the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment, the present invention relates to those 10 compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein L is C<sub>1-6</sub>alkanediyl; in particular wherein L is methylene or ethylidene; more in particular wherein L is ethylidene.

In an embodiment, the present invention relates to those 15 compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein L is methylene.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as men- 20 tioned in any of the other embodiments, wherein R<sup>2</sup> is methyl and L is ethylidene or methylene; in particular wherein R<sup>2</sup> is methyl and L is ethylidene.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as men- 25 tioned in any of the other embodiments, wherein R<sup>2</sup> is  $C_{1-4}$ alkyl and L is  $C_{1-6}$ alkanediyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein L is a covalent bond,  $-(CH_2)_{m-n}$  -Y  $-(CH_2)_n$  -, 1,2-cyclopropanediyl, C<sub>1-6</sub>alkanediyl substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  $C_{1-4}$ alkyloxy $C_{1-4}$ alkyl, or  $C_{1-6}$ alkanediyl wherein two geminal hydrogen atoms are replaced 35 by C<sub>2-6</sub>alkanediyl; in particular wherein L is a covalent bond,  $-(CH_2)_{m-n}$  -Y  $-(CH_2)_n$  -, 1,2-cyclopropanediyl, C<sub>1-6</sub>alkanediyl wherein two geminal hydrogen atoms are replaced by C<sub>2-6</sub>alkanediyl; more in particular wherein L is a covalent bond, — $(CH_2)_{m-n}$ —Y— $(CH_2)_n$ —, or 1,2-cyclopro- 40 panediyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein Ar is a ring oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthale-

wherein said ring system is optionally substituted with one or 50 more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>

C<sub>1.4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and

C<sub>1,4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as men- 60 tioned in any of the other embodiments, wherein Ar is indolyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>, C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and C<sub>1-4</sub>alkyloxy optionally substituted with one or more sub22

stituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein Ar is indolyl substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>, C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein R<sup>2</sup> is phenyl, cycloC<sub>3-7</sub>alkyl, tetrahydro-2H-pyranyl, tetrahydro-2H-thiopyranyl, piperidinyl, or C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, hydroxyl, C1\_alkyloxy and  $NR^7R^8$ .

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein R<sup>2</sup> is methyl; and wherein

L is a covalent bond, 1,2-cyclopropanediyl,  $C_{1-6}$ alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  $C_{1-4}$  alkyloxy $C_{1-4}$  alkyl,

or C<sub>1-6</sub>alkanediyl wherein two geminal hydrogen atoms are replaced by  $C_{2-6}$  alkanediyl.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein one or more of the following heterocyclic groups in the Ar definition are restricted: indolyl is restricted to indol-3-yl, oxazolyl is restricted to 2-oxazolyl, benzo[b]thienyl is restricted to benzo [b]thien-3-yl, 1,3-benzodioxolyl is restricted to 1,3-benzodioxol-5-yl, 1,4-benzodioxanyl is restricted to benzodioxan-2-yl, 3,4-dihydro-2H-1,5-benzodioxepinyl is restricted to 3,4-dihydro-2H-1,5-benzodioxepin-7-yl, pyridyl is restricted to 2-pyridinyl or 4-pyridinyl, indanyl is restricted to indan-1yl or indan-2-yl, naphthalenyl is restricted to 2-naphthalenyl; it should be understood that any of these heterocyclic groups system selected from the group consisting of indolyl, 45 may be substituted as defined in any of the other embodiments.

> In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein one or more of the following heterocyclic groups in the R<sup>2</sup> definition are restricted: tetrahydro-2H-pyranyl is restricted to tetrahydro-2H-pyran-4-yl, tetrahydro-2H-thiopyranyl is restricted to tetrahydro-2H-thiopyran-4-yl, piperidinyl is restricted to 1-piperidinyl or 4-piperidinyl;

it should be understood that any of these heterocyclic groups may be substituted as defined in any of the other embodi-

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein one or more of the following heterocyclic groups in the R<sup>0</sup> definition are restricted: piperidinyl is restricted to 1-piperidinyl, morpholinyl is restricted to 1-morpholinyl, pyrazolyl is restricted to 1-pyrazolyl, pyrrolidinyl is restricted to 1-pyrrolidinyl; it should be understood that any of these heterocyclic groups may be substituted as defined in any of the other embodiments.

In an embodiment, the present invention relates to those compounds of formula (I), or any subgroup thereof as mentioned in any of the other embodiments, wherein one or more of the heterocyclic groups in the Ar, R<sup>2</sup> and R<sup>0</sup> definitions are restricted as indicated in the 3 embodiments hereabove.

In an embodiment the compound of Formula (I) is selected from the group consisting of:

- 2-([1,1'-biphenyl]-3-ylmethyl)-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-2-[(3-methoxyphenyl)methyl]-7-(3-methyl-1H- 10 1,2,4-triazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-7-(3-methyl-1H-1,2,4-triazol-1-yl)-2-[[3-(trifluoromethyl)phenyl]-methyl]-2H-pyrido[1,2-a]pyrazine-1.6-dione.
- 3,4-dihydro-2-[(3-methoxyphenyl)methyl]-7-(4-methyl-1H- 15 imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-(phenylmethyl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[(3,4-dichlorophenyl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(trifluoromethoxy)phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[3-(trifluoromethyl)phenyl]ethyl]-2H-pyrido[1,2-a]pyrazine-1,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[3-(trifluoromethyl)phenyl]ethyl]-2H-pyrido[1,2-a]pyrazine-1, 6-dione (R or S; enantiomer A (SFC-MS)),
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[3-(trifluoromethyl)phenyl]ethyl]-2H-pyrido[1,2-a]pyrazine-1, 6-dione (S or R; enantiomer B (SFC-MS)),
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(trifluoromethyl)phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-
- 2-[[1-(4-chlorophenyl)cyclopropyl]methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1.6-dione HCl.
- 2-[[1-(4-chlorophenyl)cyclopropyl]methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-7-(5-methyl-1H-imidazol-1-yl)-2-(phenylmethyl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[(3,4-dichlorophenyl)methyl]-2,3,4,5-tetrahydro-8-(4-methyl-1H-imidazol-1-yl)-pyrido[1,2-a][1,4]diazepine-1,7-
- 2-[(3,4-dichlorophenyl)methyl]-2,3,4,5-tetrahydro-8-(5-me-50 thyl-1H-imidazol-1-yl)-pyrido[1,2-a][1,4]diazepine-1,7-
- 2-[2-(4-chlorophenyl)ethyl]-3,4-dihydro-7-(4-methyl-1Himidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1.6-dione.
- (3R)-2-[(3,4-dichlorophenyl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 2-[[4'-fluoro-2'-(trifluoromethyl) [1,1'-biphenyl]-3-yl]methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-(2-cyclopropylethyl)-3,4-dihydro-7-(2-propyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-(cyclopropylmethyl)-3,4-dihydro-7-(2-propyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,

- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[(2-methylphenyl)methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[(4-fluoro-2-methylphenyl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-di-
- 2-[[2-fluoro-3-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-(2-methylphenyl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[2-methyl-3-(trifluoromethyl)phenyl]-methyl]-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 2-[(3-cyclopropylphenyl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(4-morpholinyl)phenyl[methyl]-2H-pyrido[1,2-a]pyrazine-1,6dione,
- 20 2-(4-fluoro-2-methylphenyl)-3,4-dihydro-7-(4-methyl-1Himidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(1-piperidinyl)phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6dione.
- 25 2H-pyrido[1,2-a]pyrazine-1,6-dione, 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[2-(1-piperidinyl)phenyl]me-
  - 2H-pyrido[1,2-a]pyrazine-1,6-dione, 2-[2-fluoro-3-(trifluoromethyl)phenyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-
  - 2H-pyrido[1,2-a]pyrazine-1,6-dione, 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[2-methyl-3-(trifluoromethyl) phenyl]-
  - 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[2-methyl-3-(1-piperidinyl)phenyl]-methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[4-methyl-3-(1-piperidinyl)phenyl]-methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 40 (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[(2-methylphenyl)-methyl]-2H-pyrido[1,2-a]pyrazine-
  - (3R)-2-[(3,4-dichlorophenyl)methyl]-3,4-dihydro-3-(1-methylethyl)-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1, 2-a pyrazine-1,6-dione,
  - 2-[2-(4-chlorophenyl)cyclopropyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (TRANS),
  - (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(1-piperidinyl)-phenyl]methyl]-2H-pyrido[1,2-a] pyrazine-1,6-dione.
  - (3R)-2-[[1-(4-chlorophenyl)cyclopropyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione,
- (3S)-2-[(3,4-dichlorophenyl)methyl]-3,4-dihydro-3-methyl- 55 (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[4-methyl-3-(1-piperidinyl)phenyl]methyl]-2H-pyrido [1,2-a]pyrazine-1,6-dione .HCl,
  - (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[4-methyl-3-(1-piperidinyl)phenyl]methyl]-2H-pyrido [1,2-a]pyrazine-1,6-dione,
  - 2-([1,1'-biphenyl]-3-ylmethyl)-3,4-dihydro-7-(1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 2-([1,1'-biphenyl]-3-ylmethyl)-3,4-dihydro-7-(2-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 65 (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione,

- 2-([1,1'-biphenyl]-3-ylmethyl)-7-(2,4-dimethyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione
- 2-([1,1'-biphenyl]-3-ylmethyl)-9-chloro-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6- 5 dione.
- (3R)-2-[(3-chlorophenyl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(3-chlorophenyl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1.6-dione .HCl.
- 2-[2-[(4-chlorophenyl)methoxy]ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione.
- 2-(2-cyclohexylethyl)-3,4-dihydro-7-(4-methyl-1H-imida-zol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[(3-chlorophenyl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a|pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[1-(2-methylphenyl)-cyclopropyl]methyl]-2H-pyrido [1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[2-methyl-4-(trifluoromethoxy)-phenyl]methyl]-2H-pyrido[1,2-a] pyrazine-1.6-dione.
- 2-[(3,4-dihydro-2H-1,5-benzodioxepin-7-yl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 2-[(3-chloro-2-fluorophenyl)methyl]-3,4-dihydro-7-(4-me-thyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[2-methyl-3-(trifluoromethyl)phenyl]methyl]-2H-py-rido[1,2-a]pyrazine-1,6-dione,
- 2-[[2-chloro-3-(trifluoromethyl)phenyl]methyl]-3,4-dihy-dro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(trifluoromethoxy)-phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[(6-methyl-2-naphthalenyl)methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(3-chloro-2-fluorophenyl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione.
- 2-[[4-chloro-3-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- (3R)-2-[[3-(1,1-dimethylethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(3-chloro-2-methylphenyl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] 55 pyrazine-1,6-dione,
- 3,4-dihydro-2-[[4-methoxy-3-(trifluoromethyl)phenyl]methyl]-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 2-[2-(2,3-dihydro-1,4-benzodioxin-2-yl)ethyl]-3,4-dihydro-607-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyra-zine-1,6-dione,
- 2-[[5-(4-chlorophenyl)-2-oxazolyl]methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-(4,4-dimethylpentyl)-3,4-dihydro-7-(4-methyl-1H-imida-zol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,

- 3,4-dihydro-2-[2-(1H-indol-3-yl)ethyl]-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-methyl-4-(2,2,2-trifluoroethoxy)-2-pyridinyl]methyl]-2H-pyrido [1,2-a]pyrazine-1,6-dione,
- 2-[1-(2,3-dichlorophenyl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[1-(3,4-dichlorophenyl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(3,4-dichlorophenyl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-3-phenyl-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[4-(trifluoromethyl)-phenyl]methyl]-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- (3R)-3-cyclopropyl-2-[(3,4-dichlorophenyl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 20 (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-methyl-5-(trifluoromethyl)phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 2-[2,3-dihydro-4-(trifluoromethyl)-1H-inden-1-yl]-3,4-di-hydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
  - (3R)-3,4-dihydro-2-[[3-methoxy-5-(trifluoromethyl)phenyl] methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3S)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-(methoxymethyl)-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - (3R)-2-[[3-chloro-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 35 (3R)-3,4-dihydro-2-[[4-methoxy-3-(trifluoromethyl)phenyl] methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione,
- 2-[1-(2,3-dichlorophenyl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (R or S).
  - 2-[1-(2,3-dichlorophenyl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (S or R),
- 2-[1-(3,4-dichlorophenyl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (R or S).
- 2-[1-(3,4-dichlorophenyl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (S or R),
- 50 (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3-ethyl-3, 4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1, 2-alpyrazine-1,6-dione,
  - (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3-ethyl-3, 4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1, 2-a]pyrazine-1,6-dione .HCl,
  - (3R)-2-[[4-chloro-3-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-methyl-5-(trifluoromethyl)phenyl]-methyl]-2H-pyrido[1,2-a] pyrazine-1,6-dione,
  - 2-[(3,4-dichlorophenyl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-3-(tetrahydro-2H-pyran-4-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 65 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[1,2,3,4-tet-rahydro-7-(trifluoromethyl)-1-naphthalenyl]-2H-pyrido [1,2-a]pyrazine-1,6-dione,

- 3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2-[1,2,3,4-tet-rahydro-5-(trifluoromethyl)-1-naphthalenyl]-2H-pyrido [1,2-a]pyrazine-1,6-dione,
- 2-[[3,5-bis(trifluoromethyl)phenyl]phenylmethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- (3R)-2-[[3,4-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-7-(4-ethyl-1H-imidazol-1-yl)-3,4-dihydro-3-methyl-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-7-(4-ethyl-1H-imidazol-1-yl)-3,4-dihydro-3-methyl-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- 2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-2,3,4,5-tetrahy-dro-3-methyl-8-(4-methyl-1H-imidazol-1-yl)-pyrido[1,2-a][1,4]diazepine-1,7-dione,
- (3S)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione,
- (3S)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-7-(4-ethyl-1H-imidazol-1-yl)-3,4-dihydro-3-(methoxymethyl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3S)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-7-(4-ethyl-1H-imidazol-1-yl)-3,4-dihydro-3-(hydroxymethyl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3-[(dimethylamino)methyl]-7-(4-ethyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3S)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-(hydroxymethyl)-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[2-[3,5-bis(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1.6-dione.
- 2-[[3,5-bis(trifluoromethyl)phenyl](tetrahydro-2H-pyran-4-yl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyridol[1,2-alpyrazine-1,6-dione.
- 2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3, 4-dimethyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1, 2-a]pyrazine-1,6-dione (CIS),
- 2-[(2,3-dichlorophenyl)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3-[(dimethylamino)methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihy-dro-3-[(methylamino)-methyl]-7-(4-methyl-1H-imida-zol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3-(cyclopropylmethoxy)-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3-(cyclopropylmethoxy)-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- (3R)-3,4-dihydro-3-methyl-2-[[3-(1-methylethoxy)-5-(trif-luoromethyl)phenyl]methyl]-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-2-[[3-(1-methylethoxy)-5-(trif-luoromethyl)phenyl]-methyl]-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- (3R)-2-[[2-fluoro-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,

- (3R)-2-[[2-chloro-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[2-methyl-5-(trifluoromethyl)phenyl]methyl]-2H-py-rido[1,2-a]pyrazine-1,6-dione,
  - (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[2-methyl-5-(trifluoromethyl)phenyl]methyl]-2H-py-rido[1,2-a]pyrazine-1,6-dione .HCl,
- <sup>0</sup> (3R)-2-[2-[3,5-bis(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[4-chloro-3-(1-piperidinyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3-(ethoxymethyl)-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione;
- 20 (3R)-2-[[3-(ethoxymethyl)-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
  - (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(1-piperidinyl)-5-(trifluoromethyl)phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(1-piperidinyl)-5-(trifluoromethyl)phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- (3R)-3,4-dihydro-2-[[4-methoxy-2-methyl-5-(1-methylethyl)phenyl]methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - (3R)-3,4-dihydro-2-[[4-methoxy-2-methyl-5-(1-methyl-ethyl)phenyl]methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- 35 2-[1-[3,5-bis(trifluoromethyl)phenyl]-2-methoxyethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3, 4-dimethyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1, 2-a]pyrazine-1,6-dione (CIS A (SFC-MS)),
  - 2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3, 4-dimethyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1, 2-a]pyrazine-1,6-dione (CIS B (SFC-MS)),
- (3R)-2-[[2-chloro-5-(4-methyl-1-piperidinyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - (3R)-2-[[2-chloro-5-(4-methyl-1-piperidinyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- 50 (3R)-2-[[2-chloro-5-[4-(trifluoromethyl)-1-piperidinyl]phe-nyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imi-dazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - (3R)-2-[[2-chloro-5-[4-(trifluoromethyl)-1-piperidinyl]phe-nyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imi-dazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
  - (3R)-2-[[2-chloro-5-(1-pyrrolidinyl)phenyl]methyl]-3,4-di-hydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione,
  - (3R)-2-[[2-chloro-5-(1-pyrrolidinyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
  - (3R)-2-[[3-(3,5-dimethyl-1H-pyrazol-1-yl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 65 (3R)-2-[[3-(cyclopropylmethyl)-5-(trifluoromethyl)phenyl] methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione

- (3R)-2-[[3-(cyclopropylmethyl)-5-(trifluoromethyl)phenyl] methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- 2-[1-[3,5-bis(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione (1R or 1S, 3R) (diastereomer A (SFC-MS)).
- 2-[1-[3,5-bis(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione (1S or 1R, 3R) (diastereomer B (SFC-MS)).
- (3R)-2-[(2,2-difluoro-1,3-benzodioxol-5-yl)methyl]-3,4-di-hydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[4-methyl-3-(1-pyrazol-1-yl)phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[4-methyl-3-(1-pyrazol-1-yl)phenyl]methyl]-2H-py-rido[1,2-a]pyrazine-1,6-dione .HCl,
- (3R)-2-[(3-bromobenzo[b]thien-5-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3-cyclopropyl-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3-cyclopropyl-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- 2-[[1-[3,5-bis(trifluoromethyl)phenyl]cyclopropyl]methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione, and
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-7-(4-chloro-1H-imidazol-1-yl)-3,4-dihydro-3-methyl-2H-py-rido[1,2-a]pyrazine-1,6-dione,
- 2-[2-[3,5-bis(trifluoromethyl)phenyl]-1-methylethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1.6-dione.
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(1-piperidinyl)-4-(trifluoromethyl)phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione HCl salt,
- 2-[[1-[2-chloro-3-(trifluoromethyl)phenyl]cyclopropyl]methyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[2-[3,5-bis(trifluoromethyl)phenyl]propyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[1-[2-fluoro-5-(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-(1-hydroxyethyl)-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(2,2,2-trifluoroethoxy)-5-(trifluoromethyl)phenyl] methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- (3R)-2-[(5-chlorobenzo[b]thien-3-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-60a]pyrazine-1,6-dione,
- 2-[[1-[3,5-bis(trifluoromethyl)phenyl]cyclobutyl]methyl]-3, 4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1, 2-a]pyrazine-1,6-dione,
- (3R)-2-[[4-chloro-2-fluoro-5-(1-methylethoxy)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,

- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3,8-dimethyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 3,4-dihydro-2-[[3-(1-methylethoxy)-5-(trifluoromethyl) phenyl]methyl]-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-9-chloro-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-9-fluoro-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-8-bromo-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-2-[2-[3-(1-methylethoxy)-5-(tri-fluoromethyl)phenyl]ethyl]-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(3-methyl-1H-1,2,4-triazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[1-[3,5-bis(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione (1RS, 3R),
- 25 (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(trifluoromethyl)-5-[2-(trifluoromethyl)-4-morpholinyl]phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6dione HCl salt,
- (3R)-2-[[3-bromo-5-(trifluoromethoxy)phenyl]methyl]-3,4dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione .HCl,
  - (3R)-2-[[1-[3,5-bis(trifluoromethyl)phenyl]cyclopropyl] methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- 35 (3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-8-chloro-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 3,4-dihydro-2-[1-[3-methoxy-5-(trifluoromethyl)phenyl] ethyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R). HCl,
  - 3,4-dihydro-2-[1-[3-methoxy-5-(trifluoromethyl)phenyl] ethyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R) .HCl,
  - 3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[3-(trifluoromethoxy)phenyl]ethyl]-2H-pyrido[1,2-a] pyrazine-1,6-dione (1S or 1R, 3R) .HCl,
  - 3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[3-(trifluoromethoxy)phenyl]ethyl]-2H-pyrido[1,2-a] pyrazine-1,6-dione (1R or 1S, 3R) .HCl,
- 50 (3R)-2-[[3-bromo-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
  - 2-[1-[3-ethoxy-5-(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione (1S or 1R, 3R) .HCl,
  - 2-[1-[3-ethoxy-5-(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione (1R or 1S, 3R) .HCl,
  - 3,4-dihydro-2-[1-[3-methoxy-5-(trifluoromethyl)phenyl] propyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R) .HCl,
  - 3,4-dihydro-2-[1-[3-methoxy-5-(trifluoromethyl)phenyl] propyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R) .HCl,
- 65 (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(trifluoromethoxy)-5-(trifluoromethyl)phenyl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,

- (3R)-2-[[3-fluoro-5-(trifluoromethoxy)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- (3R)-2-[(5-chloro-1H-indol-3-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(5-chloro-1-methyl-1H-indol-3-yl)methyl]-3,4-di-hydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione,
- 2-[1-[3-(cyclopropylmethyl)-5-(trifluoromethyl)phenyl] ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R) HCl
- 2-[1-[3-(cyclopropylmethyl)-5-(trifluoromethyl)phenyl] ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-15 yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R) .HCl.
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[3-(2,2,2-trifluoroethyl)-5-(trifluoromethyl)phenyl] methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- (3R)-2-[(3,4-difluorophenyl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-2-[(3-methoxyphenyl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(3,5-dichlorophenyl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 2-[1-[3-chloro-5-(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione (1S or 1R, 3R; diasteromer A (SFC-MS)),
- 2-[1-[3-chloro-5-(trifluoromethyl)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione (1R or 1S, 3R; diasteromer B (SFC-MS));
- (3R)-3,4-dihydro-3-methyl-2-[[2-(1-methylethoxy)-6-(trif-luoromethyl)-4-pyridinyl]methyl]-7-(4-methyl-1H-imi-dazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
- 2-[1-[3-fluoro-5-(trifluoromethoxy)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione (1S or 1R, 3R) .HCl,
- 2-[1-[3-fluoro-5-(trifluoromethoxy)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione (1R or 1S, 3R) .HCl,
- (3R)-2-[(6-bromo-1H-indol-3-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)- 50 2-(phenylmethyl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(6-chloro-1H-indol-3-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(6-chloro-1-methyl-1H-indol-3-yl)methyl]-3,4-di-hydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(6-bromo-1-methyl-1H-indol-3-yl)methyl]-3,4-di-hydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-(2-phenylethyl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[1-[3-fluoro-5-(trifluoromethoxy)phenyl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione (1RS, 3R) .HCl,
- 2-(2,3-dihydro-1H-inden-2-yl)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (R or S; enantiomer A (SFC-MS)),

- 2-(2,3-dihydro-1H-inden-2-yl)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (S or R; enantiomer B (SFC-MS)),
- 2-[2,3-dihydro-6-(trifluoromethyl)-1H-inden-1-yl]-3,4-di-hydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
- 2-[2,3-dihydro-6-(trifluoromethyl)-1H-inden-1-yl]-3,4-di-hydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-py-rido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
- 10 2-[1-(5-chloro-6-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
  - 2-[1-(5-chloro-6-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
  - 2-[1-(5-chloro-6-fluoro-1-methyl-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
  - 2-[1-(5-chloro-6-fluoro-1-methyl-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
  - 2-[1-(5-chloro-1-ethyl-6-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
- 25 (3R)-2-[[6-bromo-1-(1-methylethyl)-1H-indol-3-yl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - (3R)-3,4-dihydro-2-[(7-iodo-1H-indol-3-yl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
  - 2-[1-(5-chloro-1-ethyl-6-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
  - 2-[1-[5-chloro-6-fluoro-1-(1-methylethyl)-1H-indol-3-yl] ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
  - 2-[1-[5-chloro-6-fluoro-1-(1-methylethyl)-1H-indol-3-yl] ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
- (3Ř)-2-[(5-chloro-6-fluoro-1H-indol-3-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - (3R)-2-[(5-chloro-6-fluoro-1-methyl-1H-indol-3-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- <sup>5</sup> (3R)-2-[(6-bromo-1-phenyl-1H-indol-3-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 2-[1-(5-chloro-6-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyra-zine-1,6-dione (R or S; enantiomer A (SFC-MS)),
  - 2-[1-(5-chloro-6-ĥuoro-1H-indol-3-yl)ethyl]-3,4-diĥydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyra-zine-1,6-dione (S or R; enantiomer B (SFC-MS)),
- 2-[1-(5-chloro-6-fluoro-1-methyl-1H-indol-3-yl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (R or S; enantiomer A (SFC-MS)),
- 2-[1-(5-chloro-6-fluoro-1-methyl-1H-indol-3-yl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (S or R; enantiomer B (SFC-MS)),
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[2-methyl-5-(trifluoromethyl)-1H-indol-3-yl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[1,2-dimethyl-5-(trifluoromethyl)-1H-indol-3-yl] methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 55 (3R)-2-[[1-ethyl-2-methyl-5-(trifluoromethyl)-1H-indol-3-yl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imida-zol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,

- (3R)-3,4-dihydro-2-[(5-methoxy-1H-indol-3-yl)methyl]-3methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[5-(trifluoromethyl)-1H-indol-3-yl]methyl]-2H-pyrido 5 [1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-2-[(6-methoxy-1H-indol-3-yl)methyl]-3methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 2-[[6-(trifluoromethyl)-1H-indol-3-yl]methyl]-2H-pyrido [1,2-a]pyrazine-1,6-dione,
- 2-[1-(3,5-dichlorophenyl)ethyl]-3,4-dihydro-3-methyl-7-(4methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6dione (1R or 1S, 3R),
- 2-[1-(3,5-dichlorophenyl)ethyl]-3,4-dihydro-3-methyl-7-(4methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6dione (1S or 1R, 3R),
- (3R)-2-[2-(5-chloro-2-methoxyphenyl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-20 alpyrazine-1,6-dione,
- (3R)-2-[2-[2-fluoro-5-(trifluoromethyl)phenyl]ethyl]-3,4dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[2-(2,3-dichlorophenyl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione .HCl,
- (3R)-2-[2-(2,5-dichlorophenyl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[2-[2-methyl-3-(trifluoromethyl)phenyl]ethyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[1-methyl-6-(trifluoromethyl)-1H-indol-3-yl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[[1-ethyl-5-(trifluoromethyl)-1H-indol-3-yl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[1-methyl-5-(trifluoromethyl)-1H-indol-3-yl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3,4-dihydro-2-[[5-methoxy-1-(1-methylethyl)-1H-indol-3-yl]methyl]-3-methyl-7-(4-methyl-1H-imidazol-1yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[1-(5-chloro-7-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2alpyrazine-1,6-dione (1R or 1S, 3R),
- 2-[1-(5-chloro-7-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2alpyrazine-1,6-dione (1S or 1R, 3R),
- 2-[1-(5-chloro-7-fluoro-1-methyl-1H-indol-3-yl)ethyl]-3,4dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
- (3R)-2-[[1-ethyl-6-(trifluoromethyl)-1H-indol-3-yl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[1-(5-chloro-7-fluoro-1-methyl-1H-indol-3-yl)ethyl]-3,4dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
- 2-[1-(3,5-dichlorophenyl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (S or R; enantiomer A (SFC-MS)).
- 2-[1-(3,5-dichlorophenyl)ethyl]-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (R or S; enantiomer B (SFC-MS)),
- 2-[1-(5-chloro-1-ethyl-7-fluoro-1H-indol-3-yl)ethyl]-3,4dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),

- 2-[1-(5-chloro-1-ethyl-7-fluoro-1H-indol-3-yl)ethyl]-3,4dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
- 2-[1-[1-ethyl-5-(trifluoromethyl)-1H-indol-3-yl]ethyl]-3,4dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
- 2-[1-[1-ethyl-5-(trifluoromethyl)-1H-indol-3-yl]ethyl]-3,4dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
- (3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)- 10 3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[1-methyl-5-(trifluoromethyl)-1H-indol-3-yl]ethyl]-2Hpyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
  - 3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[1-methyl-5-(trifluoromethyl)-1H-indol-3-yl]ethyl]-2Hpyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
  - 3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[5-(trifluoromethyl)-1H-indol-3-yl]ethyl]-2H-pyrido[1,2a)pyrazine-1,6-dione (1R or 1S, 3R),
  - 3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[5-(trifluoromethyl)-1H-indol-3-yl]ethyl]-2H-pyrido[1,2alpyrazine-1,6-dione (1S or 1R, 3R),
  - (3R)-2-[[7-fluoro-1-methyl-5-(trifluoromethyl)-1H-indol-3yl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 25 (3R)-2-[[7-fluoro-5-(trifluoromethyl)-1H-indol-3-yl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
    - (3R)-2-[[6-fluoro-5-(trifluoromethyl)-1H-indol-3-yl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
    - (3R)-2-[(5-bromo-6-fluoro-1H-indol-3-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
    - (3R)-2-[[6-fluoro-1-methyl-5-(trifluoromethyl)-1H-indol-3yl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
    - (3R)-2-[(5-bromo-6-fluoro-1-methyl-1H-indol-3-yl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[1-[6-(trifluoromethyl)-1H-indol-3-yl]ethyl]-2H-pyrido[1,2a]pyrazine-1,6-dione (1R or 1S, 3R),
    - 2-[1-[1-ethyl-6-(trifluoromethyl)-1H-indol-3-yl]ethyl]-3,4dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2Hpyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
  - 45 (3R)-2-[3-(Cyclopropylmethoxy)-4-methylbenzyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl,
    - (3R)-2-[3-(Cyclopropylmethoxy)-4-methylbenzyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
    - (3R)-2-[3,5-Bis(trifluoromethyl)benzyl]-7-(1H-imidazol-1yl)-3-methyl-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-
  - (3R)-2-[(5-Cyclopropyl-6-fluoro-1-methyl-1H-indol-3-yl) methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
    - (3R)-2-[(6-fluoro-1-methyl-1H-indol-3-yl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - 2-[1-(5-Chloro-1-ethyl-1H-indol-3-yl)ethyl]-3-methyl-7-(4methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a] pyrazine-1,6-dione (1S or 1R, 3R),
  - 2-[1-(5-Chloro-1H-indol-3-yl)ethyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
  - 65 2-[1-(5-Chloro-1H-indol-3-yl)ethyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),

- 2-{1-[1-Ethyl-6-(trifluoromethyl)-1H-indol-3-yl]ethyl}-3methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione (1R or 1S, 3R),
- (3R)-2-[(5-Cyclopropyl-1-methyl-1H-indol-3-yl)methyl]-3methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(5-Cyclopropyl-1-ethyl-1H-indol-3-yl)methyl]-3methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione,
- 2-[1-(5-Chloro-1-cyclopropyl-6-fluoro-1H-indol-3-yl) ethyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R,
- 2-[1-(5-Chloro-1-ethyl-1H-indol-3-yl)ethyl]-3-methyl-7-(4methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a] pyrazine-1,6-dione (1R or 1S, 3R),
- (3R)-2-[(1-Ethyl-5-methyl-1H-indol-3-yl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-[(6-Chloro-1-ethyl-5-methoxy-1H-indol-3-yl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3-Methyl-7-(4-methyl-1H-imidazol-1-yl)-2-{[5-(trifluoromethoxy)-1H-indol-3-yl]methyl}-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-3-Methyl-7-(4-methyl-1H-imidazol-1-yl)-2- $\{[1$ -methyl-5-(trifluoromethoxy)-1H-indol-3-yl]methyl}-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione. HCl,
- (3R)-3-Methyl-7-(4-methyl-1H-imidazol-1-yl)-2-{[1-methyl-5-(trifluoromethoxy)-1H-indol-3-yl]methyl}-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[(5,6-Dichloro-1-methyl-1H-indol-3-yl)methyl]-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- (3R)-2-[(5,6-Dichloro-1-methyl-1H-indol-3-yl)methyl]-3methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione,
- 2-[1-(5-Chloro-6-methoxy-1-methyl-1H-indol-3-yl)ethyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
- (3R)-2-{[5-Fluoro-7-(trifluoromethyl)-1H-indol-3-yl]methyl}-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- (3R)-2-{[5-Fluoro-1-methyl-6-(trifluoromethyl)-1H-indol-3-yl]methyl}-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3, 45 2-[1-(5-Chloro-6-methoxy-1H-indol-3-yl)ethyl]-3-methyl-4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-[(5,7-Dichloro-1-methyl-1H-indol-3-yl)methyl]-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a] pyrazine-1,6-dione,
- 2-[(5,7-Dichloro-1H-indol-3-yl)methyl]-7-(4-methyl-1Himidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,
- (3R)-2-{[5-Fluoro-1-methyl-7-(trifluoromethyl)-1H-indol-3-yl]methyl}-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3, 4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione
- (3R)-3-Methyl-7-(4-methyl-1H-imidazol-1-yl)-2-{[5-methyl-1-(2,2,2-trifluoroethyl)-1H-indol-3-yl]methyl}-3,4dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 2-{[6-fluoro-1-methyl-5-(trifluoromethyl)-1H-indol-3-yl] methyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione,
- 2-{[1-ethyl-6-fluoro-5-(trifluoromethyl)-1H-indol-3-yl]methyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione,
- 2-{1-[6-fluoro-1-methyl-5-(trifluoromethyl)-1H-indol-3-yl] ethyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione (1R or 1S; enantiomer A (SFC-MS)),

- 2-{1-[6-fluoro-1-methyl-5-(trifluoromethyl)-1H-indol-3-yl] ethyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione (1S or 1R; enantiomer B (SFC-MS)),
- 5 2-{1-[6-fluoro-5-(trifluoromethyl)-1H-indol-3-yl]ethyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1, 2-a]pyrazine-1,6-dione (1R or 1S; enantiomer A (SFC-MS)),
  - 2-{1-[6-fluoro-5-(trifluoromethyl)-1H-indol-3-yl]ethyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1, 2-a]pyrazine-1,6-dione (1S or 1R; enantiomer B (SFC-MS)).
  - (3R)-2-[(6-Chloro-5-methoxy-1H-indol-3-yl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - (3R)-3-Methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[(5-methyl-1H-indol-3-yl)methyl]-3,4-dihydro-2H-pyrido[1,2a)pyrazine-1,6-dione,
  - (3R)-2-[(5,6-Dichloro-1H-indol-3-yl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1, 2-a]pyrazine-1,6-dione,
  - (3R)-2-[(5-Cyclopropyl-1H-indol-3-yl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1, 2-a]pyrazine-1,6-dione,
- (3R)-2-[(5-chloro-4-fluoro-1-methyl-1H-indol-3-yl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
  - (3R)-2-{[4-fluoro-1-methyl-5-(trifluoromethyl)-1H-indol-3-yl]methyl}-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3, 4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione,
- 30 7-(4-methyl-1H-imidazol-1-yl)-2-{1-[1-methyl-5-(trifluoromethyl)-1H-indol-3-yl]ethyl}-3,4-dihydro-2H-pyrido [1,2-a]pyrazine-1,6-dione (1R or 1S; enantiomer A (SFC-
  - 7-(4-methyl-1H-imidazol-1-yl)-2-{1-[1-methyl-5-(trifluoromethyl)-1H-indol-3-yl]ethyl}-3,4-dihydro-2H-pyrido [1,2-a]pyrazine-1,6-dione (1S or 1R; enantiomer B (SFC-
  - 2-{1-[1-ethyl-6-(trifluoromethyl)-1H-indol-3-yl]ethyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1, 2-a]pyrazine-1,6-dione (1R or 1S; enantiomer A (SFC-
  - 2-{1-[1-ethyl-6-(trifluoromethyl)-1H-indol-3-yl]ethyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1, 2-a]pyrazine-1,6-dione (1S or 1R; enantiomer B (SFC-
- 7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido [1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),
- 7-(4-Methyl-1H-imidazol-1-yl)-2-{1-[5-(trifluoromethyl)-1H-indol-3-yl]ethyl}-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1,6-dione, 50
  - 7-(4-methyl-1H-imidazol-1-yl)-2-{1-[6-(trifluoromethyl)-1H-indol-3-yl]ethyl}-3,4-dihydro-2H-pyrido[1,2-a]pyra-
- 2-[(5,6-dichloro-1H-indol-3-yl)methyl]-7-(4-methyl-1Himidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-a]pyrazine-1, 6-dione,
  - 2-{[6-fluoro-5-(trifluoromethyl)-1H-indol-3-yl]methyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1, 2-a]pyrazine-1,6-dione,
- 2-{1-[6-fluoro-5-(trifluoromethyl)-1H-indol-3-yl]ethyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1, 2-a|pyrazine-1,6-dione.
  - 2-{1-[6-fluoro-1-methyl-5-(trifluoromethyl)-1H-indol-3-yl] ethyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2Hpyrido[1,2-a]pyrazine-1,6-dione,
- 65 7-(4-methyl-1H-imidazol-1-yl)-2-{1-[1-methyl-5-(trifluoromethyl)-1H-indol-3-yl]ethyl}-3,4-dihydro-2H-pyrido [1,2-c]pyrazine-1,6-dione,

2-{1-[1-ethyl-6-(trifluoromethyl)-1H-indol-3-yl]ethyl}-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1, 2-c]pyrazine-1,6-dione,

(3R)-2-{[5-fluoro-6-(trifluoromethyl)-1H-indol-3-yl]methyl}-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-c]pyrazine-1,6-dione,

(3R)-2-{[4-fluoro-5-(trifluoromethyl)-1H-indol-3-yl]methyl}-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-c]pyrazine-1,6-dione,

(3R)-2-[(5-chloro-4-fluoro-1H-indol-3-yl)methyl]-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-c]pyrazine-1,6-dione,

2-[1-(5-chloro-1-ethyl-1H-indol-3-yl)ethyl]-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-c]pyrazine-1,6-dione (1R or 1S; enantiomer A (SFC-MS)),

2-[1-(5-chloro-1-ethyl-1H-indol-3-yl)ethyl]-7-(4-methyl-1H-imidazol-1-yl)-3,4-dihydro-2H-pyrido[1,2-c]pyrazine-1,6-dione (1S or 1R; enantiomer B (SFC-MS)), tautomers and stereoisomeric forms thereof,

and the free bases, the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment the compound of Formula (I) is selected from the group consisting of:

(3R)-2-[(3,4-dichlorophenyl)methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione,

(3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3-ethyl-3, 4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1, 2-a]pyrazine-1,6-dione .HCl,

(3R)-2-[[3,5-bis(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido [1,2-a]pyrazine-1,6-dione,

(3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[4-methyl-3-(1-piperidinyl)phenyl]methyl]-2H-pyrido [1,2-a]pyrazine-1,6-dione .HCl,

(3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[4-methyl-3-(1-piperidinyl)phenyl]methyl]-2H-pyrido [1,2-a]pyrazine-1,6-dione,

2-([1,1'-biphenyl]-3-ylmethyl)-3,4-dihydro-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione, and

(3R)-2-[[3-(cyclopropylmethoxy)-5-(trifluoromethyl)phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione .HCl, tautomers and stereoisomeric forms thereof,

and the free bases, the pharmaceutically acceptable addition salts, and the solvates thereof.

In an embodiment the compound of Formula (I) is (3R)-2-[[3,5-bis(trifluoromethyl)-phenyl]methyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a] pyrazine-1,6-dione.

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In an embodiment the compound of Formula (I) is selected from the group consisting of:

2-[1-[1-ethyl-5-(trifluoromethyl)-1H-indol-3-yl]ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),

2-[1-(5-chloro-1-ethyl-7-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R),

tautomers and stereoisomeric forms thereof, and the free bases, the pharmaceutically acceptable addition

salts, and the solvates thereof.

In an embodiment the compound of Formula (I) is selected from the group consisting of:

2-[1-(5-chloro-1-ethyl-6-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione (1S or 1R, 3R)

(3R)-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2-[[1-methyl-5-(trifluoromethyl)-1H-indol-3-yl]methyl]-2H-pyrido[1,2-a]pyrazine-1,6-dione tautomers and stereoisomeric forms thereof,

and the free bases, the pharmaceutically acceptable addition salts, and the solvates thereof.

All possible combinations of the above-indicated interesting embodiments are considered to be embraced within the scope of this invention.

Preparation of the Compounds

The present invention also encompasses processes for the preparation of compounds of Formula (I), intermediates and subgroups thereof. In the reactions described, it can be necessary to protect reactive functional groups, for example hydroxy, amino, or carboxy groups, where these are desired in the final product, to avoid their unwanted participation in the reactions. Conventional protecting groups can be used in accordance with standard practice, for example, see T. W. Greene and P. G. M. Wuts in "Protective Groups in Organic Chemistry", John Wiley and Sons, 1999.

The compounds of Formula (I) and the subgroups thereof can be prepared by a succession of steps as described hereunder and as described in the specific examples. They are generally prepared from starting materials which are either commercially available or prepared by standard means obvious to those skilled in the art. The compounds of the present invention can be also prepared using standard synthetic processes commonly used by those skilled in the art of organic chemistry.

The skilled person will realize that in some reactions microwave heating may be used instead of conventional heating to shorten the overall reaction time.

The general preparation of some typical examples is shown below. All variables are defined as mentioned hereabove unless otherwise is indicated.

Experimental Procedures—Scheme 1

Scheme 1

$$R^{9} \longrightarrow R^{3}$$
 $R^{9} \longrightarrow R^{3}$ 
 $R^{9} \longrightarrow R^{$ 

-continued

R9 R3

BnO 
$$R^{10}$$
 + PGO  $R^{10}$  + P

Experimental Procedure 1

An intermediate of formula (III), wherein all the variables are defined as described in the scope of the invention, can be 30 prepared via nucleophilic substitution by an intermediate of formula (II-a) on an appropriate electrophile, such as for example an alkyl halide, such as for example an alkyl iodide, with methods known to the person skilled in the art, such as for example refluxing the mixture of nucleophile and electrophile in the presence or absence of solvent. Inert atmosphere may enhance the reaction outcome.

Alternatively, intermediate (III) can be obtained by reductive amination, starting from the appropriate aminoalcohol (II-b), in the presence of the desired carbonyl compound, such as for example a ketone or an aldehyde. The reaction can be typically performed in the presence of a suitable solvent, such as MeOH (methanol) and a reducing agent, such as NaBH $_4$  (sodium borohydride) or NaCNBH $_3$  (sodium cyanoborohydride). Pre-stirring of the mixture in the absence of the reducing agent under heating, and subsequent addition of the reducing agent at lower temperature, can enhance the reaction outcome

Alternatively, intermediate (III) can be obtained by  $_{50}$  manipulation of any suitable precursor by methods known to the person skilled in the art, such as reduction of the corresponding  $\alpha$ -aminoacid, for example by using borane-methyl sulphide in the presence of a suitable solvent, such as THF (tetrahydrofuran). Precooling of the reaction mixture, followed by heating after the addition of all the reagents, may enhance the reaction outcome.

# Experimental Procedure 2

An intermediate of formula (V), wherein all the variables are defined as described in the scope of the invention, can be 60 obtained via protection of the alcohol functionality of intermediate (III). The protection can be for example a silylation, that can be performed in the presence of a suitable solvent, such as DCM (dichloromethane), an additive, such as imidazole, and a silylating agent, such as TBSCl (tert-butyldimethylsilyl chloride) or TMSCl (trimethylsilyl chloride), following standard conditions known to the person skilled in the art.

Alternatively, intermediate (V) can be obtained by reductive amination of an appropriate amine (II-a) with a carbonyl intermediate such as (IV), where for example PG (the protecting group) can be tert-butyldimethylsilyl. Typical conditions involve stirring of the reagents in a suitable solvent, such as DCE (1,2-dichloroethane), in the presence of a reducing agent, such as NaBH(OAc) $_3$  (sodium triacetoxyborohydride). The person skilled in the art will notice that intermediate (V) can also be obtained via standard reductive amination conditions, starting from an intermediate of structure (II-b), where R is the desired protecting group (PG). Experimental Procedure 3

An intermediate of formula (VII), wherein

PG is a protecting group;

and all the other variables are defined as described in the scope of the invention, can be obtained via acylation of intermediate (V) with an intermediate of structure (VI), where

R<sup>9</sup> is hydrogen or bromine;

R<sup>10</sup> is hydroxyl or chlorine.

Structure (VI) is hereby named (VI-a) when R<sup>10</sup> is hydroxyl, and (VI-b) when R<sup>10</sup> is a chlorine. Acylation using intermediate (VI-a) can be performed for example under classical peptide synthesis conditions. Typically, the reaction requests stirring of the starting materials (V) and (VI-a) in the presence of a base, such as DIPEA (diisopropylethyl amine) and a peptide coupling reagent, such as HBTU (O-benzotriazole-N,N,N',N'-tetramethyl-uronium-hexafluoro-phosphate), in a suitable solvent, such as DMF (N,N-dimethyl

phate), in a suitable solvent, such as DMF (N,N-dimethyl formamide).

Alternatively, acylation can be achieved by reacting intermediate (V) with an intermediate of formula (VI-b). The reaction can be performed for example by stirring the starting materials in the presence of a base, such as DIPEA, in a suitable solvent, such as DMF.

Experimental Procedure 4

An intermediate of formula (VIII), wherein

R<sup>9</sup> is hydrogen or bromine;

and all the other variables are defined as described in the scope of the invention, can be obtained by deprotection of intermediate (VII), by methods known to the person skilled in

the art. In the case of a silyl protecting group, for example, one standard method would be treating intermediate (VII), dissolved in a suitable solvent, such as THF, with a fluoride source, such as TBAF (tetrabutylammonium fluoride).

Alternatively, an intermediate of formula (VIII) can be 5 obtained by direct acylation of a suitable aminoalcohol of structure (III) with an acid of structure (VI-a). The reaction can be performed for example under peptide coupling conditions, in the presence of a base, such as DIPEA, and a peptide coupling reagent, such as HBTU, in a suitable solvent, such as 10 DMF.

Experimental Procedures—Scheme 2

Experimental Procedure 5

An intermediate of formula (IX), wherein

R<sup>9</sup> is hydrogen or bromine;

and all the other variables are defined as described in the scope of the invention, can be obtained by debenzylation of intermediate (VIII), using standard methods known to the person skilled in the art. For example, the benzylation can be achieved by stirring a solution of intermediate (VIII) in a suitable solvent, such as MeOH or MeOH/THF, and in the presence of a hydrogenation catalyst, such as 10% Pd/C (palladium on carbon), under hydrogen atmosphere.

$$\begin{array}{c} & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\$$

Alternatively, intermediate (IX) can be obtained by amide synthesis starting from a suitable ester, such as intermediate (XI), where R<sup>11</sup> is for example a methyl group. Typical conditions involve stirring a solution of the ester in a suitable solvent, such as MeOH, in the presence of a desired aminoalcohol of structure (III) under reflux. Alternatively, starting as well from intermediate (XI), intermediate (IX) can be obtained by using a 2-step method. First, ester (XI) can be saponified to give intermediate (XII), where M is a metal. The reaction can be performed for example by adding an hydrox- 10 ide, such as LiOH (lithium hydroxide), to a solution of ester (XI) in a suitable polar solvent or in a mixture of miscible solvents of which one is highly polar, such as THF and water. Heating the reaction mixture can enhance the reaction outcome. In the second step, intermediate (XII) can be reacted 15 with an aminoalcohol of structure (III), to afford intermediate (IX). Typically, peptide coupling conditions can be applied, such as stirring the starting material, dissolved in a suitable solvent, such as DMF, in the presence of a peptide coupling agent, such as HBTU. The skilled in the art will appreciate 20 that when a base, such as DIPEA, is present in the mixture, the reaction affords directly the cyclised intermediate (XVII). Heating the reaction mixture can enhance the reaction outcome.

Alternatively, intermediate (IX) can be obtained starting 25 from acid (X), using for example standard peptide coupling conditions, such as stirring intermediate (X) and the desired aminoalcohol (III), dissolved in a suitable solvent, such as DMF, in the presence of a peptide coupling reagent, such as HBTU.

Alternatively, intermediate (IX) can be obtained by using a 3-step synthesis starting from intermediate (VIII). First, the free alcohol functionality can be protected using standard protection methods, such as for example acylation to the ester. Typical conditions would be for example treating intermedi- 35 ate (VIII) with a suitable acylating agent, such as a combination of acetic anhydride and DMAP (dimethylaminopyridine), in the presence of a base, such as Et<sub>3</sub>N (triethylamine) in a suitable inert solvent, such as DCM. The so obtained intermediate (XIII) can subsequently undergo debenzylation 40 using standard deprotection methods, such as stirring in a suitable solvent, such as MeOH, under hydrogen atmosphere in the presence of a hydrogenation catalyst, such as 10% Pd/C. Pyridone intermediate (XIV) can be finally converted into intermediate (IX) by using one of the available deprotec- 45 tion methods for the chosen protecting group. In the case of protection of the alcohol as an ester, saponification using a base, such as NaOH (sodium hydroxide) in a suitable solvent, such as MeOH, can afford the desired free alcohol (IX). The skilled in the art will recognize that this method is valuable 50 when the alcohol functionality present in intermediate (VIII) could be liable under debenzylation conditions.

Experimental Procedure 6

An intermediate of formula (XVII), wherein R<sup>9</sup> is hydrogen or bromine;

and all the other variables are defined as described in the scope of the invention, can be obtained via intramolecular cyclization, for example by applying Mitsunobu conditions to intermediate (IX). The reaction can be performed by treating a solution of intermediate (IX) in a suitable inert and dry 60 solvent, such as THF, with an azadicarboxylate species, such as DIAD (diisopropyl azodicarboxylate), in the presence of a phosphine, such as triphenylphosphine, under inert atmosphere. Precooling of the solution may be used.

Alternatively, starting from intermediate (VIII) a 3-step 65 method can be used. First, the free hydroxyl function in intermediate (VIII) can be converted into a suitable leaving

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group. For example, intermediate (XV-a), where LG=chlorine, can be obtained under mild conditions by dissolving intermediate (VIII) in a suitable solvent, such as DCM, and treating it with a chlorinating agent, such as thionyl chloride. Precooling of the solution before addition of the chlorinating agent can enhance the outcome of the reaction. Intermediate (XV) can then undergo debenzylation to give intermediate (XVI), using standard methods compatible with the presence of the leaving group. In the case of intermediate (XV-a), for example, debenzylation can be achieved by treating the intermediate, dissolved in a suitable and inert solvent, such as DCM, with a Lewis acid such as BBr<sub>3</sub> (boron tribromide). Precooling of the reaction mixture before addition of the Lewis acid can enhance the reaction outcome. The person skilled in the art will notice that for some cases where R<sup>2</sup> is  $C_{1-4}$ alkyl substituted with  $C_{1-4}$ alkyloxy and/or Y is O, alternative methods to the one suggested should be considered, to avoid side reactions. Finally, intermediate (XVI) can be processed to intermediate (XVII) by using standard substitution conditions. For example, starting from intermediate (XVI-a), where LG=chlorine, the ring closure can be achieved by treating the substrate, dissolved in a suitable solvent, such as DMF, with a base, such as NaH (sodium hydride). Precooling of the reaction and a level of dilution high enough to avoid intermolecular reactions can enhance the reaction outcome. Experimental Procedures—Scheme 3

Br 
$$R^3$$
 Br  $R^3$  Br  $R^3$  Br  $R^3$   $R^3$ 

Experimental Procedure 7

An intermediate of formula (XVIII-b), wherein all the variables are defined as described in the scope of the invention, can be obtained from an intermediate of structure (XVII-b), where all the variables are defined as mentioned hereabove, with the exception of the residual -L-R¹, defined for structure (XVII-b) as any kind of protecting group suitable for an amidic nitrogen, such as, but not restricted to, a benzyl group (L=CH<sub>2</sub>, R¹=phenyl) or a PMB group (p-methoxybenzyl, L=CH<sub>2</sub>, R¹=p-methoxyphenyl). Intermediate (XVII-b) can be converted into intermediate (XVIII-b) by means of deprotection methods known to the person skilled in the art. For

example, when L=CH $_2$ , R $^1$ =phenyl, deprotection can be achieved by treating intermediate (XVII-b), dissolved in a suitable solvent, such as dry toluene, with a strong acid, such as TfOH (trifluoromethansulfonic acid). Heating the reaction mixture under stirring can enhance the reaction outcome. Experimental Procedure 8

An intermediate of formula (XVII-b), wherein all the variables are defined as described in the scope of the invention, can be obtained from intermediate (XVIII-b) by means of any manipulation known to the person skilled in the art for the functionalization of an amidic nitrogen. Two general 10 examples are reported:

#### EXAMPLE 1

When

 $R^1$  is Ar;

L is a covalent bond;

functionalization can be achieved for example by means of a copper catalyzed C—N coupling. Standard conditions, such as stirring a mixture of intermediate (XVIII-b), dissolved in a suitable solvent, such as DMF, in the presence of a base, such as K<sub>3</sub>PO<sub>4</sub> (potassium phosphate), a ligand, such as N,N'-dimethyl-1,2-cyclohexanediamine, an aryl halide and a copper catalyst, such as CuI, could be used. Degassing the reaction mixture with an inert gas, such as N<sub>2</sub> or argon, and heating the reaction mixture to high temperatures, such as reflux temperature, may enhance the reaction outcome.

#### EXAMPLE 2

When

L is one of the variables described in the scope of the invention.

with the exception of L being a covalent bond if R<sup>1</sup> is Ar; and with the exception of m and n simultaneously having 35 value 3:

functionalization can be achieved for example by treating intermediate (XVIII-b), dissolved in a suitable and inert solvent, such as DMF, with a base, such as NaH, followed by addition of an electrophile. Precooling of the reaction mixture can enhance the reaction outcome.

Experimental Procedure 9

An intermediate of formula (XVIII-a), wherein all the variables are defined as described in the scope of the invention, can be obtained from an intermediate of structure (XVII-a), where all the variables are defined as mentioned hereabove, with the exception of

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the residual -L-R $^1$ , defined for structure (XVII-a) as any kind of protecting group suitable for an amidic nitrogen, such as, but not restricted to, a benzyl group (L=CH $_2$ , R $^1$ =phenyl) or a PMB group (p-methoxybenzyl, L=CH $_2$ , R $^1$ =p-methoxyphenyl).

Intermediate (XVII-a) can be converted into intermediate (XVIII-a) by means of deprotection methods known to the person skilled in the art. For example, when L=CH<sub>2</sub>, R<sup>1</sup>=phenyl, deprotection can be achieved by treating intermediate (XVII-a), dissolved in a suitable solvent, such as dry toluene, with a strong acid, such as TfOH. Heating the reaction mixture under stirring can enhance the reaction outcome. Alternatively, an intermediate of formula (XVIII-a) can be obtained starting from an intermediate of formula (XVIII-b) 15 by means of direct bromination. Different brominating agents can be used. For example, the reaction can be performed by dissolving intermediate (XVIII-b) in a mixture of solvents such as DCM/AcOH (acetic acid) and adding bromine to the mixture, or by adding NBS (N-bromosuccinimide) to a solution of intermediate (XVIII-b) in an appropriate solvent, such as acetonitrile. The reaction mixture may be stirred under heating and inert atmosphere.

Experimental Procedure 10

An intermediate of formula (XVII-a), wherein all the variables are defined as described in the scope of the invention, can be obtained from intermediate (XVIII-a) by means of any manipulation known to the person skilled in the art for the functionalization of an amidic nitrogen. One general example is reported:

# EXAMPLE 1

When

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L is one of the variables described in the scope of the invention,

with the exception of L being a covalent bond if R<sup>1</sup> is Ar; and with the exception of m and n simultaneously having value 3;

functionalization can be achieved for example by treating intermediate (XVIII-a), dissolved in a suitable and inert solvent, such as DMF, with a base, such as NaH, followed by an electrophile. Precooling of the reaction mixture can enhance the reaction outcome.

Alternatively, intermediate (XVII-a) can be obtained by direct bromination of intermediate (XVII-b), for example by adding bromine to a solution of intermediate (XVII-b), dissolved in a mixture of solvents such as DCM/AcOH. Experimental Procedures—Scheme 4

Br 
$$R^3$$
  $R^4$   $R^4$   $R^3$   $R^3$   $R^4$   $R^3$   $R^4$   $R$ 

# Experimental Procedure 11

A compound of formula (I), wherein all the variables are defined as described in the scope of the invention, can be obtained for example by copper catalyzed C—N coupling. Standard conditions involve stirring of intermediate (XVII-a) in the presence of a copper catalyst, such as CuI, a base, such as  $Cs_2CO_3$  (cesium carbonate), the coupling partner, such as for example 4-methylimidazole, and a ligand, such as N,N'-dimethyl-1,2-cyclohexanediamine, in a suitable solvent, such as DMF. Degassing the reaction mixture with an inert gas, such as  $N_2$  or argon, and heating the reaction mixture to high temperatures, such as reflux temperature, may enhance the reaction outcome.

Alternatively, a compound of formula (I), where R<sup>5</sup> is restricted to hydrogen, can be obtained by palladium catalyzed C—N coupling. Typically, an intermediate of formula (XVII-a) is stirred and heated in the presence of a base, such as K<sub>3</sub>PO<sub>4</sub>, a palladium source, such as Pd<sub>2</sub>(dba)<sub>3</sub> (tris(dibenzylideneacetone)dipalladium(0)), a ligand, such as 2-di-tert-butylphosphino-3,4,5,6-tetramethyl-2',4',6'-triisopropyl-1, 1'-biphenyl and the desired imidazole, in the presence of a solvent or a mixture of solvents, such as toluene/dioxane. Premixing of the catalyst and the ligand followed by heating 55 before addition of the remaining reagents, degassing of the solution and heating can enhance the reaction outcome.

Alternatively, a compound of formula (I) can be obtained via a 5-step synthesis. In the first step, intermediate (XVII-a) can be converted into intermediate (XX), where PG is a mono or divalent nitrogen protecting group. For example, when PG=acetyl, the reaction can be performed using known amide coupling methodologies. For example, acetamide can be reacted with intermediate (XVII-a) in the presence of a base, such as  $K_3PO_4$ , a palladium source, such as  $Pd_2(dba)_3$ , a 65 ligand, such as  $Pd_3(dba)_3$ , a 67 ligand, such as  $Pd_3(dba)_3$ , a 68 ligand, such as  $Pd_3(dba)_3$ , a 69 ligand, such as

as dry THF. Degassing of the reaction mixture during the set-up with an inert gas, such as N2 or argon, anhydrous conditions, and the use of high temperatures, such as reflux temperature, can enhance the reaction outcome. In the second step, intermediate (XX) can be converted into the free amine intermediate (XXI) by using any deprotection method tolerated by the other functionalities present in the molecule. For example, when PG in intermediate (XX)=acetyl, an acidic hydrolysis, using for example HCl (hydrochloric acid), in a suitable solvent, such as MeOH, can be used. In the third step, the amino group in intermediate (XXI) can be acylated to give intermediate (XXII). For example, if R<sup>5</sup> in compound (XVII) represents hydrogen, formylation of intermediate (XXI) can be obtained by adding to intermediate (XXI), dissolved in a suitable inert solvent, such as THF, a formylating agent, such as a mixture of acetic anhydride and formic acid. Stirring of the reaction under heating can enhance the reaction outcome. In the fourth step, intermediate (XXII) can be converted to the cyclization precursor (XXIII) with methodologies known to the person skilled in the art and depending on the desired functionalities X and R<sup>4</sup>. For example, if in compound (XVII) X=CH and R<sup>4</sup>=alkyl, the reaction can be performed by adding the desired α-haloketone, such as for example 1-bromo-2butanone, to a mixture of intermediate (XXII), and a base, such as K<sub>2</sub>CO<sub>3</sub>, in a suitable solvent, such as DMF. If the halogen of the  $\alpha$ -haloketone is different from iodine, the reaction can be improved by means of an in-situ Filkenstein reaction, performed by adding an iodine salt, such as KI, to the reaction mixture. Finally, intermediate (XXIII) can be converted into compound (I) by means of a classical imida-

zole synthesis. Diketo precursor (XXIII) can be cyclized into

desired compound (I) in the presence of a nitrogen source,

such as ammonium acetate, and an acid, such as AcOH.

Heating the reaction to reflux temperature can enhance the

reaction outcome.

Alternatively, when the residual -L-R<sup>1</sup> in compound (I) corresponds to any kind of protecting group suitable for an amidic nitrogen, such as, but not restricted to, a benzyl group (L=CH<sub>2</sub>, R<sup>1</sup>=phenyl) or a PMB group (p-methoxybenzyl, L=CH<sub>2</sub>, R<sup>1</sup>=p-methoxyphenyl), the compound can be further converted via a two-step method to generate other structures that can be described as well with the general formula (I). In the first step, compound (I) can be converted into intermediate (XXIV) by means of deprotection methods known to the person skilled in the art. For example, when L=CH<sub>2</sub>, R<sup>1</sup>=pmethoxyphenyl, deprotection can be achieved by treating compound (I), dissolved in a suitable solvent, such as dry toluene, with a strong acid, such as TfOH. Heating the reaction mixture under stirring can enhance the reaction outcome. In the second step, intermediate (XXIV) can be converted to a compound of general formula (I), by means of known N-functionalization methods.

For example, when L is one of the variables described above:

with the exception of L being a covalent bond if  $R^1$  is Ar; and with the exception of m and n simultaneously having value 3;

one possibility would be treating intermediate (XXIV), dissolved in a suitable and inert solvent, such as DMF, with a base, such as NaH, followed by an electrophile. Precooling of the reaction mixture and anhydrous conditions can <sup>30</sup> enhance the reaction outcome.

Alternatively, intermediate (XVII-a) wherein R³ is restricted to halo (halo=Cl, Br, I), hereby called intermediate (XVII-b), may be obtained starting from intermediate (XVII-a) wherein R³ is restricted to hydrogen, hereby called (XVII-a1), via a halogenation reaction. For example, if halo is Cl in intermediate (XVII-b), the reaction can be performed by treating intermediate (XVII-a1), dissolved in a suitable solvent, such as DMF, with a chlorine source, such as NCS 40 (N-chlorosuccinimide).

Scheme 4a

Br halo

$$Z$$
 N  $Z$  N  $Z$  N  $Z$  N  $Z$  N  $Z$  N  $Z$  Solution (XVII-a1)

(XVII-a1)

 $Z$  N  $Z$  N  $Z$  N  $Z$  N  $Z$  N  $Z$  Solution (XVII-b)

Alternatively, a compound of formula (I) bearing an  $R^1$  group which can undergo further manipulation, could be converted into other compounds described as well with the general formula (I), by mean of one or several subsequent chemical transformations known to the person skilled in the art. For example, when  $R^1$ =indole, the indolic nitrogen could undergo methylation, for example by treating the compound, dissolved in a suitable and inert solvent, such as DMF, with a base, such as NaH, followed by a methylating agent, such as 65 iodomethane. Precooling of the reaction mixture and anhydrous conditions can enhance the reaction outcome.

Experimental Procedures—Scheme 5

Experimental Procedure 12

A compound of formula (I), wherein

 $R^2$  is  $C_{1-4}$ alkyl substituted with hydroxyl;

and all the other variables are defined as described in the scope of the invention, hereby named a compound of for-

(I-b)

mula (XXV), can be obtained from a compound of structure (I), where R<sup>2</sup> is restricted to C<sub>1.4</sub>alkyl substituted with a protected hydroxyl group, such as for example a methoxy group (PG=methyl), hereby named compound (I-a). The conversion of the ether to the alcohol can be achieved for example by treating compound (I-a), dissolved in a suitable solvent, such as DCM, with a Lewis acid, such as BBr<sub>3</sub>.

reaction can be performed by adding dimethylamine to a solution of intermediate (XXVI), dissolved in a suitable solvent, such as DMF. Precooling of the reaction mixture before addition, followed by heating after the addition is complete, can enhance the reaction outcome.

Experimental Procedures—Scheme 6

$$CI \xrightarrow{N} CI \xrightarrow{PG} OCH_3$$

$$CI \xrightarrow{N} CI$$

$$CI \xrightarrow$$

The person skilled in the art will notice that for some cases <sup>40</sup> Experimental Procedure 15 where R<sup>2</sup> is C<sub>1-4</sub>alkyl substituted with C<sub>1-4</sub>alkyloxy and/or Y is O, PG=alkyl should be avoided, because of possible side reactions during the deprotection.

Experimental Procedure 13

An intermediate of structure (XXVI), wherein R<sup>2</sup> is C<sub>1-4</sub>alkyl substituted with a leaving group;

and all the other variables are defined as described in the scope of the invention, can be obtained starting from compound (XXV) by means of conversion of the hydroxyl 50 group into a leaving group, for which different methodologies, known to the person skilled in the art, are available. For example, if LG=Ms (mesylate), intermediate (XXVI) can be obtained by treating compound (XXV), dissolved in a suitable solvent, such as DCM, with a base, such as 55 DIPEA, and a suitable mesylating agent, such as methansulphonyl chloride. Precooling of the solution of compound (XXV) before addition of the other reagents can enhance the reaction outcome.

Experimental Procedure 14

A compound of formula (I), wherein  $R^2$  is  $C_{1-4}$ alkyl substituted with  $NR^7R^8$ :

and all the other variables are defined as described in the scope of the invention, hereby named compound (I-b), can be obtained starting from intermediate (XXVI), by nucleophilic substitution using a mono or disubstituted amine. For example, if  $R^7 = R^8 = methyl$  in compound (I-b), the

A compound of formula (I-d), wherein R<sup>3</sup> is fluorine, chlorine or bromine;

and all the other variables are defined as described in the scope of the invention, can be obtained via a multi-step synthesis, starting from the commercially available 2,6-dichloro-4-methoxypyridine (XXVII).

In the first step, one of the chlorine atoms can undergo substitution with an alkoxy group, following conditions known to the person skilled in the art, such as treating the substituted pyridine (XXVII), dissolved in a suitable inert solvent, such as THF, in the presence of a base, such as NaH, with a suitable electrophile. PG has to be a moiety which can be further removed with simple manipulations, such as, for example, a benzyl or a p-methoxybenzyl group. The so-obtained intermediate (XXVIII) can then be converted into intermediate (XXIX), where M is a metal, by mean for example of a cross-coupling reaction to give the corresponding 2-vinyl pyridine: the newly installed double bond can then 60 be oxidized, for example by using potassium permanganate, to obtain the carboxyl functionality. Salt (XXIX) can then undergo a series of reactions, under conditions similar to the ones described in Scheme 2, to afford the bicyclic scaffold (XXX), which can undergo selective halogenation, by using for example NIS (N-iodosuccinimide), to give intermediate (XXXI), where halo is a halogen, preferably iodine. Under the conditions described in Scheme 4, intermediate (XXXI)

can then be converted to compound (I-c). The methoxy moiety of compound (I-c) can then finally be displaced to afford compound (I-d), using a suitable halogen oxychloride, such

as POCl<sub>3</sub> (phosphorus oxychloride) in the presence of a suitable inert solvent, such as acetonitrile. Experimental Procedures—Scheme 7

# Scheme 6

$$(XXXII) \qquad O \qquad O \qquad O \qquad N$$

$$(XXXIII) \qquad (XXXIII) \qquad (XXXIV)$$

$$O = \bigcap_{N \to \mathbb{R}^{12}} \bigcap_{N \to$$

-continued 
$$\mathbb{R}^4$$
 $\mathbb{R}^4$ 
 $\mathbb{R}^4$ 
 $\mathbb{R}^3$ 
 $\mathbb{R}^5$ 
 $\mathbb{R}^5$ 
 $\mathbb{R}^5$ 
 $\mathbb{R}^5$ 
 $\mathbb{R}^2$ 
 $\mathbb{R}^12$ 
 $\mathbb{R}^{12}$ 
 $\mathbb{R}^{12}$ 
 $\mathbb{R}^{12}$ 
 $\mathbb{R}^{12}$ 
 $\mathbb{R}^{12}$ 

Experimental Procedure 16

A compound of formula (I-e), wherein

 $R^{12}$  describes the pattern of substituents allowed on the indolyl ring, as defined in the scope of the invention, or groups from which the allowed substituents can be obtained by mean of simple manipulations, such as for example cross-coupling reactions or dehalogenation reactions;

 $L^1$  is  $CH_2$  or  $CH(CH_3)$ ;

and all the other variables are defined as described in the scope of the invention, can be obtained via a multi-step syn- 35 thesis, starting from indole (XXXII), which can be commercially available or obtained by methods known to the person skilled in the art.

In the first step, indole (XXXII) can be functionalized on position 3 with a carbonyl group by mean of methods reported 40 in the literature, such as for example a Vilsmeier-Haack reaction, to give intermediate (XXXIII). The indol nitrogen of intermediate (XXXIII) can then be protected with a protecting group, such as a tert-butoxycarbonyl, under conditions known to the person skilled in the art, to give intermediate 45 (XXXIV).

The carbonyl group in intermediate (XXXIV) can be subsequently reduced to the corresponding alcohol by treating it with a reducing agent, such as for example NaBH<sub>4</sub>, in the presence of a suitable solvent, such as for example MeOH, to 50 give intermediate (XXXV). The alcohol group in intermediate (XXXV) can then be converted into a leaving group, via a reaction for which different methodologies, known to the person skilled in the art, are available. For example, if LG=Ms (mesylate), intermediate (XXXVII) can be obtained by treat- 55 ing intermediate (XXXV), dissolved in a suitable solvent, such as DCM, with a base, such as DIPEA, and a suitable mesylating agent, such as methanesulphonyl chloride. Alternatively, if LG=Cl, intermediate (XXXVII) can be obtained by treating intermediate (XXXV), dissolved in a suitable 60 solvent, such as DCM, in the present of a chlorinating agent, such as thionyl chloride. Precooling of the reaction mixture can enhance the reaction outcome.

Alternatively, intermediate (XXXIV) can be treated with a organometallic species such as for example methylmagnesium chloride, in the presence of a suitable solvent, such as THF, to convert the carbonyl compound into intermediate

(XXXVI). Precooling of the reaction mixture can enhance the reaction outcome. The alcohol group in intermediate (XXXVI) can then be converted into a leaving group, via a reaction for which different methodologies, known to the person skilled in the art, are available. For example, if LG=Ms (mesylate), intermediate (XXXVIII) can be obtained by treating intermediate (XXXVII), dissolved in a suitable solvent, such as DCM, with a base, such as DIPEA, and a suitable mesylating agent, such as methanesulphonyl chloride. Alternatively, if LG=Cl, intermediate (XXXVIII) can be obtained by treating intermediate (XXXVII), dissolved in a suitable solvent, such as DCM, in the present of a chlorinating agent, such as thionyl chloride. Precooling of the reaction mixture can enhance the reaction outcome.

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Intermediate (XXXIX) can be obtained by reacting intermediate (XXIV) where R<sup>3</sup> is H, with the suitable intermediate (XXXVII) or (XXXVIII), depending on which L' is desired in the final compound. Standard reaction conditions would involve the use of a base, such as NaH, in the presence of a suitable inert solvent, such as DMF. Precooling of the reaction mixture can enhance the reaction outcome.

The protected intermediate (XXXIX) can be finally converted into final compound (I-e) by mean of standard deprotection conditions. The person skilled in the art will appreciate that intermediate (XXXIX), when R<sup>12</sup> describes groups from which the substituents on the indolyl ring defined in the scope can be obtained by mean of simple manipulations, is normally the intermediate of choice to perform the requested manipulations.

Starting materials can be obtained commercially or can be prepared by those skilled in the art.

In order to obtain the HCl salt forms of the compounds, several procedures known to those skilled in the art can be used. In a typical procedure, for example, the free base can be dissolved in DIPE or Et<sub>2</sub>O and subsequently, a 6 N HCl solution in 2-propanol or a 1 N HCl solution in Et<sub>2</sub>O can be added dropwise. The mixture typically is stirred for 10 minutes after which the product can be filtered off. The HCl salt usually is dried in vacuo.

Where necessary or desired, any one or more of the following further steps in any order may be performed:

Compounds of Formula (I) and any subgroup thereof may be converted into further compounds of Formula (I) and any subgroup thereof, using procedures known in the art.

It will be appreciated by those skilled in the art that in the processes described above the functional groups of interme-5 diate compounds may need to be blocked by protecting groups. In case the functional groups of intermediate compounds were blocked by protecting groups, they can be deprotected after a reaction step.

In all these preparations, the reaction products may be 10 isolated from the reaction medium and, if necessary, further purified according to methodologies generally known in the art such as, for example, extraction, crystallization, trituration and chromatography. In particular, stereoisomers can be isolated chromatographically using a chiral stationary phase 15 such as, for example, Chiralpak® AD (amylose 3,5 dimethylphenyl carbamate) or Chiralpak® AS, both purchased from Daicel Chemical Industries, Ltd, in Japan, or by Supercritical Fluid Chromatography (SFC).

The chirally pure forms of the compounds of Formula (I) 20 form a preferred group of compounds. It is therefore that the chirally pure forms of the intermediates and their salt forms are particularly useful in the preparation of chirally pure compounds of Formula (I). Also enantiomeric mixtures of the intermediates are useful in the preparation of compounds of 25 Formula (I) with the corresponding configuration. Pharmacology

It has been found that the compounds of the present invention modulate the  $\gamma$ -secretase activity. The compounds according to the invention and the pharmaceutically acceptable compositions thereof therefore may be useful in the treatment or prevention of AD, TBI, dementia pugilistica, MCI, senility, dementia, dementia with Lewy bodies, cerebral amyloid angiopathy, multi-infarct dementia, Down's syndrome, dementia associated with Parkinson's disease and 35 dementia associated with beta-amyloid; preferably AD.

The compounds according to the present invention and the pharmaceutically acceptable compositions thereof may be useful in the treatment or prevention of a disease or condition selected from the group consisting of AD, TBI, dementia 40 pugilistica, MCI, senility, dementia, dementia with Lewy bodies, cerebral amyloid angiopathy, multi-infarct dementia, Down's syndrome, dementia associated with Parkinson's disease and dementia associated with beta-amyloid.

As used herein, the term "modulation of  $\dot{\gamma}$ -secretase activity" refers to an effect on the processing of APP by the  $\gamma$ -secretase-complex. Preferably it refers to an effect in which the overall rate of processing of APP remains essentially as without the application of said compounds, but in which the relative quantities of the processed products are changed, 50 more preferably in such a way that the amount of the A $\beta$ 42-peptide produced is reduced. For example a different Abeta species can be produced (e.g. Abeta-38 or other Abeta peptide species of shorter amino acid sequence instead of Abeta-42) or the relative quantities of the products are different (e.g. the 55 ratio of Abeta-40 to Abeta-42 is changed, preferably increased)

It has been previously shown that the  $\gamma$ -secretase complex is also involved in the processing of the Notch-protein. Notch is a signaling protein which plays a crucial role in developmental processes (e.g. reviewed in Schweisguth F (2004) Curr. Biol. 14, R129). With respect to the use of  $\gamma$ -secretase modulators in therapy, it seems particularly advantageous not to interfere with the Notch-processing activity of the  $\gamma$ -secretase activity in order to avoid putative undesired side-effects. 65 While  $\gamma$ -secretase inhibitors show side effects due to concomitant inhibition of Notch processing,  $\gamma$ -secretase modu-

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lators may have the advantage of selectively decreasing the production of highly aggregatable and neurotoxic forms of  $A\beta$ , i.e.  $A\beta42$ , without decreasing the production of smaller, less aggregatable forms of  $A\beta$ , i.e.  $A\beta38$  and without concomitant inhibition of Notch processing. Thus, compounds are preferred which do not show an effect on the Notch-processing activity of the  $\gamma$ -secretase-complex.

As used herein, the term "treatment" is intended to refer to all processes, wherein there may be a slowing, interrupting, arresting, or stopping of the progression of a disease, but does not necessarily indicate a total elimination of all symptoms.

The invention relates to compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof, and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for use as a medicament.

The invention also relates to compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for use in the modulation of  $\gamma$ -secretase activity.

The invention also relates to compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof, and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for use in the treatment or prevention of diseases or conditions selected from the group consisting of AD, TBI, dementia pugilistica, MCI, senility, dementia, dementia with Lewy bodies, cerebral amyloid angiopathy, multi-infarct dementia, Down's syndrome, dementia associated with Parkinson's disease and dementia associated with beta-amyloid.

In an embodiment, said disease or condition is preferably AD.

The invention also relates to compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof, and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for use in the treatment of said diseases.

The invention also relates to compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof, and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for the treatment or prevention of said diseases.

The invention also relates to compounds according to the general formula (I), the tautomers and the stereoisomeric forms thereof, and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for the treatment or prevention, in particular treatment, of  $\gamma$ -secretase mediated diseases or conditions.

The invention also relates to the use of compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof, and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for the manufacture of a medicament.

The invention also relates to the use of compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for the manufacture of a medicament for the modulation of  $\gamma$ -secretase activity.

The invention also relates to the use of compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for the manufacture of a medicament for the treatment or prevention of any one of the disease conditions mentioned hereinbefore.

The invention also relates to the use of compounds according to the general Formula (I), the tautomers and the stereoisomeric forms thereof and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, for the manufacture of a medicament for the treatment of any one of 5 the disease conditions mentioned hereinbefore.

In the invention, particular preference is given to compounds of Formula (I), or any subgroup thereof with a IC<sub>50</sub> value for the inhibition of the production of A $\beta$ 42-peptide of less than 1000 nM, preferably less than 100 nM, more preferably less than 50 nM, even more preferably less than 20 nM as determined by a suitable assay, such as the assay used in the Examples below.

The compounds of Formula (I) can be administered to of any one of the diseases mentioned hereinbefore.

In view of the utility of the compounds of Formula (I), there is provided a method of treating warm-blooded animals, including humans, suffering from or a method of preventing warm-blooded animals, including humans, to suffer from any 20 one of the diseases mentioned hereinbefore.

Said methods comprise the administration, i.e. the systemic or topical administration, preferably oral administration, of an effective amount of a compound of Formula (I), the tautomers and the stereoisomeric forms thereof and the phar- 25 maceutically acceptable acid or base addition salts and the solvates thereof, to warm-blooded animals, including humans.

The present invention also concerns to the use of compounds of Formula (I) for the modulation of γ-secretase activ- 30 ity resulting in a decrease in the relative amount of A $\beta$ 42peptides produced.

An advantage of the compounds or a part of the compounds of the present invention may be their enhanced CNS-penetra-

Those of skill in the treatment of such diseases could determine the effective therapeutic daily amount from the test results presented hereinafter. An effective therapeutic daily amount would be from about 0.005 mg/kg to 50 mg/kg, in particular 0.01 mg/kg to 50 mg/kg body weight, more in 40 particular from 0.01 mg/kg to 25 mg/kg body weight, preferably from about 0.01 mg/kg to about 15 mg/kg, more preferably from about 0.01 mg/kg to about 10 mg/kg, even more preferably from about 0.01 mg/kg to about 1 mg/kg, most preferably from about 0.05 mg/kg to about 1 mg/kg body 45 weight. The amount of a compound according to the present invention, also referred to here as the active ingredient, which is required to achieve a therapeutically effect will of course, vary on case-by-case basis, for example with the particular compound, the route of administration, the age and condition 50 of the recipient, and the particular disorder or disease being

A method of treatment may also include administering the active ingredient on a regimen of between one and four intakes per day. In these methods of treatment the compounds 55 according to the invention are preferably formulated prior to administration. As described herein below, suitable pharmaceutical formulations are prepared by known procedures using well known and readily available ingredients.

The compounds of the present invention, that can be suit- 60 able to treat or prevent Alzheimer's disease or the symptoms thereof, may be administered alone or in combination with one or more additional therapeutic agents. Combination therapy includes administration of a single pharmaceutical dosage formulation which contains a compound of Formula 65 (I) and one or more additional therapeutic agents, as well as administration of the compound of Formula (I) and each

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additional therapeutic agents in its own separate pharmaceutical dosage formulation. For example, a compound of Formula (I) and a therapeutic agent may be administered to the patient together in a single oral dosage composition such as a tablet or capsule, or each agent may be administered in separate oral dosage formulations.

While it is possible for the active ingredient to be administered alone, it is preferable to present it as a pharmaceutical composition.

Accordingly, the present invention further provides a pharmaceutical composition comprising a pharmaceutically acceptable carrier and, as active ingredient, a therapeutically effective amount of a compound according to Formula (I).

The carrier or diluent must be "acceptable" in the sense of mammals, preferably humans for the treatment or prevention 15 being compatible with the other ingredients of the composition and not deleterious to the recipients thereof.

> For ease of administration, the subject compounds may be formulated into various pharmaceutical forms for administration purposes. The compounds according to the invention, in particular the compounds according to Formula (I), the tautomers and the stereoisomeric forms thereof and the pharmaceutically acceptable acid or base addition salts and the solvates thereof, or any subgroup or combination thereof may be formulated into various pharmaceutical forms for administration purposes. As appropriate compositions there may be cited all compositions usually employed for systemically administering drugs.

To prepare the pharmaceutical compositions of this invention, an effective amount of the particular compound as the active ingredient is combined in intimate admixture with a pharmaceutically acceptable carrier, which carrier may take a wide variety of forms depending on the form of preparation desired for administration. These pharmaceutical compositions are desirable in unitary dosage form suitable, in particular, for administration orally, rectally, percutaneously, by parenteral injection or by inhalation. For example, in preparing the compositions in oral dosage form, any of the usual pharmaceutical media may be employed such as, for example, water, glycols, oils, alcohols and the like in the case of oral liquid preparations such as suspensions, syrups, elixirs, emulsions and solutions; or solid carriers such as starches, sugars, kaolin, diluents, lubricants, binders, disintegrating agents and the like in the case of powders, pills, capsules and tablets. Because of their ease in administration, tablets and capsules represent the most advantageous oral dosage unit forms in which case solid pharmaceutical carriers are obviously employed. For parenteral compositions, the carrier will usually comprise sterile water, at least in large part, though other ingredients, for example, to aid solubility, may be included. Injectable solutions, for example, may be prepared in which the carrier comprises saline solution, glucose solution or a mixture of saline and glucose solution. Injectable solutions, for example, may be prepared in which the carrier comprises saline solution, glucose solution or a mixture of saline and glucose solution. Injectable solutions containing compounds of Formula (I) may be formulated in an oil for prolonged action. Appropriate oils for this purpose are, for example, peanut oil, sesame oil, cottonseed oil, corn oil, soybean oil, synthetic glycerol esters of long chain fatty acids and mixtures of these and other oils. Injectable suspensions may also be prepared in which case appropriate liquid carriers, suspending agents and the like may be employed. Also included are solid form preparations that are intended to be converted, shortly before use, to liquid form preparations. In the compositions suitable for percutaneous administration, the carrier optionally comprises a penetration enhancing agent and/or a suitable wetting agent, optionally combined

with suitable additives of any nature in minor proportions, which additives do not introduce a significant deleterious effect on the skin. Said additives may facilitate the administration to the skin and/or may be helpful for preparing the desired compositions. These compositions may be administered in various ways, e.g., as a transdermal patch, as a spoton, as an ointment. Acid or base addition salts of compounds of Formula (I) due to their increased water solubility over the corresponding base or acid form, are more suitable in the preparation of aqueous compositions.

It is especially advantageous to formulate the aforementioned pharmaceutical compositions in unit dosage form for ease of administration and uniformity of dosage. Unit dosage form as used herein refers to physically discrete units suitable as unitary dosages, each unit containing a predetermined quantity of active ingredient calculated to produce the desired therapeutic effect in association with the required pharmaceutical carrier. Examples of such unit dosage forms are tablets (including scored or coated tablets), capsules, pills, powder packets, wafers, suppositories, injectable solutions or suspensions and the like, and segregated multiples thereof.

Since the compounds according to the invention are potent orally administrable compounds, pharmaceutical compositions comprising said compounds for administration orally are especially advantageous.

In order to enhance the solubility and/or the stability of the compounds of Formula (I) in pharmaceutical compositions, it can be advantageous to employ  $\alpha$ -,  $\beta$ - or  $\gamma$ -cyclodextrins or their derivatives, in particular hydroxyalkyl substituted cyclodextrins, e.g. 2-hydroxypropyl- $\beta$ -cyclodextrin or sulfobutyl- $\beta$ -cyclodextrin. Also co-solvents such as alcohols may improve the solubility and/or the stability of the compounds according to the invention in pharmaceutical compositions.

Depending on the mode of administration, the pharmaceutical composition will preferably comprise from 0.05 to 99% by weight, more preferably from 0.1 to 70% by weight, even more preferably from 0.1 to 50% by weight of the compound of Formula (I), and, from 1 to 99.95% by weight, more preferably from 30 to 99.9% by weight, even more preferably from 50 to 99.9% by weight of a pharmaceutically acceptable carrier, all percentages being based on the total weight of the composition.

The following examples illustrate the present invention.

# **EXAMPLES**

Hereinafter, the term "DCM" means dichloromethane; "MeOH" means methanol; "LCMS" means Liquid Chromatography/Mass spectrometry; "HPLC" means high-performance liquid chromatography; "sol." means solution; "sat." means saturated; "aq." means aqueous; "r.t." means room temperature; "CO" means carbon monoxide; "AcOH" means acetic acid; "TFA" means trifluoroacetic acid; "m.p." means 55 melting point; "N2" means nitrogen; "RP" means reversed phase; "min" means minute(s); "h" means hour(s); "EtOAc" means ethyl acetate; "Et<sub>3</sub>N" means triethylamine; "EtOH" means ethanol; "eq." means equivalent; "r.m." means reaction mixture(s); "DIPE" means diisopropyl ether; "THF" means tetrahydrofuran; "DMF" means N,N-dimethyl formamide; 'iPrOH" means 2-propanol; "NH3" means ammonia; "SFC' means Supercritical Fluid Chromatography; "TBAF" means tetrabutylammonium fluoride; "OR" means optical rotation; 65 "DIPEA" means diisopropylethylamine; "TfOH" means trifluoromethanesulfonic acid; "v/v" means volume/volume %;

"Et<sub>2</sub>O" means diethyl ether; "Cs<sub>2</sub>CO<sub>3</sub>" means cesium carbonate; "DIAD" means diisopropyl azodicarboxylate; "DMAP" means 4-dimethylamino-pyridine; "HBTU" means 0-Benzotriazole-N,N,N',N'-tetramethyl-uronium-

hexafluoro-phosphate; "CO2" means carbon dioxide; "iPrNH<sub>2</sub>" means isopropylamine; "EDTA" means ethylenediaminetetraacetic acid; "HCl" means hydrochloric acid; "K<sub>2</sub>CO<sub>3</sub>" means potassium carbonate; "K<sub>3</sub>PO<sub>4</sub>" means potassium phosphate; "KOH" means potassium hydroxide; "MgSO<sub>4</sub>" means magnesium sulphate; "Na<sub>2</sub>SO<sub>4</sub>" means sodium sulphate; "NaBH4" means sodium borohydride; "LiAlH<sub>4</sub>" means lithium aluminium hydride; "NaHCO<sub>3</sub>" means sodium hydrogencarbonate; "NaOH" means sodium hydroxide; "MgCO3" means magnesium carbonate; "NCS" means N-chlorosuccinimide; "NIS" means N-iodosuccinimide; "NH<sub>4</sub>Cl" means ammonium chloride; "NaCNBH<sub>3</sub>" means sodium cyanoborohydride; "Pd/C" means palladium on carbon; "Pd2(dba)3" means tris(dibenzylideneacetone)dipalladium(0); "PdCl<sub>2</sub>(dppf)" means [1,1-bis(diphenylphosphino)ferrocene|dichloropalladium(II); "Xantphos" means 4,5-Bis(diphenylphosphino)-9,9-dimethylxanthene;

25 "RuPhos" means 2-dicyclohexylphosphino-2',6'-diisopropoxybiphenyl; "Pd(OAc)<sub>2</sub>" means palladium (II) acetate; "X-Phos" means 2-dicyclohexyl-phosphino-2',4',6'-triisopropylbiphenyl; "dppf" means 1,1-bis(diphenylphosphino) ferrocene; "EDTA" means ethylenediaminetetraacetic acid; "TBSC1" means tert-butyldimethylsilyl chloride; "tBuOH" means tert-butanol; "w/v %" means percent weight to volume; "MsCl" means methansulphonyl chloride; "TLC" means thin layer chromatography and "TPP" means triphenylphospine.

# A. Preparation of the Intermediates

# Example A1

# a) Preparation of Intermediate 1

3,5-Bis(trifluoromethyl)benzaldehyde (1.7 mL, 10.2 mmol) was dissolved in MeOH (51 mL), then (R)-(-)-2-amino-1-butanol (1.05 mL, 11.22 mmol) and NaHCO<sub>3</sub> (0.7 g, 20.4 mmol) were added. The r.m. was stirred at 80° C. for 2 h, and was then cooled to 25° C. NaBH<sub>4</sub> (0.386 g, 10.2 mmol) was added portionwise by keeping the temperature at 25° C. The mixture was stirred at 25° C. for 1 h, then quenched with HCl 2N (pH=1) and NaHCO<sub>3</sub> (pH=7-8). MeOH was evaporated in vacuo, then EtOAc was added. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo to yield a white solid. The crude intermediate 1 (R-enantiomer) was used as such in the next step (quantitative yield).

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To a suspension of intermediate 1 (crude material,  $10.2 \, \text{mmol}$ ) and imidazole (2.1 g,  $30.6 \, \text{mmol}$ ) in DCM ( $30 \, \text{mL}$ ) was added TBSC1 ( $1.6 \, \text{g}$ ,  $10.71 \, \text{mmol}$ ), and the r.m. was stirred at r.t. for 4 h. DCM was added and the organic layer was washed with aq. sat. NaHCO<sub>3</sub>, then collected, dried over MgSO<sub>4</sub>, filtered and evaporated under reduced pressure. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo to yield intermediate 2 (R-enantiomer) as a colorless oil ( $3.7 \, \text{g}$ , 84%).

# c) Preparation of Intermediate 3

HBTU (4.2 g, 11.2 mmol) was added to a stirred solution of 6-(benzyloxy)pyridine-2-carboxylic acid (1.97 g, 8.61 mmol), DIPEA (1.93 mL, 11.2 mmol) and intermediate 2 (3.7 g, 8.61 mmol) in DMF (40 mL). The r.m. was stirred at r.t. overnight, then aq. sat. NaHCO $_3$  was added and the mixture was extracted with EtOAc. The combined extracts were dried over MgSO $_4$  and concentrated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo to yield intermediate 3 as an oil (4.42 g, 80%; R-enantiomer).

# d) Preparation of Intermediate 4

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TBAF (3.3 g, 10.34 mmol) was added to a solution of intermediate 3 (4.42 g, 6.9 mmol) in THF (21 mL). The r.m. was stirred at r.t. for 1 h, then diluted with water and extracted with EtOAc. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents were evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 50/50). The desired fractions were collected and concentrated in vacuo to yield intermediate 4 as a colourless oil (2.4 g, 66%; R-enantiomer).

# e) Preparation of Intermediate 5

$$F_3C$$
  $HO$   $HO$   $CF_3$   $N$   $N$ 

10% Pd/C (0.240 g) was added to a solution of intermediate
25 4 (2.4 g, 4.56 mmol) in MeOH (14 mL) at 0° C. The mixture
was hydrogenated (atmospheric pressure) at r.t. for 2 h. The
catalyst was filtered through diatomaceous earth and the solvent evaporated in vacuo to yield a colorless oil. The crude
was used such in the next reaction step (quantitative yield;
30 R-enantiomer).

# f) Preparation of Intermediate 6

DIAD (1.3 mL, 6.84 mmol) was added to a stirred solution of intermediate 5 (crude material, 4.56 mmol) and TPP (1.8 g, 6.84 mmol) in dry THF (14 mL) under  $N_2$  at 0° C. The mixture was stirred at r.t. for 4 h. The solvents were then evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 50/50). The desired fractions were collected and concentrated in vacuo to afford intermediate 6 (quantitative yield; R-enantiomer).

# g) Preparation of Intermediate 7

$$\begin{array}{c|c} F_3C & & O \\ \hline \\ CF_3 & & N \end{array}$$

Bromine (0.280 mL, 5.47 mmol) was added dropwise slowly to a stirred solution of intermediate 6 (4.56 mmol) in

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DCM/AcOH 4:1 (40 mL) under N<sub>2</sub>. The mixture was stirred at r.t. overnight, then diluted with aq. sat. NaHCO<sub>3</sub> and extracted with DCM. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 50/50). The desired fractions were collected and concentrated in vacuo to yield intermediate 7 as an oil (1.26 g, 55%; R-enantiomer).

# Example A2

# a) Preparation of Intermediate 8

p-Anisaldehyde (7.4 mL, 61 mmol) was dissolved in  $_{25}$  MeOH (300 mL), then D-alaninol (5.0 g, 66 mmol) and NaHCO $_{3}$  (10.2 g, 121 mmol) were added and the reaction stirred at 80° C. for 2 h. The r.m. was then cooled to 25° C. NaBH $_{4}$  (2.3 g, 61 mmol) was added portionwise while keeping the temperature below 25° C. The mixture was stirred at 25° C. for 1 additional h, then quenched with HCl 2N (pH=1) and NaHCO $_{3}$  (pH=7-8). MeOH was evaporated in vacuo, then EtOAc was added. The organic layer was separated, dried over MgSO $_{4}$ , filtered and the solvents evaporated in vacuo to yield intermediate 8 as a white solid (quantitative yield; R-enantiomer).

# b) Preparation of Intermediate 9

HBTU (2.15 g, 5.67 mmol) was added to a stirred solution of 6-(benzyloxy)pyridine-2-carboxylic acid (1 g, 4.36 mmol), DIPEA (0.98 mL, 5.68 mmol) and intermediate 8 (0.852 g, 4.36 mmol) in DMF (12 mL). The mixture was stirred at r.t. overnight. Sat. aq. NaHCO $_3$  was added and the mixture was extracted with EtOAc. The combined extracts were dried over MgSO $_4$  and concentrated in vacuo. The crude for product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 50/50). The desired fractions were

collected and concentrated in vacuo to yield intermediate 9 as an oil (1.51 g, 85%; R-enantiomer).

#### c) Preparation of Intermediate 10

10% Pd/C (0.187 g) was added to a solution of intermediate 9 (1.87 g, 4.60 mmol) in MeOH (20 mL) at 0° C. The mixture was hydrogenated (atmospheric pressure) at r.t. for 6 h. The catalyst was filtered through diatomaceous earth and the solvent evaporated in vacuo to yield a colorless oil. The crude intermediate 10 was used as such in the next reaction step (quantitative yield; R-enantiomer).

# d) Preparation of Intermediate 11

$$\bigcap_{N \in \mathbb{N}} \bigcap_{N \in \mathbb{N}} \bigcap_{$$

DIAD (1.36 mL, 6.90 mmol) was added to a stirred solution of intermediate 10 (crude material, 4.60 mmol) and TPP (1.8 g, 6.86 mmol) in dry THF (20 mL) under  $N_2$ . The mixture was stirred at r.t. overnight. The solvents were then evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 100/0). The desired fractions were collected and concentrated in vacuo to afford intermediate 11 as a white solid (927 mg, 68% over two steps; R-enantiomer).

# e) Preparation of Intermediate 12

Lithium hydroxide monohydrate (0.766 g, 18.25 mmol) was added portionwise to a stirred solution of methyl 3-bromo-2-hydroxy-6-pyridinecarboxylate (3.85 g, 16.6 mmol) in a mixture of THF (66 mL) and water (17 mL). The mixture was stirred at 60° C. for 24 h, then the solvent was

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**68** h) Preparation of Intermediate 15

evaporated in vacuo. The crude intermediate 12 was dried in vacuo and used as such in the next reaction step (quantitative yield; R-enantiomer).

# f) Preparation of Intermediate 13

$$\bigcap_{O} \bigcap_{N \in \mathcal{N}} \bigcap_{N \in \mathcal{N}} \operatorname{Br}$$

Bromine (0.17 mL, 3.32 mmol) was added dropwise slowly to a stirred solution of intermediate 11 (0.825 g, 2.76 mmol) in DCM/AcOH 4:1 (15 mL) under  $N_2$ . The mixture was stirred at r.t. overnight, then diluted with aq. sat. NaHCO $_3$  and extracted with DCM. The organic layer was separated, dried over MgSO $_4$ , filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 50/50). The 25 desired fractions were collected and concentrated in vacuo to yield intermediate 13 as an oil (530 mg, 51%; R-enantiomer).

#### f1) Alternative Preparation of Intermediate 13

HBTU (16.2 g, 42.66 mmol) was added portionwise to a stirred solution of intermediate 12 (crude material, 28.44 mmol), intermediate 8 (5.55 g, 28.44 mmol) and DIPEA (7.3 mL, 42 mmol) in DMF (24 mL). The mixture was stirred at r.t. 35 for 14 h, then 0.5 additional eq. of HBTU and DIPEA were added. The mixture was stirred at r.t. for 4 h, then poured into aq. sat. NaHCO $_3$  and extracted with EtOAc. The organic layer was separated, dried over MgSO $_4$ , filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 50/50). The desired fractions were collected and concentrated in vacuo to yield intermediate 13 as an oil (7.03 g, 65% over two steps; R-enantiomer).

# g) Preparation of Intermediate 14

Trifluoromethanesulfonic acid (0.5 mL, 5.62 mmol) was added to a stirred solution of intermediate 13 (0.53 g, 1.4 60 mmol) in toluene (5 mL). The mixture was stirred at reflux for 2 h, then diluted with NaOH 1M to pH=8 and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/DCM 0/100 to 100/0). The desired fractions were collected and concentrated in 65 vacuo to yield intermediate 14 as a solid in quantitative yield (R-enantiomer).

Intermediate 14 (1.58 g, 6.14 mmol) and sodium hydride (60% in mineral oil, 0.270 g, 6.76 mmol) were dissolved in DMF (18 mL) at 0° C., then 3,5-bis(trifluoromethyl)benzyl bromide (1.24 mL, 6.76 mmol) was added. The mixture was stirred at r.t. for 4 h. EtOAc and water were added. The organic phase was separated and dried over MgSO<sub>4</sub>, filtered and concentrated to dryness. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 70/30). The desired fractions were collected and concentrated in vacuo to yield intermediate 15 as an oil (2.8 g, 94%; R-enantiomer).

# i) Preparation of Intermediate 16

$$\bigcap_{NH} \bigcap_{N} \bigcap_{R} \bigcap_{F_3C} CF_3$$

 $\rm K_3PO_4$  (0.721 g, 3.40 mmol),  $\rm Pd_2(dba)_3$  (0.062 g, 0.07 mmol) and Xantphos (0.068 g, 0.12 mmol) were added to a stirred solution of intermediate 15 (0.821 g, 1.70 mmol) in dry THF (5 mL) at r.t., while  $\rm N_2$  was bubbled through the solution. After 10 min, acetamide (0.11 g, 1.86 mmol) was added and the mixture was stirred at r.t. for another 10 min, then heated for 3 h at 90° C. The reaction was then cooled to r.t., sat. aq. NaHCO\_3 and EtOAc added, the phases were separated and the aq. phase was extracted once more with EtOAc. The combined organic layers were dried over MgSO\_4, filtered and evaporated. The crude was purified by flash column chromatography (silica; EtOAc/heptane 0/100 to 70/30). The desired fractions were collected and concentrated in vacuo to yield intermediate 16 as a pale brown solid (0.708 g, 90%; R-enantiomer).

# 1) Preparation of Intermediate 17

$$H_2N$$
 $N$ 
 $R$ 
 $F_3C$ 
 $CF_3$ 

HCl (6N in iPrOH, 0.853 mL, 5.12 mmol) was added to a solution of intermediate 16 (0.787 g, 1.71 mmol) in MeOH (10 mL) at r.t. The r.m. was stirred overnight at r.t. The solvent

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was evaporated, and sat. aq. NaHCO<sub>3</sub> and EtOAc were added. The phases were separated and the aq. phase extracted once more with EtOAc. The combined organic phases were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude intermediate 17 was used as such in the subsequent reaction step (quantitative yield; R-enantiomer).

### m) Preparation of Intermediate 18

$$\bigcap_{NH} \bigcap_{N} \bigcap_{R} \bigcap_{F_3C} CF_3$$

Acetic anhydride (0.614 mL, 6.50 mmol) was added dropwise to formic acid (0.967 mL, 25.65 mmol) at r.t. and stirred for 30 min at the same temperature. To this solution was added dropwise intermediate 17 (1.71 mmol, crude material) in THF (10 mL). The r.m. was then stirred 16 h at 60° C. After this time the reaction was partitioned between water and EtOAc, the phases were separated, the aq. phase extracted once more with EtOAc, and the combined organic phases were dried (MgSO<sub>4</sub>), filtered and evaporated. The crude intermediate 18 was used as such in the next reaction step (quantitative yield; R-enantiomer).

### n) Preparation of Intermediate 19

$$\bigcap_{O} \bigcap_{N} \bigcap_{N} \bigcap_{N} \bigcap_{F_3C} CF_3$$

1-Bromo-2-butanone (0.436 mL, 4.27 mmol) was added dropwise to a stirred suspension of intermediate 18 (1.71 mmol, crude material),  $\rm K_2\rm CO_3$  (0.827 g, 5.99 mmol) and potassium iodide (28 mg, 0.17 mmol) in DMF (5 mL) at r.t. The mixture was stirred for 16 h at the same temperature, then water and EtOAc were added and the phases separated. The aq. phase was extracted once more with EtOAc, the combined organic phases were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude intermediate 19 was used as such in the subsequent reaction step (quantitative yield; R-enantiomer).

### Example A3

### a) Preparation of Intermediate 20

To a suspension of 4-methyl-3-piperidin-1-yl-benzoic acid (2.46 g, 11.2 mmol) in dry THF (22 mL) was added boranemethyl sulfide (2M in THF, 11.2 mL, 22.4 mmol) at  $0^{\circ}$  C. The r.m. was heated at  $50^{\circ}$  C. overnight, then cooled. Aq. sat. NaHCO $_3$  was added carefully. The r.m. was extracted with DCM. The organic layer was dried, filtered and concentrated to dryness. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo to yield intermediate 20 as an oil (2.3 g, 94%).

### b) Preparation of Intermediate 21

Thionyl chloride (0.497 mL, 0.81 mmol) was added to a solution of intermediate 20 (0.7 g, 3.4 mmol) in DCM (12 mL) at 0° C. The mixture was stirred at r.t. for 2 h, then concentrated in vacuo. The crude intermediate 21 was used as such in the next reaction step as a white solid (quantitative yield).

### c) Preparation of Intermediate 22

Intermediate 14 (0.3 g, 1.17 mmol) and sodium hydride (60% in mineral oil, 0.094 g, 2.34 mmol) were dissolved in DMF (5 mL) at 0° C., then intermediate 21 (crude material, 1.28 mmol) was added. The mixture was stirred at r.t. overnight. EtOAc and water were added. The organic phase was separated and dried over MgSO<sub>4</sub>, filtered and concentrated to dryness. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo to yield intermediate 22 as an oil (0.219 g, 42%; R-enantiomer).

### Example A4

### a) Preparation of Intermediate 23

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3,4-Dichlorobenzaldehyde (10 g, 57.142 mmol) was dissolved in MeOH (300 mL), then D-alaninol (4.7 g, 62.85 mmol) and NaHCO $_3$  (9.599 g, 114.28 mmol) were added and the reaction stirred at 80° C. for 2 h. The r.m. was then cooled to 25° C. NaBH $_4$  (2.171 g, 57.142 mmol) was added portionwise while keeping the temperature below 25° C. The mixture was stirred at 25° C. for 1 h, then quenched with water (100 mL). MeOH was evaporated in vacuo, then the residual extracted with DCM (3×100 mL). The organic layers were collected, washed with water (50 mL) and evaporated in vacuo to afford intermediate 23 (13 g, 97%; R-enantiomer).

### b) Preparation of Intermediate 24

HBTU (1.07 g, 2.83 mmol) was added to a stirred solution of 6-benzyloxy-2-pyridinecarboxylic acid (0.5 g, 2.18 mmol), DIPEA (0.407 mL, 2.83 mmol) and intermediate 23 (0.511 g, 2.18 mmol) in DMF (10 mL). The mixture was stirred at r.t. 4 h. Sat. aq. NaHCO $_3$  was added, followed by  $^{35}$  extraction with EtOAc. The combined extracts were dried over MgSO $_4$  and concentrated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo to yield intermediate 24 as an oil  $^{40}$  (0.690 g, 71%; R-enantiomer).

### c) Preparation of Intermediate 25

Acetic anhydride (0.176 mL, 1.86 mmol) was added to a solution of intermediate 24 (0.69 g, 1.55 mmol), Et<sub>3</sub>N (0.323 mL, 2.32 mmol) and DMAP (1 mg, 0.01 mmol) in DCM (5 mL). The mixture was stirred at r.t. overnight. Aq. sat. NaHCO<sub>3</sub> was added and the mixture was extracted with DCM. The combined extracts were dried over MgSO<sub>4</sub> and concentrated in vacuo. The crude product was purified by

flash column chromatography (silica; EtOAc/hexane 0/100 to 30/70). The desired fractions were collected and concentrated in vacuo to yield intermediate 25 as a colourless oil (0.5 g, 66%; R-enantiomer).

### d) Preparation of Intermediate 26

$$\begin{array}{c|c} & & & & \\ & & \\ & & & \\ & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$$

10% Pd/C (0.05 g) was added to a solution of intermediate 25 (0.5 g, 1.02 mmol) in MeOH (5 mL). The mixture was hydrogenated (atmospheric pressure) at r.t. for 6 h. The catalyst was filtered through diatomaceous earth and the solvent evaporated in vacuo to yield a colourless oil. The crude intermediate 26 was used such in the next step (quantitative yield; R-enantiomer).

### e) Preparation of Intermediate 27

NaOH 1M (1.12 mL, 1.12 mmol) was added to a solution of intermediate 26 (crude material, 1.02 mmol) in MeOH (5 mL). The mixture was stirred for 30 min at r.t. Aq. sat. NaHCO<sub>3</sub> was added and the aq. layer was extracted with DCM. The organic layer was dried over MgSO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The crude intermediate 27 was used as such in the next reaction step (quantitative yield; R-enantiomer).

### f) Preparation of Intermediate 28

DIAD (0.303 mL, 1.53 mmol) was added to a stirred solution of intermediate 27 (crude material, 1.02 mmol) and TPP (0.401 g, 1.53 mmol) in dry THF (5 mL) under  $N_2$  at 0° C. The mixture was stirred at r.t. overnight. The solvents were then evaporated in vacuo. The crude product was purified by flash

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MeOH/DCM 0/100 to 10/90). The desired fractions were collected and concentrated in vacuo to afford intermediate 30 (23.2 g, 93%).

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column chromatography (silica; EtOAc/hexane 0/100 to 100/0). The desired fractions were collected and concentrated in vacuo to yield intermediate 28 as an oil (0.226 g, 66% over 3 steps; R-enantiomer).

### 2) Preparation of Intermediate 29

$$Br$$
 $N$ 
 $R$ 
 $Cl$ 
 $Cl$ 

Bromine (0.041 mL, 0.8 mmol) was added dropwise slowly to a stirred solution of intermediate 28 (0.226 g, 0.67 20 mmol) in 5 mL of DCM/AcOH 4:1 under N<sub>2</sub>. The mixture was stirred at r.t. overnight, then diluted with sat. aq. NaHCO<sub>3</sub> and extracted with DCM. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents were evaporated in vacuo. The crude product was purified by flash column 25 chromatography (silica; EtOAc/DCM 0/100 to 100/0). The desired fractions were collected and concentrated in vacuo to yield intermediate 29 as a colourless oil (0.193 g, 69%; R-enantiomer).

### g1) Alternative Preparation of Intermediate 29

HBTU (8.2 g, 21.58 mmol) was added portionwise to a stirred solution of intermediate 12 (crude material, 16.6 mmol), intermediate 23 (3.89 g, 16.6 mmol) and DIPEA (4.3 mL, 24.9 mmol) in DMF (50 mL). The mixture was stirred at r.t. for 14 h, then 0.5 additional eq. of HBTU and DIPEA were added. The mixture was stirred at r.t. for 4 h, then poured into aq. sat. NaHCO3 and extracted with EtOAc. The organic layer was separated, dried over MgSO4, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/DCM 0/100 to 50/50). The desired fractions were collected and concentrated in vacuo to yield the intermediate 29 as a pale yellow solid (1.89 g, 27% over two steps; R-enantiomer).

### Example A5

### a) Preparation of Intermediate 30

N-Benzylethanolamine (26.3 mL, 182.84 mmol) was added to a mixture of methyl 2-hydroxy-6-pyridinecarboxylate (14 g, 91.42 mmol) and MeOH (92 mL). The r.m. was 65 stirred under reflux, the solvent was evaporated and the crude product purified by flash column chromatography (silica;

### b) Preparation of Intermediate 31

DIAD (19.05 mL, 96.117 mmol) was added to a stirred solution of intermediate 30 (17.449 g, 64.078 mmol) and TPP (25.211 g, 96.117 mmol) in dry THF (193 mL) under  $N_2$ . The mixture was stirred at r.t. for 2 h. The solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/DCM 0/100 to 100/0). The desired fractions were collected and concentrated in vacuo to give intermediate 31 as a white solid (quantitative).

### c) Preparation of Intermediate 32

$$\bigcap_{N} \bigcap_{N} \bigcap_{Bn}$$

Bromine (0.665 mL, 12.96 mmol) was added dropwise slowly to a stirred solution of intermediate 31 (2.75 g, 10.8 mmol) in DCM/AcOH 4:1 (50 mL) under N<sub>2</sub>. The mixture was stirred at r.t. overnight, then diluted with aq. sat. NaHCO<sub>3</sub> and extracted with DCM. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents were evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/DCM 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo to yield intermediate 32 as a yellow solid (quantitative).

### d) Preparation of Intermediate 33

Trifluoromethanesulfonic acid (5.05 g, 15.15 mmol) was added to a stirred solution of intermediate 32 (5.4 mL, 60.6 mmol) in dry toluene (50 mL). The mixture was stirred at reflux for 24 h, then diluted with sat.  $NH_3$  and the solvents were evaporated in vacuo. The crude product was purified by flash column chromatography (silica; MeOH/DCM 0/100 to

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6/94). The desired fractions were collected and concentrated in vacuo to yield intermediate 33 as a white solid (quantitative).

### e) Preparation of Intermediate 34

Intermediate 33 (0.57 g, 2.34 mmol) and sodium hydride 20 (60% as a dispersion in mineral oil, 0.103 g, 2.57 mmol) were dissolved in DMF (15 mL) at 0° C., then 3-phenylbenzyl bromide (0.58 g, 2.34 mmol) was added. The mixture was stirred at r.t. overnight. EtOAc and water were added. The 25 organic phase was separated and dried over MgSO<sub>4</sub>, filtered and concentrated to dryness. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 50/50). The desired fractions were collected and 30 concentrated in vacuo to yield intermediate 34 as a white solid (0.653 g, 98%).

### Example A6

### a) Preparation of Intermediate 35

$$\bigcap_{O} \bigcap_{N \in \mathcal{N}} \bigcap_{N \in \mathcal{N$$

K<sub>3</sub>PO<sub>4</sub> (0.731 g, 3.44 mmol), Pd<sub>2</sub>(dba)<sub>3</sub> (63 mg, 0.06 mmol) and Xantphos (69 mg, 0.12 mmol) were added to a stirred solution of intermediate 13 (0.65 g, 1.72 mmol) in dry THF (5 mL) at r.t., while  $N_2$  was bubbled through the mixture. After 10 min, acetamide (0.112 g, 1.89 mmol) was added and the mixture was stirred for another 10 min, then stirred for 3 h at 90° C. in a closed vessel. The reaction was then cooled to r.t. and sat. NaHCO<sub>3</sub> and EtOAc were added. The phases were EtOAc, the combined organics were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude was purified by flash column chromatography (silica; EtOAc/heptane 0/100 to 90/10). The desired fractions were collected and concentrated in vacuo to 65 yield intermediate 35 as a pale yellow foam (0.335 g, 56%; R-enantiomer).

b) Preparation of Intermediate 36

$$\bigcap_{N} \bigcap_{N} \bigoplus_{N \in \mathbb{N}} H_{2N}$$

HCl (6N in 2-propanol, 0.471 mL, 2.82 mmol) was added to a solution of intermediate 35 (0.335 g, 0.94 mmol) in MeOH (5 mL) at r.t. and the mixture was stirred overnight. The solvent was then evaporated, sat. NaHCO3 and EtOAc were added, the phases were separated, the aqueous phase was extracted once more and the combined organics were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude intermediate 36 was used as such in the next reaction step (quantitative yield; R-enantiomer).

### c) Preparation of Intermediate 37

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Acetic anhydride (0.339 mL, 3.58 mmol) was added dropwise to formic acid (0.533 mL, 14.13 mmol) at r.t. and stirred for 30 min at the same temperature. To this solution was added dropwise intermediate 36 (crude material, 0.94 mmol) in THF (6 mL). The r.m. was stirred 16 h at 60° C., then water and EtOAc were added. The phases were separated and the aqueous phase was extracted once more. The combined organics were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude intermediate 37 was used as such in the next reaction step (quantitative yield; R-enantiomer).

### d) Preparation of Intermediate 38

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Chloroacetone (0.188 mL, 2.35 mmol) was added dropseparated, the aqueous phase extracted once more with 60 wise to a stirred suspension of intermediate 37 (crude material, 0.94 mmol), K<sub>2</sub>CO<sub>3</sub> (0.456 g, 3.29 mmol) and KI (16 mg, 0.09 mmol) in DMF (3 mL) at r.t. The mixture was stirred for 16 h, then water and EtOAc were added. The phases were separated and the aq. phase was extracted once more. The combined organics were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude intermediate 38 was used as such to the next reaction step (quantitative yield; R-enantiomer).

# **78** Example A7

### a) Preparation of Intermediate 41

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NH<sub>4</sub>OAc (0.362 g, 4.70 mmol) was added to a stirred solution of intermediate 38 (crude material, 0.94 mmol) in AcOH (2 mL) at r.t. and the mixture was stirred for 1 h at reflux. The reaction was then cooled to r.t. and poured into water at 0° C. 50% NaOH was added slowly until basic pH. The product was extracted with EtOAc (×2). The combined organics were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude was purified by flash column chromatography (silica; DCM-MeOH (9:1, v/v)/DCM 0/100 to 100/0). The desired fractions were collected and concentrated in vacuo to yield intermediate 39 as a sticky brown oil (0.2 g, 57% over 4 steps; R-enantiomer).

### f) Preparation of Intermediate 40

$$\begin{array}{c|c}
N & O \\
N & NH
\end{array}$$

TfOH (0.186 mL, 2.11 mmol) was added to a stirred solution of intermediate 39 (0.2 g, 0.52 mmol) in dry toluene (2.5  $\,^{40}$  mL) at r.t. and the mixture was stirred for 2 h at reflux. The solvent was then evaporated. NaOH 1M was added to pH=8 and the solvents were evaporated. The crude was triturated with DCM-MeOH (9:1, v/v), dried over MgSO\_4, filtered and evaporated. The crude intermediate 40 was used as such in the  $\,^{45}$  next reaction step (quantitative yield; R-enantiomer).

### f1) Alternative Preparation of Intermediate 40

In a first vial equipped with a magnetic stir bar and a screw 50 cap septum, a solution of Pd<sub>2</sub>(dba)<sub>3</sub> (55 mg, 0.06 mmol) and 2-di-tert-butylphosphino-3,4,5,6-tetramethyl-2',4',6'-triisopropy 1-1,1'-biphenyl (60 mg, 0.125 mmol) in dioxane (1.45 mL) and toluene (7.1 mL) was flushed with N<sub>2</sub> and then stirred at 120° C. for 3 min. A second vial, equipped with a 55 magnetic stir bar and a screw cap septum, was charged with 4-methyl-imidazole (452 mg, 5.5 mmol) and  $K_3PO_4$  (2.12 g, 10 mmol), then with intermediate 14 (1.312 g, 5 mmol) and also flushed with N2. The premixed catalyst solution was added by syringe to the second vial. The r.m. was heated at 60 120° C. for 5 h. The reaction was cooled to r.t, diluted with DCM, washed with brine and neutralized with NH<sub>4</sub>Cl. The solvents were evaporated until dryness, then the residue dissolved in MeOH and silica was added in order to purify the product (silica; silicaMeOH/DCM 5/95 to 20/80). The fractions were collected to give intermediate 40 (1.33 g, 98%; R-enantiomer).

NCS (195 mg, 1.46 mmol) was added portionwise to a solution of intermediate 34 (0.5 g, 1.22 mmol) in DMF (5 mL) at r.t. The mixture was stirred overnight at 65° C. Sat. aq. NaHCO<sub>3</sub> and EtOAc were added, the phases were separated and the aq. phase was extracted once more with EtOAc. The combined organics were dried on MgSO<sub>4</sub>, filtered and evaporated. The crude was purified by flash column chromatography (silica; EtOAc/heptane 0/100 to 50/50). The desired fractions were collected and concentrated in vacuo to yield intermediate 41 as a pale yellow foam (451 mg, 83%).

### Example A8

### a) Preparation of Intermediate 42

Trifluoromethanesulfonic acid (4.2 mL, 47.19 mmol) was added to a stirred solution of intermediate 31 (3 g, 11.79 mmol) in dry toluene (33 mL). The mixture was stirred at reflux overnight, then diluted with NaOH 1M to pH=8 and the solvents were evaporated in vacuo. The crude product was purified by flash column chromatography (silica; MeOH/DCM 0/100 to 3/97). The desired fractions were collected and concentrated in vacuo to yield intermediate 42 as a yellow solid (1.58 g, 81%).

### b) Preparation of Intermediate 43

$$\bigcup_{N}^{\circ} \bigvee_{N}^{F}$$

CuI (0.495 g, 2.6 mmol) was added to a solution of intermediate 42 (0.855 g, 5.2 mmol), 5-fluoro-2-iodotoluene (1.6 g, 6.77 mmol), N,N-dimethyl-1,2-ethane-diamine (0.56 mL, 5.2 mmol) and  $\rm K_3PO_4$  (2.2 g, 10.4 mmol) in DMF (15 mL), while the reaction was degassed by bubbling  $\rm N_2$  through the

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solution. The mixture was then heated a  $100^{\circ}$  C. for 6 h. Water was added and the aq. layer was extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 100/0). The desired fractions were concentrated in vacuo to yield intermediate 43 as a brown solid (0.848 g, 60%).

### Example A9

### a) Preparation of Intermediate 44

Starting from 2-(o-tolylamino)ethanol, intermediate 44 <sub>25</sub> was prepared by analogy to the procedures reported for the synthesis of intermediate 4.

### b) Preparation of Intermediate 45

Thionyl chloride (1.08 mL, 14.9 mmol) was added to a solution of intermediate 44 (2.7 g, 7.44 mmol) in DCM (15 mL) at 0° C. The mixture was stirred at r.t. overnight. DCM was added and the organic layer was washed with aq. sat. NaHCO<sub>3</sub>, dried over MgSO<sub>4</sub>, filtered and evaporated under reduced pressure. The crude intermediate 45 (oil) was used as such in the next reaction step (quantitative yield).

### c) Preparation of Intermediate 46

Boron tribromide (1.4 mL, 14.88 mmol) was added to a solution of intermediate 45 (crude material, 7.44 mmol) in DCM (21 mL) at 0° C. The mixture was stirred at r.t. for 2 h, 65 then DCM and sat. aq. NaHCO<sub>3</sub> were added. The aq. layer was extracted with DCM, the organic phase was dried over

MgSO<sub>4</sub>, filtered and concentrated to dryness. The crude intermediate 46 was used such in the next reaction step (quantitative yield).

### d) Preparation of Intermediate 47

NaH (60% in mineral oil, 0.327 g, 8.2 mmol) was added portionwise to a solution of intermediate 46 (crude material, 7 44 mmol) in DMF (21 mL) at 0° C. The mixture was stirred at r.t. for 2 h. EtOAc and water were added, the organic phase was separated, dried over MgSO<sub>4</sub>, filtered and concentrated to dryness. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 100/0). The desired fractions were collected and concentrated in vacuo to yield intermediate 47 as a white solid (1.24 g, 59%).

### Example A10

### a) Preparation of Intermediate 48

A solution of 2-(benzylamino)-2-(oxan-4-yl)acetonitrile (6.05 g, 26.28 mmol) in AcOH (5 mL) and HCl 37% (50 mL) was stirred at 95° C. for 4 days. The mixture was cooled and the solvents evaporated in vacuo. The black crude was triturated with acetone to give a white solid, that was discarded. The acetone solution was concentrated in vacuo and the residual triturated with DIPE and heptane. The sticky solid was filtered and washed with Et<sub>2</sub>O, DCM and acetone. The resulting intermediate 48 (beige solid) was dried in vacuo and used as such in the next reaction step (3.77 g, 57%).

### b) Preparation of Intermediate 49

To a suspension of intermediate 48 (3.77 g, 13.21 mmol) in THF (21 mL) were added  $\rm Et_3N$  and borane-methyl sulfide (2M, 19.81 mL, 39.63 mmol) at  $0^{\circ}$  C. The r.m. was heated to

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50° C. for 36 h, then cooled. Aq. sat. NaHCO<sub>3</sub> was added, the reaction extracted with EtOAc, the organic layer was dried, filtered and concentrated to dryness. The crude material was used as such for the next reaction step (quantitative yield).

### c) Preparation of Intermediate 50

Starting from intermediate 49, intermediate 50 was prepared by analogy to the procedures reported for the synthesis of intermediate 6.

### d) Preparation of Intermediate 51

NaH (60% in mineral oil, 0.053 g, 1.32 mmol) and intermediate 50 (0.3 g, 1.20 mmol) were dissolved in DMF (12 mL) at 0° C. 3,4-Dichlorobenzyl bromide (0.217 mL, 1.45 mmol) was then added, and the mixture was stirred at r.t. for 4 h. EtOAc and water were added, the organic phase was separated, dried over MgSO<sub>4</sub>, filtered and concentrated to dryness. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 100/0). The desired fractions were collected and concentrated in vacuo to yield intermediate 51 as an oil (0.345 g, 71%).

### Example A11

### a) Preparation of Intermediate 52

A mixture of 2-iodoethanol (0.296 mL, 3.8 mmol) and 2-methyl-3-trifluoromethyl-aniline (1 g, 5.7 mmol) was heated at 90° C. under  $\rm N_2$  atmosphere for 6 h. The resulting solid was dissolved in EtOAc and washed with 2M aq. NaOH solution. The organic layer was dried, filtered and concentrated to dryness. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 50/50). The desired fractions were collected and concentrated in vacuo to yield intermediate 52 as an oil (0.711 g, 85%).

### b) Preparation of Intermediate 53

Starting from intermediate 52, intermediate 53 was prepared by analogy to the procedures reported for the synthesis of intermediate 2.

### c) Preparation of Intermediate 54

Thionyl chloride (0.417 mL, 4.86 mmol) was added to a solution of 6-(benzyloxy)-pyridine-2-carboxylic acid (0.724 g, 3.24 mmol) and a drop of DMF in DCM (15 mL). The mixture was stirred at r.t. for 2 h, then the solvent was removed under reduced pressure and the crude intermediate 54 used as such in the next reaction step (quantitative yield).

### d) Preparation of Intermediate 55

Intermediate 53 (crude material, 3.24 mmol) was added to a solution of intermediate 54 (crude material, 3.24 mmol) and DIPEA (0.726 mL, 4.212 mmol) in DMF (12 mL). The mixture was stirred at r.t. overnight, then aq. sat. NaHCO<sub>3</sub> was

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added the mixture extracted with EtOAc. The combined extracts were dried over  $MgSO_4$  and concentrated in vacuo. The crude was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 15/85). The desired fractions were collected and concentrated in vacuo to yield intermediate 55 as an oil (0.920 g, 52% over two steps).

### Example A12

### a) Preparation of Intermediate 56

Starting from 2',3'-dichloroacetophenone and ethanolamine, intermediate 56 was prepared by analogy to the procedures reported for the synthesis of intermediate 7.

### Example A13

### a) Preparation of Intermediate 57

Starting from 3',4'-dichloroacetophenone and ethanolamine, intermediate 57 was prepared by analogy to the procedures reported for the synthesis of intermediate 7.

### Example A14

### a) Preparation of Intermediate 58

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Starting from intermediate 33, intermediate 58 was prepared by analogy to the procedures reported for the synthesis of intermediate 34. Intermediate 58 was obtained as a crude, and was used as such in the synthesis of compounds 11a and 11b.

### Example A15

### a) Preparation of Intermediate 59

Starting from 3-amino-2-butanol (mixture of R,R and S,S) and 3,5-bis(trifluoromethyl)-benzaldehyde, intermediate 59 (cis) was prepared by analogy to the procedures reported for the synthesis of intermediate 13.

### Example A16

### a) Preparation of Intermediate 60

 $\rm Et_3N$  (4.24 mL, 30.51 mmol) was added to a solution of 2-(4-chlorophenyl)-cyclopropanecarboxylic acid (6 g, 30.51 mmol) in tBuOH (91.5 mL) at r.t. Diphenylphosphoryl azide (6.6 mL, 30.51 mmol) was added and the mixture stirred at r.t. for 30 min. The r.m. was slowly heated at 80° C. and kept at that temperature for 3 h. The solvent was evaporated in vacuo. The mixture was diluted with water and extracted with  $\rm EtOAc$ . The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica;  $\rm EtOAc/heptane~0/100~to~15/85)$ . The desired fractions were collected and concentrated in vacuo to yield intermediate 60 (trans) as a white solid (2.6 g, 32%).

Intermediate 60 (2.6 g, 9.7 mmol) and HCl (5 to 6N in 15 2-propanol, 9.7 mL) were dissolved in dioxane (9.7 mL). The mixture was stirred at r.t. overnight. Sat. aq. NaHCO $_3$  was added carefully. The reaction was extracted with EtOAc. The organic layer was dried, filtered and concentrated to dryness. 20 The crude intermediate 61 (trans) was used as such in the next reaction step (quantitative yield).

### c) Preparation of Intermediate 62

Sodium triacetoxyborohydride (2.5 g, 11.64 mmol) was added to a stirred solution of intermediate 61 (crude material, 9.7 mmol) and tert-butyldimethylsilyloxy-acetaldehyde (2.03 mL, 10.67 mmol) in 1,2-dichloroethane (30 mL). The reaction was stirred at r.t. overnight. The crude product was dissolved with DCM and washed with aq. sat. NaHCO<sub>3</sub>. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/DCM 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo to yield intermediate 62 (trans) as a colorless oil (1.4 g, 44%).

### Example A17

### a) Preparation of Intermediate 63

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A mixture of 5-bromo-2-fluorobenzotrifluoride (4 g, 16.46 mmol), 3-formyl-phenylboronic acid (2.96 g, 19.75 mmol),

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 $PdCl_2(dppf)$ ) (0.602 g, 0.823 mmol) and  $K_2CO_3$  (7.571 g, 32.92 mmol) in dioxane/water 5:1 (50 mL) was degassed for a few min with  $N_2$ . The r.m. was then heated at  $80^{\circ}$  C. for 5 h. The dioxane was removed under vacuum and the residual extracted with DCM. The organic layer was dried over  $Na_2SO_4$ , filtered and the solvent was evaporated. The residue was purified by flash chromatography (silica; EtOAc/petroleum ether 0/100 to 1/15). The desired fractions were collected and the solvent was evaporated to give intermediate 63 (3.4 g, 77%).

### b) Preparation of Intermediate 64

A solution of intermediate 63 (3.4 g, 12.687 mmol) and MeOH (100 mL) was stirred and cooled to 0° C. NaBH<sub>4</sub> (1.04 g, 27.374 mmol) was added and the stirring was continued at 0° C. for 10 min, then the r.m. was warmed to r.t. for 2 h. After
 this time, 1 mL of water was added, and the mixture evaporated. 40 mL of sat. aq. NaHCO<sub>3</sub> were added, the solution washed with DCM, the organic layer collected, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated to give intermediate 64 (3.2 g, 93%).

### c) Preparation of Intermediate 65

A solution of intermediate 64 (2 g, 7.407 mmol) in DCM (70 mL) was cooled to  $10^{\circ}$  C. and phosphorus tribromide (4 g, 14.814 mmol) was added dropwise. The r.m. was stirred at  $-20^{\circ}$  C. for 3 h. After this time, the reaction was warmed to  $0^{\circ}$  C. and stirred 1 h, then quenched with water (50 mL), and slowly brought to pH 8 with solid  $K_2 CO_3$ . The layers were separated and the aq. layer was extracted with DCM (2×100 mL). The combined organic layers were dried over  $Na_2 SO_4$  and evaporated to give intermediate 65 as a clear oil (1.73 g, 70% yield).

### Example A18

Following a procedure similar to the one reported for the synthesis of intermediate 21, following compounds were obtained from the commercially available corresponding acids:

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Structure	Intermediate number
CI	66

$$CI$$
 $OCF_3$ 

### Example A19

### a) Preparation of Intermediate 68

$$_{\mathrm{F}_{3}\mathrm{C}}^{\mathrm{OH}}$$

3,5-Bis(trifluoromethyl)bromobenzene (2.1 mL, 10.51 mmol) was added to a stirred solution of magnesium (0.3 g, 11.39 mmol) in THF under  $\rm N_2$  atmosphere. The mixture was stirred at reflux for 2 h, then was added to a stirred solution of 4-formyltetra-hydropyran (1 g, 8.76 mmol) in THF under  $\rm N_2$  atmosphere at r.t. The mixture was stirred at r.t. for 1 h, then aq. sat. NH<sub>4</sub>Cl was added and the mixture was extracted with EtOAc. The combined extracts were dried over MgSO<sub>4</sub> and concentrated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 50/50). The desired fractions were collected and concentrated in vacuo to yield intermediate 68 as an oil (1.67 g, 58%).

### b) Preparation of Intermediate 69

DIAD (1.5 mL, 7.62 mmol) was added to a stirred solution of intermediate 68 (1.67 g, 5.08 mmol), phtalimide (0.812 g, 5.6 mmol) and TPP (2 g, 7.62 mmol) in dry THF and under  $\rm N_2$  at 0° C. The mixture was stirred at r.t. for 4 h, then the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo to yield intermediate 69 as an oil (1.41 g, 61%).

### c) Preparation of Intermediate 70

Hydrazine hydrate (64%, 0.734 mL, 15.41 mmol) was added to a stirred solution of intermediate 69 (1.41 g, 3.08 mmol) in EtOH (9 mL). The mixture was stirred at 80° C. for 4 h, then diluted with NaOH 1M and extracted with DCM. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude intermediate 70 was used as such in the reaction step (quantitative yield).

### Example A20

### a) Preparation of Intermediate 71

4-Chlorobenzoic acid (1.58 g, 10.09 mmol) was dissolved in DCM (10 mL). Oxalyl chloride (1.30 mL, 15.13 mmol) was added slowly at r.t. and the mixture was refluxed for 2 h. The r.m. was then concentrated in vacuo. The residue was dissolved in acetonitrile (50 mL), then trimethylsilyldiazomethane (2M in hexane, 5.55 mL, 11.10 mmol) was added at 0° C. followed by  $\rm Et_3N$ . The mixture was stirred at 0° C. for

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24 h. The solvent was evaporated in vacuo and the crude purified by flash column chromatography (silica; EtOAc/heptane 0/100 to 25/75) to afford intermediate 71 (0.503 g, 28%).

### b1) Preparation of Intermediate 72

Intermediate 71 (0.503 g, 2.78 mmol) was added in small portions to a solution of chloroacetonitrile (11.2 mL, 177 mmol) and boron trifluoride etherate (0.560 mL, 9.5 mmol) at  $-5^{\circ}$  C. under stirring. The reaction proceeded with vigorous evolution of  $N_2$  gas and gave a grey precipitate. After the  $N_2$  evolution ceased, the r.m. was cooled to  $-15^{\circ}$  C. to complete the precipitation. The precipitate was separated by filtration, washed with Et<sub>2</sub>O, neutralized with aq. sat. NaHCO<sub>3</sub> and extracted with Et<sub>2</sub>O (×2). The organic layer was dried over MgSO<sub>4</sub>, filtered and concentrated in vacuo to afford intermediate 72 (0.602 g, 95%).

### Example A21

### a) Preparation of Intermediate 73

Thionyl chloride (1.394 mL. 19.109 mmol) was added to a solution of (R)-Amino-cyclopropyl-acetic acid (1 g, 8.686 mmol) in MeOH (26 mL) at 0° C. The mixture was stirred at 80° C. for 16 h. The mixture was concentrated to dryness. Toluene was added and the mixture was concentrated to dryness again to yield a colourless oil containing intermediate 73, used in the subsequent reaction step without further purification (quantitative yield; R-enantiomer).

### b) Preparation of Intermediate 74

A mixture of 3,4-dichlorobenzoylchloride (2.001 g, 9.55 60 mmol) in DCM was added slowly dropwise to a solution of intermediate 73 (crude material, 8.686 mmol) and  $\rm Et_3N$  (3.632 mL, 26.058 mmol) in DCM (26 mL total amount). The mixture was stirred at r.t. for 15 h. The crude r.m. was diluted with aq. sat. NaHCO $_3$  and extracted with DCM. The organic 65 layer was separated, dried over MgSO $_4$ , filtered and the solvents evaporated in vacuo. The crude product was purified by

flash column chromatography (silica; EtOAc/hexane 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo to yield intermediate 74 as a white solid (1.136 g, 44%; R-enantiomer).

### c) Preparation of Intermediate 75

LiAlH<sub>4</sub> (0.143 g, 3.760 mmol) was added portionwise to a mixture of intermediate 74 (1.136 g, 3.760 mmol) in dry THF (11.4 mL) at 0° C. The mixture was stirred at r.t. for 16 h, then 0.5 eq of LiAlH<sub>4</sub> were added. The mixture was stirred at 0° C. and MgSO<sub>4</sub> with water were added portionwise at 0° C. The mixture was filtered and the solvents evaporated in vacuo. The crude material was used as such in the next reaction step (quantitative yield; R-enantiomer).

### d) Preparation of Intermediate 76

Borane dimethylsulfide (2M in THF, 4.7 mL, 9.4 mmol)
45 was added to a suspension of intermediate 75 (crude material,
3.760 mmol) in dry THF (5.7 mL) at 0° C. The r.m. was heated
at 60° C. for 16 h, then cooled down. Aq. sat. NaHCO<sub>3</sub> was
added carefully. The reaction was extracted with DCM. The
organic layer was dried, filtered and concentrated to dryness.
50 The crude product was purified by flash column chromatography (silica; EtOAc/hexane 0/100 to 30/70). The desired
fractions were collected and concentrated in vacuo to yield
intermediate 76 as an oil (98%; R-enantiomer).

### Example A22

### a) Preparation of Intermediate 77

-continued

DIPEA (4.87 mL, 27.9 mmol) was added to an ice-cooled sol. of 1-(4-chlorophenyl)-1-cyclopropane methanol (1.7 g, 9.3 mmol) in DCM (23 mL). Methanesulphonyl chloride (0.72 mL, 9.3 mmol) was added and stirring was continued for 1 h, while letting the reaction reach r.t. The r.m. was quenched by aq. sat. NaHCO<sub>3</sub>. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and evaporated in vacuo. The crude intermediate 77 was used as such in the next reaction step (quantitative yield).

By using an analogous procedure to the one reported for the synthesis of intermediate 77, starting from the corresponding known or commercially available alcohols, following intermediates were obtained:

Structure	Int. number
	78
$F_3C$ $O$ $O$ $S$ $O$	79
	80
CI	81
CI—N—N	82
$CI$ $N$ $CF_3$	83

Structure

Int. number

84

$$F_{5}C$$

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65

### -continued

-continued				
Structure	Int. number			
	144			
CI	193			
F <sub>3</sub> C O S	194			
CI	195			
CI	196			
F O S O O S	197			
	198			

## a) Preparation of Intermediate 88

DIAD (2.885 mL, 14.555 mmol) was added to a stirred solution of 3-hydroxy-5-(trifluoromethyl)benzoic acid (1.2 g, 5.822 mmol), iPrOH (0.981 mL, 12.808 mmol) and TPP (3.82 g, 14.555 mmol) in dry THF (25 mL) under N<sub>2</sub> at 5° C. The mixture was stirred at r.t. for 12 h, then concentrated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/heptane 0/100 to 10/90). The desired fractions were concentrated in vacuo to yield a colourless oil (1.55 g, 91%).

### b) Preparation of Intermediate 89

NaOH 1M (8.01 mL, 8.01 mmol) was added to a stirred solution of intermediate 88 (1.55 g, 5.340 mmol) in MeOH (10 mL). The mixture was stirred at 55° C. for 12 h. The solvent was evaporated in vacuo. The mixture was diluted with water and acidified with HCl 2N until pH=2 and extracted with EtOAc. The organic layer was separated, dried (MgSO<sub>4</sub>), filtered and the solvents evaporated in vacuo. The crude intermediate 89 was used as such for the next reaction step (1.280 g, 95%).

### c) Preparation of Intermediate 90

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By following an analogous procedure as described for the synthesis of intermediate 21, intermediate 90 was obtained starting from intermediate 89.

### Example A24

### a) Preparation of Intermediate 91

By following an analogous procedure as described for the synthesis of intermediate 20 and intermediate 77, intermediate 91 was obtained starting from the commercially available  $^{25}$  acid.

### Example A25

### a) Preparation of Intermediate 92

$$\begin{array}{c} O \\ O \\ O \\ CF_3 \end{array} \qquad \qquad 40$$

By following an analogous procedure as described for the synthesis of intermediate 21, intermediate 92 was obtained 45 starting from the known alcohol (69%).

### b) Preparation of Intermediate 93

RuPhos (78 mg, 0.168 mmol),  $Pd_2(dba)_3$  (77 mg, 0.084 mmol),  $K_2CO_3$  (4.54 mg, 3.36 mmol) and potassium cyclopropyltrifluoroborate (297 mg, 2.01 mmol) were added to a stirred solution of intermediate 92 (424 mg, 1.678 mmol) in 65 toluene (19 mL) and water (1 mL) at r.t. while  $N_2$  was bubbled throughout the mixture. The reaction was then stirred for 15 h

at 120° C. After cooling to r.t.sat. NaHCO<sub>3</sub> and EtOAc were added. The aq. phase was extracted with EtOAc. The combined organics were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude was purified by flash column chromatography (silica; EtOAc/heptane 0/100 to 10/90). The desired fractions were collected and concentrated in vacuo to yield intermediate 93 as an oil (110 mg, 25%, 63% purity).

### c) Preparation of Intermediate 94

$$\bigcup_{CF_3}^{O} \bigcup_{S}$$

By following an analogous procedure as described for the synthesis of intermediate 89, 64 and 77, intermediate 94 was obtained starting from intermediate 93.

### Example A26

### a) Preparation of Intermediate 95

Bromine (0.16 mL, 3.11 mmol) was added to a solution of 1-benzothiophene-5-carboxylic acid (0.5 g, 2.8 mmol) in AcOH (15 mL). The solution was then stirred at r.t. for 4 h. Additional bromine was added and the solution stirred at r.t. for additional 16 h. The solution was then poured in water (90 mL) with vigorous stirring and the resulting solids were collected by suction filtration, washed with water and dried, to afford intermediate 95 in quantitative yield.

### b) Preparation of Intermediate 96

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By following analogous procedures as described for the synthesis of intermediates 64 and 77, intermediate 96 was obtained starting from intermediate 95.

### Example A27

### a) Preparation of Intermediate 97

NaHCO $_3$  (0.889 g, 10.59 mmol) was added to a sol. of methyl 3-bromo-5-(trifluoro-methyl)benzoate (1 g, 3.53 mmol), cyclopropylboronic acid (0.334 g, 3.89 mmol) and PdCl $_2$ (dppf) $_2$  (144 mg, 0.177 mmol) in dioxane (8 mL) and water (3 mL), while N $_2$  was bubbled through the mixture. The <sup>25</sup> solution was stirred at 100° C. overnight, then diluted with water and extracted with DCM. The organic layer was separated, dried (MgSO $_4$ ), filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/heptane 0/100 to 40/60). The <sup>30</sup> fractions were collected and evaporated in vacuo to yield intermediate 97 (0.35 g, 41%).

### b) Preparation of Intermediate 98

By following analogous procedures as described for the synthesis of intermediate 94, intermediate 98 was obtained starting from intermediate 97.

### Example A28

### a) Preparation of Intermediate 99

$$F_3C$$
  $O$   $O$   $C_1$ 

By following analogous procedures as described for the synthesis of intermediate 21, intermediate 99 was obtained starting from the known corresponding alcohol.

### Example A29

### a) Preparation of Intermediate 100

Ethyl iodide (0.5 mL, 6.12 mmol) was added to a mixture of 3-bromo-5-(trifluoro-methyl)benzyl alcohol (1.3 g, 5.10 mmol), sodium hydride (60% in mineral oil, 0.23 g, 5.61 mmol) in DMF (15 mL) at 0° C. The mixture was stirred at r.t. for 1 h, then diluted with sat. NaHCO $_3$  and extracted with EtOAc. The organic layer was washed with NaHCO $_3$  solution, dried (MgSO $_4$ ), filtered and the solvents were evaporated in vacuo. The crude product was purified by flash chromatography (silica; EtOAc/heptane 0/100 to 20/80). The desired fractions were collected and concentrated in vacuo, to yield intermediate 100 (1 g, 69%).

### b) Preparation of Intermediate 101

Buthyllithium (1.6 M in hexanes, 2.65 mL, 4.24 mmol) was added to a solution of intermediate 100 (1 g, 3.53 mmol) in dry THF (8 mL) at -78° C. The mixture was stirred at the same temperature for 1 hour, then an excess of CO<sub>2</sub> was added and the mixture was allowed to warm to rt. The mixture was stirred at r.t. overnight, then washed with 1N HCl and extracted with EtOAc. The organic layer was separated, dried (MgSO<sub>4</sub>), filtered and the solvents were evaporated in vacuo. The crude product was purified by flash column chromatography (silica; EtOAc/heptane 0/100 to 30/70), to yield intermediate 101 (0.285 g, 73%).

### c) Preparation of Intermediate 102

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By following a synthetic route similar to the one described for the synthesis of intermediate 20 and intermediate 77, intermediate 102 was obtained starting from intermediate 101.

### Example A30

### Preparation of Intermediate 103

$$\begin{array}{c|c} & & & & \\ & & & & \\ & & & & \\ & & \\ & & & \\ & & & \\ & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & &$$

DIPEA (0.034 mL, 0.196 mmol) was added to an ice-cooled solution of compound 13 (approximately 0.098 mmol) in DCM (5 mL). Methanesulphonyl chloride (0.009 mL, 0.118 mmol) was added and stirring was continued for 2 h, while letting the reaction to come to r.t. The r.m. was then quenched with sat. aq. NaHCO<sub>3</sub>. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and evaporated in vacuo. The crude material was used as such in the next reaction step (quantitative yield; S-enantiomer).

### Example A31

### Preparation of Intermediates 104 and 105

intermediate 104
$$\begin{array}{c} O \\ \hline \\ N \\ \hline \\ O \end{array}$$

$$\begin{array}{c} RorS \\ \hline \\ F_3C \end{array}$$

By following a synthetic route similar to the one described for the synthesis of intermediate 15, starting from intermediate 14 and intermediate 85, a crude mixture was obtained. The crude material was subsequently purified by flash column

chromatography (silica; EtOAc/heptane 0/100 to 20/80), to yield intermediate 104 and intermediate 105.

### Example A32

### Preparation of Intermediate 106

By following a synthetic route similar to the one described for the synthesis of intermediate 64 and intermediate 77, intermediate 106 was obtained starting from the commercially available aldehyde.

### Example A33

### a) Preparation of Intermediate 155

$$F_{3}C$$
 $O$ 
 $O$ 

DIAD (0.86 mL, 4.34 mmol) was added dropwise to a solution of methyl 2-hydroxy-6-(trifluoromethyl)isonicoti- nate (640 mg, 2.894 mmol), iPrOH (191 mg, 3.183 mmol) and TPP (1.14 g, 4.341 mmol) in THF (15 mL) at 0° C. while  $\rm N_2$  was bubbled through the solution. The r.m. was stirred overnight at r.t., then the solvent was evaporated. The crude was purified by flash column chromatography (silica; heptanes/EtOAc 100/0 to 90/10). The desired fractions were collected and concentrated in vacuo to yield intermediate 155 as a yellow oil (759 mg, 99%).

### b) Preparation of Intermediate 107

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By following an analogous procedure as described for the synthesis of intermediate 90, intermediate 107 was obtained starting from intermediate 155.

### Example A34

### a) Preparation of Intermediate 108

To a mixture of 3-methoxy-5-(trifluoromethyl)-phenylboronic acid (2 g, 9.09 mmol) and  $Cs_2CO_3$  (11.85 g, 36.387 mmol) dissolved in dioxane (50 mL) were added  $Pd_2(dba)_3$  20 (416 mg, 0.45 mmol) and Xantphos (894 mg, 1.546 mmol) while bubbling  $N_2$  through the solution. Then, 2-iodo-1,1,1-trifluoroethane (1.792 mL, 18.19 mmol) was added. The r.m. was stirred at r.t. for 1 min, then water (3 mL) was added. The mixture was further stirred at 80° C. for 12 h. After cooling, 25 the r.m. was extracted with EtOAc and water. The organic phase was washed with brine, dried over MgSO<sub>4</sub>, filtered and the solvent was concentrated in vacuo. The crude product was purified by flash column chromatography (silica; hexane/ EtOAc 100/0 to 95/5). The desired fractions were collected and concentrated in vacuo to yield intermediate 108 (1.53 g, 49%, 75% purity).

### b) Preparation of Intermediate 109

Boron tribromide (2.254 mL, 23.40 mmol) was added to a solution of intermediate 108 (1.51 g, 5.849 mmol) in DCM (10 mL) at 5° C. The mixture was stirred at r.t. for 4 h under N<sub>2</sub>, then it was diluted with DCM and washed with sat. aq. NaHCO<sub>3</sub>. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; heptanes/EtOAc 100/0 to 70/30). The desired fractions were collected and concentrated in vacuo to yield intermediate 109 (1.17 g, 82%, 95% purity).

### c) Preparation of Intermediate 110

### 102

Triflic anhydride (0.958 mL, 5.77 mmol) was added to a stirred solution of intermediate 109 (1.17 g, 4.809 mmol), Et<sub>3</sub>N (0.869 mL, 6.25 mmol) and DMAP (58 mg, 0.48 mmol) in DCM (25 mL) at  $-15^{\circ}$  C. The solution was stirred at  $-15^{\circ}$  C. for 30 min and then warmed to r.t. over 2 h. The r.m. was then poured into a sat. sol. of NH<sub>4</sub>Cl, the phases were separated and the aq. layer was extracted (×2) with DCM. The combined organic fractions were washed, dried over MgSO<sub>4</sub> and filtered. The crude material was purified by flash column chromatography (silica; hexane/EtOAc 100/0 to 90/10). The desired fractions were collected and concentrated in vacuo to yield intermediate 110 as a colourless oil (1.21 g, 66%, 95% purity).

### d) Preparation of Intermediate 111

A mixture of intermediate 110 (580 mg, 1.542 mmol), Pd(OAc)<sub>2</sub> (7 mg, 0.03 mmol), dppf (34 mg, 0.06 mmol), Et<sub>3</sub>N (0.6 mL, 4.626 mmol), EtOH (10 mL) and dioxane (10 mL) was heated under CO atmosphere (6 atm) at 95° C. for 18 h. The r.m. was then diluted with aq. sat. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude material was purified by flash column chromatography (silica; heptanes/EtOAc 100/0 to 80/20). The desired fractions were collected and concentrated in vacuo to yield intermediate 111 as a colourless oil (372 mg, 79%).

### e) Preparation of Intermediate 112

By following a synthetic route similar to the one described for the synthesis of intermediate 89, intermediate 76 and intermediate 21, intermediate 112 was obtained starting from intermediate 111.

### Example A35

### a) Preparation of Intermediate 113

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Methylmagnesium bromide (6.546 mL, 9.16 mmol) was added to a solution of 3-(cyclopropylmethyl)-5-(trifluoromethyl)benzonitrile (688 mg 3.06 mmol) in toluene (9 mL). The r.m. was stirred at 80° C. for 1 h, then treated with 10% aq. HCl and stirred at r.t. for 1 h. The phases were separated and the aq. layer was washed with EtOAc and brought to basic conditions with sat. aq. NaHCO $_3$ . The resulting slurry was extracted with EtOAc and the organic phase was dried over MgSO $_4$ , filtered and evaporated in vacuo. The crude product was purified by flash column chromatography (silica; hexane/ EtOAc 100/0 to 85/15). The desired fractions were collected and concentrated in vacuo to yield intermediate 113 (533 mg, 72%) as a colourless oil.

### b) Preparation of Intermediate 114

By following a synthetic route similar to the one described for the synthesis of intermediate 64 and intermediate 77, intermediate 114 was obtained starting from intermediate 113.

### Example A36

### Preparation of Intermediate 115

By following a synthetic route similar to the one described for the synthesis of intermediate 21, intermediate 115 was obtained starting from the commercially available alcohol.

### Example A37

### a) Preparation of Intermediate 116

HBTU (3.411 g, 8.994 mmol) was added portionwise to a 65 stirred solution of 3-methoxy-5-(trifluoromethyl)benzoic acid (1.32 g, 5.996 mmol), N-methoxymethylamine hydro-

chloride (0.702 g, 7.20 mmol) and DIPEA (3 mL, 18 mmol) in DMF (18 ml). The mixture was stirred at r.t. for 5 h, then poured into sat. aq. NaHCO<sub>3</sub> and extracted with EtOAc. The organic layer was washed with water, separated, dried over MgSO<sub>3</sub>, filtered and the solvents evaporated in vacuo. The

 ${
m MgSO_4}$ , filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; hexane/EtOAc 100/0 to 80/20). The desired fractions were collected and concentrated in vacuo to yield intermediate 116 as an oil (1.459 g, 92%, 92% purity).

### b) Preparation of Intermediate 117

Ethylmagnesium chloride (4.158 mL, 8.315 mmol) was added to a solution of intermediate 116 (1.459 g, 5.543 mmol) in THF (16.6 mL). The mixture was stirred at r.t. for 1 h, then diluted with water and extracted with EtOAc. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; hexane/EtOAc 100/0 to 70/30). The desired fractions were collected and concentrated in vacuo to yield intermediate 117 as a white solid (1.155 g, 89%, 94% purity).

### c) Preparation of Intermediate 118

By following a synthetic route similar to the one described for the synthesis of intermediate 64 and intermediate 77, intermediate 118 was obtained starting from intermediate 117.

By using an analogous procedure to the one reported for the synthesis of intermediate 118, starting from the corresponding known or commercially available acids, following intermediates were obtained:

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Structure	Intermediate number
O CF3	121

Example A38

### a) Preparation of Intermediate 123

$$CF_3$$

2-(Trifluoromethyl)morpholine hydrochloride (0.2 g, 1.044 mmol) was added to a stirred solution of methyl 3-bromo-5-(trifluoromethyl)benzoate (0.246 g, 0.87 mmol), X-phos (0.037 g, 0.078 mmol), Pd<sub>2</sub>(dba)<sub>3</sub> (0.03 g, 0.035 mmol) and Cs<sub>2</sub>CO<sub>3</sub> (0.85 g, 2.61 mmol) in toluene (10 mL) while bubbling N<sub>2</sub> through the solution. The mixture was stirred overnight at 100° C. in a sealed tube, then water and EtOAc were added. The aq. phase was extracted once more with EtOAc, then the combined organic phases were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude was purified 45 by flash column chromatography (silica; heptanes/EtOAc 100/0 to 90/10). The desired fractions were collected and concentrated in vacuo to yield intermediate 123 as a sticky yellow oil (50 mg, 17%).

### b) Preparation of Intermediate 124

$$\bigcap_{N} CF_3$$

### 106

By following a synthetic route similar to the one described for the synthesis of intermediate 89, intermediate 76 and intermediate 77, intermediate 124 was obtained starting from intermediate 123.

### Example A39

### a) Preparation of Intermediate 126

By following a synthetic route similar to the one described for the synthesis of intermediate 76 and intermediate 21, intermediate 126 was obtained starting from the commercially available acid.

### b) Preparation of Intermediate 127

Sodium cyanide (0.417 g, 8.518 mmol) was added dropwise to a stirred solution of intermediate 126 (1.435 g, 5.679 mmol) in DMF (25 mL) at 0° C. under  $\rm N_2$  atmosphere. The r.m. was stirred for 16 h at r.t., then aq. NaHCO<sub>3</sub> sat. was added and the mixture was extracted with EtOAc. The combined extracts were dried over MgSO<sub>4</sub> and concentrated in vacuo. The crude product was purified by flash column chromatography (silica; hexane/EtOAc 100/0 to 60/40). The desired fractions were collected and concentrated in vacuo to yield intermediate 127 as a colourless oil (702 mg, 51%).

### c) Preparation of Intermediate 128

KOH (5 M, 17.3 mL, 86.5 mmol) was added to a stirred solution of intermediate 127 (702 mg, 2.886 mmol) in EtOH (8.6 mL). The mixture was stirred at 95° C. for 16 h, then allowed to cool to r.t., acidified with concentrated HCl under ice cooling and extracted with EtOAc. The combined extracts were dried over MgSO<sub>4</sub> and concentrated in vacuo. The crude

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material was used as such for the next reaction step, and the yield considered to be quantitative.

### d) Preparation of Intermediate 129

By following a synthetic route similar to the one described intermediate 129 was obtained starting from intermediate

### Example A40

### Preparation of Intermediate 130

By following a synthetic route similar to the one described for the synthesis of intermediate 129, intermediate 130 was obtained starting from the commercially available nitrile.

### Example A41

### a) Preparation of Intermediate 131

2,2,2-Trifluoroethyl perfluorobutylsulfonate (0.636 mL, 2.725 mmol) was added to a stirred solution of 3-hydroxy-5trifluoromethyl benzoic acid methyl ester (500 mg, 2.271 mmol) and Cs<sub>2</sub>CO<sub>3</sub> (1.48 g, 4.542 mmol) in DMF (11 mL). 65 The mixture was stirred at r.t. for 3 h, then diluted with water and extracted with EtOAc. The organic layer was separated,

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dried over MgSO<sub>4</sub>, filtered and the solvent evaporated in vacuo to yield an oil, which was used without further purification in the next step.

### b) Preparation of Intermediate 132

By following a synthetic route similar to the one described for the synthesis of intermediate 76 and intermediate 77, 20 for the synthesis of intermediate 89, intermediate 76 and intermediate 77, intermediate 132 was obtained starting from intermediate 131.

### Example A42

### a) Preparation of Intermediate 134

$$\Gamma$$

NaOH 50% (7.832 g, 97.0 mmol) was added dropwise to a mixture of 2-chloro-3-(trifluoromethyl)phenylacetonitrile (2 g, 9.108 mmol), 1-bromo-2-chloroethane (2.65 mL, 31.85 mmol) and benzyltriethylammonium chloride (622 mg, 2.73 45 mmol). The mixture was stirred at 50° C. overnight, then water was added and the mixture was extracted with EtOAc. The organic layer was washed with 5% aq. HCl, then dried over MgSO<sub>4</sub>, filtered and the solvent was removed in vacuo. The product was purified by flash column chromatography (silica; hexane/EtOAc 100/0 to 80/20). The desired fractions were collected and concentrated in vacuo to yield intermediate 134 (1.52 g, 68%).

### b) Preparation of Intermediate 135

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By following a synthetic route similar to the one described for the synthesis of intermediate 128, intermediate 76 and intermediate 77, intermediate 135 was obtained starting from intermediate 134.

### Example A43

### a) Preparation of Intermediate 136

Methyl iodide (0.374 mL, 6.013 mmol) was added to a stirred solution of methyl 3-bromanyl-4-(trifluoromethyl) benzoate (1.348 g, 5.11 mmol) and K<sub>2</sub>CO<sub>3</sub> (2.77 g, 20.043 mmol) in DMF (15 mL) at r.t., and the mixture was stirred for 5 h at the same temperature. Water and EtOAc were then added. The aq. phase was extracted once more. The combined organics were dried over MgSO<sub>4</sub>, filtered and the solvent removed in vacuo. The crude was purified by flash column chromatography (silica; heptanes/EtOAc 100/0 to 95/5). The desired fractions were collected and concentrated in vacuo to yield intermediate 136 as a pale brown oil (quantitative).

### b) Preparation of Intermediate 137

Piperidine (0.748 mL, 7.568 mmol) was added to a stirred solution of intermediate 136 (1.785 g, 6.306 mmol), X-Phos (0.271 g, 0.568 mmol),  $Pd_2(dba)_3$  (0.231 g, 0.252 mmol) and  $Cs_2CO_3$  (4.11 g, 12.613 mmol) in toluene (20 mL) while  $N_2$  was bubbled through the mixture. The reaction was stirred overnight at 100° C. in a sealed tube. Water and EtOAc were then added. The aqueous phase was extracted once more, then the combined organics were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude was purified by flash column chromatography (silica; heptane/EtOAc 100/100 to 95/5). The

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desired fractions were collected and concentrated in vacuo to yield a pale yellow solid (3.623 g, 80%, 50% purity).

### c) Preparation of Intermediate 138

$$\bigcup_{0}^{N} \bigcup_{CF_{3}}$$

By following a synthetic route similar to the one described for the synthesis of intermediate 89, intermediate 76 and intermediate 77, intermediate 138 was obtained starting from 20 intermediate 137.

### Example A44

### a) Preparation of Intermediate 139

Selectfluor<sup>TM</sup> (896 mg, 2.528 mmol) was added to a solution of intermediate 14 (500 mg, 1.945 mmol) in acetonitrile (10 mL) and the mixture was stirred at 70° C. for 16 h. The solvent was then evaporated in vacuo and the crude product was purified by flash column chromatography ((silica; DCM/MeOH 100/0 to 90/10). The desired fractions were collected and concentrated in vacuo to yield intermediate 139 (295 mg, 55%).

### Example A45

Following a procedure similar to the one reported for the synthesis of intermediate 21, following intermediates were obtained from the commercially available corresponding alcohols:

Structure	Int. number
Br NO	140

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Structure	Int. number
CI	142

$$F_3C$$
  $Br$ 

Example A46

### a) Preparation of Intermediate 141

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ N & & \\ N & & & \\ N & &$$

Starting from intermediate 140 and intermediate 40, intermediate 141 (R-enantiomer) was prepared by analogy to the procedure reported for the synthesis of intermediate 34. A fraction containing the corresponding BOC-deprotected compound was also recovered.

### Example A47

### a) Preparation of Intermediate 145

NaH (60% in mineral oil, 124 mg, 3.09 mmol) was added to a solution of 4-methoxybenzylalcohol (0.427 g, 3.09 mmol) in THF (10 mL) at  $0^{\circ}$  C. The mixture was allowed to warm to r.t. and stirred for 30 min. 2,6-Dichloro-4-methoxy-

### 112

pyridine (0.5 g, 2.809 mmol) was added and the r.m. was heated at 100° C. in a sealed tube for 18 h, then allowed to cool down to r.t. The solvent was evaporated to dryness and the residue was dissolved in EtOAc and washed with water. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; DCM). The desired fractions were collected and concentrated in vacuo to yield intermediate 145 as a colourless oil (0.7 g, 89%).

### b) Preparation of Intermediate 146

Vinylboronic acid pinacol ester (4.9 mL, 28.6 mmol) was added to a stirred suspension of intermediate 145 (4 g, 14.3 mmol), Pd(OAc)<sub>2</sub> (482 mg, 2.145 mmol), S-Phos (1.761 g, 4.29 mmol) and K<sub>3</sub>PO<sub>4</sub> (9.106 g, 42.9 mmol) in a mixture of dioxane (40 mL) and EtOH (4 mL) under a N<sub>2</sub> atmosphere. The mixture was stirred at 100° C. for 18 h, then filtered through celite and the filtrate diluted with water and extracted with DCM. The organic layer was separated, washed with brine, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; DCM). The desired fractions were collected and concentrated in vacuo to yield intermediate 146 as a colourless oil, which solidified upon standing (3.47 g, 89%).

### c) Preparation of Intermediate 147

$$\bigcup_{O \subseteq O \setminus K^+}^{O}$$

Potassium permanganate (5.659 g, 35.811 mmol) was added portionwise to a cooled solution (0° C.) of intermediate 146 (3.47 g, 12.79 mmol) in acetone (24 mL) and the mixture was stirred at 0° C. for 10 min before being allowed to warm up to r.t. Stirring was continued for 1 h, then more potassium permanganate (0.6 eq.) was added at 0° C. and the mixture stirred at r.t. for 18 h. The mixture was filtered and the solid washed with water. The filtrate was concentrated to dryness. The residue was triturated with Et<sub>2</sub>O, the solid filtered and

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dried in vacuo to yield intermediate 147 as a white solid. The product was used in the following step without further purification  $(3.6\ g)$ .

### d) Preparation of Intermediate 148

$$R$$
 NH  $CF_3$ 

By following an analogous procedure as described for the synthesis of intermediate 1, intermediate 148 was obtained starting from the commercially available aldehyde (80%).

### e) Preparation of Intermediate 149

By following an analogous procedure as described for the synthesis of intermediate 3, intermediate 149 was obtained 35 starting from intermediate 147 and intermediate 148 (74%).

### f) Preparation of Intermediate 150

$$CF_3$$
 $CF_3$ 
 $CF_3$ 
 $CF_3$ 

By following an analogous procedure as described for the synthesis of intermediate 5, intermediate 150 was obtained starting from intermediate 149 (94%).

### g) Preparation of Intermediate 151

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By following an analogous procedure as described for the synthesis of intermediate 6, intermediate 151 was obtained starting from intermediate 150 (82%).

### h) Preparation of Intermediate 152

$$O$$
 $N$ 
 $R$ 
 $CF_3$ 

TFA (0.979 mL, 12.795 mmol) was added to a solution of intermediate 151 (975 mg, 2.245 mmol) and NIS (556 mg, 2.469 mmol) in acetonitrile (9.7 mL) and the r.m. was stirred at r.t. for 2 h. The reaction was quenched with a small amount of Na<sub>2</sub>SO<sub>3</sub>, diluted with DCM and NaHCO<sub>3</sub> sat. and extracted with DCM. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and evaporated to dryness. The resulting residue was purified by flash column chromatograpy (silica; DCM/EtOAc 100/0 to 90/10). The desired fractions were collected and concentrated in vacuo to yield intermediate 152 as a white solid (1 g, 80%).

### i) Preparation of Intermediate 158

$$\bigcap_{N} \bigcap_{N} \bigcap_{N} \bigcap_{CF_{3}} CF_{3}$$

Starting from intermediate 152, intermediate 158 was prepared by analogy to the procedure B3.a1 reported for the synthesis of compound 3 (75%, R-enantiomer).

### Example A48

### Preparation of Intermediate 153

$$F_3C$$
  $CF_3$   $H_2N$ 

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60

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By following an analogous procedure as described for the synthesis of intermediate 69 and intermediate 70, intermediate 153 was obtained starting from the commercially available alcohol.

### Example A49

### Preparation of Intermediate 154

$$F_3C$$
  $CF_2$   $H_2N$ 

By following an analogous procedure as described for the synthesis of intermediate 69 and intermediate 70, intermediate 154 was obtained starting from the commercially available alcohol.

### Example A50

### a) Preparation of Intermediate 157

$$- \bigvee_{N} \bigvee_{O} \bigvee_{O} \bigvee_{CF_{3}} CF_{3}$$

A mixture of oxalyl chloride (0.145 mL, 1.719 mmol) in DCM (20 mL) was stirred at -78° C. DMSO (0.182 mL) was added dropwise and the mixture was stirred at -78° C. for 10 min. A solution of compound 45 (410 mg, 0.819 mmol) in DCM (10 mL) was added dropwise and the mixture was stirred for 1 h at -78° C. DIPEA (1.41 mL, 8.19 mmol) was added dropwise and the mixture was allowed to reach r.t. under stirring. The r.m. was washed with water, dried on MgSO<sub>4</sub>, filtered and evaporated. The residue was purified via flash column chromatography (silica; DCM/MeOH 100/0 to 95/5). The pure fractions were evaporated, to afford an orange oil (370 mg, 91%).

### Example A51

### a) Preparation of Intermediate 159

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Chloro-difluoro-acetic acid methyl ester (1.38 mL, 13.082 mmol) was added to a stirred mixture of 1-bromo-3-iodo-5-trifluoromethoxybenzene (2 g, 5.451 mmol), potassium fluoride (380 mg, 6.541 mmol) and copper iodide (1.38 g, 6.541 mmol) in DMF (20 mL) in a sealed tube and under N<sub>2</sub> atmosphere. The mixture was stirred at 120° C. for 18 h, then it was diluted with water and extracted with Et<sub>2</sub>O. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; pentane). The desired fractions were collected and concentrated in vacuo (200 mbar) to yield a colourless oil (1.41 g, 83%, volatile compound).

### b) Preparation of Intermediate 160

Intermediate 159 (1.41 g, 4.563 mmol) was dissolved in 25 THF (15 mL). The solution was cooled to -78° C. under N<sub>2</sub> and butyl lithium (1.6M in hexanes, 3.1 mL, 5 mmol) was added dropwise over 15 min. Upon completion of the addition, the reaction mixture was stirred for 1 h at -78° C. Then, CO<sub>2</sub> (1.98 g, 45 mmol) was added at the same temperature.

The resulting solution was allowed to warm to r.t. for 3 h. Water and HCl 1N were added, and the mixture was extracted with EtOAc. The organic phase was separated, dried over MgSO<sub>4</sub>, filtered and concentrated in vacuo. The crude product was purified by flash column chromatography (silica; 5 heptanes/EtOAc 100/0 to 50/50). The desired fractions were collected and concentrated in vacuo to yield a beige solid (1.25 g, 38%).

### c) Preparation of Intermediate 161

By following an analogous procedure as described for the synthesis of intermediate 20 intermediate 161 was obtained starting from intermediate 160 (65%).

### Example A52

### a) Preparation of Intermediate 162

Phosphorus oxychloride (144  $\mu$ L, 1.55 mmol) was added dropwise at 5° C. to DMF (1.71 mL) and the mixture was further stirred for 5 min at 5° C. and then at r.t. for 45 min. The

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r.m. was then cooled again to  $5^{\circ}$  C. and 2-methyl-5-(trifluoromethyl)-1H-indole (220 mg, 1.11 mmol) was added portionwise. The reaction mixture was further stirred for 5 min at  $5^{\circ}$  C. and at r.t. for 45 min, then it was poured carefully onto ice and the solution was neutralized (pH=7) by the addition of a sat. NaHCO $_3$  sol. The product was extracted with EtOAc. The organic layer was washed with water, dried over MgSO $_4$  and the solvent was removed in vacuo to give a crude material, which was used in the subsequent step without further purification (320 mg).

### b) Preparation of Intermediate 163

Intermediate 162 (250 mg, crude material) was dissolved in THF (20 mL). di-Tert-butyl dicarbonate (293 mg, 1.34 mmol) and DMAP (1 mg, 11  $\mu$ mol) were added. The r.m. was stirred for 1 h at r.t., then NaHCO<sub>3</sub> sat. sol. was added. DCM 35 was added and the organic layer was separated. The organic layer was dried with MgSO<sub>4</sub>, filtered and the solvent was evaporated to give crude intermediate 163 as a brownish solid (225 mg). The crude was used as such in the next reaction. 40

### c) Preparation of Intermediate 164

Intermediate 163 (225 mg) was dissolved in MeOH (4.3 mL) and the r.m. was cooled to  $0^{\circ}$  C. by using an ice bath. 60 NaBH<sub>4</sub> (52 mg, 1.37 mmol) was then added. The reaction mixture was stirred for 1 h at r.t., then it was added to a mixture of DCM and NaHCO<sub>3</sub> sat. sol. The organic layer was separated, washed with brine, dried over MgSO<sub>4</sub>, filtered and the solvent was evaporated in vacuo to give a brownish crystalline crude (216 mg), which was used without further purification for the subsequent reaction.

### d) Preparation of Intermediate 165

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 21, intermediate 165 was obtained starting from intermediate 164.

### e) Preparation of Intermediate 166

Using experimental conditions analogous to those reported for the synthesis of intermediate 47, intermediate 166 was obtained starting from intermediate 165 and intermediate 40.

### Example A53

The following intermediates were synthesized by following an analogous synthetic sequence as reported in Example A52. For the preparation of some intermediates of the table below, only part of the synthetic sequence in A52 had to be followed, for example when one of the intermediates in the sequence was commercially available or in case a more efficient preparation for an intermediate in the sequence was well-known from literature.

	-continued			-(	120 continu	ed
Structure	Inter- me- diate num- ber	Starting material	5	Structure	Inter- me- diate num- ber	Starting material
O CI	<u> </u>	OH N O	10 15	Br Cl	229	Br
CI	169	NH	25	CI	232	ONH
O NO	181	OH N O	35 40	CI	234	
$F_3C$ $N$ $O$ $O$	F <sub>3</sub> C	O NH	45 50	CI		Cl
F <sub>3</sub> C Cl	212 F <sub>3</sub> C		<ul><li>55</li><li>60</li><li>65</li></ul>	F CI N O	239	$\Gamma$ $CF_3$

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### a) Preparation of Intermediate 175

Using experimental conditions analogous to those reported for the synthesis of intermediate 47, intermediate 175 was obtained starting from intermediate 169 and intermediate 40.

### Example A55

### a) Preparation of Intermediate 170

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 163, intermediate 170 was prepared starting from commercially available 40 5-chloro-6-fluoro-1H-indole-3-carbaldehyde.

### b) Preparation of Intermediate 171

To a stirred solution of intermediate 170 (6.34 g, 21.30 mmol) in THF (60 mL) at 0-5° C. was added methylmagnesium chloride (3 M in THF, 7.81 mL, 23.42 mmol). The reaction was stirred at 0-5° C. for 30 min and then warmed to r.t. The r.m. was then quenched with water. THF was evaporated and the product was extracted with DCM. The organic layer was dried over MgSO<sub>4</sub>, filtered and evaporated. The 65 crude was purified by flash column chromatography (silica; DCM/MeOH 100/0 to 98/2). The pure fractions were evapo-

### 122

rated and the product was crystallized from DIPE. The crystals were filtered off and dried, yielding intermediate 171 (6.68 g, quantitative yield).

### c) Preparation of Intermediate 172

Intermediate 171 (2 g, 6.38 mmol) was stirred in DCM (122 mL). MsCl (1.46 g, 12.75 mmol) was added, followed by Et<sub>3</sub>N (1.29 g, 12.75 mmol). The r.m. was stirred for 2 h at r.t. Sat. NaHCO<sub>3</sub> sol. was added and the organic layer was separated, dried over MgSO<sub>4</sub>, filtered, evaporated and co-evaporated with toluene to give a crude material containing intermediate 172, which was used without further purification for the subsequent reaction (2.5 g).

### d) Preparation of Intermediate 173

Intermediate 173 was obtained by following a synthetic procedure similar to the one reported for the synthesis of intermediate 40.

### d) Preparation of Intermediate 174

$$\begin{array}{c|c} & & & \\ & & & \\ N & & \\ N$$

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65

### e) Preparation of Intermediate 176

$$\begin{array}{c|c}
Cl & 10 \\
\hline
N & N & N
\end{array}$$
15

Using experimental conditions analogous to those reported for the synthesis of intermediate 47, intermediate 174 was obtained starting from intermediate 172 and intermediate 40. 25

### Example A56

### a) Preparation of Intermediate 177

By following a synthetic procedure similar to the one 45 reported for the synthesis of intermediate 164, starting from intermediate 170 intermediate 177 was obtained.

### b) Preparation of Intermediate 178

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By following a synthetic procedure similar to the one reported for the synthesis of intermediate 172, starting from intermediate 177 intermediate 178 was obtained.

### c) Preparation of Intermediate 179

Using experimental conditions analogous to those reported for the synthesis of intermediate 47, intermediate 179 was obtained starting from intermediate 178 and intermediate 40.

### Example A57

### a) Preparation of Intermediate 180

Using experimental conditions analogous to those reported for the synthesis of intermediate 47, intermediate 180 was obtained starting from intermediate 168 and intermediate 40.

### Example A58

### a) Preparation of Intermediate 182

Using experimental conditions analogous to those reported for the synthesis of intermediate 47, intermediate 180 was obtained starting from intermediate 181 and intermediate 40.

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(274 mg).

### c) Preparation of Intermediate 186

126

which was purified by flash column chromatography (silica;

heptanes/EtOAc 100/0 to 70/30) to give intermediate 185

### Example A59

### a) Preparation of Intermediate 184

$$CF_3$$

HN

 $R$ 

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6-(Trifluoromethyl)indan-1-ol (420 mg, 2.08 mmol) was dissolved in thionyl chloride (4.2 mL), stirred at r.t. for 5 h and then evaporated unti dryness and re-dissolved in N,N-dimethylacetamide (8 mL). D-Alaninol (0.32 mL, 4.16 mmol) and DIPEA (1.03 mL) were added, and the r.m. stirred overnight at 70° C. The r.m. was then quenched with sat. aq. NaHCO<sub>3</sub> and washed with EtOAc. The organic layer was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated until dryness to give a crude which was purified by flash column chromatography (silica; heptanes/EtOAc 100/0 to 0/100) to afford intermediate 184 (185 mg, 34%).

### b) Preparation of Intermediate 185

$$CF_3$$

HN

 $R$ 
 $Si$ 
 $Si$ 
 $Si$ 

To a stirred solution of intermediate 184 (185 mg, 0.71 mmol) in DCM (9.4 mL) was added imidazole (97 mg, 1.43 mmol) and then TBSC1 (161 mg, 1.07 mmol). The r.m. was stirred overnight at r.t. Water and DCM were then added and the phases separated. The water layer was again extracted with DCM. The combined organic layers were dried over  $Na_2SO_4$ , filtered and evaporated until dryness to give a crude

# CF<sub>3</sub>

Intermediate 54 (374 mg, crude material) was dissolved in acetonitrile (5 mL) and added in portions of 1 mL to a solution of intermediate 185 (141 mg, 0.38 mmol), DIPEA (0.65 mL, 3.78 mmol) and DMAP (231 mg, 1.89 mmol) in acetonitrile (3.5 mL) at 70° C. The r.m. was stirred at reflux for 90 min, then overnight at 70° C. The reaction was then allowed to cool to r.t. and treated with aq. NaHCO<sub>3</sub>. The mixture was extracted with EtOAc, the organic layer washed with water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated until dryness to give a crude, which was purified by flash column chromatography (silica; heptanes/EtOAc 100/0 to 50/50). After evaporation of the solvent, intermediate 186 was obtained (109 mg).

### d) Preparation of Intermediate 187

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189

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 4, intermediate 187 was obtained starting from intermediate 186.

### e) Preparation of Intermediate 188

10% Pd/C (27) mg was added to a solution of intermediate 187 (115 mg, 0.24 mmol) in MeOH (8.45 mL) at r.t. and the  $_{25}$  r.m. was hydrogenated (atmospheric pressure) for 2 h at r.t. The catalyst was filtered through Dicalite® and the filtrate was evaporated until dryness to give crude intermediate 188, which was used without further purification for the subsequent reaction (90 mg).

### f) Preparation of Intermediates 189 and 190

$$\bigcap_{O} \bigcap_{Or} \bigcap_{R} \bigcap_{OF_3} \bigcap_{CF_3} \bigcap_{CF_3}$$

Intermediate 188 (90 mg, 0.24 mmol) and TPP (87 mg, 0.33 mmol) were stirred in THF (3 mL) at r.t. and DIAD (64  $\mu$ L, 0.33 mmol) was then added dropwise via syringe. The r.m. was stirred overnight at r.t., then it was evaporated until dryness to give a crude, which was dissolved in a small amount of DCM and purified by flash column chromatography (silica; heptanes/EtOAc 95/5 to 0/100) to afford intermediate 189 (45 mg; the fraction contains also triphenylphosphine oxide) and intermediate 190 (19 mg).

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Intermediates 189 and 190 were used as such in the next reaction steps, without further purificiation.

### g) Preparation of Intermediate 191

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Bromine (4  $\mu$ L, 0.08 mmol) was added to a stirred solution of intermediate 189 (25 mg, not pure material) in DCM (0.6 mL) and AcOH (0.14 mL). The r.m. was stirred overnight at r.t. and it was then diluted with DCM and washed with a sat. NaHCO<sub>3</sub> sol. The organic layer was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated until dryness to give a crude, which was purified by flash column chromatography (silica; heptanes/EtOAc100/0 to 0/100). After collection of the fractions and evaporation of the solvent, intermediate 191 was obtained (16 mg, 79% LC-MS purity, 42%).

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 191, intermediate 192 was obtained starting from intermediate 190.

### Example A60

### a) Preparation of Intermediate 199

$$F_3C$$

Phosphorus oxychloride (703  $\mu$ L, 7.56 mmol) was added dropwise at 5° C. to DMF (8.36 mL) and the mixture was then stirred for 5 min at 5° C. and then at r.t. for 45 min. After this time the r.m. was cooled again to 5° C. and 5-trifluoromethylindole (1 g, 5.40 mmol) was added portionwise. The r.m. was further stirred for 5 min at 5° C., at r.t. for 45 min and at 60° C. for 1 h, then it was poured carefully onto ice and the

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solution was neutralized (pH=7) by the addition of a sat. NaHCO<sub>3</sub> sol. A part of the product precipitated and it was filtered off to give a first batch of intermediate 199. The aqueous filtrate was extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub> and the solvent was removed in vacuo to provide an oily residue, from which the product precipitated when triturated with water. Filtration of the precipitate provided a second batch of intermediate 199. The two batches were dried in the vacuum oven and used as such in the subsequent reaction step (1 g, 87%).

### b) Preparation of Intermediate 200

$$F_3C$$

Intermediate 199 (1 g, 4.69 mmol) was dissolved in THF (5 mL). di-Tert-butyl dicarbonate (1.251 g, 5.73 mmol) was added, followed by DMAP (5.73 mg, 0.05 mmol). The r.m. was stirred for 1 h at r.t. until LC-MS analysis showed full conversion. Sat. NaHCO<sub>3</sub> sol. was then added. Most of the product precipitated and it was filtered off and dried in the vacuum oven. DCM was added to the filtrate and the organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvent was evaporated. The residue was combined with the dried precipitate and coevaporated with toluene (×2) to afford a crude material, which was used without further purification for the subsequent reaction step (1.41 g, 96%).

### c) Preparation of Intermediate 201

$$F_3C$$
 OH

Methylmagnesium bromide (3M in THF, 8.62 mL, 25.86 mmol) was added to intermediate 200 (1.62 g, 5.17 mmol) in THF (17.3 mL) at 0° C. After 30 min, 4 more eq. of methylmagnesium bromide were added at  $-78^{\circ}$  C. After 30 min the r.m. was quenched at 0° C. with NH<sub>4</sub>Cl, brought to r.t. and poured in EtOAc/water. The organic layer was separated, the

aq. phase extracted with EtOAc, the combined organic layers dried, filtered and evaporated, to give a crude material, which was used without further purification in the subsequent reaction step (1.85 g).

### d) Preparation of Intermediate 202

$$F_3C$$
 OSO<sub>2</sub>CH<sub>3</sub>

Intermediate 201 (1.12 g, 3.40 mmol) was stirred in DCM (9.5 mL) and cooled to  $0^{\circ}$  C. Et<sub>3</sub>N (0.71 mL, 5.10 mmol) was then added, followed by methanesulphonyl chloride (0.40 mL, 5.10 mmol). After 20 min LC-MS analysis showed completion of the reaction. Sat. NaHCO<sub>3</sub> sol. was added and the organic layer was separated, dried over MgSO<sub>4</sub>, filtered, evaporated and co-evaporated with toluene to give a crude material, which was used without further purification in the subsequent reaction step (1.2 g).

### e) Preparation of Intermediate 203

$$\bigcap_{N = 1}^{N} \bigcap_{N = 1}^{N} \bigcap_{N = 1}^{N} \bigcap_{CF_3}$$

Intermediate 40 (0.659 g, 2.55 mmol) was stirred in DMF (200 mL) under  $\rm N_2$  atmosphere. NaH (60% in mineral oil, 122 mg, 3.06 mmol) was added and the mixture was stirred for 10 min. Intermediate 202 (1.2 g, crude material) was dissolved in a small amount of DMF and was added at 0° C. dropwise to the r.m. The reaction was stirred at r.t. overnight. Water was then added, the aq.layer extracted with EtOAc (×2), the organic layers collected, washed with brine, dried over MgSO<sub>4</sub>, filtered and the solvent evaporated, to give a crude which was used without further purification in the subsequent reaction step.

### a) Preparation of Intermediate 204

O F CI

132

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 202, intermediate 204 was obtained starting from commercially available 5-chloro-7-fluoroindole.

### Example A62

Using experimental conditions analogous to those reported for the synthesis of intermediate 47, starting from intermediate 14 and the corresponding mesylate, the following intermediates were obtained:

Intermediate number	Structure	Mesylate
205	O $O$ $O$ $O$ $O$ $O$ $O$ $O$ $O$ $O$	Intermediate 194
206	CI $CI$ $CI$ $CI$ $CI$ $CI$	Intermediate 195
207	CI $CI$ $CI$ $R$ $R$	Intermediate 196
208	$\bigcap_{N} F$ $CF_3$	Intermediate 197

45

50

### -continued

Intermediate number	Structure	Mesylate
209	$\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	Intermediate 198

Example A63

a) Preparation of Intermediate 210

# 20 O CF<sub>3</sub> O S O 30

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 172, intermediate 210 was obtained starting from commercially available 6-trifluoromethyl-1H-indole-3-carboxaldehyde.

### b) Preparation of Intermediate 211

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 203, intermediate 211 was obtained starting from intermediate 210.

### Example A64

### a) Preparation of Intermediate 213

$$\bigcap_{N \in \mathbb{N}} \bigcap_{N \in \mathbb{N}} \bigcap_{N \in \mathbb{N}} \bigcap_{CF_3} \bigcap_{CF_3$$

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 203, intermediate 213 was obtained starting from intermediate 40 and intermediate 212. Intermediate 213 was obtained together with impurities and was used as such in the next reaction step (see B37).

### Example A65

### a) Preparation of Intermediate 218

$$F_3C - \bigvee_F NH_2$$

2-Fluoro-6-iodo-4-(trifluoromethyl)-benzenamine (4.1 g, 13.44 mmol) was stirred in THF (29 mL) under N<sub>2</sub>. Bis (triphenylphosphine)palladium(II) dichloride (511 mg, 0.73 mmol), CuI (277 mg, 1.46 mmol), Et<sub>3</sub>N (29 mL, 210.01 mmol) and trimethylsilylacetylene (1.584 g, 16.13 mmol)
 were added and the mixture was stirred at r.t. for 30 min. EtOAc was added and the mixture was filtered over Dicalite®. The filtrate was evaporated and the residue was

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35

purified by flash column chromatography (silica; heptane/EtOAC 100/0 to 99/1) to afford intermediate 218 (2.77 g, 75%).

### b) Preparation of Intermediate 219

$$F_3C$$
  $NH_2$ 

Intermediate 218 (2.75 g, 10.03 mmol),  $K_2\mathrm{CO}_3$  (1.37 g, 10.03 mmol) and MeOH (60 mL) were stirred for 30 min. The solid  $K_2\mathrm{CO}_3$  was filtered off over Dicalite® and the filtrate was evaporated to yield intermediate 219 (2.04 g, quantitative).

### c) Preparation of Intermediate 220

$$F_3C$$
  $NH$ 

Intermediate 219 (2.04 g, 10.02 mmol), potassium tert-butoxide (2.25 g, 20.04 mmol) and 1-methyl-2-pyrrolidinone (30 mL) were stirred at r.t. for 24 h under  $\rm N_2$ . Water and HCl <sup>40</sup> (1N) were added until acidic pH, and then the product was extracted with DIPE. The organic layer was dried over MgSO<sub>4</sub>, filtered and evaporated. The residue was purified by flash column chromatography (silica; heptane/EtOAc 100/0 to 99/1) to afford intermediate 220 (1.15 g, 56%).

### d) Preparation of Intermediate 221

$$\begin{array}{c|c}
CF_3 \\
\hline
N \\
N \\
O
\end{array}$$

$$\begin{array}{c}
CF_3 \\
F
\end{array}$$

$$\begin{array}{c}
F \\
O
\end{array}$$

$$\begin{array}{c}
60 \\
\end{array}$$

By following a synthetic procedure similar to the one 65 described in example A52, intermediate 221 was obtained starting from intermediate 220 and intermediate 40.

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Example A66

### a) Preparation of Intermediate 222

Dihydropyran (0.655 mL, 7.162 mmol) and p-toluenesul-phonic acid (56.7 mg, 0.298 mmol) were added to a solution of 3-bromo-4-methylbenzyl alcohol (1.2 g, 5.968 mmol) in DCM (30 mL) at r.t. The mixture was stirred overnight at r.t., then diluted with DCM, washed with sat. NaHCO $_3$  (×2), dried over MgSO $_4$ , filtered and the solvent was evaporated. The crude product was purified by flash column chromatography (silica; hexane/EtOAc 100/0 to 80/20). The desired fractions were collected and concentrated in vacuo to yield a colourless oil (1.4 g, 82%).

### b) Preparation of Intermediate 223

In a vial, to a solution of intermediate 222 (1.4 g, 4.909 mmol) in cyclopropanemethanol (5 mL) were added  $\rm K_3PO_4$  (2.085 g, 9.82 mmol), 8-hydroxyquinoline (71 mg, 0.491 mmol) and CuI (187 mg, 0.982 mmol), while the mixture was degassed by bubbling  $\rm N_2$ . The mixture was stirred at 110° C. for 10 h, then diluted with water, extracted with EtOAc, dried over MgSO<sub>4</sub>, filtered and the solvent evaporated. The crude product was purified by flash column chromatography (silica; hexane/EtOAc 100/0 to 90/10). The desired fractions were collected and concentrated in vacuo to afford intermediate 223 (1.15 g).

### c) Preparation of Intermediate 224

HCl (1M, 0.208 mL) was added to a solution of intermediate 223 (1.15 g) in MeOH (6 mL) at r.t. and the mixture was

stirred for 2 h at the same temperature. The solvent was then evaporated, sat. NaHCO3 added and the product was extracted with EtOAc (x2). The combined organic phases were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude was purified by flash column chromatography (silica; heptanes/EtOAc 100/0 to 30/70). The desired fractions were collected and concentrated in vacuo to yield a sticky yellow oil (200 mg).

### d) Preparation of Intermediate 225

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 21, intermediate 225 was obtained starting from intermediate 224.

### Example A67

The following intermediates were synthesized by following an analogous synthetic sequence as reported in Example  $_{15}$  A60. For the preparation of some intermediates of the table below, only part of the synthetic sequence in A60 had to be followed, for example when one of the intermediates in the sequence was commercially available or in case a more efficient preparation for an intermediate in the sequence was well-known from literature. Separation into diastereoisomers, unless otherwise indicated, was obtained by standard flash column chromatography.

Structure	Intermediate number	Starting materials
	226	Cl NH NH Intermediate 40

### -continued

Structure	Intermediate number	Starting materials
$O \longrightarrow O$ $O \longrightarrow $	228	Cl NH NH Intermediate 40
$\bigcap_{N = -\infty}^{O} \bigcap_{N = -\infty}^{N} \bigcap_{N = -\infty}^{F} \bigcap_{CF_3}^{F}$	243	F <sub>3</sub> C NH Intermediate 173
$\bigcap_{N=0}^{O}\bigcap_{N}\bigcap_{N=0}^{O}\bigcap_{N}\bigcap_{N=0}^{O}\bigcap_{N}\bigcap_{N=0}^{O}\bigcap_{N}\bigcap_{N}\bigcap_{N}\bigcap_{N}\bigcap_{N}\bigcap_{N}\bigcap_{N}\bigcap_{N$	244	Intermediate 202 Intermediate 173
ON NOTICE STATE OF ST	245	Intermediate 210 Intermediate 173

### -continued

Structure	Intermediate number	Starting materials
	262*	Cl NH NH Intermediate 173

\*the reaction was run in the presence of 0.7 equivalents of 18-crown-6

# Example A68

By using experimental conditions analogous to those reported for the synthesis of intermediate 47, the following 25 intermediates were obtained:

Intermediate number	Structure	Starting materials
230	DO NO Br	Intermediate 229 Intermediate 40

# -continued

Intermediate number	Structure	Starting materials
235		Intermediate 234 Intermediate 40

### Example A69

### a) Preparation of Intermediate 231

A pressure tube was charged with intermediate 230 (780 mg, 1.377 mmol), cyclopropylboronic acid (154 mg, 1.798 mmol), tricyclohexylphosphine (39 mg, 0.139 mmol) and <sup>20</sup> toluene (7 mL). Then, Pd(OAc)<sub>2</sub> (32 mg, 0.07 mmol) and

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ground  $\rm K_3PO_4$  (1.023 g, 4.819 mmol) were added. Finally, water (0.4 mL) was added and the tube was capped and placed in a preheated oil bath of 120° C. and stirring was continued for 3 h. The r.m. was cooled and diluted with water (30 mL). The layers were separated and the aq. layer was extracted with toluene (50 mL). The combined organic layers were treated with brine (20 mL), dried over MgSO<sub>4</sub>, filtered and concentrated in vacuo. The residue was purified by flash column chromatography (silica; DCM/MeOH 100/0 to 98/2) to afford intermediate 231 (410 mg, 56%).

### Example A70

The following intermediates were synthesized by following an analogous synthetic sequence as reported in Example A56. For the preparation of some intermediates of the table below, only part of the synthetic sequence in A56 had to be followed, for example when one of the intermediates in the sequence was commercially available or in case a more efficient preparation for an intermediate in the sequence was well-known from literature.

Structure	Intermediate number	Starting materials
O N O O O O O O O O O O O O O O O O O O	236 F <sub>2</sub>	ONH Intermediate 40

# -continued

-continued						
Structure	Intermediate number	Starting materials				
O O O CI	238	C1 NH Intermediate 173				
O CI O CI O CI	242	Cl NH NH Intermediate 173				
$\bigcap_{N=0}^{O}\bigcap_{N}\bigcap_{F}\bigcap_{CF_{3}}$	260	Intermediate 259 Intermediate 173				
$O \longrightarrow O$ $N \longrightarrow O$ $N \longrightarrow V$ $N \longrightarrow $	261	CI NH Intermediate 173				

20

35

55

Example A71

a) Preparation of Intermediate 246

$$NH_2$$
 $CF_3$ 
 $CF_3$ 

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 20, intermediate 15 245 was obtained starting from commercially available 3-amino-4-[3,5-bis(trifluoromethyl)phenyl]butanoic acid (661 mg).

b) Preparation of Intermediate 247

$$NH_2$$
 $CF_3$ 
 $CF_3$ 
 $CF_3$ 
 $CF_3$ 

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 2, intermediate 247 was obtained starting from intermediate 246 (0.57 g).

c) Preparation of Intermediate 248

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 6, intermediate 248 was obtained starting from methyl 5-bromo-6-oxo-1,6-dihydropyridine-2-carboxylate and benzyl alcohol (6.942 g, quantitative yield).

d) Preparation of Intermediate 249

Using experimental conditions similar to those reported for  $_{65}$  the synthesis of intermediate 128, intermediate 249 was obtained starting from intermediate 248 (2.94 g).

e) Preparation of Intermediate 250

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 3, intermediate 250 was obtained starting from intermediate 247 and intermediate 249 (571 mg).

f) Preparation of Intermediate 251

By following a synthetic procedure similar to the one reported for the synthesis of intermediate 34, intermediate 251 was obtained starting from intermediate 250 (130 mg).

g) Preparation of Intermediate 252

$$\bigcap_{\mathrm{OH}} \bigcap_{\mathrm{OH}} \bigcap_{\mathrm$$

35

55

### h) Preparation of Intermediate 253

$$Br$$
 $O$ 
 $N$ 
 $F_3C$ 
 $CF_3$ 

By following a synthetic procedure similar to the one <sup>20</sup> reported for the synthesis of intermediate 6, intermediate 253 was obtained starting from intermediate 252 (45 mg).

### Example A72

# a) Preparation of Intermediate 254

$$F_3C$$
 $NO_2$ 

(Trifluoromethyl)trimethylsilane (5.807 mL, 39.285 mmol) was added to a stirred suspension of 1-iodo-2-fluoro-4-methyl-5-nitro-benzene (5.52 g, 19.642 mmol), copper iodide (5.611 g, 29.464 mmol) and potassium fluoride (2.282 g, 39.285 mmol) in dry DMF (75 mL). The resulting solution was stirred at 70° C. Reaction allowed to go on overnight, then cooled down, diluted with EtOAc, washed with sat. 45 NH<sub>4</sub>Cl, water and brine. The aq. phase was filtered through celite because some insoluble salts created an emulsion: some additional org. phase collected from the filtrate. The org. phase was collected, dried and the solvent was evaporated. The residue was adsorbed on silica and purified via flash 50 column chromatography, to give an orange oil. The material was used as such in the subsequent reaction step (3.18 g, 73%).

### b) Preparation of Intermediate 255

$$F_3C$$
 $N$ 
 $NO_2$ 
 $NO_2$ 
 $NO_2$ 
 $NO_2$ 

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N,N-Dimethylformamide dimethyl acetal (9.359 mL, 70.453 mmol) was added to a solution of intermediate 254 (3.144 g, 14.091 mmol) in dry DMF (11 mL), and the r.m. was stirred at 100° C. Straight after the addition the reaction turns from red to dark green. After 1 h GC-MS shows complete conversion to the desired product. Reaction allowed to cool down, then poured into water: a dark red ppt is formed, which was dried in vacuo at 40° C. overnight and then in vacuo at r.t. over the weekend. The material was used without further purification for the subsequent reaction (3.344 g, 85%).

### c) Preparation of Intermediate 256

Intermediate 255 (3.344 g, 12.02 mmol), silica gel (1.45 g) and iron powder (3.393 g, 132.371 mmol) in toluene (49 mL) and AcOH (30 mL) were heated at 90° C. for 90 min, then the mixture was cooled to r.t., diluted with EtOAc and filtered on silica gel. The filtrate was evaporated and the resulting residue dissolved in DCM and washed with NaHCO<sub>3</sub> sat., then with water and brine. The org. layer was dried over MgSO<sub>4</sub>, filtered and evaporated to give a dark oil, used without further elaboration in the subsequent reaction step (2.587 g, quantitative).

### d) Preparation of Intermediate 257

By following a synthetic procedure similar to the one reported in Example A52, intermediate 257 was obtained starting from intermediate 256 and intermediate 40.

### Example A73

### a) Preparation of Intermediate 258

$$NH_2$$
 $CF_3$ 
 $E/Z$  mixture

40

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To a solution of 3-fluoro-2-iodo-4-(trifluoromethyl)-benzenamine (1.933 g, 6.337 mmol) and 2-(2-ethoxyethenyl)-4, 4,5,5-tetramethyl-1,3,2-dioxaborolane (1.632 g, 8.238 mmol) in DMF (13 mL) was added lithium hydroxide monohydrate (0.798 g, 19.011 mmol) while  $\rm N_2$  was bubbled 5 through the reaction mixture. Pd(dppf)Cl $_2$  (0.155 g, 0.190 mmol) was added and the stirring continued at 70° C. overnight. The reaction mixture was partitioned between water and EtOAc and the org. phase was dried over MgSO $_4$ , filtered and concentrated in vacuo. The crude product was purified by 10 flash column chromatography (silica; hexane/EtOAc 100/0 to 80/20). The desired fractions were collected and concentrated in vacuo to give the desired product (1.249 g, 79%).

### b) Preparation of Intermediate 259

$$CF_3$$
 $F$ 
 $NH$ 

Intermediate 258 (1.249 g, 5.012 mmol) was dissolved in AcOH (28.6 mL) in a sealed tube. The resulting mixture was then heated at reflux (125° C.) for 2 h. The solvent was evaporated in vacuo and the residual AcOH were removed by azeotropic evaporation with toluene. The crude product was purified by flash column chromatography (silica; hexane/ EtOAc 100/0 to 70/30). The desired fractions were collected and concentrated in vacuo to give the desired product (840 mg, 83%).

# B. Preparation of the Final Compounds

### Example B1

### a) Preparation of Compound 1

$$- \bigvee_{N}^{O} \bigvee_{N}^{N} \bigvee_{F_3C}^{CF_3} \cdot_{HCl}$$

CuI (0.096 g, 0.506 mmol) was added to a suspension of 55 intermediate 7 (1.26 g, 2.53 mmol),  $Cs_2CO_3$  (2.1 g, 6.32 mmol), 4-methylimidazole (0.415 g, 5.06 mmol) and N,N-dimethyl-1,2-cyclohexanediamine (0.080 mL, 0.506 mmol) in DMF (25 mL), while the reaction was degassed by bubbling  $N_2$  through the solution. The mixture was heated at  $106^\circ$  60 C. overnight. Water was then added and the aq. layer was extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The crude product was purified by flash column chromatography (silica; MeOH/DCM 0/100 to 3/97). The 65 desired fractions were collected and concentrated in vacuo. The residue was dissolved in EtOAc, and HCl 4N was added

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(1 eq). Compound 1 (R-enantiomer) was obtained pure after recrystallization from DIPE (0.177 g, 13%).

### Example B2

### a) Preparation of Compound 2

CuI (0.118 g, 0.62 mmol) was added to a solution of intermediate 29 (1.293 g, 3.1 mmol), 4-methylimidazole (0.255 g, 3.1 mmol), Cs<sub>2</sub>CO<sub>3</sub> (2.12 g, 6.51 mmol) and N,N'-dimethyl-1,2-cyclohexanediamine (0.098 mL, 0.62 mmol) in DMF (15 mL), while the reaction was degassed by bubbling N<sub>2</sub> through the solution. The mixture was then heated at 110° C. overnight. Water was added and the aq. layer was extracted with EtOAc. The phases were separated and the organic layer was dried over MgSO<sub>4</sub>, filtered and the solvent removed under reduced pressure. The crude product was purified by flash column chromatography (silica; MeOH/DCM 0/100 to 3/97).

The desired fractions were collected and concentrated in vacuo to afford compound 2 (0.750 g, 58%; R-enantiomer).

### Example B3

### a) Preparation of Compound 3

$$\bigcap_{N} \bigcap_{N} \bigcap_{R} \bigcap_{F_3C} CF_3$$

CuI (0.221 g, 1.16 mmol) was added to a suspension of intermediate 15 (2.8 g, 5.8 mmol), Cs<sub>2</sub>CO<sub>3</sub> (4.7 g, 14.5 mmol), 4-methylimidazole (0.952 g, 11.6 mmol) and N,N-dimethyl-1,2-cyclohexanediamine (0.183 mL, 1.16 mmol) in DMF (58 mL), while the reaction was degassed by bubbling N<sub>2</sub> through the solution. The mixture was heated at 106° C. overnight. Water was then added and the aq. layer was extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The crude product was purified by flash column chromatography (silica; MeOH/DCM 0/100 to 3/97). The desired fractions were collected and concentrated in vacuo to afford compound 3 (0.604 g, 21%; R-enantiomer).

### a1) Alternative Preparation of Compound 3

In a first vial equipped with a magnetic stir bar and a screw cap septum, a solution of Pd<sub>2</sub>(dba)<sub>3</sub> (6 mg, 0.006 mmol) and 2-di-tert-butylphosphino-3,4,5,6-tetramethyl-2',4',6'-triiso-propy 1-1,1'-biphenyl (6 mg, 0.0125 mmol) in dioxane (0.17

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25

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mL) and toluene (0.83 mL) was flushed with  $N_2$  and then stirred at  $120^{\circ}$  C. for 3 min. A second vial, equipped with a magnetic stir bar and a screw cap septum, was charged with 4-methylimidazole (45 mg, 0.55 mmol) and  $K_3PO_4$  (213 mg, 1 mmol), then with intermediate 15 (242 mg, 0.5 mmol) and also flushed with  $N_2$ . The premixed catalyst solution was added by syringe to the second vial. The r.m. was heated at  $120^{\circ}$  C. for 5 h. The reaction was cooled to r.t, diluted with EtOAc, washed with brine, dried over MgSO<sub>4</sub> and concentrated in vacuo. The crude product was purified by flash column chromatography (silica MeOH/DCM 0/100 to 3/97). The fractions were collected and the solvent was evaporated to give compound 3 (170 mg, 70%; R-enantiomer).

### Example B4

### a) Preparation of Compound 4

CuI (0.041 g, 0.21 mmol) was added to a solution of intermediate 22 (0.476 g, 1.07 mmol),  $Cs_2CO_3$  (0.871 g, 2.68 mmol), 4-methylimidazole (0.176 g, 2.16 mmol) and N,N'-dimethyl-1,2-cyclohexanediamine (0.038 mL, 0.214 mmol) in DMF (5 mL), while the reaction was degassed by bubbling  $N_2$  through the solution. The mixture was heated at 110° C. overnight. Water was then added and the aq. layer was extracted with EtOAc. The organic layer was dried (MgSO<sub>4</sub>), filtered and the solvent was removed under reduced pressure. The crude product was purified by flash column chromatography (silica; MeOH/DCM 0/100 to 3/97). The desired fractions were collected and concentrated in vacuo. The product was dissolved in EtOAc (0.5 mL) and HC1 4M was added (1.0 eq). Compound 4 was obtained pure after recrystallization from DIPE (127 mg, 25%; R-enantiomer).

### Example B5

### a) Preparation of Compound 5

CuI (0.102 g, 0.538 mmol) was added to a solution of intermediate 34 (1.1 g, 2.69 mmol),  $Cs_2CO_3$  (1.84 g, 5.65 mmol), 4-methylimidazole (0.22 g, 2.69 mmol) and N,N- 65 dimethyl-1,2-cyclohexanediamine (0.085 mL, 0.538 mmol) in DMF (10 mL), while the reaction was degassed by bub-

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bling  $N_2$  through the solution. The mixture was heated at  $110^\circ$  C. overnight. Water was then added and the aq. layer was extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The crude product was purified by flash column chromatography (silica; MeOH/DCM 0/100 to 3/97). The desired fractions were collected and concentrated in vacuo, to afford compound 5 (0.5 g, 45%).

### Example B6

## a) Preparation of Compound 6

NaH (60% in mineral oil, 0.036 g, 0.91 mmol) was added to a stirred solution of intermediate 40 (crude material, 0.83 mmol) in DMF (2.5 mL) at 0° C. under N<sub>2</sub> atmosphere. The mixture was stirred for 15 min and then intermediate 99 (crude material, 0.83 mmol) was added. The r.m. was stirred for 4 h at r.t., then it was quenched with water and extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, filtered and evaporated. The crude material was purified by flash column chromatography (silica; DCM-MeOH (9:1, v/v)/ DCM, 0/100 to 35/65). The desired fractions were collected and concentrated in vacuo. The product (0.07 g) was dissolved in EtOAc (2 mL), then HCl 4M in dioxane was added to obtain the hydrochloride salt. The solvent was evaporated and the product was triturated with DIPE to yield the compound 6 as a pale brown solid (0.048 g, 69% over two steps; R-enantiomer).

### Example B7

### a) Preparation of Compound 7

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & &$$

Ammonium acetate (659 mg, 8.55 mmol) was added to a stirred solution of intermediate 19 (1.71 mmol, crude material) in AcOH (3.5 mL) at r.t. The mixture was stirred for 1 h at reflux, then cooled to r.t. and poured into ice/water. NaOH 50% aq. was added slowly until basic pH. The product was

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extracted with EtOAc (×2). The combined organic phases were dried over MgSO<sub>4</sub>, filtered and evaporated. The crude was purified by flash column chromatography (silica; EtOAc/heptane 0/100 to 90/10). The desired fractions were collected and concentrated in vacuo. The product was dissolved in EtOAc, then 4M HCl in dioxane was added to obtain the HCl salt. The solvent was evaporated and the product was triturated with DIPE to yield compound 7 as a white solid (35% over 4 steps; R-enantiomer).

mmol) in dry DMF (30 mL), while the reaction was degassed by bubbling N<sub>2</sub> through the solution. The mixture was heated at 110° C. for 24 h. Water was then added and the aq. layer was extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The crude product was purified by flash column chromatography (silica; MeOH in DCM 0/100 to 3/97). The desired fractions were collected and concentrated in vacuo. over 4 steps; R-enantiomer).

### Example B8

### a) Preparation of Compound 8

$$\bigcap_{N}\bigcap_{O}\bigcap_{N}\bigcap_{F_{3}C}CF_{3}$$

Dimethyl amine (40% solution in water, 0.062 mL, 0.49 mmol) was added to a solution of intermediate 103 (crude 30 material, 0.098 mmol) in DMF (5 mL) at 0° C. The mixture was stirred at 80° C. overnight, then 5 additional eq. of dimethylamine were added. The mixture was stirred at 80° C. overnight. EtOAc and water were added. The organic phase was separated and dried over MgSO<sub>4</sub>, filtered and concentrated to dryness. The crude product was purified by flash column chromatography (silica; MeOH/DCM 0/100 to 3/97). The desired fractions were collected and concentrated in vacuo. The solid was triturated from DIPE to afford compound 8 (0.010 g, 19% over two steps; R-enantiomer).

### Example B9

a) Preparation of Compounds 9 (Racemic Mixture), 9a (R or S Enantiomer) and 9b (S or R Enantiomer)

CuI (0.120 g, 0.633 mmol) was added to a solution of intermediate 57 (1.316 g, 3.163 mmol),  $Cs_2CO_3$  (2.576 g, 7.907 mmol), 4-methylimidazole (0.519 g, 6.325 mmol) and N,N'-dimethyl-1,2-cyclohexanediamine (0.1 mL, 0.633

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mmol) in dry DMF (30 mL), while the reaction was degassed by bubbling N<sub>2</sub> through the solution. The mixture was heated at 110° C. for 24 h. Water was then added and the aq. layer was extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The crude product was purified by flash column chromatography (silica; MeOH in DCM 0/100 to 3/97). The Compound 9 was obtained pure after recrystallization from 5% EtOAc in DIPE (435 mg, 33%, rac), and subsequently separated into its enantiomers by preparative SFC (Chiralpak Diacel OJ 20×250 mm; mobile phase (CO<sub>2</sub>, MeOH with 0.2% iPrNH<sub>2</sub>)). The fractions containing the separated enantiomers were collected, evaporated, dissolved again in MeOH and the solvent evaporated again, to afford the desired pure compounds. Yield: 165 mg of compound 9a (40%; R or S enantiomer) and 161 mg of compound 9b (39%; S or R enantiomer).

### Example B10

a) Preparation of Compounds 10 (Racemic Mixture), 10a (R or S Enantiomer) and 10b (S or R Enantiomer)

CuI (73 mg, 0.384 mmol) was added to a solution of intermediate 56 (0.8 g, 1.92 mmol), Cs<sub>2</sub>CO<sub>3</sub> (1.56 g, 4.8 mmol), 4-methylimidazole (0.315 g, 3.84 mmol) and N,N-dimethyl-1,2-cyclohexanediamine (0.06 mL, 0.384 mmol) in dry DMF (19 mL), while the reaction was degassed by bubbling N<sub>2</sub> through the solution. The mixture was heated at 106° C. overnight. Water was then added and the aq. layer was 55 extracted with EtOAc. The organic layer was dried over MgSO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The crude product was purified by flash column chromatography (silica; MeOH in DCM 0/100 to 3/97). The desired fractions were collected and concentrated in vacuo. Compound 10 (racemic) was obtained pure after recrystallization from DIPE (138 mg, 17%, rac), and subsequently separated into its enantiomers by preparative SFC (Chiralcel Diacel OD 20×250 mm; mobile phase (CO<sub>2</sub>, MeOH with 0.2% iPrNH<sub>2</sub>)). The fractions containing the separated enantiomers were collected, evaporated, dissolved again in MeOH and the solvent evaporated again, to afford the desired pure

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compounds. Yield: 46 mg of compound 10a (39%; R or S enantiomer) and 49 mg of compound 10b (41%; S or R enantiomer).

### Example B11

a) Preparation of Compounds 11 (Racemic), 11a (R or S Enantiomer) and 11b (S or R Enantiomer)

$$\bigcap_{N} \bigcap_{N} \bigcap_{CF_3}$$

4-Methylimidazole (246 mg, 3 mmol), copper iodide (38 mg, 0.2 mmol) and Cs<sub>2</sub>CO<sub>3</sub> (977 mg, 3 mmol) were added to the crude r.m. containing intermediate 58 (crude material, 1 mmol in 20 mL of DMF). N<sub>2</sub> was bubbled through the mixture for 5 min before this was heated in a sealed flask at 110° C. under N<sub>2</sub> atmosphere for 48 h. The mixture was cooled to r.t. and partitioned between DCM (50 mL) and water (50 mL). Because the aqueous phase was cloudy, EDTA (5.6 g, 15 mmol) was added and the biphasic mixture was shaken for 5 min, until the aq. phase became clear. The organic layer was separated and the aqueous layer was extracted with DCM (2×50 mL). The combined organic layers were washed with brine, dried (MgSO<sub>4</sub>), filtered and the solvent was evaporated providing the racemic compound 11 as a brown oil, that was separated into enantiomers by preparative HPLC (Chiralpak Diacel OD 20×250 mm, mobile phase (24% MeOH with 0.2% iPrNH<sub>2</sub>/76% CO<sub>2</sub> hold 12.0 min, then from 24 to 50% MeOH with 0.2% iPrNH2 at 15% rate and hold 2.0 min at 50%)). Yield: 112 mg of compound 11a (beige solid, 27%; R or S enantiomer; enantiomer A (SFC-MS)) and 109 mg of compound 11b (beige solid, 26%; S or R enantiomer; enantiomer B (SFC-MS)).

### Example B12

### a) Preparation of Compound 12

$$\bigcup_{N} \bigcup_{O} \bigcup_{N} \bigcup_{O} \bigcup_{F_{3}C} CF_{3}$$

Starting from (S)-2-amino-3-methoxy-1-propanol and 3,5-bis(trifluoromethyl)benzaldehyde, compound 12

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(S-enantiomer) was prepared by analogy to the procedures reported for the synthesis of compound 7.

### Example B13

a) Preparation of Compound 13

Boron tribromide (0.093 mL, 0.98 mmol) was added to a solution of compound 12 (0.052 g, 0.098 mL) in DCM (2 mL). The mixture was stirred at r.t. for 5 h, then sat. aq. NaHCO $_3$  was added. The product was extracted with DCM (×2), the combined organics were dried over MgSO $_4$ , filtered and evaporated, yielding compound 13 (S-enantiomer).

### Example B14

a) Preparation of Compounds 14 (Racemic), 14a (Cis A) and 14b (Cis B)

$$- \bigvee_{N} \bigvee_{O} \bigvee_{N} \bigvee_{F_{3}C} CF_{3}$$

Starting from intermediate 59, compound 14 was obtained following an analogous synthetic route as described for compound 7. The mixture was subsequently separated into enantiomers by preparative SFC (Chiralpak Diacel OD 20×250 mm, mobile phase (CO<sub>2</sub>, MeOH with 0.2% iPrNH<sub>2</sub>)). Yield: 23 mg of compound 14a (49%; R,S or S,R enantiomer; cis A) and 19 mg of compound 14b (40%; S,R or R,S enantiomer; ocis B).

### Example B15

a) Preparation of Compound 111

$$\begin{array}{c|c} O & & \\ \hline & N & \\$$

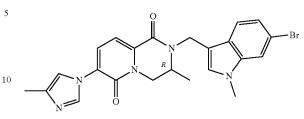
162 Example B18

S-diastereoisomer) was prepared by analogy to the procedure B3.a1, reported for the synthesis of compound 3. The product was subsequently dissolved in EtOAc and treated with a small excess of HCl (4M in dioxane) to obtain the hydrochloride salt (.HCl). The solvent was then evaporated and the residual

triturated with DIPE to give a white solid (38%) (1R or 1S,

3R; diastereomer A (SFC-MS)).

# a) Preparation of Compound 177



Compound 125 (170 mg, 0.365 mmol) was dissolved in

DMF (13 mL) under  $N_2$  atmosphere, then NaH (60% in mineral oil, 17 mg, 0.413 mmol) was added. The r.m. was stirred at r.t. for 10 min, then methyl iodide (0.025 mL, 0.402 mmol) was added. The r.m. was stirred at r.t. for 2 h. An additional amount of methyl iodide (15  $\mu L$ ; 0.65 eq) was added, and the reaction stirred overnight. 11  $\mu L$  More methyl iodide (11  $\mu L$ ; 0.5 eq) was added and the r.m. stirred for 3 h. Subsequently, 14 mg of NaH was added (0.95 eq) and the reaction was

allowed to stir for an additional 3 h, then the solvent was evaporated and the r.m. added dropwise to a mixture of DCM and water, and the mixture allowed to agitate for 20 min. The organic layer was then separated, dried over MgSO<sub>4</sub> and filtered. The solvent was evaporated in vacuo to give a crude which was purified via flash column chromatography (silica;

### a) Preparation of Compound 112

 $\begin{array}{c|c} & & & & \\ & &$ 

Starting from intermediate 105, compound 112 was prepared by analogy to the procedure B3.a1 reported for the 30 synthesis of compound 3. The product was subsequently dissolved in EtOAc and treated with a small excess of HCl (4M in dioxane) to obtain the hydrochloride salt (.HCl). The solvent was then evaporated and the residual triturated with 35 DIPE to give a white solid (43%; 15 or 1R, 3R; diastereomer B (SFC-MS)).

### Example B19

40%; R-enantiomer).

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MeOH/DCM 0/100 5/95): the desired fractions were col-

lected, the solvent removed in vacuo and the residue was

suspended in DIPE, the precipitate was filtered off and dried in vacuo at 40° C. overnight to afford compound 177 (69 mg,

### Example B17

### a) Preparation of Compound 179

$$\begin{array}{c|c} O & & & \\ \hline & N & & \\ N & & & \\ N & & & \\ \end{array}$$

Starting from intermediate 142 and intermediate 40, compound 179 (R-enantiomer) was prepared by analogy to the procedure reported for the synthesis of intermediate 34. The deprotected form was obtained as the sole product after flash column chromatography (25%; R-enantiomer).

Intermediate 141 (470 mg, 0.83 mmol) was dissolved in DCM (9.4 mL) at r.t. under  $\rm N_2$  atmosphere. TFA (1 mL, 13.067 mmol) was added and the r.m. stirred overnight at r.t., then added dropwise to a mixture DCM/NaHCO $_{\rm 3}$  (sat. aq.) and stirred for 20 min. The organic layer was separated, washed with brine, dried over MgSO $_{\rm 4}$ , filtered and evaporated in vacuo to give a crude, which was suspended in DIPE, under stirring. The solid so obtained was filtered and dried in vacuo at 45° C., to afford compound 125 (351 mg, 58%; R-enantiomer).

### Example B20

### a) Preparation of Compounds 175, 127 and 126

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Starting from intermediate 40 and intermediate 106, compound 175 was obtained following an analogous synthetic route as described for compound 6, in the presence of 1 eq. of lithium bromide. The mixture was subsequently separated into diastereoisomers by flash column chromatography (silica; DCM/EtOAc 100/0 to 20/80). The desired fractions were collected and concentrated in vacuo. The separated diastereoisomers were dissolved in DCM and 4M HCl in dioxane was added (1 eq.) to obtain the HCl salt. The solvent was evaporated and the products triturated with Et<sub>2</sub>O. Yield: 75 mg of compound 127 (8%; R,S or R,R diastereoisomer, con-

### Example B21

taining 4% of compound 126) and 27 mg of compound 126

(3%; R,R or R,S diastereoisomer).

### a) Preparation of Compounds 130 and 129

$$\begin{array}{c} 20 \\ 129 \\ \\ N \end{array}$$

Starting from intermediate 40 and intermediate 143, a mixture of compound 130 and compound 129 was obtained as a HCl salt form following an analogous synthetic route as described for compound 6 (Example B6), in the presence of 1 eq. of lithium bromide. The mixture was subsequently separated into diastereoisomers by preparative SFC (Chiralpak 45 Diacel AD 30×250 mm, mobile phase (CO<sub>2</sub>, EtOH with 0.2% iPrNH<sub>2</sub>)). Yield: 105 mg of compound 130 (23%; 1S or 1R, 3R; diastereomer A (SFC-MS)) and 182 mg of compound 129 (41%; 1R or 1S, 3R; diastereomer B (SFC-MS)).

### Example B22

### a) Preparation of Compound 150

$$\begin{array}{c|c} CI & CF_3 \\ \hline \\ N & O \end{array}$$

Phosphorous oxychloride (1 eq.) was added to a stirred solution of intermediate 158 (45 mg, 0.0875 mmol) in aceto-

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nitrile (0.9 mL). The mixture was heated at 90° C. in a sealed tube for 18 h, then more phosphorous oxychloride (1 eq.) was added and the mixture was heated at 90° C. for 18 h. Another eq. of phosphorous oxychloride was added and the reaction heated at 90° C. for additional 18 h, then diluted with DCM and washed with sat. NaHCO $_3$ . The organic layer was separated, dried over MgSO $_4$ , filtered and concentrated in vacuo. The residue was purified by flash column chromatography (silica; EtOAc). The desired fractions were collected and concentrated in vacuo. Yield: 21 mg of compound 150 (46%, R-enantiomer).

### Example B23

### a) Preparation of Compound 167

Methylmagnesium bromide (1.6M in THF, 0.85 mL, 1.184 mmol) was added portionwise to a stirred solution of intermediate 157 (370 mg, 0.742 mmol) in dry THF (15 mL) under N<sub>2</sub> at 5° C. The mixture was stirred at 5° C. for 3 h, then additional methylmagnesium bromide (0.37 mL, 0.519 mmol) was added portionwise under N<sub>2</sub> at 5° C. The r.m. was stirred at r.t. overnight, then it was diluted with sat. NH<sub>4</sub>Cl and extracted with EtOAc. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and the solvents evaporated in vacuo. The crude product was purified by flash column chromatography (silica; DCM/MeOH 100/0 to 95/5). The desired fractions were collected and concentrated in vacuo to yield a yellow oil. The product was purified by reversed phase chromatography twice, using (81% water with 0.1% TFA/19% MeCN-MeOH to 45% water with 0.1% TFA/55% MeCN-MeOH) and then (75% water (25 mM NH<sub>4</sub>HCO<sub>3</sub>)/25% MeCN-MeOH to 0% water (25 mM NH<sub>4</sub>HCO<sub>3</sub>)/100% MeCN-MeOH). Yield: 9 mg of compound 167 (2%).

### Example B24

### a) Preparation of Compound 181

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-continued

At r.t. and under  $N_2$  atmosphere, NaH (60% in mineral oil, 9 mg, 0.22 mmol) was added to a solution of compound 180 (94 mg, 0.2 mmol) in DMF (3.2 mL). The r.m. was stirred at r.t. for 15 min, then methyl iodide (14  $\mu$ L, 0.22 mmol) was added. The r.m. was stirred for 90 min, then 3 drops of water were added. DMF was removed in vacuo. The residue was dissolved in DCM and the organic layer was washed with water, dried over MgSO<sub>4</sub> and the solvent was removed in vacuo. The residue was crystallized from Et<sub>2</sub>O. The crystals were filtered off and dried, yielding compound 181 as a white solid (60 mg, 62%).

### Example B25

### a) Preparation of Compounds 183 and 184

Starting from intermediate 174, a mixture of compound <sup>40</sup> 183 and compound 184 was obtained following an analogous synthetic route as described for compound 125 (Example B17). The mixture was subsequently separated into enantiomers by preparative SFC (Chiralcel Diacel OD 20×250 mm, mobile phase (CO<sub>2</sub>, EtOH with 0.2% iPrNH<sub>2</sub>)). Yield: 93 mg of compound 183 (9%; R or S; enantiomer A (SFC-MS)) and 90 mg of compound 184 (8%; S or R; enantiomer B (SFC-MS)).

### Example B26

### a) Preparation of Compounds 185 and 186

A mixture of compounds 183 and 184 (see B25.a) was processed following a synthetic route similar to the one described for the synthesis of compound 181 (Example B24) to afford a mixture of compound 185 and compound 186. The mixture was subsequently separated into enantiomers by preparative SFC (Chiralcel Diacel OD 20×250 mm, mobile phase (CO<sub>2</sub>, EtOH with 0.2% iPrNH<sub>2</sub>)). Yield: 139 mg of compound 185 (38%; R or S; enantiomer A (SFC-MS)) and 135 mg of compound 186 (37%; S or R; enantiomer B (SFC-MS)).

### Example B27

### a) Preparation of Compound 198

Starting from intermediate 167, compound 198 was obtained following an analogous synthetic route as described for compound 6 (Example B6).

### Example B28

### a) Preparation of Compound 199

$$\begin{array}{c|c} & & & \\ & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$$

Compound 125 (125 mg, 0.27 mmol) was dissolved in DCM (6 mL). The solution was stirred for 5 min, then phenylboronic acid (66 mg, 0.54 mmol), copper acetate (97 mg, 0.54 mmol), molecular sieves, pyridine (45  $\mu$ L, 0.56 mmol) and Et<sub>3</sub>N (75  $\mu$ L, 0.54 mmol) were added and the r.m. was

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vacuo before being purified by preparative HPLC (RP Sun-Fire Prep C18 OBD-10  $\mu$ m, 30×150 mm; Mobile phase: 0.25% NH<sub>4</sub>HCO<sub>3</sub> solution in water+5% acetonitrile/acetonitrile), to yield compound 204 (6 mg, 75%).

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# stirred overnight at r.t. After night, 2 eq. of phenylboronic acid and 2 eq. of copper acetate were added. The r.m. was stirred again overnight at rt. After this time the reaction mixture was filtered through celite. Water and brine were added to the filtrate, and all solvents were evaporated under reduced pressure. The residue was dissolved in EtOAc, filtered and the filtrate was washed with brine. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered off and the solvent was evaporated in vacuo to afford an intermediate (95 mg), which was purified by flash column chromatography (silica; DCM/MeOH 100/0 to 50/50). The fractions containing the product were collected and evaporated, to afford compound 199 (16 mg, 11%).

### Example B29

### a) Preparation of Compound 203

Starting from intermediate 183, compound 203 was obtained following an analogous synthetic route as described for compound 6 (Example B6).

### Example B30

### a) Preparation of Compound 204

$$\bigcap_{N}\bigcap_{O}\bigcap_{S}\bigcap_{Or}\bigcap_{S}\bigcap_{CF_{3}}$$

In a first vial equipped with a magnetic stir bar and a screw cap septum, a solution of Pd<sub>2</sub>(dba)<sub>3</sub> (3 mg, 0.004 mmol) and 2-di-tert-butylphosphino-3,4,5,6-tetramethyl-2',4',6'-triisopropy 1-1,1'-biphenyl (3 mg, 0.007 mmol) in dioxane (0.1 55 mL) and toluene (0.45 mL) was flushed with N2 and then stirred at 120° C. for 3 min. A second vial, equipped with a magnetic stir bar and a screw cap septum, was charged with 4-methylimidazole (3 mg, 0.04 mmol) and K<sub>3</sub>PO<sub>4</sub> (8 mg, 0.04 mmol), then with intermediate 191 (8 mg, 0.02 mmol) and 60 also flushed with N2. The premixed catalyst solution was added by syringe to the second vial. The r.m. was heated at 120° C. for 5 h. The reaction was cooled to r.t., diluted with EtOAc, washed with brine, and the organic layer was filtered over an Isolute® HM-N column filter (modified form of 65 diatomaceous earth) to remove the water. The solvent was removed under a stream of N<sub>2</sub> and the crude concentrated in

# Example B31

## a) Preparation of Compounds 212 and 213

$$\begin{array}{c} & & & \\$$

Intermediate 203 (1.2 g) was dissolved in DCM (24 mL).

TFA (4.84 mL) was added and the r.m. was stirred until completion of the reaction. The solvent was then evaporated, EtOAc and sat. aq. NaHCO<sub>3</sub> sat. were added. The organic layer was separated, washed with brine and dried over MgSO<sub>4</sub>. After filtration, the solvent was evaporated in vacuo to give a crude material (808 mg). 90 mg of the crude were purified by preparative SFC (Chiralpak Diacel AD 30×250 mm, mobile phase (CO<sub>2</sub>, MeOH with 0.2% iPrNH<sub>2</sub>)). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again. Yield: 40 mg of compound 212 (1R or 1S, 3R; diastereoisomer A (SFC-MS), white powder) and 7 mg of compound 213 (1S or 1R, 3R; diastereoisomer B (SFC-MS), white powder).

### Example B32

### a) Preparation of Compounds 214 and 215

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$$\bigcap_{N=0}^{N}\bigcap_{N=0}^{N}\bigcap_{SorR}^{N}\bigcap_{SorR}$$

Starting from a mixture of compound 212 and compound 213, compound 214 and compound 215 were obtained as a mixture following an analogous synthetic route as described for compound 181 (Example B24). The mixture was separated into the single diastereoisomers by preparative SFC (Chiralpak Diacel AD 30×250 mm, mobile phase (CO $_2$ , EtOH with 0.2% iPrNH $_2$ )). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again, Yield: 86 mg of compound 214 (1R or 1S, 3R; diastereoisomer A (SFC-MS)) and 17 mg of compound 215 (1S or 1R, 3R; diastereoisomer B (SFC-MS)).

### Example B33

### a) Preparation of Compounds 216 and 217

$$\bigcap_{N=1}^{N}\bigcap_{SorR}^{N}\bigcap_{SorR}^{CF_{3}}$$

A mixture of compound 212 and compound 213 (358 mg, 0.76 mmol) was dissolved in DMF (8.7 mL) under  $N_2$  flow, NaH (60% in mineral oil, 38 mg, 0.95 mmol) was added and the r.m. was stirred for 15 min. Iodoethane (69  $\mu$ L, 0.86 60 mmol) was added and the r.m. was stirred at r.t. for 90 min, then aq. sat. NH<sub>4</sub>Cl was added. The mixture was extracted with EtOAc, the combined organic layers were dried over MgSO<sub>4</sub>, filtered and the solvents were evaporated in vacuo to give a crude, which was triturated with DIPE, dried in the 65 vacuum oven overnight and then purified by preparative SFC (Chiralpak Diacel AD 30×250 mm, mobile phase (CO<sub>2</sub>,

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EtOH with 0.2% iPrNH $_2$ )). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again. Yield: 134 mg of compound 216 (1R or 1S, 3R; diastereoisomer A (SFC-MS)) and 26 mg of compound 217 (1S or 1R, 3R; diastereoisomer B (SFC-MS)).

### Example B34

### a) Preparation of Compound 222

Starting from compound 201, compound 222 was obtained following an analogous synthetic route as described for compound 181 (Example B24). The crude material was purified by preparative HPLC (RP SunFire Prep C18 OBD-10 µm, 30×150 mm; Mobile phase: 0.25% NH<sub>4</sub>HCO<sub>3</sub> solution in water, acetonitrile), to yield compound 222 (27 mg, 33%).

### Example B35

### a) Preparation of Compound 225

Starting from compound 203, compound 225 was obtained following an analogous synthetic route as described for compound 181 (Example B24). The crude material was purified by preparative HPLC (RP SunFire Prep C18 OBD-10 µm, 30×150 mm; Mobile phase: 0.25% NH<sub>4</sub>HCO<sub>3</sub> solution in water, acetonitrile), to yield compound 225 (15 mg, 29%).

### Example B36

### a) Preparation of Compound 208 and 209

$$Cl$$
 $Cl$ 
 $Cl$ 
 $Ror S$ 

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a) Preparation of Compound 162

$$Cl$$
 $N$ 
 $N$ 
 $SorR$ 
 $N$ 

Starting from intermediate 173 and intermediate 193, a mixture of compound 208 and compound 209 was obtained following an analogous procedure as described for compound 6 (Example B6), in the presence of 1 eq. of lithium bromide. The mixture was subsequently separated into enantiomers by preparative SFC (Chiralcel Diacel OD 20×250 mm, mobile phase (CO<sub>2</sub>, MeOH with 0.2% iPrNH<sub>2</sub>). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again. Yield: 92 mg of compound 208 (13%; R or S; enantiomer B (SFC-MS)) and 86 mg of compound 209 (13%; S or R; enantiomer A (SFC-MS)).

### Example B37

### a) Preparation of Compounds 236 and 237

Starting from intermediate 225, compound 162 was

obtained following an analogous procedure as described for

### Example B39

### a) Preparation of Compound 174 and Compound 242

$$\begin{array}{c} 236 \\ 35 \\ \hline \\ N \\ \hline \\ CF_3 \\ \end{array} \qquad 40$$

Starting from intermediate 213, a mixture of compound 236 and compound 237 (derived from an impurity present in intermediate 213) was obtained following an analogous procedure as described for compound 125 (Example B17). The mixture was subsequently separated into enantiomers by preparative SFC (Chiralcel Diacel OJ 20×250 mm, mobile phase (CO<sub>2</sub>, MeOH with 0.4% iPrNH<sub>2</sub>). The desired fractions were collected and evaporated. Yield: 121 mg of compound 236 (17%) and 190 mg of compound 237 (26%).

A pressure tube was charged with compound 239 (175 mg, 0.35 mmol), cyclopropylboronic acid (40 mg, 0.46 mmol), tricyclohexylphosphine (10 mg, 0.04 mmol) and toluene (1.7 mL). Then, Pd(OAc)<sub>2</sub> (8 mg, 0.02 mmol) and ground K<sub>3</sub>PO<sub>4</sub> (261 mg, 1.23 mmol) were added. Finally, water (0.1 mL) was added and the tube was capped and placed in a preheated oil bath of 120° C. and stirring was continued for 2 h. More cyclopropylboronic acid, Pd(OAc)<sub>2</sub> and tricyclohexylphosphine were added, and the mixture was stirred overnight at 90° C. The r.m. was then cooled and diluted with water (5 mL). The layers were separated and the aqueous layer was

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extracted with toluene (5 mL). The combined organic layers were treated with brine (10 mL), dried over MgSO<sub>4</sub>, filtered and concentrated in vacuo. The resulting residue was resubmitted to the same reaction conditions, using the same 5 amounts of equivalents. The crude obtained was then purified by flash column chromatography (silica; DCM/MeOH 100/0 to 94/6). The product fractions were collected and purified again by Prep HPLC (RP SunFire Prep C18 OBD-10  $\mu m$  30×150 mm, mobile phase: 0.25% NH<sub>4</sub>HCO<sub>3</sub> solution in water, MeOH), to yield compound 174 (13 mg, 8%) and compound 242 (2 mg, 1%).

### Example B40

### a) Preparation of Compound 244

Starting from compound 234, compound 244 was obtained following an analogous procedure as described for compound 181 (Example B24). The r.m. was purified by Prep HPLC (RP SunFire Prep C18 OBD-10  $\mu$ m, 30×150 mm, mobile phase: 0.25% NH<sub>4</sub>HCO<sub>3</sub> solution in water, acetonitrile). After evaporation of the solvent, the product was dissolved in 40 MeOH and the solvent was removed under reduced pressure (×2). The residue was dissolved in DCM/heptane and the solvents were evaporated under nitrogen flow. The residue was dried in vacuo at 50° C. 48 h. Yield: 145 mg of compound 244 (46%).

### Example B41

### a) Preparation of Compound 246

Starting from compound 277, compound 246 was obtained 65 following an analogous procedure as described for compound 181 (Example B24). The r.m. was purified by Prep HPLC (RP

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SunFire Prep C18 OBD-10  $\mu$ m, 30×150 mm, mobile phase: 0.25% NH<sub>4</sub>HCO<sub>3</sub> solution in water, acetonitrile). Yield: 55 mg of compound 246 (44%).

### Example B42

### a) Preparation of Compound 247

Starting from compound 277, compound 247 was obtained following an analogous procedure as described for compound 181 (Example B24). The r.m. was purified by Prep HPLC (RP SunFire Prep C18 OBD-10  $\mu$ m, 30×150 mm, mobile phase: 0.25% NH<sub>4</sub>HCO<sub>3</sub> solution in water, acetonitrile). Yield: 48 mg of compound 247 (36%).

### Example B43

### a) Preparation of Compound 248

To a suspension of compound 188 (100 mg, 0.22 mmol), cyclopropylboronic acid (38 mg, 0.44 mmol) and sodium carbonate (47 mg, 0.44 mmol) in 1,2-dichloroethane (350 μL) was added a suspension of copper acetate (41 mg, 0.23 mmol) and 4,4'-di-tert-butyl-2,2'-bipyridyl (62 mg, 0.23 mmol) in hot 1,2-dichloroethane (600 μL, 50° C.). The mixture was warmed to 70° C. and stirred for 6 h and 30 min under air. The r.m. was then cooled to r.t. and sat. NH<sub>4</sub>Cl sol. was added, followed by water. The org. layer was separated and the aqlayer was extracted with DCM (×3). The combined org. layers were washed with brine, dried over MgSO<sub>4</sub>, and concentrated in vacuo to afford a crude, which was purified by flash column chromatography (silica; DCM/MeOH 100/0 to 95/5). The fractions containing the product were collected and concentrated in vacuo. Yield: 58 mg of compound 248 (53%).

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### Example B44

### a) Preparation of Compound 251

Starting from compound 279, compound 251 was obtained following an analogous procedure as described for compound 181 (Example B24). The r.m. was purified by Prep HPLC (RP Vydac Denali C18-10  $\mu$ m, 200 g, 5 cm, mobile phase: 0.25% NH<sub>4</sub>HCO<sub>3</sub> solution in water, acetonitrile). Yield: 100 mg of compound 251 (44%).

### Example B45

## a) Preparation of Compound 256

Starting from compound 282, compound 256 was obtained following an analogous procedure as described for compound 181 (Example B24). The r.m. was purified by Prep SFC  $^{40}$  (Chiralcel Diacel OJ 20×250 mm, mobile phase: CO2, MeOH with 0.2% iPrNH2). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again. Yield: 14 mg of compound 256 (24%).

### Example B46

### a) Preparation of Compounds 284 (Racemic Mixture), 267 (R or S Enantiomer) and 268 (S or R Enantiomer)

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Starting from intermediate 243, compound 284 was obtained following an analogous procedure as described for compound 125 (Example B17). The r.m. was purified by Prep SFC (Chiralcel Diacel AD 30×250 mm, mobile phase:  $\rm CO_2$ , EtOH with 0.2% iPrNH<sub>2</sub>). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again. Yield: 40 mg of compound 267 and 44 mg of compound 268.

### Example B47

a) Preparation of Compounds 285 (Racemic Mixture), 265 (R or S Enantiomer) and 266 (S or R Enantiomer)

Starting from compound 284, compound 285 was obtained following an analogous procedure as described for compound 181 (Example B24). The r.m. was purified by Prep SFC (Chiralcel Diacel AD 30×250 mm, mobile phase: CO<sub>2</sub>, EtOH with 0.2% iPrNH<sub>2</sub>). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again. Yield: 135 mg of compound 265 (19%) and 130 mg of compound 266 (18%).

### Example B48

a) Preparation of Compounds 287 (Racemic Mixture), 271 (R or S Enantiomer) and 272 (S or R Enantiomer)

50

177

Starting from compound 286, compound 287 was obtained following an analogous procedure as described for compound 181 (Example B24). The r.m. was purified by Prep SFC (Chiralcel Diacel OD 20×250 mm, mobile phase:  $\rm CO_2$ , EtOH with 0.4% iPrNH<sub>2</sub>). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again. Yield: 63 mg of compound 271 (25%) and 69 mg of compound 272 (27%).

### Example B49

### a) Preparation of Compound 262

At r.t. and under N<sub>2</sub>, NaH (60% in mineral oil, 21 mg, 0.523 mmol) was added to a solution of compound 278 (210 mg, 0.523 mmol) in DMF (0.49 mL). The r.m. was stirred at r.t. for 15 min and 2,2,2-trifluoroethyl perfluorobutylsulfonate (220 mg, 0.575 mmol) was added. The r.m. was stirred for 1.5 h, then DMF was removed under reduced pressure. The residue was dissolved in DCM and a few drops of MeOH and the org. solution was washed with water, dried through Extrelut® and the solvent was evaporated to afford a crude, which was purified by flash column chromatography (silica; DCM/MeOH 100/0 to 96/4). The fractions containing the product were repurified by Prep HPLC (RP Vydac Denali C18-10 µm, 200 g, 5 cm, mobile phase: 0.25% NH<sub>4</sub>HCO<sub>3</sub> solution in water, acetonitrile) Yield: 48 mg of compound 262 (19%).

### Example B50

a) Preparation of Compounds 289 (Racemic Mixture), 273 (R or S Enantiomer) and 274 (S or R Enantiomer)

178

Starting from compound 288, compound 289 was obtained following an analogous procedure as described for compound 181 (Example B24). The r.m. was purified by Prep SFC (Chiralcel Diacel AD 30×250 mm, mobile phase: CO<sub>2</sub>, EtOH with 0.2% iPrNH<sub>2</sub>). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again. Yield: 23 mg of compound 273 (27%) and 23 mg of compound 274 (27%).

### Example B51

a) Preparation of Compounds 294 (Racemic Mixture), 295 (R or S Enantiomer) and 296 (S or R Enantiomer)

Starting from compound 293, compound 294 was obtained following an analogous procedure as described for compound 181 (Example B24). The r.m. was purified by Prep SFC (Chiralcel Diacel OD 20×250 mm, mobile phase: CO<sub>2</sub>, EtOH with 0.2% iPrNH<sub>2</sub>). The desired fractions were collected, evaporated, dissolved in MeOH and evaporated again. Yield: 96 mg of compound 295 (27%) and 100 mg of compound 296 (28%).

By using analogous reaction protocols as described in the foregoing examples, the compounds listed in Tables 1a, 1b, 1c, 1d, 1e, 1f and 1g have been prepared. 'Co. No.' means compound number. 'cb' means covalent bond. 'Pr.' refers to the Example number in analogy to which protocol the compound was synthesized.

In case no specific stereochemistry is indicated for a stereocenter of a compound, this means that the compound was obtained as a mixture of the R and the S enantiomers. In case no salt form is indicated, the compound was obtained as a free base.

B3\* refers to a reaction protocol analogue to B3, but no ligand was used and the reaction was performed under microwave conditions.

TABLE 1a

			$\mathbb{R}^2$ $\mathbb{N}$	\ <sub>N</sub>				
			O R	$\sum_{i=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{j=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{j=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{j$	−R <sup>4</sup>			
Co. No.	Pr.	$\mathbb{R}^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
15	В3	7		Н	Н	CH <sub>3</sub>	Н	
16	В3	Н	<u></u>	Н	Н	Н	(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	
17	В3	Н	<b>√</b> `	Н	Н	Н	(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	
18	В3	Н		Н	Н	$\mathrm{CH_3}$	Н	
19	В3		$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
20	В3			CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
21	В3		$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
22	ВЗ		$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
23	В3		$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
24	В3	CI	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R-enantiomer •HCl
25	В3	Cl	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	

TABLE 1a-continued

			II O	$\mathbb{R}^5$	- K '			
Co. No.	Pr.	R <sup>1</sup>	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
26	В3	CI	(CH <sub>2</sub> ) <sub>2</sub>	Н	Н	CH <sub>3</sub>	Н	
27	В3	CI	$\mathrm{CH_2O}(\mathrm{CH_2})_2$	Н	Н	CH <sub>3</sub>	Н	
28	В3	Cl		$\mathrm{CH}_3$	Н	CH <sub>3</sub>	Н	R enantiomer
29a	В3	CI		Н	Н	CH <sub>3</sub>	Н	•HCl
29	В3	CI	```	Н	Н	CH <sub>3</sub>	Н	
30	B1	CI	$\triangle$	Н	Н	CH <sub>3</sub>	Н	TRANS
2	B2	CI	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
31	B1	CI	$\mathrm{CH}_2$		Н	CH <sub>3</sub>	Н	R enantiomer
32	B1	CI	$\mathrm{CH}_2$		Н	CH <sub>3</sub>	Н	
33	B1	CI	CH <sub>2</sub>		Н	CH <sub>3</sub>	Н	R enantiomer

			$\mathbb{R}^{1}$ $\mathbb{R}^{2}$ $\mathbb{N}$ $\mathbb{N}$ $\mathbb{N}$	N N	-R <sup>4</sup>			
Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
34	В2	CI	$\mathrm{CH}_2$	CH(CH <sub>3</sub> ) <sub>2</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
35	В3	Cl	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
36	В3	CI	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
37	В3*	CI	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	S enantiomer
9	В9	CI	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	racemic mixture
38	В3	CI	$\mathrm{CH}_2$	н	Н	CH <sub>3</sub>	Н	
9a	В9	CI	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	R or S enantiomer OR: -121.40° (589 nm; 20° C.; 0.500 w/v %; DMF)
9Ь	В9	CI	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	S or R enantiomer
10	B10	CI	$\mathrm{CH}(\mathrm{CH}_3)$	Н	Н	CH <sub>3</sub>	Н	racemic mixture

			$R^1$ $R^2$ $N$	N N	—R⁴			
Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$R^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
10a	B10	CI	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	R or S enantiomer OR: +193.67° (589 nm; 20° C.; 0.300 w/v %; DMF)
10b	B10	CI	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	S or R enantiomer OR: -168.07° (589 nm; 20° C.; 0.285 w/v %; DMF)
39	В3	CI	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
40	ВЗ	CI	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
41	ВЗ	F	cb	Н	Н	CH <sub>3</sub>	Н	
42	В3	F <sub>3</sub> C	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
11	B11	F <sub>3</sub> C	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	racemic mixture
11b	B11	F <sub>3</sub> C	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	S or R enantiomer; enantiomer B (SFC-MS)
11a	B11	F <sub>3</sub> C	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	R or S enantiomer; enantiomer A (SFC-MS)

Co. No.	Pr.	$\mathbb{R}^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
53	В3	F <sub>3</sub> C CF <sub>3</sub>	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
54	В7	F <sub>3</sub> C CF <sub>3</sub>	OCH	Н	Н	CH <sub>3</sub>	Н	
55	В3	F <sub>3</sub> C	cb	Н	Н	CH <sub>3</sub>	Н	
56	В3	F <sub>3</sub> C	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
57	В1	F <sub>3</sub> C	cb	Н	Н	CH <sub>3</sub>	Н	
58	В3	F <sub>3</sub> C	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
59	В3	$CF_3$	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R enantiomer
60	В3	F <sub>3</sub> C	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	

Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
61	B1	$CF_3$	CH <sub>2</sub>	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
62	B1	Cl CF3	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	СН3	Н	R enantiomer
63	В3	Cl CF <sub>3</sub>	CH₂	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
64	В3	F <sub>3</sub> C	$\mathrm{CH}_2$	н	Н	CH <sub>3</sub>	Н	
65	В3	F <sub>3</sub> C	Н	Н	CH <sub>3</sub>	Н		
66	В3	H <sub>3</sub> CO	CH <sub>2</sub>	Н	Н	CH <sub>3</sub>	Н	
67	В3	$CF_3$	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
68	В3	F <sub>3</sub> C	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer

В7

TABLE 1a-continued

 $\mathrm{CH}_2$ 

 $\mathrm{CH}_3$ 

Η

 $\mathrm{CH}_3$ 

Η

R enantiomer •HCl

		$\mathbb{R}^2$ $\mathbb{N}$ $\mathbb{N}$ $\mathbb{R}^4$							
Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	R <sup>5</sup>	Salt forms/Stereo- chemistry/Optical Rotation (OR)	
6	В6	F <sub>3</sub> C O	CH₂	СН3	Н	CH <sub>3</sub>	Н	R-enantiomer •HCl	
78	В7	O CF <sub>3</sub>	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R-enantiomer •HCl	
79	В3	F <sub>3</sub> C	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R-enantiomer	
80	B6	CI N	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R-enantiomer •HCl	
81	В6	$CI$ $N$ $CF_3$	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R-enantiomer •HCl	

$R^{1}$ $R^{2}$ $N$ $N$ $R^{4}$								
Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	R <sup>5</sup>	Salt forms/Stereo- chemistry/Optical Rotation (OR)
82	В6	CI	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R-enantiomer •HCl
83	В6	N N	$\mathrm{CH}_2$	СН3	Н	$\mathrm{CH}_3$	Н	R-enantiomer
84	В3		$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
4	B4	N. C.	$\mathrm{CH}_2$	СН3	Н	СН3	Н	R enantiomer •HCl
4a	В4		$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R enantiomer
85	В6	C <sub>I</sub>	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer

			K						
Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)	
86	В6	$\bigcap_{N}$	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer •HCl	
87	В3		$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	CH <sub>3</sub>		
88	В5		CH₂	Н	Cl	CH <sub>3</sub>	Н		
5	В5		$\mathrm{CH}_2$	н	Н	СН3	Н		
90	В3	F F F	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н		
91	В3	N O	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н		

$$R^1$$
 $R^2$ 
 $N$ 
 $N$ 
 $R^4$ 

Co. No.	Pr.	$\mathbb{R}^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
92	В3	$F_3C$ $CF_3$	(CH <sub>2</sub> ) <sub>2</sub>	Н	Н	СН3	Н	
93	В3	0	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
94	В3		$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
95	В3	$\bigcap_{N \in \mathcal{N}} \mathbb{R}$	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
96	В3		$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
97	В3	$\bigcap_{N}$	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
98	B1	N	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
99	В3	$F_3\dot{C}$	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	

		$R^{1}$ $N$ $N$ $N$ $N$ $R^{4}$						
Co. No.	Pr.	$\mathbb{R}^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	R <sup>5</sup>	Salt forms/Stereo- chemistry/Optical Rotation (OR)
100	В3		$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
101	В3		$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
102	B1		cb	Н	Н	CH <sub>3</sub>	Н	
103	В3	0	(CH <sub>2</sub> ) <sub>2</sub>	Н	Н	CH <sub>3</sub>	Н	
104	B1		(CH <sub>2</sub> ) <sub>2</sub>	Н	Н	СН3	Н	
105	B1	F <sub>3</sub> C	cb	Н	Н	CH <sub>3</sub>	Н	
106	B1	CF <sub>3</sub>	cb	Н	Н	CH <sub>3</sub>	Н	

			$\mathbb{R}^2$	$\mathbb{R}^5$	-R <sup>4</sup>			
Co. No.	Pr.	$\mathbb{R}^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
107	B1	F <sub>3</sub> C	cb	Н	Н	CH <sub>3</sub>	Н	
108	В3		$\mathrm{CH}_2$	Н	Н	Н	CH <sub>3</sub>	
109	В3		$ m CH_2$	Н	Н	Н	Н	
12	B12	F <sub>3</sub> C CF <sub>3</sub>	$\mathrm{CH}_2$	$\mathrm{CH_2OCH_3}$	Н	CH <sub>2</sub> CH <sub>3</sub>	Н	S enantiomer
110	В6	F <sub>3</sub> C	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R-enantiomer •HCl
111	B15	$F_3C$ $CF_3$	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R); diastereomer A (SFC-MS) •HCl
112	B16	F <sub>3</sub> C CF <sub>3</sub>	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R); diastereomer B (SFC-MS) •HC1

$R^{1}$ $R^{2}$ $N$ $N$ $R^{4}$								
Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	R <sup>5</sup>	Salt forms/Stereo- chemistry/Optical Rotation (OR)
113	В3	F O F	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
114	B6		$\mathrm{CH}_2$	СН3	Н	СН3	Н	R-enantiomer •HCl
115	В6	S Br	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	СН3	Н	R enantiomer
116	B6	F <sub>3</sub> C	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R-enantiomer •HCl
117	В3	F <sub>3</sub> C		Н	Н	CH <sub>3</sub>	Н	
118	В3	F <sub>3</sub> C CF <sub>3</sub>	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	Cl	Н	R enantiomer
175	B20	F OCF3	CH(CH <sub>3</sub> ) (RS)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1RS, 3R) •HCl

$$\begin{array}{c|c}
R^1 & & \\
R^2 & & \\
\end{array}$$

$$\begin{array}{c|c}
R^3 & & \\
N & & \\
\end{array}$$

$$\begin{array}{c|c}
R^4 & & \\
\end{array}$$

Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
176	В6		(CH <sub>2</sub> ) <sub>2</sub>	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
177	B18	Br	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
178	B18	CI	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
179	B19	Cl	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
124	В6		CH <sub>2</sub>	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
125	B17	Br	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
126	B20	$F$ $OCF_3$	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R); •HCl
127	B20	$F$ $OCF_3$	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R); •HCl
128	В6	$O$ $CF_3$	CH₂	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer •HCl

Salt forms/Stereo-

TABLE 1a-continued

$$R^1$$
 $R^2$ 
 $R^2$ 
 $R^3$ 
 $R^4$ 
 $R^2$ 
 $R^4$ 
 $R^2$ 
 $R^2$ 
 $R^4$ 

$$R^1$$
 $R^2$ 
 $N$ 
 $N$ 
 $R^3$ 
 $R^4$ 

Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
172	В3	$F_3C$ $CF_3$	CH <sub>2</sub> CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	
180	B17	$F_3C$	CH <sub>2</sub>	CH <sub>3</sub>	Н	$\mathrm{CH}_3$	Н	R enantiomer
181	B24	$F_3C$	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
182	B24	$F_{3}C$	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	СН3	Н	R enantiomer
183	B25	F NH	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	R or S enantiomer; enantiomer A (SFC-MS)
184	B25	FNH	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	S or R enantiomer; enantiomer B (SFC-MS)
185	B26	F	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	R or S enantiomer; enantiomer A (SFC-MS)
186	B26	F	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	S or R enantiomer; enantiomer B (SFC-MS)

$$R^1$$
 $R^2$ 
 $R^3$ 
 $R^4$ 
 $R^5$ 

Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
187	B17	F	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R)
188	B17	FNH	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
189	B24	F	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R)
190	B24	F	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
191	B24	F N	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R)
192	B24	F N	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	СН3	Н	(1S or 1R, 3R)
193	B24	F_N_N_	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R)

Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
194	B24	F N	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
195	B17	FNH	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
196	B24	F	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R enantiomer
197	B17	NH	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	СН3	Н	R enantiomer
198	B27	$F_3C$	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
199	B28	Br N	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
200	B24	Br	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer

Co. No.	Pr.	$\mathbb{R}^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
211	B20	Cl	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	СН3	Н	(1S or 1R, 3R)
212	B31	$F_3C$	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R)
213	B31	$F_3C$	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
214	B32	$F_3C$	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R) OR: -45.62° (589 nm; 20° C.; 0.445 w/v %; DMF)
215	B32	$F_3C$	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
216	B33	$F_{3}C$	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R) OR: -31.67° (589 nm; 20° C.; 0.42 w/v %; DMF)
217	B33	$F_{3}C$	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
218	B17	CI NH	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	СН3	Н	(1R or 1S, 3R)

$$R^{1} \xrightarrow{L} N \xrightarrow{O} N \xrightarrow{R^{3}} N \xrightarrow{N} R^{4}$$

Co.	Pr.	$R^1$	L	$\mathbb{R}^2$	$R^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
219	B17	F NH	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
220	B24	F N	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	СН3	Н	(1R or 1S, 3R)
221	B24	CI	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	СН3	Н	(1S or 1R, 3R)
222	B34		$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R enantiomer
223	B24	CI	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R)
224	B24	CI	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)

$$R^{1}$$
 $R^{2}$ 
 $N$ 
 $N$ 
 $N$ 
 $N$ 
 $R^{2}$ 

Co. No.	Pr.	$\mathbb{R}^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
225	B35	$F_3C$	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R enantiomer
226	B24	$F_3C$	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R enantiomer
227	B24	$F_{3}C$	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R enantiomer
228	B24	F <sub>3</sub> C N	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
229	В3	F <sub>3</sub> C		CH <sub>3</sub>	Н	СН3	Н	R enantiomer
230	В3	CI		CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
231	В3	Cl		CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer •HCl

		$R^{1}$ $R^{2}$ $N$ $N$ $R^{4}$						
Co. No.	Pr.	${ m R}^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
232	В3	F CF <sub>3</sub>		СН3	Н	СН3	Н	R enantiomer
233	В3	CI		CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
234	B17	F <sub>3</sub> C NH	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R)
235	B24	F <sub>3</sub> C N	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
236	B37	$F_{3}C$ $NH$	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
237	B37	F NH	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
238	B24	$F_{3}C$	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer

B43

$$R^1$$
 $R^2$ 
 $N$ 
 $N$ 
 $N$ 
 $R^4$ 

Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$R^3$	$R^4$	$R^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
249	B24	CI	CH(CH <sub>3</sub> ) (R or S)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1R or 1S, 3R)
250	B24		$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
251	B44	CI	$\mathrm{CH}_2$	СН <sub>3</sub>	Н	СН3	Н	R enantiomer
252	B17	CF <sub>3</sub> O NH	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
253	B24	CF <sub>3</sub> O	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer; •HCl
254	B24	CI	${\rm CH}_2$	Н	Н	CH <sub>3</sub>	Н	
255	B24	CI	CH <sub>2</sub>	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer

$$R^1$$
 $R^2$ 
 $N$ 
 $N$ 
 $R^4$ 

Co. No.	Pr.	$\mathbb{R}^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$\mathbb{R}^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
256	B45	CI	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
257	B17	CF <sub>3</sub>	CH <sub>2</sub>	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
258	B24	F <sub>3</sub> C N	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
259	B24	CI	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
260	B17	CI	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
261	B24	F CF3	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
262	B49	F <sub>3</sub> C N	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R enantiomer

 $R^{1}$ 

$$R^1$$
 $R^2$ 
 $N$ 
 $N$ 
 $N$ 
 $R^4$ 

Co. No.	Pr.	$R^1$	L	$\mathbb{R}^2$	$\mathbb{R}^3$	$R^4$	$\mathbb{R}^5$	Salt forms/Stereo- chemistry/Optical Rotation (OR)
271	B48	F <sub>3</sub> C N	CH(CH <sub>3</sub> ) (R or S)	Н	Н	CH <sub>3</sub>	Н	1R or 1S enantiomer; enantiomer A (SFC-MS)
272	B48	$F_3C$	CH(CH <sub>3</sub> ) (S or R)	Н	Н	CH <sub>3</sub>	Н	1S or 1R enantiomer; enantiomer B (SFC-MS)
273	B50	F <sub>3</sub> C N	CH(CH <sub>3</sub> ) (R or S)	Н	Н	CH <sub>3</sub>	Н	1R or 1S enantiomer; enantiomer A (SFC-MS)
274	B50	F <sub>3</sub> C N	CH(CH <sub>3</sub> ) (S or R)	Н	Н	СН3	Н	1S or 1R enantiomer; enantiomer B (SFC-MS)
277	B17	NH	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
278	B17	NH	$\mathrm{CH}_2$	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
279	B17	CINH	$\mathrm{CH}_2$	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
280	B17	CI	$\mathrm{CH}_2$	$\mathrm{CH_3}$	Н	CH <sub>3</sub>	Н	R enantiomer

$$R^{1} \xrightarrow{L} N \xrightarrow{O} N \xrightarrow{R^{3}} N \xrightarrow{N} R^{4}$$

Co. No.	Pr.	R <sup>1</sup>	L	$\mathbb{R}^2$	$\mathbb{R}^3$	R <sup>4</sup>	R <sup>5</sup>	Salt forms/Stereo- chemistry/Optical Rotation (OR)
281	B17	CINH	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
282	B17	CI	CH(CH <sub>3</sub> ) (S or R)	CH <sub>3</sub> (R)	Н	CH <sub>3</sub>	Н	(1S or 1R, 3R)
283	B17	F <sub>3</sub> C NH	$\mathrm{CH}_2$	Н	Н	CH <sub>3</sub>	Н	
284	B46	$F_{3}C$ NH	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	
285	B47	$F_{3}C$	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	
286	B17	$_{\mathrm{F_{3}C}}$	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	
287	B48	$F_3C$	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	
288	B17	F <sub>3</sub> C NH	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	

			 	$\mathbb{R}^5$ N				Salt forms/Stereo-
Co. No.	Pr.	$R^1$	L	$R^2$	$R^3$	$R^4$	$R^5$	chemistry/Optical Rotation (OR)
289	B50	F <sub>3</sub> C N	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	
290	B17	F <sub>3</sub> C NH	CH <sub>2</sub>	СН3	Н	CH <sub>3</sub>	Н	R enantiomer
291	B17	$F_3C$ $F$ $NH$	CH <sub>2</sub>	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
292	B24	CI	СН2	CH <sub>3</sub>	Н	CH <sub>3</sub>	Н	R enantiomer
293	B51	CI	CH(CH <sub>3</sub> )	Н	Н	CH <sub>3</sub>	Н	CH(CH <sub>3</sub> )
295	B50	CI	CH(CH <sub>3</sub> ) (R or S)	Н	Н	СН3	Н	1R or 1S enantiomer; enantiomer A (SFC-MS)
296	B50	CI	CH(CH <sub>3</sub> ) (S or R)	Н	Н	CH <sub>3</sub>	Н	1S or 1R enantiomer; enantiomer B (SFC-MS)

35

45

50

TABLE 1b

TABLE 1d

120 B3 
$$F_3C$$
  $CH_2$   $H$   $H$   $CH_3$   $H$ 

155 B3 
$$F_3C$$
  $CH_2$   $CH_3$   $H$   $CH_3$   $H$   $CCH_3$   $C$ 

$$R^1$$
  $L$   $R^2$   $R^6$   $R^6$   $R^5$   $R^4$ 

$$\mathbb{R}^{1} \xrightarrow{L} \mathbb{N} \xrightarrow{\mathbb{N}} \mathbb{N} \mathbb{N} \xrightarrow{\mathbb{N}} \mathbb{R}^{4}$$

15	Co. No.	Pr.	$R^1$	L	$R^4$	$\mathbb{R}^5$	Stereo- chemistry
20	14	B14	F <sub>3</sub> C CF <sub>3</sub>	CH <sub>2</sub>	CH <sub>3</sub>	Н	cis

14b B14 
$$F_3C$$
  $CH_2$   $CH_3$  H cis B (SFC-MS)  $CF_3$ 

### TABLE 1e

$$R^{1}$$
 $R^{2}$ 
 $N$ 
 $N$ 
 $N$ 
 $N$ 
 $N$ 
 $N$ 

	Co. No.	Pr.	$R^1$	L	$R^2$	$\mathbb{R}^3$	$R^6$
60	89	В3		CH <sub>2</sub>	Н	Н	CH <sub>3</sub>
65							

TABLE 1g

Co. No.	Pr.	Structure
204	B30	O R O S S S CEF3 2
205	B30	O S O O O O O O O O O O O O O O O O O O

Analytical Part

LCMS (Liquid Chromatography/Mass spectrometry) General Procedure A

The LC measurement was performed using an Acquity UPLC (Ultra Performance Liquid Chromatography) (Waters) system comprising a binary pump, a sample organizer, a column heater (set at 55° C.), a diode-array detector (DAD) and a column as specified in the respective methods below. 65 Flow from the column was split to a MS spectrometer. The MS detector was configured with an electrospray ionization

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source. Mass spectra were acquired by scanning from 100 to 1000 in 0.18 seconds (sec) using a dwell time of 0.02 sec. The capillary needle voltage was 3.5 kV and the source temperature was maintained at  $140^{\circ}$  C.  $N_{\rm 2}$  was used as the nebulizer gas. Data acquisition was performed with a Waters-Micromass MassLynx-Openlynx data system.

General Procedure B

The HPLC measurement was performed using an Alliance HT 2790 (Waters) system comprising a quaternary pump with degasser, an autosampler, a column oven (set at 40° C., unless otherwise indicated), a diode-array detector (DAD) and a column as specified in the respective methods below. Flow from the column was split to a MS spectrometer. The MS detector was configured with an electrospray ionization source. Mass spectra were acquired by scanning from 100 to 1000 in 1 second using a dwell time of 0.1 second. The capillary needle voltage was 3 kV and the source temperature was maintained at 140° C. Nitrogen was used as the nebulizer gas. Data acquisition was performed with a Waters-Micromass MassLynx-Openlynx data system.

General Procedure C

The HPLC measurement was performed using an Agilent G1956A LC/MSD quadrupole coupled to an Agilent 1100 series liquid chromatography system comprising a binary pump with degasser, an autosampler, a thermostated column department, a diode-array detector and a column as specified in the respective methods below. Flow from the column was split to a MS spectrometer, which was operated with an atmospheric pressure electrospray ionization source in positive ion mode. The capillary voltage was 3 kV, the fragmentor voltage was set to 70 V, and the quadrupole temperature was maintained at 100° C. The drying gas flow and temperature values were 12.0 l/min and 350° C. respectively. Nitrogen was used as the nebulizer gas. Data acquisition was performed with an Agilent Chemstation software.

General Procedure D

The HPLC measurement was performed using an Agilent 1100 module comprising a pump, a diode-array detector (DAD) (wavelength used 220 nm), a column heater and a column as specified in the respective methods below. Flow from the column was split to a Agilent MSD Series G1946C and G1956A. MS detector was configured with API-ES (atmospheric pressure electrospray ionization). Mass spectra were acquired by scanning from 100 to 1000. The capillary needle voltage was 2500 V for positive ionization mode and 3000 V for negative ionization mode. Fragmentation voltage was 50 V. Drying gas temperature was maintained at 350° C. at a flow of 10 l/min.

General Procedure E

The UPLC (Ultra Performance Liquid Chromatography) measurement was performed using an Acquity UPLC (Waters) system comprising a sampler organizer, a binary pump with degasser, a four column's oven, a diode-array detector (DAD) and a column as specified in the respective methods below. Column flow was used without split to the MS detector. The MS detector was configured with an ESCI dual ionization source (electrospray combined with atmospheric pressure chemical ionization). Nitrogen was used as the nebulizer gas. The source temperature was maintained at 140° C. Data acquisition was performed with MassLynx-Openlynx software.

General Procedure F

The LC measurement was performed using an Acquity UPLC (Waters) system comprising a binary pump, a sample organizer, a column heater (set at 55° C.), a diode-array detector (DAD) and a column as specified in the respective methods below. All the flow from the column went to a MS

spectrometer. The MS detector was configured with an electrospray ionization source. Mass spectra were acquired by scanning from 120 to 1000 in 0.1 seconds. The capillary needle voltage was 3.0 kV and the source temperature was maintained at 150° C. Nitrogen was used as the nebulizer gas. 5 Data acquisition was performed with a Waters-Micromass MassLynx-Openlynx data system.

#### General Procedure G

Analyses were performed using a Hitachi LaChrom Elite liquid chromatography (LC) system consisting of a pump with degasser, autosampler, thermostated column compartment and UV detector. Data acquisition was performed with Agilent EZChrom Elite software.

#### LCMS Method 1

In addition to general procedure A: Reversed phase UPLC 15 (Ultra Performance Liquid Chromatography) was carried out on a bridged ethylsiloxane/silica hybrid (BEH) C18 column (1.7  $\mu$ m, 2.1×50 mm; Waters Acquity) with a flow rate of 0.8 ml/min. Two mobile phases (mobile phase A: 25 mM ammonium acetate in H<sub>2</sub>O/acetonitrile 95/5; mobile phase B: acetonitrile) were used to run a gradient condition from 95% A and 5% B to 5% A and 95% B in 1.3 minutes and hold for 0.3 minutes. An injection volume of 0.5  $\mu$ l was used. Cone voltage was 30 V for positive ionization mode and 30 V for negative ionization mode.

#### LCMS Method 2

In addition to general procedure A: Reversed phase UPLC was carried out on a bridged ethylsiloxane/silica hybrid (BEH) C18 column (1.7  $\mu m, 2.1 \times 50$  mm; Waters Acquity) with a flow rate of 0.8 ml/min. Two mobile phases (mobile 30 phase A: 0.1% formic acid in  $H_2O/methanol$  95/5; mobile phase B: methanol) were used to run a gradient condition from 95% A and 5% B to 5% A and 95% B in 1.3 minutes and hold for 0.2 minutes. An injection volume of 0.5  $\mu l$  was used. Cone voltage was  $10\,V$  for positive ionization mode and  $20\,V$  35 for negative ionization mode.

### LCMS Method 3

In addition to general procedure B: Reversed phase HPLC was carried out on an Xterra MS C18 column (3.5  $\mu m$ , 4.6× 100 mm) with a flow rate of 1.6 ml/min. Three mobile phases 40 (mobile phase A: 95% 25 mM ammoniumacetate+5% acetonitrile; mobile phase B: acetonitrile; mobile phase C: methanol) were employed to run a gradient condition from 100% A to 1% A, 49% B and 50% C in 6.5 minutes, to 1% A and 99% B in 1 minute and hold these conditions for 1 minute and 45 reequilibrate with 100% A for 1.5 minutes. An injection volume of 10  $\mu l$  was used. Cone voltage was 10 V for positive ionization mode and 20 V for negative ionization mode. LCMS Method 4

In addition to general procedure A: Reversed phase UPLC 50 (Ultra Performance Liquid Chromatography) was carried out on a bridged ethylsiloxane/silica hybrid (BEH) C18 column (1.7  $\mu$ m, 2.1×50 mm; Waters Acquity) with a flow rate of 0.8 ml/min. Two mobile phases (25 mM ammonium acetate in H<sub>2</sub>O/acetonitrile 95/5; mobile phase B: acetonitrile) were 55 used to run a gradient condition from 95% A and 5% B to 5% A and 95% B in 1.3 minutes and hold for 0.3 minutes. An injection volume of 0.5  $\mu$ l was used. Cone voltage was 10 V for positive ionization mode and 20 V for negative ionization mode.

# LCMS Method 5

In addition to general procedure D: Reversed phase HPLC was carried out on a YMC-Pack ODS-AQ,  $50\times2.0$  mm 5  $\mu$ m column with a flow rate of 0.8 ml/min. Two mobile phases (mobile phase A: water with 0.1% TFA; mobile phase B: 65 acetonitrile with 0.05% TFA) were used. First, 100% A was hold for 1 minute. Then a gradient was applied to 40% A and

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60% B in 4 minutes and hold for 2.5 minutes. Typical injection volumes of 2  $\mu$ l were used. Oven temperature was  $50^{\circ}$  C. (MS polarity: positive).

### LCMS Method 6

In addition to general procedure A: Reversed phase UPLC was carried out on a BEH C18 column (1.7  $\mu m, 2.1 \times 50$  mm; Waters Acquity) with a flow rate of 0.8 ml/min. Two mobile phases (10 mM ammonium acetate in  $\rm H_2O/acetonitrile$  95/5; mobile phase B: acetonitrile) were used to run a gradient condition from 95% A and 5% B to 5% A and 95% B in 1.3 minutes and hold for 0.3 minutes. An injection volume of 0.5  $\mu l$  was used. Cone voltage was 10 V for positive ionization mode and 20 V for negative ionization mode.

#### LCMS Method 7

In addition to general procedure B: Column heater was set at 45° C. Reversed phase HPLC was carried out on an Atlantis C18 column (3.5  $\mu$ m, 4.6×100 mm) with a flow rate of 1.6 ml/min. Two mobile phases (mobile phase A: 70% methanol+30% H<sub>2</sub>O; mobile phase B: 0.1% formic acid in H<sub>2</sub>O/methanol 95/5) were employed to run a gradient condition from 100% B to 5% B+95% A in 9 minutes and hold these conditions for 3 minutes. An injection volume of 10  $\mu$ l was used. Cone voltage was 10 V for positive ionization mode and 20 V for negative ionization mode.

#### 25 LCMS Method 8

In addition to general procedure C: Reversed phase HPLC was carried out on a YMC-Pack ODS-AQ C18 column (4.6× 50 mm; 3  $\mu m$  particle size) with a flow rate of 2.6 ml/min. A gradient run was used from 95% (water/0.1% formic acid) and 5% acetonitrile to 95% acetonitrile and 5% (water/0.1% formic acid) in 4.80 minutes and was hold for 1 minute. Acquisition ranges were set to 190-400 nm for the UV detector and 100 to 1400 m/z for the MS detector. Injection volume was 2  $\mu l$ . Column temperature was 35° C.

#### LCMS Method 9

In addition to general procedure C: Reversed phase HPLC was carried out on a Phenomenex Kinetex XB-C18 column (4.6×50 mm; 2.6  $\mu m$  particle size) at 35° C., with a flow rate of 3.0 ml/min. A gradient elution was performed from 95% (water+0.1% formic acid)/5% Acetonitrile to 5% (water+0.1% formic acid)/95% acetonitrile in 4.20 minutes, then the resulting composition was held for an additional 0.70 min. Acquisition ranges were set to 190-400 nm for the UV detector and 100 to 1200 m/z for the MS detector. The injection volume was 2  $\mu l$ .

### LCMS Method 10

In addition to general procedure D: Reversed phase HPLC was carried out on a YMC-Pack ODS-AQ,  $50\times2.0$  mm 5  $\mu m$  column with a flow rate of 0.8 ml/min. Two mobile phases (mobile phase A: water with 0.1% TFA; mobile phase B: acetonitrile with 0.05% TFA) were used. First, 90% A and 10% B was hold for 0.8 minutes. Then a gradient was applied to 20% A and 80% B in 3.7 minutes and hold for 3 minutes. Typical injection volumes of 2  $\mu l$  were used. Oven temperature was 50° C.

(MS polarity: positive).

#### LCMS Method 11

In addition to the general procedure E: Reversed phase UPLC was carried out on a RRHD Eclipse Plus-C18 column (1.8 μm, 2.1×50 mm) from Agilent, with a flow rate of 1.0 ml/min, at 50° C. without split to the MS detector. The gradient conditions used are: 95% A (0.5 g/1 ammonium acetate solution+5% acetonitrile), 5% B (acetonitrile), to 40% A, 60% B in 1.2 minutes, to 5% A, 95% B in 1.8 minutes, kept till 2.0 minutes. Injection volume 2.0 μl. Low-resolution mass spectra (single quadrupole, SQD detector) were acquired by scanning from 100 to 1000 in 0.1 seconds using an inter-

channel delay of 0.08 second. The capillary needle voltage was 3 kV. The cone voltage was 25 V for positive ionization mode and 30 V for negative ionization mode.

#### LCMS Method 12

In addition to the general procedure E: Reversed phase 5 UPLC was carried out on a RRHD Eclipse Plus-C18 column (1.8  $\mu$ m, 2.1×50 mm) from Agilent, with a flow rate of 1.0 ml/min, at 50° C. without split to the MS detector. The gradient conditions used are: 95% A (0.5 g/1 ammonium acetate solution+5% acetonitrile), 5% B (acetonitrile), to 40% A, 10 60% B in 3.8 minutes, to 5% A, 95% B in 4.6 minutes, kept till 5.0 minutes. Injection volume 2.0  $\mu$ l. Low-resolution mass spectra (single quadrupole, SQD detector) were acquired by scanning from 100 to 1000 in 0.1 seconds using an interchannel delay of 0.08 second. The capillary needle voltage was 3 kV. The cone voltage was 25 V for positive ionization mode and 30 V for negative ionization mode.

### LCMS Method 13

In addition to the general procedure F: Reversed phase UPLC was carried out on a HSS T3 column (1.8  $\mu m, 2.1 \times 100~$  20 mm; Waters) with a flow rate of 0.8 ml/min. Two mobile phases (10 mM ammonium acetate in  $H_2O$ /acetonitrile 95/5; mobile phase B: acetonitrile) were used to run a gradient condition from 100% A and 0% B to 5% A and 95% B in 2.1 minutes, then to 0% A and 100% B in 0.9 minutes, and finally 25 to 5% A and 95% B in 0.5 minutes. Typical injection volumes and cone voltages which can be determined by the skilled person were used in this method.

#### LCMS Method 14

Identical to LCMS method 13, except that general procedure A was used.  $^{30}$ 

#### LCMS Method 15

In addition to the general procedure G: Analyses were carried out on a Waters XTerra C18 column (100×4.6 mm I.D. 3.5  $\mu m$  particles) at  $40^{\circ}$  C., with a flow rate of 1.6 mL/min. A 35 gradient elution was performed as follows: from 100% of a solution of ammonium acetate (25 mM) in Water/Acetonitrile 90:10 to a mixture of Acetonitrile/Methanol 50:50 in 7.5 min; from the resulting composition to 100% Acetonitrile in 1.0 min; 100% Acetonitrile for 1.5 min; from 100% Acetonitrile 40 to 100% to 100% of a solution of ammonium acetate (25 mM) in Water/Acetonitrile 90:10 (25 mM) in 3.0 minutes. The standard injection volume was 3  $\mu L$ . Acquisition ranges were set to 200-400 nm for the UV.

### Melting Points

For compounds 3, 5, 9a, 10a, 10b, 11a, 11b, 29, 38, 42, 53, 74, 119, 120, 121, 124, 125, 129, 130, 131, 132, 133, 155, 165, 185, 186, 189, 190, 196, 200, 208, 209, 219, 221, 223, 228, 251, 263, 264, 265, and 266 melting points (m.p.) were determined with a DSC823e or DSC1 (Mettler-Toledo). The 50 m.p. of compounds 3, 9a, 10a, 10b, 129, 130, 131, 132, 133, 155, 165, 185, 186, 196 200, 208, 209, 219, 221, 223, 228, 251, 263, 264, 265 and 266 were measured with a temperature gradient of 10° C./min. The m.p. of compounds 5, 11a, 11b, 29, 38, 42, 53, 74, 119, 120, 121, 124, 125, 189 and 190 were 55 measured with a temperature gradient of 30° C./min.

For compounds 1, 2, 6, 7, 14, 18, 20-22, 24, 25, 27, 28, 30, 31, 34-36, 40 43, 46, 55-59, 63, 65, 67-70, 72, 73, 75, 80-82, 84, 86, 88, 91-93, 95-97, 100-103, 105-107, 111-115, 117, 118, 128, 134, 135, 139-145, 147, 149, 151, 160, 163, 164, 60 166, 168-172, 162 and 253 m.p. were determined in open capillary tubes on a Mettler FP62 apparatus. M.p. were measured with a temperature ranging from 50° C. to 300° C., using a gradient of 10° C./minute. The m.p. value was read from a digital display.

For compounds 26, 37 and 90 m.p. were determined with a WRS-2A melting point apparatus that was purchased from

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Shanghai Precision and Scientific Instrument Co. Ltd. M.p. were measured with a linear heating up rate of 0.2-5.0° C./minute. The maximum temperature was 300° C.

For compounds 157 and 158, m.p. were determined in open capillary tubes on a FP 81HT/FP90 apparatus (Mettler-Toledo). M.p. were measured with a temperature gradient of 1, 3, 5 or 10° C./minute. Maximum temperature was 300° C. The m.p. value was read from a digital display.

For compounds 126, 127, 229, 230, 231, 232, 233, 254, 255, 259, 260, 269 and 270 m.p. were determined in open capillary tubes on a Mettler Toledo MP50 apparatus. M.p. were measured with a temperature ranging from 50° C. to 300° C., using a gradient of 10° C./minute. The m.p. value was read from a digital display.

The results of the analytical measurements are shown in table 2a.

TABLE 2a

Retention time (R<sub>2</sub>) in min., [M + H]<sup>+</sup> peak (protonated molecule), LCMS method and m.p. (melting point in ° C.). (n.d. means not determined)

Co. No.	$R_{t}$	$[M + H]^{+}$	LCMS Method	m.p. (° C.)
1	2.62	499	8	203.6
2	2.20	418	8	185.0
3	1.05	485	6	176.6
4a	1.55	446	8	n.d.
4	1.56	446	8	n.d.
5	0.95	411	4	165.6
6	2.65	487	8	175.4
7	2.59	499	8	290.0
8	1.93	542	8	n.d.
9	2.26	418	8	n.d.
9a	0.97	417	2	206.8
10	2.11	418	8	n.d.
10a	0.91	417	2	203.8
10b	0.91	417	2 1	206.4
11a 11b	0.92 0.92	417 417	1	181.1 181.0
14	2.59	499	8	189.8
14a	1.05	499	2	n.d.
14b	1.05	499	2	n.d.
15	2.11	343	8	n.d.
18	2.19	355	8	192.6
19	4.39	335	3	n.d.
20	2.23	403	8	199.8
21	1.89	363	8	163.0
22	1.73	349	9	186.1
23	2.46	405	8	n.d.
24	2.00	383	8	184.6
25	1.90	369	8	176.7
26	3.23	383	10	184.4
27	1.99	413	8	168.0
28	2.30	423	8	157.4
29 29a	2.19 0.98	409 409	8 2	227.1 n.d.
29a 30	2.15	395	8	168.5
31	2.13	480	8	165.3
32	2.92	488	8	n.d.
33	2.42	444	8	n.d.
34	2.45	446	8	107.6
35	2.05	401	8	182.2
36	2.09	404	8	224.9
37	4.68	417	5	136-141
38	0.95	403	2	190.8
39	1.91	387	8	n.d.
40	2.19	397	8	182.1
41	1.54	353	8	n.d.
42	0.87	403	1	183.4
43	2.18	417	8 8	164.9
44 45	2.51 2.33	485 501	8	n.d. n.d.
43 46	2.33 2.44	485	8	n.a. 145.6
46 47	2.54	515	8	n.d.
48	1.90	528	8	n.d.
49	1.81	514	8	n.d.

**263**TABLE 2a-continued

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TABLE 2a-continued

TABLE 2a-continued						TABLE 2a-continued					
		in., [M + H] <sup>+</sup> peal ting point in ° C.)			-				H] <sup>+</sup> peak (protonated molecule), LCMS at in ° C.). (n.d. means not determined)		
Co. No.	$\mathbf{R}_{t}$	$[M + H]^+$	LCMS Method	m.p. (° C.)	5	Co. No.	$R_t$	$[M + H]^+$	LCMS Method	m.p. (° C.)	
50	2.58	499	8	n.d.		124	1.44	349	13	136.5	
51	2.51	555	8	n.d.		125	1.69	466	14	n.d.	
52	2.83	547	8	n.d.		126	2.62	465	8	206.5	
53 54	1.09 1.90	471 515	2 9	219.0 n.d.	10	127 128	2.64 2.60	465 476	8 8	218.5 116.9	
55	1.92	407	8	227.5		129	1.90	465	14	n.d.	
56	1.99	421	8	197.6		130	1.89	465	14	n.d.	
57	1.99	403	8	248.6		131	0.99	417	6	170.7	
58	2.12	417	8	199.4		132	0.81	379	6	n.d.	
59 60	2.33 2.26	431 417	8	167.7 n.d.	15	133 134	0.85 2.56	385 499	6 8	198.1 136.0	
61	2.32	431	8	n.d.		135	2.84	485	8	213.2	
62	2.35	451	8	n.d.		136	2.81	485	8	n.d.	
63	2.38	451	8	180.4		137	1.72	436	13	n.d.	
64	2.20	437	8	n.d.		138	0.87	422	6	n.d.	
65 66	2.27 0.74	437 365	8 2	195.1 n.d.	20	139 140	1.61 1.99	451 501	9 9	195.9 115.2	
67	2.20	447	8	140.2		141	1.94	475	9	146.2	
68	2.25	447	8	132.4		142	1.96	475	9	141.4	
69	2.08	433	8	177.0		143	1.94	475	9	229.3	
70 71	2.19	435	8	108.5		144	1.94	475	9	194.4	
72	2.34 2.30	451 431	8	n.d. 208.3	25	145 146	n.d. 1.79	n.d. 447	9	204.5 n.d.	
73	1.82	367	8	227.4		147	2.38	447	8	232.9	
74	0.90	419	1	163.4		148	2.36	461	8	n.d.	
75	2.30	433	8	214.1		149	2.37	461	8	122.0	
76 77	1.77 2.01	475 435	9 9	n.d. n.d.		150 151	1.36 2.72	519 525	11 8	n.d. 292.1	
78	2.60	475	8	n.d.	30	151	2.72	511	8	n.d.	
79	2.24	433	8	n.d.	50	153	2.72	570	8	n.d.	
80	2.27	480	8	235.1		154	2.55	499	8	n.d.	
81	2.68	534	8	160.6		155	1.09	486	6	160.9	
82 83	2.51 1.96	452 443	8 8	233.6 n.d.		156 157	2.70 1.38	n.d. 563	8 11	n.d. 147.7	
84	1.23	432	8	108.2	25	158	2.95	503	12	137.1	
85	1.87	466	9	n.d.	35	159	3.22	519	12	n.d.	
86	1.97	500	9	185.3		160	2.50	461	8	171.7	
87 88	0.92	425 445	6	n.d. 136.5		161	2.94	499 459	12	n.d. 103.1	
89	2.48 4.33	335	8 7	146.5		163 164	2.48 2.76	525	8 8	213.7	
90	4.05	497	10	138-143		165	1.02	439	6	n.d.	
91	2.13	436	8	246.3	40	166	2.60	515	8	137.5	
92	2.47	485	8	242.2		167	2.42	515	8	n.d.	
93 94	1.71 2.30	407 399	8	192.2 n.d.		168 169	2.15 2.58	435 499	8 8	225.6 181.9	
95	1.54	420	8	181.6		170	n.d.	n.d.	n.d.	203.0	
96	0.33	418	8	163.6		171	2.83	500	8	130.6	
97	1.47	432	8	207.8	45	172	2.50	499	8	182.3	
98	1.39	448	8	n.d.		175	2.61	465	8	n.d.	
99 100	1.42 1.20	432 418	8 8	n.d. 167.4		176 177	0.81 0.97	363 482	6 6	n.d. n.d.	
101	2.01	375	8	121.5		178	0.94	436	6	n.d.	
102	1.47	335	8	222.2		179	0.84	422	6	n.d.	
103	1.89	407	8	176.8	50	180	1.73	470	14	n.d.	
104 105	1.62	388 429	8	n.d. 253.5		181 182	1.89 1.97	484	14	n.d.	
106	2.28 2.35	443	8	233.3 189.4		183	1.54	498 440	14 13	n.d. n.d.	
107	2.40	443	8	257.3		184	1.55	440	13	n.d.	
108	2.27	411	8	n.d.		185	1.79	454	14	258	
109	2.28	397	8	n.d.	55	186	1.79	454	14	261	
110	2.68 2.62	471 499	8	n.d. 147.3		187 188	0.88 0.89	454 454	6	n.d. n.d.	
111 112	2.65	499	8	259.0		189	0.89	468	6 6	263	
113	2.20	429	8	136.1		190	0.98	468	6	255	
114	1.88	429	8	126.9		191	1.03	482	6	n.d.	
115	2.35	483	8	101.4	60	192	1.03	482	6	n.d.	
116 117	2.55 2.57	457 511	8 8	n.d. 221.5	00	193 194	1.08 1.10	496 496	6	n.d.	
117	3.72	505	8	221.5 157.9		194	0.88	496 440	6 6	n.d. n.d.	
119	0.77	366	4	151.2		196	0.96	454	6	187	
120	1.25	404	2	194.9		197	1.71	514	14	n.d.	
121	0.92	417	1	173.4	65	198	0.89	456	6	n.d.	
122	0.90	417	1	n.d.	65	199	1.17 1.07	542	6	n.d.	
123	2.55	499	8	n.d.		200	1.07	508	6	181	

**266**TABLE 2a-continued

	$R_t$	[M + H] <sup>+</sup>	LCMS Method	m.p. (° C.)	5	Co. No.	R,	$[M + H]^{+}$	LCMS Method	m.p. (° C.)
No.		[141 + 11]			_					
201	0.74	418	6	n.d.		251	1.75	480	13	150
202	1.50	418	14	n.d.		252	2.55	472	8	n.d.
203	0.90	456	6	n.d.		253	2.81	486	8	190
204	1.74	443	13	n.d.	10	254	2.72	456	8	217
205	1.73	443	13	n.d.		255	2.79	470	8	265
206	0.87	375	6	n.d.		256	1.66	480	13	n.d.
207	0.87	375	6	n.d.		257	1.65	474	14	n.d.
208	1.74	417	13	206		258	1.73	488	14	n.d.
209	1.74	417	13	206		259	2.86	456	8	230
210	5.92	431	3	n.d.	15	260	2.52	442	8	208
211	6.03	431	3	146		261	1.79	488	14	n.d.
212	1.72	470	14	n.d.		262	1.75	484	13	n.d.
213	1.71	470	14	n.d.		263	0.93	474	6	223
214	1.77	484	13	n.d.		264	0.98	488	6	193
215	1.77	484	13	n.d.		265	0.92	488	6	260
216	1.86	498	13	n.d.	20	266	0.96	488	6	258
217	1.87	498	13	n.d.	20	267	1.53	474	14	n.d.
218	0.91	454	6	n.d.		268	1.59	474	14	n.d.
219	0.91	454	6	263		269	2.58	474	8	138
220	1.08	482	6	n.d.		270	2.72	488	8	138
221	1.09	482	6	194		271	0.95	470	6	n.d.
222	1.80	460	14	n.d.		272	0.95	470	6	n.d.
223	1.01	468	6	243	25	273	1.02	484	6	n.d.
224	1.03	468	6	n.d.		274	1.84	484	13	n.d.
225	1.03	484	6	n.d.		276	0.87	436	6	n.d.
226	1.80	470	14	n.d.		277	0.88	428	6	n.d.
227	1.03	484	6	n.d.		278	0.81	402	6	n.d.
228	0.98	470	6	161		279	0.81	452	6	n.d.
229	2.62	445	8	129	30	282	0.82	466	6	n.d.
230	2.48	431	8	192		283	0.86	460	6	n.d.
231	2.45	431	8	218		284	0.89	474	6	n.d.
232	2.43	449	8	148		285	0.96	488	6	n.d.
233	2.37	427	8	193		286	0.86	456	6	n.d.
234	0.91	470	6	n.d.		287	0.96	470	6	n.d.
235	1.05	498	6	n.d.	35	288	0.88	456	6	n.d.
236	1.66	474	14	n.d.	33	289	1.02	484	6	n.d.
237	1.62	484	14	n.d.		293	0.82	422	6	n.d.
238	0.98	488	6	n.d.		295	0.97	450	6	n.d.
239	0.96	498	6	n.d.		296	0.97	450	6	n.d.
240	0.93	474	6	n.d.						
241	1.03	488	6	n.d.						
162	n.d.	n.d.	n.d.	171	40	NMR				
173	1.00	471	6	n.d.		For a m	umber of	compounds,	<sup>1</sup> H NMR	spectra
174	1.00	460	6	n.d.				Avance III wi		
242	0.88	420	6	n.d.						
243	1.01	464	6	n.d.				DPX-400 spec		
244	1.05	498	6	n.d.				PX-360 opera		
245	0.86	436	6	n.d.	45			ectrometer op		
246	0.97	442	6	n.d.				(deuterated		
247	1.04	456	6	n.d.						
248	1.93	494	13	n.d.				d DMSO, dim		
249	1.00	464	6	n.d.		vent. Chen	nical shift	s (δ) are repo	orted in part	s per m
250	0.96	430	6	n.d.				methylsilane		

TABLE 2b

	<sup>1</sup> H NMR results
Co. No.	<sup>1</sup> H NMR result
2	$\begin{array}{l} (300 \text{ MHz, DMSO-d}_{6})  \delta \text{ ppm } 0.84 \text{ (t, J} = 7.4 \text{ Hz, 3 H), } 1.24\text{-}1.44 \text{ (m, 1 H),} \\ 1.63\text{-}1.79 \text{ (m, 1 H), } 2.35 \text{ (br. s, 3 H), } 3.84\text{-}4.00 \text{ (m, 2 H), } 4.62 \text{ (d, J} = 15.5 \text{ Hz, 1 H),} \\ 4.84 \text{ (dd, J} = 15.7, 3.3 \text{ Hz, 1 H), } 5.24 \text{ (d, J} = 15.7 \text{ Hz, 1 H), } 7.22 \text{ (d, J} = 7.7 \text{ Hz, 1 H),} \\ 1.87\text{-}7.88\text{-}7.91 \text{ (m, 1 H), } 8.02\text{-}8.18 \text{ (m, 4 H), } 9.54 \text{ (d, J} = 1.4 \text{ Hz, 1 H)} \\ (300 \text{ MHz, DMSO-d}_{6})  \delta \text{ ppm } 1.12 \text{ (d, J} = 6.6 \text{ Hz, 3 H), } 2.16 \text{ (s, 3 H),} \\ 3.89\text{-}4.02 \text{ (m, 2 H), } 4.42 \text{ (d, J} = 15.4 \text{ Hz, 1 H), } 4.60 \text{ (d, J} = 11.8 \text{ Hz, 1 H), } 5.01 \text{ (d, J} = 15.4 \text{ Hz, 1 H),} \\ 1.19\text{-} 1.3 \text{ (d, J} = 7.8 \text{ Hz, 1 H), } 7.36\text{-}7.46 \text{ (m, 2 H), } 7.60\text{-}7.70 \text{ (m, 2 H),} \end{array}$
3	$\begin{array}{l} 7.80\text{-}7.85\ (m,1\ H),\ 8.28\ (s,1\ H)\\ (300\ MHz,\ DMSO-d_6)\ \delta\ ppm\ 1.15\ (d,\ J=6.5\ Hz,\ 3\ H),\ 2.17\ (d,\ J=0.7\ Hz,\ 3\ H),\\ 3.89\text{-}4.13\ (m,\ 2\ H),\ 4.61\ (dd,\ J=13.7,\ 2.3\ Hz,\ 2\ H),\ 5.18\ (d,\ J=15.7\ Hz,\ 1\ H),\\ 7.15\ (d,\ J=7.8\ Hz,\ 1\ H),\ 7.43\ (t,\ J=1.2\ Hz,\ 1\ H),\ 7.83\ (d,\ J=7.7\ Hz,\ 1\ H),\\ 8.02\text{-}8.17\ (m,\ 3\ H),\ 8.28\ (d,\ J=1.2\ Hz,\ 1\ H) \end{array}$

#### <sup>1</sup>H NMR results Co. No. 1H NMR result $(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.04 \text{ (d, J} = 6.2 \text{ Hz}, 3 \text{ H)}, 1.55 \text{ (br. s., 2 H)},$ 1.65 (br. s., 4 H), 2.23 (s, 3 H), 2.31 (s, 3 H), 2.80 (t, J = 9.1 Hz, 4 H), 3.80-3.98 (m, 2 H), 4.34 (d, J = 14.8 Hz, 1 H), 4.61 (dd, J = 13.7, 1.6 Hz, 1 H), 5.01 (d, J = 14.8 Hz, 1 H), 6.92-7.03 (m, 2 H), 7.14 (d, J = 7.7 Hz, 1 H), 7.22 (d, J = 7.8 Hz, 1 H), 7.78 (s, 1 H), 8.05 (d, J = 7.8 Hz, 1 H), 9.26 (s, 1 H)(300 MHz, CHLOROFORM-d) δ ppm 1.21 (d, J = 6.7 Hz, 3 H), 1.52-1.64 (m, 2 H), 1.65-1.80 (m, 4 H), 2.29 (s, 6 H), 2.72-2.93 (m, 4 H), 3.66 (dd, J = 14.1, 4.1 Hz, 1 H), 3.75-3.89 (m, 1 H), 4.05 (d, J = 14.7 Hz, 1 H), 4.76 (dd, J = 14.0, 2.1 Hz, 1 H), 5.36 (d, J = 14.7 Hz, 1 H), 6.81-6.98 (m, 2 H), 7.15 (d, J = 7.7 Hz, 2 H), 7.32 (d, J = 7.6 Hz, 1 H), 7.48 (d, J = 7.7 Hz, 1 H), 8.26 (br. s., 1 H) $(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 2.16 \text{ (s, 3 H)}, 3.71 \text{ (t, J = 6.3 Hz, 2 H)}, 4.27 \text{ (t, J = 6.3 Hz, 2 Hz, 2 H)}, 4.27 \text{ (t, J = 6.3 Hz, 2 Hz, 2 Hz, 2 Hz)}, 4.27 \text{ (t, J = 6.3 Hz, 2 Hz,$ J = 6.3 Hz, 2 H), 4.80 (s, 2 H), 7.15 (d, J = 7.7 Hz, 1 H), 7.27-7.55 (m, 6 H), 7.55-7.75 (m, 4 H), 7.81 (d, J = 7.7 Hz, 1 H), 8.26 (s, 1 H)(300 MHz, DMSO-d<sub>6</sub>) δ ppm 0.15-0.42 (m, 2 H), 0.42-0.72 (m, 2 H), 1.13 (d, J = 6.3 Hz, 3 H), 1.16-1.28 (m, 1 H), 2.33 (s, 3 H), 3.71-4.10 (m, 4 H),1.15 (d, J = 0.5 Hz, 3 H), 1.10-1.26 (lll, 1 H), 2.53 (8, 3 H), 5.71-4.10 (lll, 4 H), 4.48 (d, J = 15.4 Hz, 1 H), 4.60 (d, J = 12.4 Hz, 1 H), 5.07 (d, J = 15.5 Hz, 1 H), 7.02-7.42 (lll, 4 H), 7.86 (s, 1 H), 8.10 (d, J = 7.7 Hz, 1 H), 9.48 (s, 1 H) (300 MHz, DMSO-d<sub>c</sub>) & ppm 1.17 (d, J = 6.5 Hz, 3 H), 1.26 (t, J = 7.6 Hz, 3 H), 2.71 (q, J = 7.4 Hz, 2 H), 3.93-4.18 (lll, 2 H), 4.57-4.69 (lll, 2 H), 5.20 (d, J = 15.5 Hz, 1 H), 7.25 (d, J = 7.7 Hz, 1 H), 7.89 (d, J = 0.7 Hz, 1 H), 7 8.03-8.16 (m, 4 H), 9.48 (s, 1 H) $(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.57 \text{ (d, J} = 7.1 \text{ Hz, 3 H)}, 2.16 \text{ (s, 3 H)},$ 3.35-3.49 (m, 1 H), 3.67 (ddd, J = 13.3, 8.6, 4.1 Hz, 1 H), 3.93-4.21 (m, 1 H), 4.30 (ddd, J = 14.1, 6.8, 4.1 Hz, 1 H), 5.59-5.90 (m, 1 H), 7.14 (d, J = 7.8 Hz, 1 H), $7.27-7.50 \; (m, 2 \; H), \\ 7.50-7.69 \; (m, 2 \; H), \\ 7.81 \; (d, J = 7.7 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; H), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz, 1 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\ 8.26 \; (d, J = 1.0 \; Hz), \\$ 1 H). 9a The NMR spectrum of 9a matches the one of compound 9 9b The NMR spectrum of 9b matches the one of compound 9 $\,$ 10 $(300 \text{ MHz}, DMSO-d_6) \delta \text{ ppm } 1.57 \text{ (d, J} = 7.0 \text{ Hz}, 3 \text{ H)}, 2.15 \text{ (br. s, 3 H)},$ 3.19-3.30 (m, 1 H), 3.52-3.74 (m, 1 H), 4.08-4.27 (m, 2 H), 5.85 (q, J = 7.0 Hz, 1 H), 7.11 (d, J = 7.7 Hz, 1 H), 7.36-7.51 (m, 2 H), 7.60 (dd, J = 7.8, 1.4 Hz, 1 H), $7.66~(\mathrm{dd},\,\mathrm{J}=8.0,\,1.6~\mathrm{Hz},\,1~\mathrm{H}),\,7.79~(\mathrm{d},\,\mathrm{J}=7.8~\mathrm{Hz},\,1~\mathrm{H}),\,8.25~(\mathrm{d},\,\mathrm{J}=1.2~\mathrm{Hz},\,1~\mathrm{H})$ 10a The NMR spectrum of 10a matches the one of compound 10 10h The NMR spectrum of 10b matches the one of compound 10 $(360 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.61 \text{ (d, J} = 7.0 \text{ Hz, 3 H)}, 2.15 \text{ (s, 3 H)},$ 3.18-3.32 (m, 1 H), 3.69 (ddd, J = 13.2, 8.4, 4.0 Hz, 1 H), 4.08 (ddd, J = 12.4, 8.4, 4.0 Hz, 1 H), 4.29 (ddd, J = 14.1, 6.8, 3.7 Hz, 1 H), 5.87 (q, J = 7.0 Hz, 1 H), 7.14 (d, J = 7.7 Hz, 1 H), 7.43 (br. s., 1 H), 7.59-7.73 (m, 4 H), 7.81 (d, J = 7.7 Hz, 1 H), 8.27 (br. s., 1 H) 11b The NMR spectrum of 11b matches the one of compound 11a The NMR spectrum of 14 matches the one of compound 14a 14a $(600 \text{ MHz}, \text{CHLOROFORM-d}) \delta \text{ ppm } 1.31 \text{ (d, J = 6.6 Hz, 3 H)}, 1.39 \text{ (d, J = 7.0 Hz, })$ 3 H), 2.29 (d, J = 0.9 Hz, 3 H), 4.17 (qd, J = 6.9, 3.2 Hz, 1 H), 4.82 (d, J = 16.0 Hz, 1 H), 5.10 (qd, J = 6.6, 3.2 Hz, 1 H), 5.13 (d, J = 16.0 Hz, 1 H), 7.15 (s, 1 H), 7.36 (d, J = 7.6 Hz, 1 H), 7.49 (d, J = 7.8 Hz, 1 H), 7.72 (s, 2 H), 7.83 (s, 1 H),14b The NMR spectrum of 14b matches the one of compound 14a (300 MHz, DMSO-d<sub>6</sub>) δ ppm 0.88 (s, 9 H), 1.12-1.22 (m, 2 H), 1.47-1.60 (m, 2 H), 2.16 (s, 3 H), 3.45 (t, J = 7.3 Hz, 2 H), 3.63-3.76 (m, 2 H), 4.19-4.28 (m, 2 H), 7.08 (d, J = 7.7 Hz, 1 H), 7.38-7.43 (m, 1 H), 7.79 (d, J = 7.7 Hz, 1 H), 8.22-8.28 (m, 1 H) (300 MHz, DMSO-d<sub>6</sub>) δ ppm 0.77-1.03 (m, 2 H), 1.04-1.37 (m, 4 H), 18 1.46 (q, J = 7.0 Hz, 2 H), 1.54-1.82 (m, 5 H), 2.16 (s, 3 H), 3.50 (t, J = 7.6 Hz, 2 H),3.69 (dd, J = 6.8, 5.0 Hz, 2 H), 4.11-4.40 (m, 2 H), 7.07 (d, J = 7.8 Hz, 1 H), 7.40 (s, 1 H), 7.78 (d, J = 7.7 Hz, 1 H), 8.14-8.36 (m, 1 H) $(400 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 2.15 \text{ (s, 3 H)}, 3.64 \text{ (t, J} = 6.1 \text{ Hz, 2 H)}, 4.22 \text{ (t, J)}$ 19 J = 5.7 Hz, 2 H), 4.71 (s, 2 H), 7.12 (d, J = 7.7 Hz, 1 H), 7.20-7.45 (m, 6 H), 7.70 (d, J = 7.7 Hz, 1 H), 8.17 (br. s., 1 H) 20 $(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 0.69 + 0.89 \text{ (m, 2 H)}, 1.01 \text{ (d, J = 6.7 Hz, 3 H)},$ 1.04 (d, J = 6.0 Hz, 2 H), 2.15 (s, 3 H), 2.48 (s, 3 H), 3.08 (d, J = 13.2 Hz, 1 H),3.39 (dd, J = 14.1, 4.1 Hz, 1 H), 3.46-3.68 (m, 1 H), 4.21 (br. s., 1 H), 4.51 (d, J = 14.2 Hz, 1 H, 7.01 (d, J = 7.7 Hz, 1 H, 7.06-7.20 (m, 3 H), 7.30 (d, J = 6.3 Hz, $1~{\rm H}),\,7.39~({\rm s},\,1~{\rm H}),\,7.76~({\rm d},\,{\rm J}=7.7~{\rm Hz},\,1~{\rm H}),\,8.23~({\rm s},\,1~{\rm H})$ 21 $(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.09 \text{ (d, J = 6.5 Hz, 3 H), } 2.16 \text{ (s, 3 H), } 2.30 \text{ (s, 3 H)}$ H), 3.77-3.86 (m, 1 H), 3.86-3.96 (m, 1 H), 4.39 (d, J = 15.5 Hz, 1 H), 4.64 (dd, J = 13.9, 1.6 Hz, 1 H), 5.04 (d, J = 15.5 Hz, 1 H), 7.14 (d, J = 7.7 Hz, 1 H), 7.17-7.33 (m, 4 H), 7.43 (s, 1 H), 7.83 (d, J = 7.7 Hz, 1 H), 8.28 (s, 1 H) 23 $(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.09 \text{ (d, J} = 6.5 \text{ Hz}, 3 \text{ H)}, 1.29 \text{ (s, 9 H)}, 2.18 \text{ (s, 3 H)}$ H), 3.78-3.90 (m, 1 H), 3.90-4.00 (m, 1 H), 4.43 (d, J = 15.1 Hz, 1 H), 4.63 (dd, J = 13.8, 2.0 Hz, 1 H), 5.03 (d, J = 15.3 Hz, 1 H), 7.11-7.22 (m, 2 H),7.24-7.37 (m, 2 H), 7.39 (s, 1 H), 7.47 (s, 1 H), 7.85 (d, J = 7.7 Hz, 1 H), 8.40 (s, 1 H)

 $(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.13 \text{ (d, J = 6.5 Hz, 3 H), } 2.31 \text{ (s, 3 H),}$ 

J = 7.7 Hz, 1 H), 9.26 (s, 1 H)

1 H), 7.22 (d, J = 7.7 Hz, 1 H), 7.28-7.57 (m, 4 H), 7.78 (s, 1 H), 8.05 (d,

3.90-4.10 (m, 2 H), 4.46 (d, J = 15.4 Hz, 1 H), 4.60 (d, J = 12.0 Hz, 1 H), 5.03 (d, J = 15.4 Hz,

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	<sup>1</sup> H NMR results
Co. No.	<sup>1</sup> H NMR result
25	(300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 2.16 (s, 3 H), 3.64-3.74 (m, 2 H), 4.21-4.31 (m, 2 H), 4.72 (s, 2 H), 7.15 (d, J = 7.7 Hz, 1 H), 7.28-7.47 (m, 5 H), 7.81 (d, J = 7.8 Hz, 1 H), 8.26 (a, 1 H)
29	1 H), 8.26 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 0.78-0.89 (m, 2 H), 0.95-1.04 (m, 2 H), 2.15 (s, 3 H), 3.55 (t, J = 5.6 Hz, 2 H), 3.73 (s, 2 H), 3.97-4.16 (m, 2 H), 6.96 (d, J = 7.7 Hz, 1 H), 7.28-7.35 (m, 2 H), 7.35-7.43 (m, 3 H), 7.74 (d, J = 7.7 Hz, 1 H), 8.20 (a, J H)
29a	8.22 (s, 1 H) (400 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 0.83-0.90 (m, 2 H), 0.97-1.03 (m, 2 H), 2.29 (s, 3 H), 3.52-3.60 (m, 2 H), 3.76 (s, 2 H), 4.02-4.12 (m, 2 H), 7.04 (d, $J$ = 7.7 Hz, 1 H), 7.28 (m, $J$ = 8.5 Hz, 2 H), 7.37 (m, $J$ = 8.5 Hz, 2 H), 7.80 (br. s., 1 H),
30	7.93 (d, J = 7.7 Hz, 1 H), 9.18 (br. s., 1 H) (300 MHz, DMSO- $^{4}$ <sub>6</sub> ) δ ppm 1.42 (q, J = 7.1 Hz, 1 H), 1.48-1.61 (m, 1 H), 2.21 (s, 3 H), 2.39 (ddd, J = 9.8, 6.6, 3.6 Hz, 1 H), 3.02-3.14 (m, 1 H), 3.77 (t, J = 6.0 Hz, 2 H), 4.14-4.30 (m, 1 H), 4.30-4.44 (m, 1 H), 7.16 (d, J = 7.7 Hz, 1 H), 7.33 (d, J = 8.5 Hz, 2 H), 7.42 (d, J = 8.5 Hz, 2 H), 7.47 (s, 1 H), 7.86 (d,
31	J = 7.8 Hz, 1 H), 8.32 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.17 (s, 3 H), 4.18 (d, J = 15.4 Hz, 1 H), 4.30 (dd, J = 14.2, 4.5 Hz, 1 H), 4.99 (dd, J = 14.2, 2.2 Hz, 1 H), 5.18-5.32 (m, 2 H), 7.15-7.23 (m, 2 H), 7.27 (d, J = 7.7 Hz, 1 H), 7.32-7.47 (m, 5 H), 7.65 (d, J = 8.2 Hz,
34	1 H), 7.70 (d, J = 2.1 Hz, 1 H), 7.86 (d, J = 7.7 Hz, 1 H), 8.21-8.27 (m, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 0.71 (d, J = 6.5 Hz, 3 H), 0.97 (d, J = 6.6 Hz, 3 H), 1.87 (dq, J = 13.1, 6.7 Hz, 1 H), 2.16 (s, 3 H), 3.66-3.88 (m, 2 H), 4.31 (d, J = 15.3 Hz, 1 H), 4.94 (d, J = 14.3 Hz, 1 H), 5.17 (d, J = 15.4 Hz, 1 H), 7.09 (d, J = 7.7 Hz, 1 H), 7.33-7.48 (m, 2 H), 7.61 (d, J = 8.2 Hz, 1 H), 7.69 (s, 1 H),
35	7.81 (d, J = 7.6 Hz, 1 H), 8.27 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.15 (d, J = 6.5 Hz, 3 H), 2.16 (s, 3 H), 3.88-4.12 (m, 2 H), 4.54 (d, J = 15.7 Hz, 1 H), 4.64 (dd, J = 13.5, 1.7 Hz, 1 H), 5.04 (d, J = 15.7 Hz, 1 H), 7.12 (d, J = 7.7 Hz, 1 H), 7.17-7.31 (m, 1 H), 7.43 (t, J = 6.5 Hz, 2 H), 7.47 (f, J = 1.1 H), 7.17 (f, J = 1.1 H), 7.17 (f, J = 1.1 H), 7.18 (f, J = 6.5 Hz, 2 Hz, 2 H), 7.18 (f, J = 6.5 Hz, 2
36	2 H), 7.49-7.61 (m, 1 H), 7.82 (d, J = 7.7 Hz, 1 H), 8.29 (br. s., 1 H) (300 MHz, DMSO- $\frac{1}{6}$ 0 8 ppm 2.16 (s, 3 H), 3.75 (t, J = 5.6 Hz, 2 H), 4.32 (t, J = 5.6 Hz, 2 H), 4.82 (s, 2 H), 7.13 (d, J = 7.7 Hz, 1 H), 7.28-7.50 (m, 3 H), 7.62 (dd, J = 6.9, 2.3 Hz, 1 H), 7.82 (d, J = 7.7 Hz, 1 H), 8.28 (s, 1 H)
37	$ \begin{array}{l} (400 \text{ MHz, CHLOROFORM-d}) \ \delta \ ppm \ 1.26 \ (d, J=6.5 \ Hz, 3 \ H), \ 2.31 \ (br.  s., 3 \ H), \ 3.69-3.90 \ (m, 2 \ H), \ 4.15 \ (d, J=15.1 \ Hz, 1 \ H), \ 4.77 \ (d, J=13.8 \ Hz, 1 \ H), \ 5.27 \ (d, J=15.3 \ Hz, 1 \ H), \ 7.19 \ (d, J=6.8 \ Hz, 2 \ H), \ 7.33 \ (d, J=7.5 \ Hz, 1 \ H), \ \end{array} $
38	7.40-7.58 (m, 3 H), 8.29 (br. s., 1 H) (360 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.15 (s, 3 H), 3.66-3.74 (m, 2 H), 4.21-4.31 (m, 2 H), 4.71 (s, 2 H), 7.13 (d, J = 7.7 Hz, 1 H), 7.35 (dd, J = 8.2, 2.4 Hz, 1 H),
39	7.42 (s, 1 H), 7.59-7.67 (m, 2 H), 7.81 (d, $J = 8.1$ Hz, 1 H), 8.26 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 2.16 (s, 3 H), 3.68-3.79 (m, 2 H), 4.23-4.33 (m, 2 H), 4.80 (s, 2 H), 7.12 (d, $J = 7.7$ Hz, 1 H), 7.19-7.28 (m, 1 H), 7.36-7.45 (m, 2 H), 7.50-7.60 (m, 1 H), 7.80 (d, $J = 7.7$ Hz, 1 H), 8.24-8.29 (m, 1 H)
40	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.11 (d, J = 6.6 Hz, 3 H), 2.17 (s, 3 H), 2.35 (s, 3 H), 3.78-3.88 (m, 1 H), 3.93 (dd, J = 14.1, 4.1 Hz, 1 H), 4.44 (d, J = 15.7 Hz, 1 H), 4.64 (dd, J = 14.0, 2.1 Hz, 1 H), 5.09 (d, J = 15.7 Hz, 1 H), 7.14 (d, J = 7.7 Hz, 1 H), 7.23 (t, J = 7.7 Hz, 1 H), 7.31 (d, J = 6.7 Hz, 1 H), 7.36-7.46 (m, 2 H), 7.83 (d, J = 7.8 Hz, 1 H), 8.28 (d, J = 1.2 Hz, 1 H)
42	(360 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.15 (s, 3 H), 3.64-3.75 (m, 2 H), 4.21-4.30 (m, 2 H), 4.81 (s, 2 H), 7.14 (d, J = 7.7 Hz, 1 H), 7.42 (s, 1 H), 7.57-7.71 (m, 3 H), 7.73 (s, 1 H), 7.81 (d, J = 8.1 Hz, 1 H), 8.26 (s, 1 H)
43	$(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.14 \text{ (d, J} = 6.6 \text{ Hz}, 3 \text{ H)}, 2.16 \text{ (s, 3 H)}, \\ 3.88\text{-}4.09 \text{ (m, 2 H)}, 4.47\text{-}4.71 \text{ (m, 2 H)}, 5.10 \text{ (d, J} = 15.7 \text{ Hz}, 1 \text{ H)}, 7.15 \text{ (d, J} = 7.7 \text{ Hz}, 1 \text{ H)}, \\ 7.44 \text{ (s, 1 H)}, 7.61 \text{ (d, J} = 8.1 \text{ Hz}, 2 \text{ H)}, 7.73 \text{ (d, J} = 8.2 \text{ Hz}, 2 \text{ H)}, 7.84 \text{ (d, J} = 7.7 \text{ Hz}, 1 \text{ H)}, \\ 8.28 \text{ (d, J} = 1.1 \text{ Hz}, 1 \text{ H)}$
44	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.15 (d, J = 6.6 Hz, 3 H), 2.16 (s, 3 H), 3.96 (dd, J = 13.8, 4.2 Hz, 1 H), 4.00-4.13 (m, 1 H), 4.53-4.66 (m, 2 H), 5.18 (d, J = 15.5 Hz, 1 H), 7.15 (d, J = 7.8 Hz, 1 H), 7.44 (s, 1 H), 7.83 (d, J = 7.8 Hz, 1 H), 8.02-8.16 (m, 3 H), 8.29 (s, 1 H)
45	(300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 2.16 (s, 3 H), 3.39-3.60 (m, 2 H), 3.83-3.98 (m, 2 H), 4.64 (d, J = 15.8 Hz, 1 H), 4.78-4.90 (m, 1 H), 5.07-5.17 (m, 1 H), 5.22 (d, J = 15.9 Hz, 1 H), 7.09 (d, J = 7.7 Hz, 1 H), 7.43 (t, J = 1.2 Hz, 1 H), 7.79 (d, J = 7.8 Hz, 1 H), 8.05 (s, 1 H), 8.12 (s, 2 H), 8.28 (d, J = 1.2 Hz, 1 H)
48	39 (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.14 (s, 6 H), 2.22 (s, 3 H), 2.29 (d, J = 7.4 Hz, 2 H), 3.93 (dd, J = 14.0, 4.3 Hz, 1 H), 4.05-4.21 (m, 1 H), 4.76 (d, J = 15.8 Hz, 1 H), 4.95 (d, J = 13.2 Hz, 1 H), 5.19 (d, J = 15.5 Hz, 1 H), 7.18 (d, J = 7.7 Hz, 1 H), 7.50 (s, 1 H), 7.87 (d, J = 7.7 Hz, 1 H), 8.11 (s, 1 H), 8.18 (s, 2 H), 8.34 (d, J = 1.1 Hz, 1 H)
50	$(300 \text{ MHz}, \text{DMSO-d}_6)  \delta  \text{ppm}  1.13  (\text{d},  \text{J} = 6.7  \text{Hz},  3  \text{H}),  2.16  (\text{s},  3  \text{H}), \\ 3.07\text{-}3.19  (\text{m},  2  \text{H}),  3.37\text{-}3.51  (\text{m},  1  \text{H}),  3.77  (\text{dd},  \text{J} = 14.1,  4.2  \text{Hz},  1  \text{H}),  3.97\text{-}4.17  (\text{m},  2  \text{H}), \\ 4.62  (\text{dd},  \text{J} = 14.0,  2.3  \text{Hz},  1  \text{H}),  7.04  (\text{d},  \text{J} = 7.7  \text{Hz},  1  \text{H}),  7.38\text{-}7.45  (\text{m},  1  \text{H}), \\ 4.62  (\text{dd},  \text{J} = 14.0,  2.3  \text{Hz},  1  \text{H}),  7.04  (\text{d},  \text{J} = 7.7  \text{Hz},  1  \text{H}),  7.38\text{-}7.45  (\text{m},  1  \text{H}), \\ 4.62  (\text{dd},  \text{J} = 14.0,  2.3  \text{Hz},  1  \text{H}),  7.04  (\text{d},  \text{J} = 7.7  \text{Hz},  1  \text{H}),  7.38\text{-}7.45  (\text{m},  1  \text{H}), \\ 4.62  (\text{dd},  \text{J} = 14.0,  2.3  \text{Hz},  1  \text{H}),  7.04  (\text{d},  \text{J} = 7.7  \text{Hz},  1  \text{H}),  7.38\text{-}7.45  (\text{m},  1  \text{H}), \\ 4.62  (\text{dd},  \text{J} = 14.0,  2.3  \text{Hz},  1  \text{H}),  7.04  (\text{d},  \text{J} = 7.7  \text{Hz},  1  \text{H}),  7.38\text{-}7.45  (\text{m},  1  \text{H}),  7.38\text{-}7.45  (\text{m},  1  \text{H}), \\ 4.62  (\text{dd},  \text{J} = 14.0,  2.3  \text{Hz},  1  \text{H}),  7.38\text{-}7.45  (\text{m},  1 $
53	H), 7.80 (d, J = 7.8 Hz, 1 H), 7.96 (s, 1 H), 8.08 (s, 2 H), 8.27 (d, J = 1.1 Hz, 1 H) (360 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.15 (s, 3 H), 3.67-3.85 (m, 2 H), 4.17-4.37 (m, 2 H), 4.89 (s, 2 H), 7.15 (d, J = 7.7 Hz, 1 H), 7.43 (br. s., 1 H), 7.81 (d, J = 7.7 Hz, 1 H), 7.98-8.15 (m, 3 H), 8.27 (br. s., 1 H)

	<sup>1</sup> H NMR results
Co. No.	<sup>1</sup> H NMR result
55	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.17 (s, 3 H), 4.11-4.19 (m, 2 H), 4.44 (t, J = 5.6 Hz, 2 H), 7.21 (d, J = 7.8 Hz, 1 H), 7.46 (s, 1 H), 7.50-7.60 (m, 1 H),
57	7.78-7.96 (m, 3 H), 8.32 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.17 (s, 3 H), 2.31 (s, 3 H), 3.84-3.98 (m, 1 H), 4.08-4.22 (m, 1 H), 4.25-4.40 (m, 1 H), 4.61 (dt, J = 14.1, 4.6 Hz, 1 H), 7.17 (d, J = 7.7 Hz, 1 H), 7.45 (s, 1 H), 7.48-7.60 (m, 1 H), 7.68 (d, J = 7.8 Hz, 1 H),
59	7.70 (d, J = 7.8 Hz, 1 H), 7.84 (d, J = 7.8 Hz, 1 H), 8.30 (s, 1 H) (300 MHz, DMSO- $d_6$ ) $\delta$ ppm 1.13 (d, J = 6.5 Hz, 3 H), 2.16 (s, 3 H), 2.40 (s, 3 H), 3.84-4.02 (m, 2 H), 4.46 (d, J = 15.4 Hz, 1 H), 4.61 (d, J = 11.8 Hz, 1 H), 5.09 (d, J = 15.4 Hz, 1 H), 7.15 (d, J = 7.7 Hz, 1 H), 7.43 (s, 1 H), 7.50 (d, J = 5.5 Hz,
60	2 H), 7.54 (s, 1 H), 7.83 (d, J = 7.7 Hz, 1 H), 8.28 (d, J = 1.1 Hz, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 2.16 (s, 3 H), 2.40 (s, 3 H), 3.63-3.76 (m, 2 H), 4.20-4.33 (m, 2 H), 4.77 (s, 2 H), 7.15 (d, J = 7.7 Hz, 1 H), 7.38-7.44 (m, 1 H), 7.44-7.55 (m, 3 H), 7.81 (d, J = 7.7 Hz, 1 H), 8.22-8.29 (m, 1 H)
61	$\begin{array}{l} (300 \text{ MHz}, \text{DMSO-d}_6) \ \delta \text{ ppm 1.14} \ (d, \ J=6.6 \ \text{Hz}, \ 3 \ \text{H}), \ 2.17 \ (s, \ 3 \ \text{H}), \ 2.41 \ (s, \ 3 \ \text{H}), \ 3.89 \ (\text{br. s.}, \ 1 \ \text{H}), \ 4.01 \ (dd, \ J=14.2, \ 4.0 \ \text{Hz}, \ 1 \ \text{H}), \ 4.50 \ (d, \ J=15.9 \ \text{Hz}, \ 1 \ \text{H}), \ 4.65 \ (dd, \ J=14.1, \ 2.1 \ \text{Hz}, \ 1 \ \text{H}), \ 5.10 \ (d, \ J=15.9 \ \text{Hz}, \ 1 \ \text{H}), \ 7.14 \ (d, \ J=7.7 \ \text{Hz}, \ 1 \ \text{H}), \ 7.36-7.48 \ (m, \ 2 \ \text{H}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ 1 \ \text{H}), \ 7.66 \ (d, \ J=7.8 \ \text{Hz}, \ 1 \ \text{H}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ 1 \ \text{Hz}, \ 1 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ 1 \ \text{Hz}, \ 1 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ 1 \ \text{Hz}, \ 1 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ 1 \ \text{Hz}, \ 1 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ 1 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ 1 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ 1 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ 1 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ J=7.6 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ J=7.6 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ J=7.6 \ \text{Hz}), \ 7.61 \ (d, \ J=7.6 \ \text{Hz}, \ J=7.6 \ \text{Hz}), \ 7.61 \ (d, \ \ J=7.6 \ \text{Hz}, \ \ J=7.6 \ \text{Hz}), \ 7.61 \ (d, \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$
62	7.84 (d, J = 7.7 Hz, 1 H), 8.29 (d, J = 1.1 Hz, 1 H) (300 MHz, CHLOROFORM-d) δ ppm 1.28 (d, J = 6.6 Hz, 3 H), 2.30 (s, 3 H), 3.70-3.92 (m, 2 H), 4.23 (d, J = 15.1 Hz, 1 H), 4.80 (dd, J = 14.0, 2.3 Hz, 1 H), 5.31 (d, J = 15.1 Hz, 1 H), 7.16 (s, 1 H), 7.33 (d, J = 7.7 Hz, 1 H), 7.44-7.57 (m, 3 H), 7.62-7.69 (m, 1 H), 8.26 (d, J = 1.1 Hz, 1 H)
63	(300 MHz, CHLOROFORM-d) & ppm 1.30 (d, J = 6.7 Hz, 3 H), 2.31 (s, 3 H), 3.75-3.92 (m, 2 H), 4.24 (d, J = 15.3 Hz, 1 H), 4.82 (dd, J = 13.7, 1.9 Hz, 1 H), 5.36 (d, J = 15.3 Hz, 1 H), 7.17 (s, 1 H), 7.35 (d, J = 7.7 Hz, 1 H), 7.46-7.58 (m, 3 H), 7.62 (s, 1 H), 8.28 (s, 1 H)
64	(300 MHz, DMSO-d <sub>6</sub> ) 8 ppm 2.17 (s, 3 H), 3.73-3.87 (m, 2 H), 4.28-4.41 (m, 2 H), 4.86 (s, 2 H), 7.14 (d, J = 7.8 Hz, 1 H), 7.43 (s, 1 H), 7.56 (t, J = 7.8 Hz, 1 H), 7.74 (d, J = 7.7 Hz, 1 H), 7.80-7.86 (m, 2 H), 8.28 (s, 1 H)
65	(300 MHz, DMSO-d <sub>6</sub> ) 8 ppm 2.16 (s, 3 H), 3.66-3.79 (m, 2 H), 4.18-4.33 (m, 2 H), 4.79 (s, 2 H), 7.14 (d, J = 7.7 Hz, 1 H), 7.42 (s, 1 H), 7.64-7.76 (m, 2 H), 7.81 (d, J = 7.7 Hz, 1 H), 7.85 (s, 1 H), 8.26 (s, 1 H)
67	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.10 (d, J = 6.6 Hz, 3 H), 2.16 (s, 3 H), 3.84 (dd, J = 14.0, 4.1 Hz, 1 H), 3.89 (s, 3 H), 3.91-4.04 (m, 1 H), 4.43 (d, J = 15.0 Hz, 1 H), 4.61 (dd, J = 14.0, 2.2 Hz, 1 H), 4.99 (d, J = 15.0 Hz, 1 H), 7.14 (d, J = 7.7 Hz, 1 H), 7.26 (d, J = 9.2 Hz, 1 H), 7.43 (s, 1 H), 7.57-7.74 (m, 2 H), 7.82 (d, J = 7.8 Hz, 1 H), 7.82 (d, J = 7.8 Hz, 1 H), 7.82 (d, J = 7.8 Hz, 1 H), 7.82 (d, J = 9.2 Hz, 1 H), 7.83 (s, 1 H), 7.57-7.74 (m, 2 H), 7.82 (d, J = 7.8 Hz, 1 H), 7.83 (d, J = 7.8 Hz
68	1 H), 8.27 (d, J = 1.2 Hz, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.13 (d, J = 6.5 Hz, 3 H), 2.16 (s, 3 H), 3.85 (s, 3 H), 3.87-3.95 (m, 1 H), 3.95-4.04 (m, 1 H), 4.48 (d, J = 15.5 Hz, 1 H), 4.61 (dd, J = 13.8, 2.0 Hz, 1 H), 5.08 (d, J = 15.3 Hz, 1 H), 7.15 (d, J = 7.7 Hz, 1 H), 7.18 (s, 1 H), 7.25 (s, 1 H), 7.32 (s, 1 H), 7.43 (s, 1 H), 7.83 (d, J = 7.7 Hz, 1 H), 8.38 (d, J = 1.3 Hz, 1 H)
69	8.28 (d, J = 1.2 Hz, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.16 (s, 3 H), 3.61-3.73 (m, 2 H), 3.89 (s, 3 H), 4.16-4.29 (m, 2 H), 4.71 (s, 2 H), 7.14 (d, J = 7.7 Hz, 1 H), 7.26 (d, J = 8.5 Hz, 1 H), 7.41 (d, J = 7.7 Hz, 1 H), 7.26 (d, J = 8.5 Hz, 1 H), 7.41 (d, J = 7.7 Hz, 1 H), 8.36 (s, 1 H)
70	H), 7.41 (s, 1 H), 7.57-7.67 (m, 2 H), 7.80 (d, $J = 7.7$ Hz, 1 H), 8.26 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 1.15 (d, $J = 6.5$ Hz, 3 H), 2.16 (s, 3 H), 3.93 (dd, $J = 13.8$ , 3.5 Hz, 1 H), 4.05 (br. s., 1 H), 4.57 (d, $J = 15.5$ Hz, 1 H), 4.64 (d, $J = 15.7$ Hz, 1 H), 5.06 (d, $J = 15.5$ Hz, 1 H), 7.13 (d, $J = 7.7$ Hz, 1 H), 7.44 (s, 1 H), 7.51 (t, $J = 9.1$ Hz, 1 H), 7.64-8.01 (m, 3 H), 8.28 (s, 1 H)
71	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.16 (d, J = 6.3 Hz, 3 H), 2.16 (s, 3 H), 3.92-4.11 (m, 2 H), 4.58 (d, J = 16.1 Hz, 1 H), 4.62-4.71 (m, 1 H), 5.10 (d, J = 15.9 Hz, 1 H), 7.14 (d, J = 7.8 Hz, 1 H), 7.44 (s, 1 H), 7.70-7.89 (m, 4 H), 8.29 (s, 1 H)
72	(300 MHz, DMSO-d <sub>6</sub> ) 8 ppm 1.13 (d, J = 6.5 Hz, 3 H), 2.35 (s, 3 H), 2.40 (s, 3 H), 3.82-4.02 (m, 2 H), 4.54 (d, J = 15.7 Hz, 1 H), 4.64 (d, J = 12.2 Hz, 1 H), 5.09 (d, J = 15.7 Hz, 1 H), 7.25 (d, J = 7.8 Hz, 1 H), 7.48 (d, J = 7.8 Hz, 1 H), 7.54-7.70 (m, 2 H), 7.88 (s, 1 H), 8.11 (d, J = 7.7 Hz, 1 H), 9.52 (s, 1 H)
74	(360 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.15 (s, 3 H), 3.64-3.74 (m, 2 H), 4.21-4.30 (m, 2 H), 4.77 (s, 2 H), 7.14 (d, J = 7.7 Hz, 1 H), 7.31 (d, J = 8.1 Hz, 1 H), 7.34-7.44 (m, 3 H), 7.47-7.55 (m, 1 H), 7.81 (d, J = 7.7 Hz, 1 H), 8.26 (d, J = 1.5 Hz, 1 H)
75	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.16 (s, 3 H), 2.34 (s, 3 H), 3.64 (t, J = 5.7 Hz, 2 H), 4.28 (t, J = 5.7 Hz, 2 H), 4.28 (t, J = 5.7 Hz, 2 H), 4.73 (s, 2 H), 7.10-7.21 (m, 2 H), 7.25 (s, 1 H), 7.36 (d, J = 8.4 Hz, 1 H), 7.42 (s, 1 H), 7.81 (d, J = 7.8 Hz, 1 H), 8.26 (s, 1 H)
76	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.07-1.25 (m, 6 H), 2.35 (s, 3 H), 3.52 (q, J = 7.0 Hz, 2 H), 3.89-4.09 (m, 2 H), 4.51-4.69 (m, 4 H), 5.12 (d, J = 15.5 Hz, 1 H), 7.25 (d, J = 7.7 Hz, 1 H), 7.58-7.72 (m, 3 H), 7.87 (s, 1 H), 8.12 (d, J = 7.7 Hz, 1 H), 9.51 (d, J = 1.4 Hz, 1 H)
77	$\begin{array}{l} (300 \text{ MHz}, \text{DMSO-d}_6) \ \delta \text{ ppm 1.07-1.21} \ (m, 6 \ \text{H}), \ 2.27 \ (s, 3 \ \text{H}), \ 2.34 \ (s, 3 \ \text{H}), \\ 3.16\text{-}3.23 \ (m, 1 \ \text{H}), \ 3.57 \ (s, 3 \ \text{H}), \ 3.67\text{-}3.90} \ (m, 5 \ \text{H}), \ 4.41 \ (d, J = 14.8 \ \text{Hz}, 1 \ \text{H}), \\ 4.55\text{-}4.74 \ (m, 1 \ \text{H}), \ 4.98 \ (d, J = 14.8 \ \text{Hz}, 1 \ \text{H}), \ 6.83 \ (s, 1 \ \text{H}), \ 7.11 \ (s, 1 \ \text{H}), \end{array}$
78	7.25 (d, J = 7.7 Hz, 1 H), 7.87 (s, 1 H), 8.11 (d, J = 7.8 Hz, 1 H), 9.51 (s, 1 H) (300 MHz, DMSO- $d_6$ ) $\delta$ ppm 1.13 (d, J = 6.5 Hz, 3 H), 1.28 (d, J = 5.9 Hz, 6 H), 2.34 (s, 3 H), 3.85-3.99 (m, 1 H), 3.99-4.10 (m, 1 H), 4.50 (d, J = 15.4 Hz, 1 H), 4.61 (dd, J = 13.8, 1.7 Hz, 1 H), 4.76 (dt, J = 12.0, 6.0 Hz, 1 H), 5.06 (d, J = 15.4 Hz, 1 H), 7.15 (s, 1 H), 7.19-7.28 (m, 2 H), 7.29 (s, 1 H), 7.87 (s, 1 H), 8.11 (d, J = 7.7 Hz, 1 H), 9.51 (s, 1 H)

	<sup>1</sup> H NMR results
Co. No.	<sup>1</sup> H NMR result
79	(300 MHz, DMSO- $d_6$ ) $\delta$ ppm 1.11 (d, J = 6.6 Hz, 3 H), 2.16 (s, 3 H), 3.87-4.05 (m, 2 H), 4.50 (d, J = 15.4 Hz, 1 H), 4.60 (d, J = 11.7 Hz, 1 H), 5.04 (d, J = 15.4 Hz, 1 H), 7.13 (d, J = 7.7 Hz, 1 H), 7.30 (d, J = 7.0 Hz, 1 H), 7.36-7.46 (m, 3 H), 7.46-7.55 (m, 1 H), 7.82 (d, J = 7.7 Hz, 1 H), 8.27 (s, 1 H)
83	$ \begin{array}{l} (300 \text{ MHz}, \text{DMSO-}4_6) \ \delta \text{ ppm 1.18} \ (d,  J=6.5 \text{ Hz},  3 \text{ H}),  2.22 \ (s,  3 \text{ H}),  2.24 \ (s,  3 \text{ H}),  2.35 \ (s,  3 \text{ H}),  3.96 \ (dd,  J=14.0,  4.1 \text{ Hz},  1 \text{ H}),  4.01 \cdot 4.13 \ (m,  1 \text{ H}),  4.57 \ (d,  J=15.3 \text{ Hz},  1 \text{ H}),  4.68 \ (dd,  J=13.9,  2.2 \text{ Hz},  1 \text{ H}),  5.13 \ (d,  J=15.3 \text{ Hz},  1 \text{ H}),  6.13 \ (s,  1 \text{ H}),  7.20 \ (d,  J=7.7 \text{ Hz},  1 \text{ H}),  7.40 \cdot 7.59 \ (m,  5 \text{ H}),  7.89 \ (d,  J=7.7 \text{ Hz},  1 \text{ H}),  \end{array} $
84	8.33 (d, J = 1.2 Hz, 1 H) (300 MHz, DMSO- $d_6$ ) $\delta$ ppm 1.15 (d, J = 6.5 Hz, 3 H), 1.50-1.76 (m, 6 H), 2.22 (s, 3 H), 3.12-3.26 (m, 4 H), 3.82-3.92 (m, 1 H), 3.92-4.01 (m, 1 H), 4.36 (d, J = 15.0 Hz, 1 H), 4.68 (dd, J = 13.8, 2.1 Hz, 1 H), 5.06 (d, J = 14.8 Hz, 1 H), 6.80 (d, J = 7.4 Hz, 1 H), 6.91 (dd, J = 8.0, 2.2 Hz, 1 H), 6.97 (s, 1 H),
85	7.13-7.30 (m, 2 H), 7.48 (s, 1 H), 7.88 (d, $J$ = 7.7 Hz, 1 H), 8.33 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 1.10 (d, $J$ = 6.6 Hz, 3 H), 1.48-1.60 (m, 2 H), 1.60-1.74 (m, 4 H), 2.16 (s, 3 H), 2.85-3.02 (m, 4 H), 3.77-3.99 (m, 2 H), 4.36 (d, $J$ = 15.3 Hz, 1 H), 4.61 (dd, $J$ = 13.7, 1.8 Hz, 1 H), 5.01 (d, $J$ = 15.1 Hz, 1 H), 7.02 (dd, $J$ = 8.2, 1.9 Hz, 1 H), 7.09-7.18 (m, 2 H), 7.37 (d, $J$ = 8.1 Hz, 1 H), 7.01 (d. 1 = 8.2, 1.9 Hz, 1 H), 7.02 (d. 1 = 8.2, 1.9 Hz, 1 H), 7.02 (d. 1 = 8.2, 1.9 Hz, 1 H), 7.03 (d. 1 = 8.2, 1.9 Hz, 1 H), 7.04 (d. 1 = 8.2, 1.9 Hz, 1 H), 7.05 (d. 1 = 8.2, 1.9 Hz, 1 H), 7.07 (d. 1 = 8.2, 1.9 Hz, 1 H), 7.08 (d. 1 = 8.2, 1.9 Hz, 1 H), 7.09 (d. 1 = 8.2, 1.9 Hz, 1 Hz), 7
88	7.40-7.46 (m, 1 H), 7.83 (d, J = 7.7 Hz, 1 H), 8.28 (d, J = 1.1 Hz, 1 H) (300 MHz, DMSO- $d_6$ ) $\delta$ ppm 2.16 (s, 3 H), 3.60-3.80 (m, 2 H), 4.11-4.38 (m, 2 H), 4.79 (s, 2 H), 7.32-7.72 (m, 10 H), 7.92 (s, 1 H), 8.36 (s, 1 H)
89	(360 MHz, DMSO- $d_6$ ) $\delta$ ppm 2.08 (s, 3 H), 3.66 (t, $J = 5.7$ Hz, 2 H), 4.21 (t, $J = 5.7$ Hz, 2 H), 4.73 (s, 2 H), 6.77 (br. s., 1 H), 7.13 (d, $J = 7.7$ Hz, 1 H),
91	7.26-7.43 (m, 5 H), 7.62 (br. s., 1 H), 7.75 (d, J = 7.3 Hz, 1 H) (300 MHz, DMSO-d <sub>0</sub> ) δ ppm 2.04-2.17 (m, 3 H), 3.76-3.95 (m, 2 H), 4.18-4.34 (m, 2 H), 4.87 (s, 2 H), 7.07 (d, J = 7.7 Hz, 1 H), 7.36 (t, J = 1.3 Hz, 1 H), 7.43-7.52 (m, 2 H), 7.60-7.70 (m, 3 H), 7.74 (d, J = 7.8 Hz, 1 H), 8.21 (d, J = 1.1 Hz, 1 H)
92	$(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 2.18 \text{ (s, 3 H), 3.13 (t, J} = 7.2 \text{ Hz, 2 H),} \\ 3.61\text{-}3.97 \text{ (m, 4 H), 4.15\text{-}4.33 (m, 2 H), 7.03 (d, J} = 7.7 \text{ Hz, 1 H), 7.43 (br. s., 1 H),} \\$
93	7.80 (d, J = 7.7 Hz, 1 H), 7.98 (s, 1 H), 8.08 (s, 2 H), 8.29 (br. s., 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.99-2.21 (m, 5 H), 3.56-3.77 (m, 2 H), 4.13 (q, J = 5.9 Hz, 4 H), 4.19-4.37 (m, 2 H), 4.68 (s, 2 H), 6.84-7.02 (m, 3 H), 7.12 (d, J = 7.7 Hz, 1 H), 7.42 (br. s., 1 H), 7.80 (d, J = 7.7 Hz, 1 H), 8.26 (br. s., 1 H)
94	(300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 2.16 (s, 3 H), 2.47 (s, 3 H), 3.59-3.80 (m, 2 H), 4.27 (t, J = 5.8 Hz, 2 H), 4.87 (s, 2 H), 7.18 (d, J = 7.8 Hz, 1 H), 7.27-7.51 (m, 3
97	H), 7.68 (s, 1 H), 7.75-7.93 (m, 4 H), 8.26 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.54 (br. s., 2 H), 1.59-1.72 (m, 4 H), 2.16 (s, 3 H), 2.19 (s, 3 H), 2.76 (br. s., 4 H), 3.49-3.60 (m, 2 H), 4.18-4.30 (m, 2 H), 4.70 (s, 2 H), 6.93 (d, J = 7.4 Hz, 1 H), 7.00 (d, J = 7.7 Hz, 1 H), 7.08-7.20 (m, 2 H), 4.70 (s, 2 H), 6.93 (d, J = 7.4 Hz, 1 H), 7.00 (d, J = 7.7 Hz, 1 H), 7.08-7.20 (m, 2 Hz), 4.70 (m,
98	H), 7.42 (s, 1 H), 7.81 (d, J = 7.8 Hz, 1 H), 8.26 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.12-2.17 (m, 3 H), 2.18 (s, 3 H), 3.66-3.81 (m, 2 H), 4.23-4.35 (m, 2 H), 4.85 (s, 2 H), 4.91 (q, J = 8.8 Hz, 2 H), 7.06-7.14 (m, 2 H), 7.41 (t, J = 1.2 Hz, 1 H), 7.80 (d, J = 7.8 Hz, 1 H), 8.26 (d, J = 1.1 Hz, 1 Hz,
99	H), 8.30 (d, J = 5.6 Hz, 1 H) (300 MHz, DMSO- $d_6$ ) $\delta$ ppm 1.42-1.59 (m, 2 H), 1.59-1.74 (m, 4 H), 2.16 (s, 3 H), 2.21 (s, 3 H), 2.69-2.88 (m, 4 H), 3.46-3.72 (m, 2 H), 4.12-4.32 (m, 2 H), 4.64 (s, 2 H), 6.89 (d, J = 7.6 Hz, 1 H), 6.95 (s, 1 H), 7.13 (dd, J = 7.8, 4.5 Hz, 2 H), 7.41 (s, 1 H), 7.80 (d, J = 7.7 Hz, 1 H), 8.25 (s, 1 H)
103	211), 7.41 (s, 1 H), 7.40 (d, 3 = 7.7 Hz, 1 H), 6.25 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.86-2.02 (m, 2 H), 2.16 (s, 3 H), 3.62 (dt, J = 13.6, 6.9 Hz, 1 H), 3.71-3.85 (m, 3 H), 3.93 (dd, J = 11.3, 7.4 Hz, 1 H), 4.18-4.30 (m, 3 H), 4.34 (dd, J = 11.3, 2.1 Hz, 1 H), 6.78-6.91 (m, 4 H), 7.08 (d, J = 7.7 Hz, 1 H), 7.41 (s, 1 H), 7.79 (d, J = 7.7 Hz, 1 H), 8.25 (d, J = 1.1 Hz, 1 H)
104	(300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 2.18 (s, 3 H), 2.89-3.04 (m, 2 H), 3.54-3.63 (m, 2 H), 3.68 (t, J = 7.6 Hz, 2 H), 3.92-4.05 (m, 2 H), 6.62 (dd, J = 9.1, 1.3 Hz, 1 H), 6.95 (dd, J = 6.9, 1.2 Hz, 1 H), 7.04-7.13 (m, 1 H), 7.18 (td, J = 7.5, 1.1 Hz, 1 H), 7.23 (s, 1 H), 7.35 (d, J = 8.0 Hz, 1 H), 7.55 (dd, J = 9.1, 6.9 Hz, 1 H), 7.70 (d, J = 7.7 Hz, 1 H), 7.91 (d, J = 1.2 Hz, 1 H), 11.62 (s, 1 H)
105	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.04-2.15 (m, 1 H), 2.16 (s, 3 H), 2.40-2.48 (m, 1 H), 2.97-3.28 (m, 3 H), 3.54 (ddd, J = 13.0, 8.5, 4.1 Hz, 1 H), 4.06-4.23 (m, 1 H), 4.24-4.37 (m, 1 H), 6.18 (t, J = 8.4 Hz, 1 H), 7.18 (d, J = 7.8 Hz, 1 H), 7.43 (s, 1 H), 7.44-7.52 (m, 1 H), 7.63 (t, J = 8.8 Hz, 2 H), 7.84 (d, J = 7.7 Hz, 1 H), 8.27 (d, J = 1.1 Hz, 1 H)
108	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.19 (s, 3 H), 3.72 (t, J = 5.8 Hz, 2 H), 4.23 (t, J = 5.8 Hz, 2 H), 4.80 (s, 2 H), 6.87 (s, 1 H), 7.08-7.17 (m, 2 H), 7.31-7.42 (m, 2 H), 7.42-7.53 (m, 3 H), 7.57-7.71 (m, 4 H), 7.74 (d, J = 7.6 Hz, 1 H)
109	2 H), 7.42-7.35 (III, 3 H), 7.37-7.71 (III, 4 H), 7.74 (II, 3 = 7.0 Hz, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 3.65-3.80 (m, 2 H), 4.20-4.37 (m, 2 H), 4.80 (s, 2 H), 7.08 (s, 1 H), 7.17 (d, J = 7.7 Hz, 1 H), 7.31-7.43 (m, 2 H), 7.48 (t, J = 7.4 Hz, 3 H), 7.55-7.76 (m, 5 H), 7.88 (d, J = 7.7 Hz, 1 H), 8.33 (s, 1 H)
111	$(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.20 \text{ (d, J} = 6.6 \text{ Hz}, 3 \text{ H)}, 1.73 \text{ (d, J} = 7.1 \text{ Hz}, 3 \text{ H)}, \\ 2.33 \text{ (s, 3 H)}, 3.71 \text{ (dd, J} = 14.0, 3.7 \text{ Hz}, 1 \text{ H)}, 4.06-4.19 \text{ (m, 1 H)}, \\ 4.53-4.69 \text{ (m, 1 H)}, 5.74 \text{ (q, J} = 6.9 \text{ Hz}, 1 \text{ H)}, 7.19 \text{ (d, J} = 7.7 \text{ Hz}, 1 \text{ H)}, 7.81-7.89 \text{ (m, 1 H)}, \\ 8.02-8.17 \text{ (m, 4 H)}, 9.40-9.51 \text{ (m, 1 H)}$

	<sup>1</sup> H NMR results
Co. No.	<sup>1</sup> H NMR result
112	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 0.82 (d, J = 6.5 Hz, 3 H), 1.79 (d, J = 7.0 Hz, 3 H),
	2.33 (s, 3 H), 3.95 (dd, J = 14.0, 3.8 Hz, 1 H), 4.27-4.48 (m, 1 H), 4.65 (d, J = 12.8 Hz, 1 H), 5.64 (g, J = 6.8 Hz, 1 H), 7.21 (d, J = 7.7 Hz, 1 H), 7.85 (s, 1
440	H), 7.99-8.28 (m, 4 H), 9.47 (s, 1 H)
119	$(600 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 2.37 \text{ (s, 3 H)}, 3.64-3.68 \text{ (m, 2 H)}, 3.75 \text{ (3 H)}, 4.27-4.30 \text{ (m, 2 H)}, 4.69 \text{ (s, 2 H)}, 6.87 \text{ (ddd, J} = 8.2, 2.6, 1.0 Hz, 1 H),$
	6.90-6.92 (m, 2 H), 7.25 (d, J = 7.8 Hz, 1 H), 7.28 (t, J = 7.8 Hz, 1 H), 8.13 (d, J = 7.9 Hz,
120	1 H), 9.40 (d, J = 0.6 Hz, 1 H) (600 MHz, DMSO-d <sub>6</sub> ) δ ppm 2.37 (s, 3 H), 3.71-3.74 (m, 2 H),
120	4.30-4.32 (m, 2 H), 4.82 (s, 2 H), 7.26 (d, J = 7.9 Hz, 1 H), 7.60-7.63 (m, 1 H), 7.67 (d,
121	J = 7.9 Hz, 2 H), 7.73 (s, 1 H), 8.13 (d, J = 7.9 Hz, 1 H), 9.40 (s, 1 H) (400 MHz, CHLOROFORM-d) δ ppm 2.00 (quin, J = 6.4 Hz, 2 H), 2.29 (s, 3
121	H), 3.34 (t, J = 6.1 Hz, 2 H), 4.33 (br. s., 2 H), 4.70 (s, 2 H), 6.79 (d, J = 7.5 Hz, 1
	H), 7.11 (br. s., 1 H), 7.21 (d, J = 8.0 Hz, 1 H), 7.41 (d, J = 7.5 Hz, 1 H),
122	7.43-7.51 (m, 2 H), 8.16 (br. s., 1 H) (360 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 1.90 (quin, J = 6.2 Hz, 2 H), 2.07 (s, 3 H),
	3.36-3.41  (m, 2 H), 4.17  (br. s., 2 H), 4.70  (s, 2 H), 6.70  (d, J = 7.7  Hz, 1  H), 6.75  (s, 2.1)
176	1 H), 7.38 (dd, J = 8.4, 1.8 Hz, 1 H), 7.60 (s, 1 H), 7.62-7.74 (m, 3 H) (360 MHz, CHLOROFORM-d) δ ppm 1.15 (d, J = 6.6 Hz, 3 H), 2.28 (d, J = 1.1 Hz,
	3 H), 2.94-3.07 (m, 2 H), 3.11-3.21 (m, 1 H), 3.36-3.51 (m, 2 H),
	4.28-4.38 (m, 1 H), 4.60 (dd, J = 13.9, 2.2 Hz, 1 H), 7.13 (t, J = 1.5 Hz, 1 H), 7.20-7.26 (m, 4 H), 7.27-7.37 (m, 2 H), 7.45 (d, J = 7.7 Hz, 1 H), 8.22 (d, J = 1.5 Hz,
	1 H)
177	(360 MHz, CHLOROFORM-d) & ppm 1.19 (d, J = 7.0 Hz, 3 H), 2.28 (s, 3 H), 3.49 (dd, J = 14.1, 4.2 Hz, 1 H), 3.77 (s, 3 H), 3.82-3.95 (m, 1 H), 4.28 (d,
	J = 15.0 Hz, 1 H), 4.72 (dd, J = 14.3, 2.2 Hz, 1 H), 5.48 (d, J = 15.0 Hz, 1 H),
	7.12 (d, J = 8.8 Hz, 2 H), 7.18-7.25 (m, 1 H), 7.33 (d, J = 7.7 Hz, 1 H), 7.43-7.58 (m, 3 H), 8.22 (s, 1 H)
179	$(360 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.62 \text{ (d, J} = 6.6 \text{ Hz}, 3 \text{ H)}, 2.65 \text{ (d, J} = 1.1 \text{ Hz}, 3 \text{ H)},$
	4.03-4.16 (m, 1 H), 4.48-4.60 (m, 1 H), 5.06 (d, J = 14.6 Hz, 1 H), 5.20 (dd, J = 14.3, 2.2 Hz, 1 H), 5.80 (dd, J = 15.0, 0.7 Hz, 1 H), 7.51 (dd, J = 8.6, 2.0 Hz, 1
	H), 7.72 (d, J = 7.7 Hz, 1 H), 7.82 (t, J = 1.3 Hz, 1 H), 7.95 (d, J = 1.8 Hz, 1 H),
	8.03-8.09 (m, 1 H), 8.18 (d, J = 8.8 Hz, 1 H), 8.24 (d, J = 7.7 Hz, 1 H), 8.77 (d, J = 1.5 Hz, 1 H), 11.03 (br. s., 1 H)
124	$(360 \text{ MHz}, \text{CHLOROFORM-d}) \delta \text{ ppm } 1.23 \text{ (d, J} = 6.6 \text{ Hz, 3 H)}, 2.24-2.36 \text{ (m,}$
	3 H), 3.71 (dd, J = 14.1, 4.2 Hz, 1 H), 3.77-3.90 (m, 1 H), 4.15 (d, J = 15.0 Hz, 1 H), 4.75 (dd, J = 14.1, 2.4 Hz, 1 H), 5.38 (d, J = 15.0 Hz, 1 H), 7.15 (s, 1 H),
	7.29-7.44 (m, 6 H), 7.48 (d, J = 7.7 Hz, 1 H), 8.20-8.30 (m, 1 H)
125	(400 MHz, CHLOROFORM-d) δ ppm 1.17 (d, J = 6.9 Hz, 3 H), 2.28 (s, 3 H), 3.50 (dd, J = 14.1, 4.0 Hz, 1 H), 3.81-3.90 (m, 1 H), 4.33 (d, J = 14.9 Hz, 1 H),
	5.30 (dd, J = 14.1, 4.0 Hz, 1 H), 5.81-3.90 (III, 1 H), 4.33 (d, J = 14.9 Hz, 1 H), 4.70 (dd, J = 14.1, 2.0 Hz, 1 H), 5.49 (d, J = 14.9 Hz, 1 H), 7.13 (s, 1 H),
	7.18-7.26 (m, 2 H), 7.34 (d, J = 7.7 Hz, 1 H), 7.46-7.57 (m, 3 H), 8.19-8.29 (m, 1 H), 9.05 (br. s., 1 H)
126	$(300 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 0.80 \text{ (d, J = 6.6 Hz, 3 H), 1.74 (d, J = 7.0 Hz, 3 H),}$
	2.39 (s, 3 H), 4.00 (dd, J = 14.0, 3.7 Hz, 1 H), 4.32 (br. s., 1 H), 4.68 (d, J = 13.6 Hz, 1 H), 5.67 (g, J = 6.8 Hz, 1 H), 7.27 (d, J = 7.8 Hz, 1 H), 7.36 (s, 1 H),
	7.39-7.53 (m, 2 H), 7.92 (s, 1 H), 8.16 (d, J = 7.7 Hz, 1 H), 9.58 (d, J = 1.0 Hz, 1 H)
127	(300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 1.24 (d, J = 6.5 Hz, 3 H), 1.71 (d, J = 7.1 Hz, 3 H),
	2.41 (s, 3 H), 3.76 (dd, J = 14.0, 3.6 Hz, 1 H), 3.95-4.14 (m, 1 H), 4.65 (d, J = 13.6 Hz, 1 H), 5.73 (q, J = 7.0 Hz, 1 H), 7.25 (d, J = 7.7 Hz, 1 H),
120	7.31-7.50 (m, 3 H), 7.94 (s, 1 H), 8.18 (d, J = 7.7 Hz, 1 H), 9.62 (d, J = 1.0 Hz, 1 H)
129	(360 MHz, CHLOROFORM-d) δ ppm 0.75 (d, J = 7.0 Hz, 3 H), 1.73 (d, J = 7.3 Hz, 3 H), 2.28 (s, 3 H), 3.71 (dd, J = 14.1, 3.8 Hz, 1 H), 3.92-4.02 (m, 1 H),
	4.87 (dd, J = 13.9, 1.8 Hz, 1 H), 6.00 (q, J = 7.0 Hz, 1 H), 7.14 (s, 1 H), 7.31 (d, J = 7.7 Hz, 1 H), 7.40 (d, J = 7.7 Hz, 1 H), 7.65 (-1.1 H), 7.65 (-1.1 Hz, 1 Hz), 7.65 (-1.1 Hz), 7.65 (-1.1 Hz, 1 Hz), 7.65 (-1.1 Hz), 7.65 (-1.1 Hz, 1 Hz), 7.65 (-1.1 Hz), 7.65 (-
	J = 7.7 Hz, 1 H), 7.49 (d, J = 7.7 Hz, 1 H), 7.58 (s, 1 H), 7.60 (s, 1 H), 7.65 (s, 1 H), 8.25 (s, 1 H)
130	$(360 \text{ MHz}, \text{CHLOROFORM-d}) \delta \text{ ppm } 1.32 \text{ (d, J} = 6.6 \text{ Hz}, 3 \text{ H)}, 1.72 \text{ (d, J} = 7.3 \text{ Hz}, 1.00 \text{ Hz})$
	3 H), 2.28 (s, 3 H), 3.38 (dd, J = 14.3, 3.7 Hz, 1 H), 3.64-3.76 (m, 1 H), 4.77 (dd, J = 13.9, 1.5 Hz, 1 H), 6.06 (q, J = 7.2 Hz, 1 H), 7.15 (s, 1 H), 7.30 (d,
	J = 7.7  Hz, 1  H), 7.45-7.52  (m, 2 H), 7.56  (s, 1 H), 7.61  (s, 1 H), 8.23  (s, 1 H)
131	(400 MHz, CHLOROFORM-d) δ ppm 1.27 (d, J = 6.9 Hz, 3 H), 2.29 (d, J = 1.2 Hz, 3 H), 3.76-3.87 (m, 2 H), 4.12 (d, J = 15.3 Hz, 1 H), 4.78 (dd, J = 13.9, 2.2 Hz,
	1 H), 5.27 (d, J = 15.7 Hz, 1 H), 7.15 (t, J = 1.2 Hz, 1 H), 7.22 (d, J = 2.0 Hz, 2
	H), 7.32 (d, J = 7.7 Hz, 1 H), 7.34 (t, J = 1.8 Hz, 1 H), 7.48 (d, J = 7.7 Hz, 1 H),
132	8.25 (d, J = 1.2 Hz, 1 H) (360 MHz, CHLOROFORM-d) δ ppm 1.23 (d, J = 6.6 Hz, 3 H), 2.29 (d, J = 0.7 Hz,
102	3 H), 3.71 (dd, J = 14.1, 4.2 Hz, 1 H), 3.77-3.87 (m, 4 H), 4.10 (d, J = 14.6 Hz,
	1 H), 4.76 (dd, J = 14.1, 2.4 Hz, 1 H), 5.37 (d, J = 15.0 Hz, 1 H),
	6.81-6.93 (m, 3 H), 7.15 (t, J = 1.1 Hz, 1 H), 7.27-7.35 (m, 2 H), 7.48 (d, J = 7.7 Hz, 1 H), 8.24 (d, J = 1.5 Hz, 1 H)
133	$(360 \text{ MHz}, \text{CHLOROFORM-d}) \delta \text{ ppm } 1.24 \text{ (d, J} = 7.0 \text{ Hz}, 3 \text{ H)}, 2.29 \text{ (s, 3 H)},$
	3.75 (dd, J = 14.1, 4.2 Hz, 1 H), 3.79-3.89 (m, 1 H), 4.16 (d, J = 15.0 Hz, 1 H), 4.78 (dd, J = 14.3, 2.6 Hz, 1 H), 5.24 (d, J = 14.6 Hz, 1 H), 7.02-7.11 (m, 1 H),
	7.11-7.22 (m, 3 H), 7.32 (d, J = 7.7 Hz, 1 H), 7.48 (d, J = 7.7 Hz, 1 H),
	8.21-8.28 (m, 1 H)

	<sup>1</sup> H NMR results
Co. No.	<sup>1</sup> H NMR result
135	(300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 0.12-0.29 (m, 2 H), 0.39-0.55 (m, 2 H), 0.63 (d, J = 6.5 Hz, 3 H), 0.88-1.07 (m, 1 H), 1.70 (d, J = 7.3 Hz, 3 H), 2.33 (s, 3 H), 2.62 (d, J = 6.9 Hz, 2 H), 3.90 (dd, J = 14.0, 3.4 Hz, 1 H), 4.25 (br. s., 1 H), 4.61 (d, J = 13.6 Hz, 1 H), 5.60-5.92 (m, 1 H), 7.23 (d, J = 7.7 Hz, 1 H), 7.59 (s, 2 H),
136	7.66 (s, 1 H), 7.85 (s, 1 H), 8.09 (d, J = 7.8 Hz, 1 H), 9.49 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) & ppm 0.09-0.32 (m, 2 H), 0.35-0.54 (m, 2 H), 0.89-1.08 (m, 1 H), 1.19 (d, J = 6.5 Hz, 3 H), 1.68 (d, J = 6.9 Hz, 3 H), 2.34 (s, 3 H), 2.62 (d, J = 6.7 Hz, 2 H), 3.62 (dd, J = 14.0, 3.1 Hz, 1 H), 3.93 (br. s., 1 H), 4.60 (d, J = 13.6 Hz, 1 H), 5.69-5.82 (m, 1 H), 7.21 (d, J = 7.7 Hz, 1 H), 7.57 (br. s., 2
137	H), 7.62 (br. s., 1 H), 7.86 (s, 1 H), 8.11 (d, $J = 7.8$ Hz, 1 H), 9.49 (s, 1 H) (360 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 1.05 (d, $J = 6.6$ Hz, 3 H), 2.15 (s, 3 H), 3.64 (dd, $J = 14.3$ , 4.0 Hz, 1 H), 3.78 (s, 3 H), 3.90-4.03 (m, 1 H), 4.51 (d, $J = 15.0$ Hz, 1 H), 4.59 (dd, $J = 14.1$ , 1.6 Hz, 1 H), 5.10 (d, $J = 15.0$ Hz, 1 H), 7.17 (d, $J = 7.3$ Hz, 2 H), 7.41 (s, 1 H), 7.47 (d, $J = 8.8$ Hz, 1 H), 7.55 (s, 1 H), 7.69 (d, $J = 1.8$ Hz, 1 H), 7.82 (d, $J = 7.7$ Hz, 1 H), 8.25 (s, 1 H)
138	$\begin{array}{l} 11, 12, 13, 13, 13, 13, 13, 13, 14, 14, 14, 14, 14, 14, 14, 14, 14, 14$
139	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.17 (d, J = 6.6 Hz, 3 H), 2.38 (s, 3 H), 3.80 (dd, J = 14.3, 4.1 Hz, 1 H), 3.95-4.05 (m, 1 H), 4.61 (d, J = 14.7 Hz, 1 H), 4.66-4.77 (m, 1 H), 5.20 (d, J = 14.6 Hz, 1 H), 7.27 (d, J = 7.7 Hz, 1 H), 7.33-7.49 (m, 2 H), 7.58-7.69 (m, 1 H), 7.89 (s, 1 H), 8.13 (d, J = 7.8 Hz, 1 H), 9.50 (s, 1 H)
140	(300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 1.19 (d, J = 6.5 Hz, 3 H), 2.38 (s, 3 H), 3.93-4.20 (m, 2 H), 4.52-4.78 (m, 2 H), 5.17 (d, J = 15.7 Hz, 1 H), 7.28 (d, J = 7.7 Hz, 1 H), 7.80 (d, J = 8.0 Hz, 2 H), 7.90 (d, J = 5.8 Hz, 2 H), 8.15 (d, J = 7.7 Hz, 1 H), 9.56 (s, 1 H)
141	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 0.55 (d, J = 6.2 Hz, 3 H), 0.91 (t, J = 6.9 Hz, 3 H), 2.03-2.42 (m, 5 H), 3.76-4.01 (m, 4 H), 4.24 (br. s., 1 H), 4.61 (d, J = 13.9 Hz, 1 H), 5.53 (t, J = 7.5 Hz, 1 H), 7.17-7.30 (m, 2 H), 7.33 (br. s., 1 H), 7.36 (br. s., 1 H), 7.83 (s, 1 H), 8.08 (d, J = 7.7 Hz, 1 H), 9.43 (s, 1 H)
142	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 0.94 (t, J = 6.8 Hz, 3 H), 1.16 (d, J = 6.2 Hz, 3 H), 2.03-2.29 (m, 2 H), 2.33 (s, 3 H), 3.52 (dd, J = 13.9, 2.7 Hz, 1 H), 3.87 (s, 3 H), 3.92 (br. s., 1 H), 4.58 (d, J = 13.6 Hz, 1 H), 5.43 (t, J = 7.6 Hz, 1 H), 7.08-7.26 (m, 2 H), 7.30 (br. s., 1 H), 7.35 (s, 1 H), 7.83 (s, 1 H), 8.09 (d, J = 7.6 Hz, 1 H), 9.41 (s, 1 H)
143	3.7.1 (8, 1 II) (300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 0.66 (d, J = 6.6 Hz, 3 H), 1.35 (t, J = 6.9 Hz, 3 H), 1.69 (d, J = 7.1 Hz, 3 H), 2.34 (s, 3 H), 3.90 (dd, J = 14.1, 3.2 Hz, 1 H), 4.14 (q, J = 6.9 Hz, 2 H), 4.29 (br. s., 1 H), 4.62 (d, J = 13.7 Hz, 1 H), 5.72 (q, J = 7.0 Hz, 1 H), 7.21 (br. s., 1 H), 7.23 (d, J = 7.7 Hz, 1 H), 7.28-7.44 (m, 2 H), 7.85 (s, 1 H), 8.10 (d, J = 7.8 Hz, 1 H), 9.49 (s, 1 H)
144	(300 MHz, DMSO-d <sub>c</sub> ) \( \delta\) pm 1.19 (d, J = 6.5 Hz, 3 H), 1.35 (t, J = 6.9 Hz, 3 H), 1.67 (d, J = 7.1 Hz, 3 H), 2.35 (s, 3 H), 3.62 (dd, J = 14.0, 3.4 Hz, 1 H), 3.98 (br. s., 1 H), 4.14 (q, J = 6.9 Hz, 2 H), 4.60 (d, J = 13.6 Hz, 1 H), 5.65-5.75 (m, 1 H), 7.18 (br. s., 1 H), 7.21 (d, J = 7.7 Hz, 1 H), 7.26 (s, 1 H), 7.31 (s, 1 H), 7.87 (s, 1 H), 8.11 (d, J = 7.7 Hz, 1 H), 9.51 (s, 1 H)
145	(300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 1.21 (d, J = 6.3 Hz, 3 H), 2.39 (s, 3 H), 3.96-4.18 (m, 2 H), 4.51-4.72 (m, 2 H), 5.16 (d, J = 15.5 Hz, 1 H), 7.30 (d, J = 7.7 Hz, 1 H), 7.91 (s, 1 H), 7.86 (s, 1 H), 8.00 (d, J = 9.5 Hz, 2 H), 8.15 (d, J = 7.7 Hz, 1 H), 9.53 (s, 1 H)
146	$(300 \text{ MHz}, \text{DMSO-d}_6)  \delta  \text{ppm}  0.68  (d,  \text{J} = 6.6  \text{Hz},  3  \text{H}),  1.74  (d,  \text{J} = 7.0  \text{Hz},  3  \text{H}), \\ 2.38  (s,  3  \text{H}),  3.96  (dd,  \text{J} = 14.0,  3.6  \text{Hz},  1  \text{H}),  4.22-4.38  (m,  1  \text{H}),  4.67  (d, \\ \text{J} = 13.3  \text{Hz},  1  \text{H}),  5.81  (q,  \text{J} = 7.0  \text{Hz},  1  \text{H}),  7.05-7.05  (m,  1  \text{H}),  7.28  (d,  \text{J} = 7.8  \text{Hz}, \\ 1  \text{H}),  7.42  (d,  \text{J} = 7.6  \text{Hz},  1  \text{H}),  7.49  (s,  1  \text{H}),  7.54-7.65  (m,  2  \text{H}),  7.89  (s,  1  \text{H}), \\ 8.13  (d,  \text{J} = 7.7  \text{Hz},  1  \text{H}),  9.48  (s,  1  \text{H})$
147	$(300 \text{ MHz, DMSO-d}_6) \delta \text{ ppm 1.19 (d, J = 6.6 \text{ Hz, 3 H), 1.67 (d, J = 7.3 \text{ Hz, 3 H),}} \\ 2.34 (s, 3 \text{ H), 3.64 (dd, J = 13.9, 4.0 \text{ Hz, 1 H), 3.85-4.00 (m, 1 H), 4.60 (d,} \\ J = 12.9 \text{ Hz, 1 H), 5.74 (q, J = 7.1 \text{ Hz, 1 H), 7.22 (d, J = 7.8 \text{ Hz, 1 H), 7.34 (d,}} \\ J = 7.6 \text{ Hz, 1 H), 7.43 (s, 1 H), 7.45-7.60 (m, 2 H), 7.87 (s, 1 H), 8.11 (d, J = 7.7 \text{ Hz,} 1 \text{ H), 9.46-9.54 (m, 1 H)} $
148	(300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 0.68 (d, J = 6.2 Hz, 3 H), 1.70 (d, J = 6.9 Hz, 3 H), 2.35 (s, 3 H), 3.80-3.99 (m, 4 H), 4.29 (br. s., 1 H), 4.62 (d, J = 13.6 Hz, 1 H), 5.72 (q, J = 7.0 Hz, 1 H), 7.06-7.29 (m, 2 H), 7.33 (br. s., 1 H), 7.35 (br. s., 1 H), 7.87 (s, 1 H), 8.11 (d, J = 7.7 Hz, 1 H), 9.53 (s, 1 H)
149	(300 MHz, DMSO-d <sub>6</sub> ) \( \delta \) pm 1.19 (d, J = 6.5 Hz, 3 H), 1.67 (d, J = 7.1 Hz, 3 H), 2.34 (s, 3 H), 3.63 (dd, J = 14.4, 3.3 Hz, 1 H), 3.86 (s, 3 H), 3.91-4.03 (m, 1 H), 4.61 (d, J = 13.7 Hz, 1 H), 5.72 (d, J = 7.2 Hz, 1 H), 7.15-7.25 (m, 2 H), 7.33 (s, 1 H), 7.27 (s, 1 H), 7.86 (s, 1 H), 8.10 (d, J = 7.7 Hz, 1 H), 9.48 (s, 1 H)
155	1 H), 7.27 (s, 1 H), 7.80 (s, 1 H), 8.10 (d, J = 7.7 Hz, 1 H), 9.48 (s, 1 H) (360 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.15 (d, J = 6.2 Hz, 3 H), 2.38 (s, 3 H), 3.93-4.03 (m, 1 H), 4.06 (ddd, J = 6.3, 4.1, 2.0 Hz, 1 H), 4.60 (d, J = 15.7 Hz, 1 H), 4.66 (dd, J = 13.7, 2.0 Hz, 1 H), 5.19 (d, J = 15.7 Hz, 1 H), 7.26 (d, J = 8.1 Hz, 1 H), 8.06 (s, 1 H), 8.10-8.19 (m, 3 H), 9.42 (s, 1 H)

	<sup>1</sup> H NMR results
Co. No.	<sup>1</sup> H NMR result
160	$\begin{array}{l} (300 \text{ MHz}, \text{DMSO-d}_6)  \delta  \text{ppm}   1.28  (\text{d},  \text{J} = 5.9  \text{Hz},  6  \text{H}),  2.16  (\text{s},  3  \text{H}), \\ 3.65\text{-}3.77  (\text{m},  2  \text{H}),  4.19\text{-}4.31  (\text{m},  2  \text{H}),  4.70\text{-}4.81  (\text{m},  3  \text{H}),  7.10\text{-}7.21  (\text{m},  3  \text{H}), \\ 7.24  (\text{s},  1  \text{H}),  7.42  (\text{t},  \text{J} = 1.2  \text{Hz},  1  \text{H}),  7.81  (\text{d},  \text{J} = 7.7  \text{Hz},  1  \text{H}),  8.26  (\text{dd},  \text{J} = 1.2,  0.3  \text{Hz}, \\ 1.21  \text{Hz},  1.21 $
163	1 H) (300 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 1.14 (d, J = 6.5 Hz, 3 H), 1.28 (d, J = 5.9 Hz, 6 H), 2.34 (s, 3 H), 3.90-4.11 (m, 2 H), 4.44-4.55 (m, 1 H), 4.57-4.72 (m, 2 H), 4.99 (d, J = 15.3 Hz, 1 H), 7.14-7.29 (m, 2 H), 7.48 (d, J = 9.5 Hz, 1 H), 7.87 (s, 1 H), 8.10 (d, J = 7.8 Hz, 1 H), 9.45-9.55 (m, 1 H)
165	$\begin{array}{l} (360 \text{ MHz}, \text{DMSO-d}_6)  \delta \text{ ppm } 1.06  (\text{d},  \text{J} = 6.2  \text{Hz},  3  \text{H}),  2.15  (\text{s},  3  \text{H}),  3.84  (\text{dd},  \text{J} = 13.9,  2.9  \text{Hz},  1  \text{H}),  3.93 - 4.10  (\text{m},  1  \text{H}),  4.56 - 4.80  (\text{m},  2  \text{H}),  5.19  (\text{d},  \text{J} = 15.4  \text{Hz},  1  \text{H}),  7.19  (\text{dd},  \text{J} = 7.7,  1.8  \text{Hz},  1  \text{H}),  7.39 - 7.50  (\text{m},  2  \text{H}),  7.83  (\text{dd},  \text{J} = 7.9,  \text{Hz}),  1.8  \text{Hz},  1  \text{Hz},  1.8  \text{Hz}$
168	1.6 Hz, 1 H), 7.94 (s, 1 H), 8.00-8.12 (m, 2 H), 8.27 (s, 1 H) (300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.62 (d, J = 7.1 Hz, 3 H), 2.15 (s, 3 H), 3.35-3.51 (m, 1 H), 3.61-3.84 (m, 1 H), 4.06-4.30 (m, 2 H), 5.93 (q, J = 7.0 Hz, 1 H), 7.11 (d, J = 7.8 Hz, 1 H), 7.33-7.55 (m, 2 H), 7.73-7.95 (m, 3 H), 8.18-8.33 (m, 1 H)
169	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.33 (d, J = 7.0 Hz, 3 H), 2.15 (s, 3 H), 3.40-3.78 (m, 4 H), 3.78-3.91 (m, 1 H), 4.10-4.18 (m, 2 H), 6.98 (d, J = 7.7 Hz, 1 H), 7.38 (s, 1 H), 7.76 (d, J = 7.8 Hz, 1 H), 7.94 (s, 1 H), 8.06 (s, 2 H), 8.23 (s, 1 H)
171	$\begin{array}{l} (300 \text{ MHz}, \text{DMSO-d}_6)  \delta  \text{ppm}  1.14  (\text{d},  \text{J} = 6.6  \text{Hz},  3  \text{H}),  1.42\text{-}1.77  (\text{m},  6  \text{H}), \\ 2.35  (\text{s},  3  \text{H}),  2.83  (\text{t},  \text{J} = 5.2  \text{Hz},  4  \text{H}),  3.90\text{-}4.09  (\text{m},  2  \text{H}),  4.50  (\text{d},  \text{J} = 15.9  \text{Hz}, \\ 1  \text{H}),  4.57\text{-}4.72  (\text{m},  1  \text{H}),  5.10  (\text{d},  \text{J} = 15.9  \text{Hz},  1  \text{H}),  7.15\text{-}7.39  (\text{m},  2  \text{H}), \\ 7.49  (\text{s},  1  \text{H}),  7.64  (\text{d},  \text{J} = 8.2  \text{Hz},  1  \text{H}),  7.81\text{-}7.96  (\text{m},  1  \text{H}),  8.13  (\text{d},  \text{J} = 7.7  \text{Hz},  1  \text{H}), \\ 9.42\text{-}9.68  (\text{m},  1  \text{H}) \end{array}$
172	(300 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.26 (d, J = 6.7 Hz, 3 H), 2.15 (s, 3 H), 3.05-3.15 (m, 2 H), 3.55-3.77 (m, 2 H), 3.80-3.97 (m, 1 H), 4.21-4.35 (m, 1 H), 4.88-5.08 (m, 1 H), 6.94 (d, J = 7.8 Hz, 1 H), 7.33-7.45 (m, 1 H), 7.74 (d, J = 7.7 Hz, 1 H), 7.91 (s, 1 H), 8.01 (s, 2 H), 8.19-8.28 (m, 1 H)
187	(360 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.13 (d, J = 6.6 Hz, 3 H), 1.66 (d, J = 7.0 Hz, 3 H), 2.11-2.18 (m, 3 H), 3.07 (dd, J = 14.1, 3.8 Hz, 1 H), 3.68-3.80 (m, 1 H), 4.49 (dd, J = 13.9, 1.5 Hz, 1 H), 6.00 (q, J = 7.0 Hz, 1 H), 7.18 (d, J = 7.7 Hz, 1 H), 7.38-7.45 (m, 2 H), 7.52 (d, J = 7.3 Hz, 1 H), 7.66 (d, J = 1.8 Hz, 1 H), 7.83 (d, J = 7.7 Hz, 1 H), 8.24 (d, J = 1.5 Hz, 1 H), 11.48 (d, J = 1.8 Hz, 1 H)
188	(360 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 0.25 (d, J = 6.6 Hz, 3 H), 1.65 (d, J = 7.0 Hz, 3 H), 2.10-2.17 (m, 3 H), 3.78 (dd, J = 13.9, 3.7 Hz, 1 H), 4.10 (ddd, J = 6.3, 3.9, 1.8 Hz, 1 H), 4.56-4.66 (m, 1 H), 6.10 (q, J = 7.1 Hz, 1 H), 7.22 (d, J = 7.7 Hz, 1 H), 7.36-7.44 (m, 2 H), 7.54 (d, J = 7.3 Hz, 1 H), 7.63 (d, J = 1.8 Hz, 1 H), 7.82 (d,
189	J = 7.7 Hz, 1 H), 8.25 (d, J = 1.1 Hz, 1 H), 11.43 (d, J = 1.8 Hz, 1 H) (360 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.13 (d, J = 7.0 Hz, 3 H), 1.64 (d, J = 7.0 Hz, 3 H), 2.14 (s, 3 H), 3.10 (dd, J = 13.7, 3.8 Hz, 1 H), 3.68-3.77 (m, 1 H), 3.80 (s, 3 H), 4.49 (dd, J = 13.7, 1.6 Hz, 1 H), 5.99 (q, J = 7.0 Hz, 1 H), 7.17 (d, J = 7.7 Hz, 1 H), 7.41 (s, 1 H), 7.52 (d, J = 7.3 Hz, 1 H), 7.62 (d, J = 10.2 Hz, 1 H), 7.66 (s, 1 H), 7.82 (d, J = 7.7 Hz, 1 H), 8.24 (d, J = 1.1 Hz, 1 H)
190	7.83 (d, J = 7.7 Hz, 1 H), 8.24 (d, J = 1.1 Hz, 1 H) (360 MHz, DMSO-d <sub>6</sub> ) δ ppm 0.29 (d, J = 6.6 Hz, 3 H), 1.65 (d, J = 7.0 Hz, 3 H), 2.15 (s, 3 H), 3.72-3.83 (m, 4 H), 4.04-4.14 (m, 1 H), 4.56-4.65 (m, 1 H), 6.07 (q, J = 6.8 Hz, 1 H), 7.22 (d, J = 8.1 Hz, 1 H), 7.41 (t, J = 1.1 Hz, 1 H), 7.55 (d, J = 7.3 Hz, 1 H), 7.58-7.65 (m, 2 H), 7.82 (d, J = 8.1 Hz, 1 H), 8.26 (d, J = 1.1 Hz, 1 H)
191	(360 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.13 (d, J = 7.0 Hz, 3 H), 1.38 (t, J = 7.3 Hz, 3 H), 1.65 (d, J = 7.3 Hz, 3 H), 2.16 (s, 3 H), 3.09 (dd, J = 13.9, 3.7 Hz, 1 H), 3.74 (dt, J = 4.6, 2.1 Hz, 1 H), 4.14-4.27 (m, 2 H), 4.50 (dd, J = 13.7, 1.6 Hz, 1 H), 5.99 (q, J = 6.8 Hz, 1 H), 7.18 (d, J = 7.7 Hz, 1 H), 7.45 (s, 1 H), 7.52 (d, J = 7.3 Hz, 1 H), 7.68 (d, J = 10.6 Hz, 1 H), 7.73 (s, 1 H), 7.85 (d, J = 7.7 Hz, 1 H), 8.35 (s, 1 H)
192	(360 MHz, DMSO-d <sub>6</sub> ) δ ppm 0.27 (d, J = 6.6 Hz, 3 H), 1.34 (t, J = 7.1 Hz, 3 H), 1.55 (d, J = 7.0 Hz, 3 H), 2.14 (s, 3 H), 3.79 (dd, J = 13.9, 3.7 Hz, 1 H), 4.05-4.15 (m, 1 H), 4.20 (q, J = 7.1 Hz, 2 H), 4.57-4.66 (m, 1 H), 6.09 (q, J = 7.0 Hz, 1 H), 7.22 (d, J = 7.7 Hz, 1 H), 7.41 (s, 1 H), 7.55 (d, J = 7.3 Hz, 1 H), 7.64-7.72 (m, 2 H), 7.82 (d, J = 7.7 Hz, 1 H), 8.22-8.29 (m, 1 H)
193	(360 MHz, DMSO-d <sub>6</sub> ) δ ppm 1.14 (d, J = 6.6 Hz, 3 H), 1.45 (d, J = 6.6 Hz, 3 H), 1.51 (d, J = 6.6 Hz, 3 H), 1.66 (d, J = 7.0 Hz, 3 H), 2.14 (d, J = 0.7 Hz, 3 H), 3.06 (dd, J = 13.9, 3.7 Hz, 1 H), 3.68-3.81 (m, 1 H), 4.50 (dd, J = 14.1, 1.6 Hz, 1 H), 4.74 (quin, J = 6.7 Hz, 1 H), 6.01 (q, J = 7.2 Hz, 1 H), 7.17 (d, J = 7.7 Hz, 1 H), 7.40 (s, 1 H), 7.51 (d, J = 7.7 Hz, 1 H), 7.71 (d, J = 11.0 Hz, 1 H), 7.79-7.86 (m, 2.1), 8.24 (d, J = 1.1 Hz, 1 H), 7.71 (d, J = 7.7 Hz, 1
194	2 H), 8.24 (d, J = 1.1 Hz, 1 H) (360 MHz, DMSO-d <sub>6</sub> ) δ ppm 0.24 (d, J = 6.6 Hz, 3 H), 1.46 (d, J = 4.4 Hz, 3 H), 1.44 (d, J = 4.4 Hz, 3 H), 1.67 (d, J = 7.0 Hz, 3 H), 2.14 (s, 3 H), 3.79 (dd, J = 13.7, 3.5 Hz, 1 H), 4.05-4.16 (m, 1 H), 4.55-4.66 (m, 1 H), 4.74 (quin, J = 6.7 Hz, 1 H), 6.11 (q, J = 7.1 Hz, 1 H), 7.21 (d, J = 7.7 Hz, 1 H), 7.41 (s, 1 H), 7.56 (d, J = 7.3 Hz, 1 H), 7.70 (d, J = 11.0 Hz, 1 H), 7.77-7.85 (m, 2 H), 8.26 (d, J = 1.1 Hz,
204	1 H) (360 MHz, CHLOROFORM-d) δ ppm 0.97 (d, J = 6.6 Hz, 3 H), 2.24 (ddd, J = 14.5, 9.3, 4.8 Hz, 1 H), 2.32 (s, 3 H), 2.61-2.73 (m, 1 H), 3.03-3.13 (m, 1 H), 3.22-3.33 (m, 1 H), 3.74-3.89 (m, 2 H), 4.82-4.88 (m, 1 H), 6.11 (dd, J = 8.8, 5.1 Hz, 1 H), 7.17 (br. s., 1 H), 7.32 (d, J = 7.7 Hz, 1 H), 7.42 (d, J = 8.4 Hz, 1 H), 7.51-7.59 (m, 3 H) 8.43 (br. s., 1 H)

	<sup>1</sup> H NMR results
Co. No.	<sup>1</sup> H NMR result
205	(400 MHz, CHLOROFORM-d, 55° C.) δ ppm 1.27-1.31 (m, 3 H), 2.21 (dd, J = 13.3, 8.5 Hz, 1 H), 2.33 (s, 3 H), 2.61-2.65 (m, 1 H), 3.03 (d, J = 8.1 Hz, 1 H), 3.08-3.22 (m, 1 H), 3.61 (dd, J = 14.1, 4.0 Hz, 1 H), 3.64-3.73 (m, 1 H), 4.76-4.86 (m, 1 H), 6.22 (t, J = 8.1 Hz, 1 H), 7.19 (s, 1 H), 7.30 (d, J = 7.7 Hz, 1 H), 7.38 (s, 1 H), 7.43 (d, J = 7.7 Hz, 1 H), 7.52 (d, J = 7.7 Hz, 1 H), 7.54-7.60 (m, 1 H), 8.45 (s, 1 H)
206	(600 MHz, CHLOROFORM-d) δ ppm 1.17 (d, J = 6.7 Hz, 3 H), 2.28 (d, J = 1.0 Hz, 3 H), 3.10 (dd, J = 16.1, 6.2 Hz, 1 H), 3.18 (dd, J = 16.1, 6.4 Hz, 1 H), 3.28-3.39 (m, 2 H), 3.69 (dd, J = 13.9, 4.0 Hz, 1 H), 3.88-3.98 (m, 1 H), 4.79-4.87 (m, 1 H), 5.25-5.33 (m, 1 H), 7.14 (t, J = 1.2 Hz, 1 H), 7.18-7.24 (m, 2 H), 7.24-7.30 (m, 3 H), 7.46 (d, J = 7.6 Hz, 1 H), 8.21-8.25 (m, 1 H)
207 210	The NMR spectrum of 207 matches the one of compound 206 (400 MHz, DMSO- $d_6$ ) $\delta$ ppm 1.17 (d, J = 6.5 Hz, 3 H), 1.62 (d, J = 6.9 Hz, 3 H), 2.15 (s, 3 H), 3.63 (dd, J = 13.9, 3.8 Hz, 1 H), 3.98 (br. s., 1 H), 4.61 (dd, J = 13.7, 1.6 Hz, 1 H), 5.64 (q, J = 6.9 Hz, 1 H), 7.09 (d, J = 7.7 Hz, 1 H), 7.42 (s, 1 H), 7.48 (d, J = 2.0 Hz, 2 H), 7.55 (t, J = 1.8 Hz, 1 H), 7.81 (d, J = 7.7 Hz, 1 H), 8.27 (s, 1 H)
211	$^{(4)}$ C(4, 1 H), $^{(4)}$ C(4), $^{(4)}$ C(5), $^{(4)}$ C(5), $^{(4)}$ C(6), $^{(4)}$ C(7), $^{(4)}$ C(7), $^{(4)}$ C(8), $^{(4)}$ C(8), $^{(4)}$ C(8), $^{(4)}$ C(9),
212	(360 MHz, CHLOROFORM-d) \( \delta\) ppm 1.23 (s, 3 H), 1.77 (d, J = 7.0 Hz, 3 H), 2.28 (s, 3 H), 3.11 (dd, J = 13.9, 3.7 Hz, 1 H), 3.70 (ddd, J = 6.5, 4.1, 1.8 Hz, 1 H), 4.57 (dd, J = 14.1, 2.0 Hz, 1 H), 6.26-6.36 (m, 1 H), 7.13 (s, 1 H), 7.37 (d, J = 7.7 Hz, 1 H), 7.42 (s, 1 H), 7.44-7.54 (m, 3 H), 7.77 (s, 1 H), 8.22 (d, J = 1.1 Hz, 1 H), 8.87 (br. s., 1 H)
213	360 MHz, CHLOROFORM-d) δ ppm 0.43 (d, J = 6.6 Hz, 3 H), 1.74 (d, J = 7.0 Hz, 3 H), 2.28 (s, 3 H), 3.65 (dd, J = 13.9, 3.7 Hz, 1 H), 3.96 (ddd, J = 6.7, 4.1, 2.0 Hz, 1 H), 4.83 (dd, J = 13.9, 1.8 Hz, 1 H), 6.40 (q, J = 7.1 Hz, 1 H), 7.12 (s, 1 H), 7.36-7.40 (m, 1 H), 7.40 (d, J = 2.6 Hz, 1 H), 7.45-7.51 (m, 3 H), 7.96 (s, 1 H), 8.15-8.29 (m, 1 H), 8.52 (bz. s 1 H)
214	(360 MHz, CHLOROFORM-d) & ppm 1.24 (d, J = 6.6 Hz, 3 H), 1.75 (d, J = 7.0 Hz, 3 H), 2.28 (d, J = 0.7 Hz, 3 H), 3.11 (dd, J = 13.9, 4.0 Hz, 1 H), 3.65-3.76 (m, 1 H), 3.88 (s, 3 H), 4.57 (dd, J = 14.1, 2.0 Hz, 1 H), 6.29 (q, J = 6.8 Hz, 1 H), 7.13 (t, J = 1.1 Hz, 1 H), 7.25 (d, J = 0.7 Hz, 1 H), 7.36 (d, J = 7.7 Hz, 1 H), 7.39-7.44 (m, 1 H), 7.47-7.53 (m, 2 H), 7.73-7.77 (m, 1 H), 8.20 (d, J = 1.1 Hz, 1
215	H) (360 MHz, CHLOROFORM-d) δ ppm 0.45 (d, J = 7.0 Hz, 3 H), 1.72 (d, J = 7.3 Hz, 3 H), 2.30 (s, 3 H), 3.65 (dd, J = 13.5, 3.7 Hz, 1 H), 3.87 (s, 3 H), 3.90-4.01 (m, 1 H), 4.82 (dd, J = 13.9, 1.8 Hz, 1 H), 6.35 (q, J = 6.8 Hz, 1 H), 7.11 (br. s., 1 H), 7.24 (s, 1 H), 7.34-7.42 (m, 2 H), 7.44-7.53 (m, 2 H), 7.92 (s, 1 H), 8.29 (s, 1 H)
216	(360 MHz, CHLOROFORM-d) δ ppm 1.24 (d, J = 6.6 Hz, 3 H), 1.54 (t, J = 7.3 Hz, 3 H), 1.75 (d, J = 7.0 Hz, 3 H), 2.28 (d, J = 0.7 Hz, 3 H), 3.12 (dd, J = 13.9, 3.7 Hz, 1 H), 3.64-3.74 (m, 1 H), 4.21-4.30 (m, 2 H), 4.57 (dd, J = 14.1, 2.0 Hz, 1 H), 6.23-6.33 (m, 1 H), 7.13 (s, 1 H), 7.30 (s, 1 H), 7.36 (d, J = 7.7 Hz, 1 H), 7.41-7.52 (m, 3 H), 7.75 (s, 1 H), 8.20 (s, 1 H)
217	(360 MHz, CHLOROFORM-d) δ ppm 0.42 (d, J = 6.6 Hz, 3 H), 1.50 (t, J = 7.3 Hz, 3 H), 1.72 (d, J = 7.0 Hz, 3 H), 2.28 (d, J = 0.7 Hz, 3 H), 3.65 (dd, J = 13.9, 3.7 Hz, 1 H), 3.95 (td, J = 4.2, 1.8 Hz, 1 H), 4.23 (q, J = 7.3 Hz, 2 H), 4.83 (dd, J = 13.7, 1.6 Hz, 1 H), 6.37 (q, J = 7.2 Hz, 1 H), 7.12 (s, 1 H), 7.29 (s, 1 H), 7.34-7.44 (m, 2 H), 7.44-7.52 (m, 2 H), 7.94 (s, 1 H), 8.22 (s, 1 H)
218	(360 MHz, DMSO- $d_6$ ) $\delta$ ppm 1.13 (d, J = 6.6 Hz, 3 H), 1.67 (d, J = 7.0 Hz, 3 H), 2.14 (s, 3 H), 3.04 (dd, J = 13.9, 3.7 Hz, 1 H), 3.71-3.80 (m, 1 H), 4.49 (dd, J = 13.9, 1.5 Hz, 1 H), 6.01 (q, J = 7.0 Hz, 1 H), 7.12 (dd, J = 11.0, 1.5 Hz, 1 H), 7.18 (d, J = 8.1 Hz, 1 H), 7.27 (d, J = 1.8 Hz, 1 H), 7.41 (s, 1 H), 7.75 (d, J = 2.2 Hz, 1 H), 7.83 (d, J = 7.7 Hz, 1 H), 8.24 (d, J = 1.1 Hz, 1 H), 11.83-12.05 (m, 1 H)
219	(360 MHz, DMSO- $d_6$ ) $\delta$ ppm 0.26 (d, J = 6.6 Hz, 3 H), 1.67 (d, J = 7.3 Hz, 3 H), 2.14 (s, 3 H), 3.79 (dd, J = 13.7, 3.5 Hz, 1 H), 4.05-4.16 (m, 1 H), 4.55-4.67 (m, 1 H), 6.10 (q, J = 6.8 Hz, 1 H), 7.11 (dd, J = 10.8, 1.6 Hz, 1 H), 7.21 (d, J = 7.7 Hz, 1 H), 7.31 (d, J = 1.8 Hz, 1 H), 7.41 (s, 1 H), 7.72 (d, J = 2.2 Hz, 1 H), 7.82 (d, J = 7.7 Hz, 1 H), 8.26 (d, J = 1.1 Hz, 1 H), 11.95 (d, J = 2.2 Hz, 1 H)
220	$\begin{array}{l} (360 \text{ MHz}, \text{DMSO-d}_6)  \delta  \text{ppm 1.14}  (\text{d, J} = 16.6  \text{Hz}, 3  \text{H}),  1.40  (\text{d, J} = 7.1  \text{Hz}, 1  \text{H}), \\ (360 \text{ MHz}, \text{DMSO-d}_6)  \delta  \text{ppm 1.14}  (\text{d, J} = 6.6  \text{Hz}, 3  \text{H}),  1.40  (\text{t, J} = 7.1  \text{Hz}, 3  \text{H}), \\ (365 \text{ (d, J} = 7.3  \text{Hz}, 3  \text{H}),  2.17  (\text{s, 3}  \text{H}),  3.08  (\text{dd, J} = 13.9, 3.7  \text{Hz}, 1  \text{H}), \\ (371 - 3.80  (\text{m, 1}  \text{H}),  4.27 - 4.37  (\text{m, 2}  \text{H}),  4.50  (\text{dd, J} = 13.7,  1.6  \text{Hz}, 1  \text{H}),  5.98  (\text{q, J} = 7.0  \text{Hz}, 1  \text{H}),  7.14  (\text{dd, J} = 12.3,  1.6  \text{Hz}, 1  \text{H}),  7.19  (\text{d, J} = 7.7  \text{Hz}, 1  \text{H}),  7.26  (\text{d, J} = 1.8  \text{Hz}, 1  \text{H}),  7.49  (\text{s, 1}  \text{H}),  7.78  (\text{s, 1}  \text{H}),  7.88  (\text{d, J} = 7.7  \text{Hz}, 1  \text{H}),  8.46  (\text{s, 1}  \text{H}),  7.49  (\text{s, 1}$
221	(360 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 0.28 (d, J = 6.6 Hz, 3 H), 1.36 (t, J = 7.1 Hz, 3 H), 1.65 (d, J = 7.3 Hz, 3 H), 2.15 (s, 3 H), 3.80 (dd, J = 14.1, 3.8 Hz, 1 H), 4.10 (dt, J = 4.5, 2.3 Hz, 1 H), 4.31 (q, J = 7.4 Hz, 2 H), 4.53-4.67 (m, 1 H), 6.07 (q, J = 7.3 Hz, 1 H), 7.13 (dd, J = 12.1, 1.8 Hz, 1 H), 7.21 (d, J = 7.7 Hz, 1 H), 7.31 (d, J = 1.5 Hz, 1 H), 7.43 (s, 1 H), 7.75 (s, 1 H), 7.83 (d, J = 7.7 Hz, 1 H), 8.32 (s, 1 H)

#### TABLE 2b-continued

	<sup>1</sup> H NMR results				
Co. No.	<sup>1</sup> H NMR result				
223	$(360 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.13 \text{ (d, J} = 6.6 \text{ Hz}, 3 \text{ H)}, 1.64 \text{ (d, J} = 7.0 \text{ Hz}, 3 \text{ H)}, \\ 2.14 \text{ (s, 3 H)}, 3.08 \text{ (dd, J} = 14.1, 3.8 \text{ Hz}, 1 \text{ H)}, 3.70\text{-}3.82 \text{ (m, 1 H)}, 3.97 \text{ (d, m)}$				
	J = 2.2 Hz, 3 H), 4.50 (dd, J = 13.9, 1.8 Hz, 1 H), 5.97 (q, J = 7.0 Hz, 1 H), 7.12 (dd, J = 12.1, 1.8 Hz, 1 H), 7.17 (d, J = 7.7 Hz, 1 H), 7.24 (d, J = 1.5 Hz, 1 H),				
224	7.41 (s, 1 H), 7.70 (s, 1 H), 7.83 (d, $J = 7.7$ Hz, 1 H), 8.25 (d, $J = 1.1$ Hz, 1 H) (360 MHz, DMSO-d <sub>6</sub> ) $\delta$ ppm 0.31 (d, $J = 6.6$ Hz, 3 H), 1.64 (d, $J = 7.0$ Hz, 3 H),				
	2.15 (s, 3 H), 3.79 (dd, J = 13.9, 3.7 Hz, 1 H), 3.96 (d, J = 1.8 Hz, 3 H), 4.03-4.13 (m, 1 H), 4.62 (d, J = 15.0 Hz, 1 H), 6.05 (q, J = 7.1 Hz, 1 H), 7.12 (dd,				
	J = 12.1, 1.8  Hz, 1  H), 7.21  (d,  J = 7.7  Hz, 1  H), 7.29  (d,  J = 1.8  Hz, 1  H), 7.42  (s,				
243	1 H), 7.67 (s, 1 H), 7.82 (d, J = 7.7 Hz, 1 H), 8.28 (s, 1 H) (360 MHz, CHLOROFORM-d) δ ppm 0.45 (d, J = 6.6 Hz, 3 H), 1.48 (t, J = 7.3 Hz,				
	3 H), 1.69 (d, J = 7.3 Hz, 3 H), 2.28 (s, 3 H), 3.63 (dd, J = 13.7, 3.8 Hz, 1 H), 3.88-4.00 (m, 1 H), 4.18 (q, J = 7.1 Hz, 2 H), 4.82 (dd, J = 14.1, 1.6 Hz, 1 H),				
	6.30 (q, J = 6.7 Hz, 1 H), 7.12 (s, 1 H), 7.18 (dd, J = 8.8, 1.8 Hz, 1 H), 7.23 (d,				
	J = 10.2 Hz, 2 H), 7.37 (d, J = 7.7 Hz, 1 H), 7.49 (d, J = 7.7 Hz, 1 H), 7.60 (d, J = 1.5 Hz, 1 H), 8.22 (s, 1 H)				
244	(360 MHz, CHLOROFORM-d) δ ppm 1.25 (d, J = 7.0 Hz, 3 H), 1.56 (t, J = 7.3 Hz, 3 H), 1.75 (d, J = 7.0 Hz, 3 H), 2.28 (s, 3 H), 3.09 (dd, J = 13.9, 3.7 Hz, 1 H),				
	3.64-3.74 (m, 1 H), 4.26 (qd, J = 7.2, 2.9 Hz, 2 H), 4.53 (dd, J = 13.9, 1.8 Hz, 1				
	H), 6.33 (q, J = 7.2 Hz, 1 H), 7.13 (s, 1 H), 7.29 (d, J = 7.3 Hz, 1 H), 7.33 (s, 1 H), 7.35 (d, J = 8.1 Hz, 1 H), 7.49 (d, J = 7.7 Hz, 1 H), 7.52 (d, J = 8.4 Hz, 1 H),				
	7.64 (s, 1 H), 8.19 (d, J = 0.7 Hz, 1 H)				
245	$(360 \text{ MHz}, \text{DMSO-d}_6) \delta \text{ ppm } 1.14 \text{ (d, J} = 6.6 \text{ Hz}, 3 \text{ H)}, 1.68 \text{ (d, J} = 7.3 \text{ Hz}, 3 \text{ H)}, 2.30 \text{ (d, J} = 0.7 \text{ Hz}, 3 \text{ H)}, 3.07 \text{ (dd, J} = 13.9, 3.7 \text{ Hz}, 1 \text{ H)}, 3.72-3.83 \text{ (m, 1 H)},$				
	4.42-4.54 (m, 1 H), 6.02 (q, J = 7.0 Hz, 1 H), 7.12 (dd, J = 8.6, 2.0 Hz, 1 H),				
	7.28 (d, J = 8.1 Hz, 1 H), 7.40 (d, J = 1.8 Hz, 1 H), 7.43 (d, J = 8.8 Hz, 1 H), 7.68 (d, J = 2.2 Hz, 1 H), 7.80 (s, 1 H), 8.08 (d, J = 7.7 Hz, 1 H), 9.34 (s, 1 H),				
276	11.47 (s, 1 H) (360 MHz, CHLOROFORM-d) δ ppm 0.45 (d, J = 7.0 Hz, 3 H), 1.71 (d, J = 7.0 Hz,				
	3 H), 2.42-2.50 (m, 3 H), 3.69 (dd, Ĵ = 13.9, 3.7 Hz, 1 H), 3.91-4.04 (m, 1 H), 4.79 (dd, J = 14.1, 1.6 Hz, 1 H), 6.30 (q, J = 6.8 Hz, 1 H), 7.17 (dd, J = 8.6, 2.0 Hz,				
	1 H), 7.25 (m, 1 H), 7.32 (d, J = 8.8 Hz, 1 H), 7.35 (d, J = 2.2 Hz, 1 H),				
	7.43 (d, J = 8.1 Hz, 1 H), 7.57 (d, J = 1.8 Hz, 1 H), 7.70 (d, J = 7.7 Hz, 1 H), 8.65 (br. s., 1 H), 9.12 (s, 1 H)				
247	(360 MHz, CHLOROFORM-d) δ ppm 0.59-0.76 (m, 2 H), 0.83-1.00 (m, 2				
	H), 1.19 (d, J = 7.0 Hz, 3 H), 1.46 (t, J = 7.3 Hz, 3 H), 1.92-2.08 (m, 1 H), 2.28 (s, 3 H), 3.54 (dd, J = 14.3, 4.0 Hz, 1 H), 3.90 (ddd, J = 6.7, 4.3, 2.2 Hz, 1 H),				
	4.14 (q, J = 7.3 Hz, 2 H), 4.30 (d, J = 15.0 Hz, 1 H), 4.64-4.77 (m, 1 H), 5.50 (d, J = 14.6 Hz, 1 H), 6.98 (dd, J = 8.4, 1.8 Hz, 1 H), 7.10-7.17 (m, 2 H), 7.26 (d,				
	J = 7.3 Hz, 1 H), 7.31-7.39 (m, 2 H), 7.48 (d, J = 7.7 Hz, 1 H), 8.22 (s, 1 H)				
248	(360 MHz, CHLOROFORM-d) 8 ppm 0.45 (d, J = 6.6 Hz, 3 H), 0.97-1.04 (m, H), 1.08-1.16 (m, 2 H), 1.66 (d, J = 7.0 Hz, 3 H), 2.28 (s, 3 H), 3.32 (tt,				
	J = 6.9, 3.7  Hz, 1  H), 3.63  (dd,  J = 13.9, 4.0  Hz, 1  H), 3.84-3.97  (m, 1 H),				
	4.82 (dd, J = 13.7, 1.6 Hz, 1 H), 6.25 (q, J = 7.0 Hz, 1 H), 7.13 (s, 1 H), 7.19 (d, J = 1.1 Hz, 1 H), 7.31 (d, J = 9.1 Hz, 1 H), 7.37 (d, J = 7.7 Hz, 1 H), 7.49 (d, J = 7.7 Hz, 1				
249	H), 7.60 (d, J = 7.3 Hz, 1 H), 8.23 (s, 1 H) (360 MHz, CHLOROFORM-d) δ ppm 1.23 (d, J = 6.6 Hz, 3 H), 1.51 (t, J = 7.1 Hz,				
2-10	3 H), 1.72 (d, J = 7.0 Hz, 3 H), 2.28 (s, 3 H), 3.17 (dd, J = 13.9, 3.7 Hz, 1 H),				
	3.65-3.76 (m, 1 H), 4.19 (qd, J = 7.3, 2.0 Hz, 2 H), 4.56 (dd, J = 13.9, 1.8 Hz, 1 H), 6.22 (q, J = 7.0 Hz, 1 H), 7.14 (s, 1 H), 7.16-7.22 (m, 2 H), 7.28 (d, J = 9.0 Hz,				
	1 H), 7.35 (d, J = 7.7 Hz, 1 H), 7.41 (d, J = 1.8 Hz, 1 H), 7.49 (d, J = 7.7 Hz, 1				
256	H), 8.20 (s, 1 H) (360 MHz, CHLOROFORM-d) δ ppm 0.47 (d, J = 6.6 Hz, 3 H), 1.66 (d, J = 7.0 Hz,				
	3 H), 2.28 (s, 3 H), 3.63 (dd, J = 13.7, 3.8 Hz, 1 H), 3.78 (s, 3 H), 3.87-3.98 (m, 4 H), 4.81 (dd, J = 13.7, 1.6 Hz, 1 H), 6.21-6.31 (m, 1 H), 6.77 (s, 1 H),				
	7.04 (s, 1 H), 7.13 (s, 1 H), 7.37 (d, J = 7.7 Hz, 1 H), 7.49 (d, J = 7.7 Hz, 1 H),				
	7.60 (s, 1 H), 8.22 (s, 1 H)				

### SFC-MS

For SFC-MS, an analytical SFC system from Berger 55 Instruments (Newark, Del., USA) was used comprising a dual pump control module (FCM-1200) for delivery of  $\rm CO_2$  and modifier, a thermal control module for column heating (TCM2100) with temperature control in the range 1-150° C. and column selection valves (Valco, VICI, Houston, Tex., 60 USA) for 6 different columns. The photodiode array detector (Agilent 1100, Waldbronn, Germany) is equipped with a high-pressure flow cell (up to 400 bar) and configured with a CTC LC Mini PAL auto sampler (Leap Technologies, Carrboro, N.C., USA). A ZQ mass spectrometer (Waters, Milford, 65 Mass., USA) with an orthogonal Z-electrospray interface is coupled with the SFC-system. Instrument control, data col-

lection and processing were performed with an integrated platform consisting of the SFC ProNTo software and Masslynx software.

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Co. No. 11a-11b: SFC-MS was carried out on a OD-H column (500×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 3 ml/min. Two mobile phases (mobile phase A: CO<sub>2</sub>; mobile phase B: MeOH containing 0.2% iPrNH<sub>2</sub>) were employed. First 30% B was hold for 18.5 min. Then a gradient was applied from 30% B to 50% B in 2 min and hold for 4.1 min. Column temperature was set at 50° C. Under these conditions, Co. No. 11a ('enantiomer A') had a shorter retention time (R<sub>t</sub>) on the column than Co. No. 11b ('enantiomer B'). The measurement was compared against the racemic mixture.

Co. No. 14a-14b: SFC-MS was carried out on a OD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 3 ml/min. Two mobile phases (mobile phase A:  $CO_2$ ; mobile phase B: MeOH containing 0.2% iPrNH<sub>2</sub>) were employed. 20% B was hold for 15 min. Column temperature 5 was set at 30° C. Under these conditions, Co. No. 14a ('Cis A') had a shorter R, on the column than Co. No. 14b ('Cis B'). The measurement was compared against the mixture of the compounds.

Co. No. 111-112: SFC-MS was carried out on a AD-H 10 column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 3 ml/min. Two mobile phases (mobile phase A:  $CO_2$ ; mobile phase B: MeOH containing 0.2% iPrNH $_2$ ) were employed. 8% B was hold for 15 min. Column temperature was set at 30° C. Under these conditions, Co. No. 111 (diastereomer A') had a shorter  $R_t$  on the column than Co. No. 112 (diastereomer B'). The measurement was compared against the mixture of the compounds.

Co. No. 129-130: SFC-MS was carried out on a AD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with 20 a flow rate of 3 ml/min. Two mobile phases (mobile phase A: CO<sub>2</sub>; mobile phase B: EtOH containing 0.2% iPrNH<sub>2</sub>) were employed. 25% B was hold for 15 min. Column temperature was set at 30° C. Under these conditions, Co. No. 130 (diastereomer A') had a shorter R<sub> $\tau$ </sub> on the column than Co. No. 129 25 (diastereomer B'). The measurement was compared against the mixture of the compounds.

Co. No. 183-184: SFC-MS was carried out on an OD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 3 ml/min. Two mobile phases (mobile phase A: 30 CO<sub>2</sub>; mobile phase B: EtOH containing 0.2% iPrNH<sub>2</sub>) were employed. 35% B was hold for 15 min. Column temperature was set at 30° C. Under these conditions, Co. No. 183 ('enantiomer A') had a shorter  $R_r$  on the column than Co. No. 184 ('enantiomer B'). The measurement was compared against 35 the racemic mixture.

Co. No. 185-186: The same SFC-MS conditions as for Co. No. 183-184 were used. Under these conditions, Co. No. 185 ('enantiomer A') had a shorter R, on the column than Co. No. 186 ('enantiomer B'). The measurement was compared 40 against the racemic mixture.

Co. No. 206-207: SFC-MS was carried out on a OJ-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 3 ml/min. Two mobile phases (mobile phase A:  $CO_2$ ; mobile phase B: MeOH containing 0.2% iPrNH $_2$ ) were 45 employed. First 25% B was hold for 18 min. Then a gradient was applied from 25% B to 50% B in 2.5 min, and 50% B was hold for 4.1 min. Column temperature was set at 30° C. Under these conditions, Co. No. 207 ('enantiomer A') had a shorter retention time ( $R_t$ ) on the column than Co. No. 206 ('enantiomer B'). The measurement was compared against the racemic mixture.

Co. No. 208-209: SFC-MS was carried out on a OD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 3 ml/min. Two mobile phases (mobile phase A: 55 CO<sub>2</sub>; mobile phase B: MeOH containing 0.2% iPrNH<sub>2</sub>) were employed. 35% B was hold for 15 min. Column temperature was set at 30° C. Under these conditions, Co. No. 209 ('enantiomer A') had a shorter retention time (R<sub>t</sub>) on the column than Co. No. 208 ('enantiomer B'). The measurement was compared against the racemic mixture.

Co. No. 212-213: SFC-MS was carried out on a AD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 3 ml/min. Two mobile phases (mobile phase A: CO<sub>2</sub>; mobile phase B: MeOH containing 0.2% iPrNH<sub>2</sub>) were 65 employed. 30% B was hold for 15 min. Column temperature was set at 40° C. Under these conditions, Co. No. 212 ('enan-

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tiomer A') had a shorter retention time  $(R_i)$  on the column than Co. No. 213 ('enantiomer B'). The measurement was compared against the racemic mixture.

Co. No. 236-237: SFC-MS was carried out on a OJ-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 5 ml/min. Two mobile phases (mobile phase A: CO<sub>2</sub>; mobile phase B: MeOH containing 0.2% iPrNH<sub>2</sub>) were employed. 25% B was hold for 7 min. Column temperature was set at 40° C. Under these conditions, Co. No. 236 had a shorter R<sub>t</sub> on the column than Co. No. 237. The measurement was compared against the mixture of the compounds.

Co. No. 265-266: SFC-MS was carried out on a AD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 5 ml/min. Two mobile phases (mobile phase A:  $CO_2$ ; mobile phase B: EtOH (containing 0.2% iPrNH<sub>2</sub>) were employed. 35% B was hold for 4 min, increased up to 50% B in 1 min and hold for 2 min at 50% B. Column temperature was set at 40° C. Under these conditions, Co. No. 265 ('enantiomer A') had a shorter  $R_t$  on the column than Co. No. 266 ('enantiomer B'). The measurement was compared against the mixture

Co. No. 267-268: SFC-MS was carried out on a AD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 5 ml/min. Two mobile phases (mobile phase A:  $CO_2$ ; mobile phase B: EtOH (containing 0.2% iPrNH<sub>2</sub>) were employed. 40% B was hold for 4 min, increased up to 50% B in 1 min and hold for 2 min at 50% B. Column temperature was set at 40° C. Under these conditions, Co. No. 267 ('enantiomer A') had a shorter R<sub>t</sub> on the column than Co. No. 268 ('enantiomer B'). The measurement was compared against the mixture.

Co. No. 271-272: SFC-MS was carried out on a OD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 5 ml/min. Two mobile phases (mobile phase A:  $\rm CO_2$ ; mobile phase B: EtOH (containing 0.2% iPrNH<sub>2</sub>) were employed. 35% B was hold for 4 min, increased up to 50% B in 1 min and hold for 2 min at 50% B. Column temperature was set at 40° C. Under these conditions, Co. No. 271 ('enantiomer A') had a shorter R<sub>t</sub> on the column than Co. No. 272 ('enantiomer B'). The measurement was compared against the mixture.

Co. No. 273-274: SFC-MS was carried out on a AD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 5 ml/min. Two mobile phases (mobile phase A:  $CO_2$ ; mobile phase B: EtOH (containing 0.2% iPrNH<sub>2</sub>) were employed. 35% B was hold for 4 min, increased up to 50% B in 1 min and hold for 2 min at 50% B. Column temperature was set at 40° C. Under these conditions, Co. No. 273 ('enantiomer A') had a shorter  $R_t$  on the column than Co. No. 274 ('enantiomer B'). The measurement was compared against the mixture.

Co. No. 295-296: SFC-MS was carried out on a OD-H column (250×4.6 mm) (Daicel Chemical Industries Ltd) with a flow rate of 5 ml/min. Two mobile phases (mobile phase A: CO<sub>2</sub>; mobile phase B: EtOH (containing 0.2% iPrNH<sub>2</sub>) were employed. 40% B was hold for 4 min, increased up to 50% B in 1 min and hold for 2 min at 50% B. Column temperature was set at 40° C. Under these conditions, Co. No. 295 ('enantiomer A') had a shorter R<sub>t</sub> on the column than Co. No. 296 ('enantiomer B'). The measurement was compared against the mixture.

Pharmacology

A) Screening of the Compounds of the Invention for  $\gamma$ -Secretase-Modulating Activity

Screening was carried out using SKNBE2 human neuroblastoma cells carrying the hAPP 695—wild type, grown in Dulbecco's Modified Eagle's Medium/Nutrient mixture F-12

Co.

No.

82

92

62

115

46

10a

23

99

2

59

118

77

76

72

123

63

68

53

81

10

84

47

29

71

79

28

38

65

121

74

113

97

61

94

34

51

36

40

45

67

70

48

9h

287

(DMEM/NUT-mix F-12) (HAM) provided by Invitrogen (cat no. 10371-029) containing 5% Serum/Fe supplemented with 1% non-essential amino acids, 1-glutamine 2 mM, Hepes 15 mM, penicillin 50 U/ml (units/ml) en streptomycin 50 μg/ml. Cells were grown to near confluency.

The screening was performed using a modification of the assay as described in Citron et al (1997) Nature Medicine 3: 67. Briefly, cells were plated in a 384-well plate at 10<sup>4</sup> cells/ well in Ultraculture (Lonza, BE12-725F) supplemented with 1% glutamine (Invitrogen, 25030-024), 1% non-essential amino acid (NEAA), penicillin 50 U/ml en streptomycin 50 μg/ml in the presence of test compound at different test concentrations. The cell/compound mixture was incubated overnight at 37° C., 5% CO<sub>2</sub>. The next day the media were assayed by two sandwich immuno-assays, for A $\beta$ 42 and A $\beta$  total.

Aβtotal and Aβ42 concentrations were quantified in the cell supernatant using the Aphalisa technology (Perkin Elmer). Alphalisa is a sandwich assay using biotinylated antibody attached to streptavidin coated donorbeads and antibody conjugated to acceptor beads. In the presence of antigen, 20 the beads come into close proximity. The excitation of the donor beads provokes the release of singlet oxygen molecules that trigger a cascade of energy transfer in the acceptor beads, resulting in light emission. To quantify the amount of A $\beta$ 42 in the cell supernatant, monoclonal antibody specific to the 25 C-terminus of Aβ42 (JRF/cAβ42/26) was coupled to the receptor beads and biotinylated antibody specific to the N-terminus of A $\beta$  (JRF/A $\beta$ N/25) was used to react with the donor beads. To quantify the amount of Aßtotal in the cell supernatant, monoclonal antibody specific to the N-terminus of AB 30 (JRF/AβN/25) was coupled to the receptor beads and biotinylated antibody specific to the mid region of Aβ (biotinylated 4G8) was used to react with the donor beads.

To obtain the values reported in Table 3, the data are calculated as percentage of the maximum amount of amyloid 35 Beta 42 measured in the absence of the test compound. The sigmoidal dose response curves were analyzed using nonlinear regression analysis with percentage of the control plotted against the log concentration of the compound. A 4-parameter equation was used to determine the IC<sub>50</sub>.

	TABLE 3			27 43	0.59 0.62	>10 >10	
Co. No.	IC50 Aβ42 (μΜ)	IC50 Aβtotal (μΜ)	45	44 14b 60 11a	0.63 0.65 0.66 0.66	>10 >10 >10 >10 >10	
112	0.07	>10		9	0.68	>10	
104	>10	>10		64	0.69	>10	
57	>10	>10		106	0.72	>10	
41	>10	>10		101	0.74	>10	
102	>10	>10	50	42	0.74	>10	
17	>10	>10	50	100	0.76	>10	
16	>10	>10		96	0.76	>10	
122	>10	>10		58	0.83	>10	
89	>10	>10		91	0.91	>6.03	
119	>10	>10		8	0.93	>10	
86	0.03	>10	55	39	0.95	>10	
117	0.04	>10	33	52	0.98	>10	
110	0.04	>10		75	0.98	>10	
50	0.04	>10		69	1.02	>10	
111	0.07	>10		35	1.02	>10	
85	0.06	>10		24	1.10	>10	
78	0.07	>10		37	1.15	>10	
6	0.07	>10	60	54	1.26	>10	
4	0.08	>10		83	1.35	>10	
4a	0.07	>10		20	1.55	>10	
90	0.07	>10		31	1.62	>10	
116	0.08	>10		33	1.66	>10	
5	0.08	>10		26	1.66	>10	
80	0.10	>10	65	25	1.70	>10	
3	0.10	>10		18	1.70	>10	

40

288

IC50

Aßtotal

(uM)

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10 >10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

>10

3.63

TABLE	3-continued

IC50

Αβ42

(µM)

0.11

0.11

0.13

0.13

0.14

0.14

0.14

0.14

0.14

0.17

0.17

0.18

0.19

0.19

0.19

0.19

0.19

0.19

0.19

0.21

0.22

0.22

0.23

0.26

0.27

0.28

0.28

0.28

0.33

0.37

0.38

0.38

0.38

0.40

0.44

0.48

0.49

0.49

0.51

0.51

0.54

0.56

0.58

0.58

0.5

290
TABLE 3-continued

	IC50	IC50			IC50	IC50
Co.	Αβ42	Aβtotal		Co.	Αβ42	Aβtotal
No.	(μΜ)	(μΜ)	5	No.	(μΜ)	(μΜ)
56	1.70	>10	,	171	0.04	>10
14a	1.78	>10		172	0.25	>10
88	1.86	>10		175	0.14	>10
114	1.91	>10		176	1.55	>10
21	1.95	>10		177	0.04	>10
30	2.04	>10	10	178	n.d.	n.d.
109	2.09	>10		179	0.07	>10
15	2.29	>10		180	0.01	6.61
107	2.45	>10		181	0.01	4.57
103	2.45	>10		182	0.01	2.51
73	2.57	>10		183	0.03	6.76
22	2.57	>10	15	184	0.65	9.12
105	2.82	>10		185	0.02	5.89
98 49	2.88 3.39	>10 >10		186 187	0.78 0.19	>10 6.92
55	3.80	>10		188	0.01	4.57
66	3.89	>10		189	0.22	10
9a	4.47	>10		190	0.01	3.31
108	4.79	>10	20	191	0.20	>10
120	6.17	>10		192	0.01	1.82
10b	6.61	>10		193	0.20	>10
93	6.76	>10		194	0.01	1.55
19	6.76	>10		195	0.03	6.46
95	7.59	>10		196	0.02	7.41
87	7.76	>10	25	197	0.06	>10
11b	7.76	>10		198	n.d.	n.d.
32	9.55	9.77		199	0.02	9.78
14	n.d.	n.d.		200	0.02	>10
13	n.d.	n.d.		201	n.d.	n.d.
12	n.d.	n.d.		202	n.d.	n.d.
124	2.34	>10	30	203	n.d.	n.d.
125	0.05	>10		204	0.12	>10
126	0.11	>10		205	0.24	>10
127	0.18	>10		206	>10	>10
128	0.09	>10		207	0.93	>10
129	0.11	>10		208	3.89	>10
130	0.16	>10	35	209	0.38	>10
131	0.20 2.14	>10 >10		210	0.26 0.09	>10 >10
132 133	1.51	>10		211 212	n.d.	n.d.
134	0.10	>10		212	n.d.	n.d.
135	0.02	>10		214	0.24	6.46
136	0.04	>10		215	0.01	1.44
137	0.03	>10	40	216	0.12	6.46
138	0.04	>10		217	0.003	0.48
139	0.95	>10		218	0.14	3.80
140	0.06	>10		219	0.02	3.09
141	0.11	>10		220	0.17	6.03
142	0.21	>10		221	0.005	0.79
143	0.06	>10	45	222	0.06	>10
144	0.13	>10		223	0.22	6.61
145	0.11	>10		224	0.01	1.86
146	0.15	>10		225	0.02	>10
147	0.33	>10		226	0.01	4.79
148	0.14	>10		227	0.01	2.19
149	0.30	>10	50	228	0.05	>10
150	n.d.	n.d.		229	0.13	>10
151	0.03	>10		230	0.21	>10
152	0.07	>10		231	0.26	>10
153	0.06	>10 >10		232	0.21	>10 >10
154 155	0.05 0.51	>10 >10		233 234	0.26 n.d.	>10 n.d.
156	0.03	>10	55	234	n.d. 0.01	n.a. 6
156	2.04	8.71		235	n.d.	n.d.
158	0.46	>10		237	n.d.	n.d.
159	0.28	>10		237	0.009	11.u. 4
160	0.19	>10		239	n.d.	n.d.
161	0.79	>10		240	n.d.	n.d.
163	0.17	>10	60	240	0.006	11.u. 4
164	0.03	>10		162	0.10	>10
165	0.02	>10		173	1.26	>10
166	0.05	>10		174	0.01	5
	1.20	>10		242	0.26	>10
167						
167 168	1.41	>10		243	0.01	2
167 168 169	1.41 0.13	>10 >10	65	243 244	0.01 0.37	2 >10

Co. No.	IC50 Aβ42 (μΜ)	IC50 Aβtotal (μΜ)
246	0.02	7
247	0.01	6
248	0.008	1
249	0.49	8
250	0.05	>10
251	0.02	6
252	0.01	5
253	0.009	5
254	0.02	6
255	0.01	3
256	0.04	4
257	0.04	>10
258	0.02	9
259	0.02	6
260	0.03	10
261	0.06	>10
262	0.02	>10
263	n.d.	n.d.

("n.d." means not determined)

### B) Demonstration of In Vivo Efficacy

#### $B-1)A\beta 42$

 $A\beta42$  lowering agents of the invention can be used to treat  $_{25}$  AD in mammals such as humans or alternatively demonstrating efficacy in animal models such as, but not limited to, the mouse, rat, or guinea pig. The mammal may not be diagnosed with AD, or may not have a genetic predisposition for AD, but may be transgenic such that it overproduces and eventually 30 deposits  $A\beta$  in a manner similar to that seen in humans afflicted with AD.

 $A\beta42$  lowering agents can be administered in any standard form using any standard method. For example, but not limited to,  $A\beta42$  lowering agents can be in the form of liquid, tablets 35 or capsules that are taken orally or by injection.  $A\beta42$  lowering agents can be administered at any dose that is sufficient to significantly reduce levels of  $A\beta42$  in the blood, blood plasma, serum, cerebrospinal fluid (CSF), or brain.

To determine whether acute administration of an A $\beta$ 42 40 lowering agent would reduce A $\beta$ 42 levels in vivo, non-transgenic rodents, e.g. mice or rats were used. Animals treated with the A $\beta$ 42 lowering agent were examined and compared to those untreated or treated with vehicle and brain levels of soluble A $\beta$ 42 and total A $\beta$  were quantitated by standard techniques, for example, using ELISA. Treatment periods varied from hours (h) to days and were adjusted based on the results of the A $\beta$ 42 lowering once a time course of onset of effect could be established.

A typical protocol for measuring A $\beta$ 42 lowering in vivo is shown but it is only one of many variations that could be used to optimize the levels of detectable A $\beta$ . For example, A $\beta$ 42 lowering compounds were formulated in 20% of Captisol® (a sulfobutyl ether of  $\beta$ -cyclodextrin) in water or 20% hydroxypropyl  $\beta$  cyclodextrin. The A $\beta$ 42 lowering agents were administered as a single oral dose or by any acceptable route of administration to overnight fasted animals. After 4 h, the animals were sacrificed and A $\beta$ 42 levels were analysed.

Blood was collected by decapitation and exsanguinations in EDTA-treated collection tubes. Blood was centrifuged at 60 1900 g for 10 minutes (min) at 4° C. and the plasma recovered and flash frozen for later analysis. The brain was removed from the cranium and hindbrain. The cerebellum was removed and the left and right hemisphere were separated. The left hemisphere was stored at –18° C. for quantitative 65 analysis of test compound levels. The right hemisphere was rinsed with phosphate-buffered saline (PBS) buffer and

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immediately frozen on dry ice and stored at -80° C. until homogenization for biochemical assays.

Mouse brains from non-transgenic animals were resuspended in 8 volumes of 0.4% DEA (diethylamine)/50 mM NaCl containing protease inhibitors (Roche-11873580001 or 04693159001) per gram of tissue, e.g. for 0.158 g brain, add 1.264 ml of 0.4% DEA. All samples were homogenized in the FastPrep-24 system (MP Biomedicals) using lysing matrix D (MPBio #6913-100) at 6 m/s for 20 seconds. Homogenates were centrifuged at 221.300×g for 50 min. The resulting high speed supernatants were then transferred to fresh eppendorf tubes. Nine parts of supernatant were neutralized with 1 part 0.5 M Tris-HCl pH 6.8 and used to quantify Aβtotal and Aβ42.

To quantify the amount of Aβtotal and Aβ42 in the soluble fraction of the brain homogenates, Enzyme-Linked-Immunosorbent-Assays were used. Briefly, the standards (a dilution of synthetic A $\beta$ 1-40 and A $\beta$ 1-42, Bachem) were prepared in 1.5 ml Eppendorf tube in Ultraculture, with final concentra-20 tions ranging from 10000 to 0.3 pg/ml. The samples and standards were co-incubated with HRPO-labelled N-terminal antibody for A $\beta$ 42 detection and with the biotinylated middomain antibody 4G8 for Aßtotal detection. 50 µl of conjugate/sample or conjugate/standards mixtures were then added to the antibody-coated plate (the capture antibodies selectively recognize the C-terminal end of Aβ42, antibody JRF/  $cA\beta42/26$ , for  $A\beta42$  detection and the N-terminus of  $A\beta$ , antibody JRF/rAβ/2, for Aβtotal detection). The plate was allowed to incubate overnight at 4° C. in order to allow formation of the antibody-amyloid complex. Following this incubation and subsequent wash steps the ELISA for Aβ42 quantification was finished by addition of Quanta Blu fluorogenic peroxidase substrate according to the manufacturer's instructions (Pierce Corp., Rockford, Ill.). A reading was performed after 10 to 15 min (excitation 320 nm/emission

For A $\beta$ total detection, a Streptavidine-Peroxidase-Conjugate was added, followed 60 min later by an additional wash step and addition of Quanta Blu fluorogenic peroxidase substrate according to the manufacturer's instructions (Pierce Corp., Rockford, Ill.). A reading was performed after 10 to 15 min (excitation 320 nm/emission 420 nm).

In this model a  $A\beta42$  lowering compared to untreated animals would be advantageous, in particular a  $A\beta42$  lowering with at least 10%, more in particular a  $A\beta42$  lowering with at least 20%.

### B-2) Aβ38

 $A\beta38$  increasing agents of the invention can be used to treat AD in mammals such as humans or alternatively demonstrating efficacy in animal models such as, but not limited to, the mouse, rat, or guinea pig. The mammal may not be diagnosed with AD, or may not have a genetic predisposition for AD, but may be transgenic such that it overproduces and eventually deposits  $A\beta$  in a manner similar to that seen in humans afflicted with AD.

 $A\beta38$  increasing agents can be administered in any standard form using any standard method. For example, but not limited to,  $A\beta38$  increasing agents can be in the form of liquid, tablets or capsules that are taken orally or by injection.  $A\beta38$  increasing agents can be administered at any dose that is sufficient to significantly increase levels of  $A\beta38$  in the blood, plasma, serum, cerebrospinal fluid (CSF), or brain.

To determine whether acute administration of an A $\beta$ 38 increasing agents would increase A $\beta$ 38 levels in vivo, nontransgenic rodents, e.g. mice or rats were used. Animals treated with the A $\beta$ 38 increasing agents were examined and compared to those untreated or treated with vehicle and brain

levels of soluble A $\beta$ 38 and total A $\beta$  were quantitated by standard techniques, for example, using ELISA. Treatment periods varied from hours (h) to days and were adjusted based on the results of the A $\beta$ 38 increase once a time course of onset of effect could be established.

A typical protocol for measuring A $\beta$ 38 increase in vivo is shown but it is only one of many variations that could be used to optimize the levels of detectable A $\beta$ . For example, A $\beta$ 38 increasing agents were formulated in 20% of Captisol® (a sulfobutyl ether of  $\beta$ -cyclodextrin) in water or 20% hydroxypropyl  $\beta$ cyclodextrin. The A $\beta$ 38 increasing agents were administered as a single oral dose or by any acceptable route of administration to overnight fasted animals. After 4 h, the animals were sacrificed and A $\beta$ 38 levels were analysed.

Blood was collected by decapitation and exsanguinations in EDTA-treated collection tubes. Blood was centrifuged at 1900 g for 10 minutes (min) at 4° C. and the plasma recovered and flash frozen for later analysis. The brain was removed from the cranium and hindbrain. The cerebellum was 20 removed and the left and right hemisphere were separated. The left hemisphere was stored at –18° C. for quantitative analysis of test compound levels. The right hemisphere was rinsed with phosphate-buffered saline (PBS) buffer and immediately frozen on dry ice and stored at –80° C. until 25 homogenization for biochemical assays.

Mouse brains from non-transgenic animals were resuspended in 8 volumes of 0.4% DEA (diethylamine)/50 mM NaCl containing protease inhibitors (Roche-11873580001 or 04693159001) per gram of tissue, e.g. for 0.158 g brain, add 1.264 ml of 0.4% DEA. All samples were homogenized in the FastPrep-24 system (MP Biomedicals) using lysing matrix D (MPBio #6913-100) at 6 m/s for 20 seconds. Homogenates were centrifuged at 221.300×g for 50 min. The resulting high speed supernatants were then transferred to fresh eppendorf tubes. Nine parts of supernatant were neutralized with 1 part 0.5 M Tris-HCl pH 6.8 and used to quantify  $\Delta\beta$  total and  $\Delta\beta38$ .

To quantify the amount of A $\beta$ total and A $\beta$ 38 in the soluble  $_{40}$ fraction of the brain homogenates, Enzyme-Linked-Immunosorbent-Assays were used. Briefly, the standards (a dilution of synthetic Aβ1-40 and Aβ1-38, ANASPEC) were prepared in 1.5 ml Eppendorf tube in Ultraculture, with final concentrations ranging from 10000 to 0.3 pg/ml. The samples and 45 standards were co-incubated with HRPO-labelled N-terminal antibody for Aβ38 detection and with the biotinylated middomain antibody 4G8 for Aßtotal detection. 50 µl of conjugate/sample or conjugate/standards mixtures were then added to the antibody-coated plate (the capture antibodies selec- 50 tively recognize the C-terminal end of Aβ38, antibody J&JPRD/Aβ38/5, for Aβ38 detection and the N-terminus of A $\beta$ , antibody JRF/rA $\beta$ /2, for A $\beta$ total detection). The plate was allowed to incubate overnight at 4° C. in order to allow formation of the antibody-amyloid complex. Following this 55 incubation and subsequent wash steps the ELISA for Aβ38 quantification was finished by addition of Quanta Blu fluorogenic peroxidase substrate according to the manufacturer's instructions (Pierce Corp., Rockford, Ill.). A reading was performed after 10 to 15 min (excitation 320 nm/emission 60

For Aβtotal detection, a Streptavidine-Peroxidase-Conjugate was added, followed 60 min later by an additional wash step and addition of Quanta Blu fluorogenic peroxidase substrate according to the manufacturer's instructions (Pierce 65 Corp., Rockford, Ill.). A reading was performed after 10 to 15 min (excitation 320 nm/emission 420 nm).

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In this model a A $\beta$ 38 increase compared to untreated animals would be advantageous, in particular a A $\beta$ 38 increase with at least 10%, more in particular a A $\beta$ 38 increase with at least 20%.

#### B-3) Results

The results are shown in Table 4 (dose 30 mg/kg oral dosing) (value for untreated animals as control (Ctrl) was set at 100):

	Co. No.	Aβ38 (% vs Ctrl)_Mean	Aβ42 (% vs Ctrl)_Mean	Aβtotal (% vs Ctrl)_Mean
	6	129	85	102
	7	80	73	94
5	10a	85	87	118
	63	133	82	124
	59	126	82	126
	65	132	104	114
	3	137	55	98
	99	92	111	113
)	22	105	106	92
	2	129	82	100
	37	83	76	98
	5	110	78	87
	23	132	122	122
	1	103	61	89
	62	160	68	98
	46	62	60	87
	92	115	94	96
	78	106	92	98
	50	110	88	91
	80	118	119	119
)	82	120	78	125

B-1b) Aβ42

 $A\beta42$  lowering agents of the invention can be used to treat AD in mammals such as humans or alternatively demonstrating efficacy in animal models such as, but not limited to, the mouse, rat, or guinea pig. The mammal may not be diagnosed with AD, or may not have a genetic predisposition for AD, but may be transgenic such that it overproduces and eventually deposits  $A\beta$  in a manner similar to that seen in humans afflicted with AD.

 $A\beta42$  lowering agents can be administered in any standard form using any standard method. For example, but not limited to,  $A\beta42$  lowering agents can be in the form of liquid, tablets or capsules that are taken orally or by injection.  $A\beta42$  lowering agents can be administered at any dose that is sufficient to significantly reduce levels of  $A\beta42$  in the blood, blood plasma, serum, cerebrospinal fluid (CSF), or brain.

To determine whether acute administration of an A $\beta$ 42 lowering agent would reduce A $\beta$ 42 levels in vivo, non-transgenic rodents, e.g. mice or rats were used. Animals treated with the A $\beta$ 42 lowering agent were examined and compared to those untreated or treated with vehicle and brain levels of soluble A $\beta$ 42, A $\beta$ 40, A $\beta$ 38, and A $\beta$ 37 were quantitated by Meso Scale Discovery's (MSD) electrochemiluminescence detection technology. Treatment periods varied from hours (h) to days and were adjusted based on the results of the A $\beta$ 42 lowering once a time course of onset of effect could be established.

A typical protocol for measuring A $\beta$ 42 lowering in vivo is shown but it is only one of many variations that could be used to optimize the levels of detectable A $\beta$ . For example, A $\beta$ 42 lowering compounds were formulated in 20% of Captisol® (a sulfo-butyl ether of  $\beta$ -cyclodextrin) in water or 20% hydroxypropyl  $\beta$ 6cyclodextrin. The A $\beta$ 42 lowering agents were administered as a single oral dose or by any acceptable route of administration to overnight fasted animals. After 4 h, the animals were sacrificed and A $\beta$ 42 levels were analysed.

Blood was collected by decapitation and exsanguinations in EDTA-treated collection tubes. Blood was centrifuged at 1900 g for 10 minutes (min) at 4° C. and the plasma recovered and flash frozen for later analysis. The brain was removed from the cranium and hindbrain. The cerebellum was removed and the left and right hemisphere were separated. The left hemisphere was stored at –18° C. for quantitative analysis of test compound levels. The right hemisphere was rinsed with phosphate-buffered saline (PBS) buffer and immediately frozen on dry ice and stored at –80° C. until homogenization for biochemical assays.

Mouse brains from non-transgenic animals were resuspended in 8 volumes of 0.4% DEA (diethylamine)/50 mM NaCl containing protease inhibitors (Roche-11873580001 or 04693159001) per gram of tissue, e.g. for 0.158 g brain, add 1.264 ml of 0.4% DEA. All samples were homogenized in the FastPrep-24 system (MP Biomedicals) using lysing matrix D (MPBio #6913-100) at 6 m/s for 20 seconds. Homogenates were centrifuged at 20800×g for 5 min and supernatants collected. Supernatants were centrifuged at 221.300×g for 50 min. The resulting high speed supernatants were then transferred to fresh eppendorf tubes. Nine parts of supernatant were neutralized with 1 part 0.5 M Tris-HCl pH 6.8 and used to quantify Aβ.

To quantify the amount of  $A\beta42$ ,  $A\beta40$ ,  $A\beta38$ , and  $A\beta37$  in the soluble fraction of the brain homogenates, simultaneous specific detection of Aβ42, Aβ40, Aβ38, and Aβ37 was performed using MSD's electro-chemiluminescence multiplex detection technology. In this assay purified monoclonal anti- 30 bodies specific for Abeta37 (JRD/Aβ37/3), Abeta38 (J&JPRD/Aβ38/5), Abeta40 (JRF/cAβ40/28), and Abeta42 (JRF/cAβ42/26) were coated on MSD 4-plex plates. Briefly, the standards (a dilution of synthetic Aβ42, Aβ40, Aβ38, and Aβ37) were prepared in 1.5 ml Eppendorf tube in Ultracul- 35 ture, with final concentrations ranging from 10000 to 0.3 pg/m. The samples and standards were co-incubated with Sulfo-tag labelled JRF/rAβ/2 antibody to the N-terminus of Aβ as detector antibody. 50 μl of conjugate/sample or conjugate/standards mixtures were then added to the antibody- 40 coated plate. The plate was allowed to incubate overnight at 4° C. in order to allow formation of the antibody-amyloid complex. Following this incubation and subsequent wash steps the assay was finished by adding read buffer according to the manufacturer's instructions (Meso Scale Discovery, 45 Gaitherburg, Md.).

The SULFO-TAG emits light upon electrochemical stimulation initiated at the electrode. MSD Sector instrument SI6000 was used for signal read-out.

In this model a A $\beta$ 42 lowering compared to untreated 50 animals would be advantageous, in particular a A $\beta$ 42 lowering with at least 10%, more in particular a A $\beta$ 42 lowering with at least 20%.

B-2b) Aβ38

 $A\beta38$  increasing agents of the invention can be used to treat  $\,^{55}$  AD in mammals such as humans or alternatively demonstrating efficacy in animal models such as, but not limited to, the mouse, rat, or guinea pig. The mammal may not be diagnosed with AD, or may not have a genetic predisposition for AD, but may be transgenic such that it overproduces and eventually  $\,^{60}$  deposits  $\,^{60}$  in a manner similar to that seen in humans afflicted with AD.

 $A\beta38$  increasing agents can be administered in any standard form using any standard method. For example, but not limited to,  $A\beta38$  increasing agents can be in the form of 65 liquid, tablets or capsules that are taken orally or by injection.  $A\beta38$  increasing agents can be administered at any dose that

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is sufficient to significantly increase levels of A $\beta$ 38 in the blood, plasma, serum, cerebrospinal fluid (CSF), or brain.

To determine whether acute administration of an A $\beta$ 38 increasing agents would increase A $\beta$ 38 levels in vivo, nontransgenic rodents, e.g. mice or rats were used. Animals treated with the A $\beta$ 38 increasing agents were examined and compared to those untreated or treated with vehicle and brain levels of soluble A $\beta$ 42, A $\beta$ 40, A $\beta$ 38, and A $\beta$ 37 were quantitated by MSD electrochemiluminescence detection technology. Treatment periods varied from hours (h) to days and were adjusted based on the results of the A $\beta$ 38 increase once a time course of onset of effect could be established.

A typical protocol for measuring A $\beta$ 38 increase in vivo is shown but it is only one of many variations that could be used to optimize the levels of detectable A $\beta$ . For example, A $\beta$ 38 increasing agents were formulated in 20% of Captisol® (a sulfobutyl ether of  $\beta$ -cyclodextrin) in water or 20% hydroxypropyl  $\beta$ 29 cyclodextrin. The A $\beta$ 38 increasing agents were administered as a single oral dose or by any acceptable route of administration to overnight fasted animals. After 4 h, the animals were sacrificed and A $\beta$ 38 levels were analysed.

Blood was collected by decapitation and exsanguinations in EDTA-treated collection tubes. Blood was centrifuged at 1900 g for 10 minutes (min) at 4° C. and the plasma recovered and flash frozen for later analysis. The brain was removed from the cranium and hindbrain. The cerebellum was removed and the left and right hemisphere were separated. The left hemisphere was stored at –18° C. for quantitative analysis of test compound levels. The right hemisphere was rinsed with phosphate-buffered saline (PBS) buffer and immediately frozen on dry ice and stored at –80° C. until homogenization for biochemical assays.

Mouse brains from non-transgenic animals were resuspended in 8 volumes of 0.4% DEA (diethylamine)/50 mM NaCl containing protease inhibitors (Roche-11873580001 or 04693159001) per gram of tissue, e.g. for 0.158 g brain, add 1.264 ml of 0.4% DEA. All samples were homogenized in the FastPrep-24 system (MP Biomedicals) using lysing matrix D (MPBio #6913-100) at 6 m/s for 20 seconds. Homogenates were centrifuged at 20800×g for 5 min and supernatants collected. Supernatants were centrifuged at 221.300×g for 50 min. The resulting high speed supernatants were then transferred to fresh eppendorf tubes. Nine parts of supernatant were neutralized with 1 part 0.5 M Tris-HCl pH 6.8 and used to quantify  ${\rm A}{\beta}$ .

To quantify the amount of Aβ42, Aβ40, Aβ38, and Aβ37 in the soluble fraction of the brain homogenates, simultaneous specific detection of Aβ42, Aβ40, Aβ38, and Aβ37 was performed using MSD's electro-chemiluminescence multiplex detection technology. In this assay purified monoclonal antibodies specific for Abeta37 (JRD/Aβ37/3), Abeta38 (J&JPRD/Aβ38/5), Abeta40 (JRF/cAβ40/28), and Abeta42 (JRF/cAβ42/26) were coated on MSD 4-plex plates. Briefly, the standards (a dilution of synthetic A $\beta$ 42, A $\beta$ 40, A $\beta$ 38, and Aβ37) were prepared in 1.5 ml Eppendorf tube in Ultraculture, with final concentrations ranging from 10000 to 0.3 pg/m. The samples and standards were co-incubated with Sulfo-tag labelled JRF/rAβ/2 antibody to the N-terminus of Aβ as detector antibody. 50 μl of conjugate/sample or conjugate/standards mixtures were then added to the antibodycoated plate. The plate was allowed to incubate overnight at 4° C. in order to allow formation of the antibody-amyloid complex. Following this incubation and subsequent wash steps the assay was finished by adding read buffer according to the manufacturer's instructions (MesoScale Discovery, Gaitherburg, Md.).

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The SULFO-TAG emits light upon electrochemical stimulation initiated at the electrode. MSD Sector instrument SI6000 was used for signal read-out.

In this model a  $A\beta38$  increase compared to untreated animals would be advantageous, in particular a  $A\beta38$  increase with at least 10%, more in particular a  $A\beta38$  increase with at least 20%.

### B-3b) Results

The results are shown in Table 5 (dose 30 mg/kg oral dosing) (value for untreated animals as control (Ctrl) was set at 100):

	Αβ40	Αβ42	Αβ38
Co. No.	(% vs Ctrl)_Mean	(% vs Ctrl)_Mean	(% vs Ctrl)_Mean
3	85	51	133
92	100	87	124
6	108	73	118
78	114	89	100
50	112	99	160
80*	164	148	133
82	125	103	118
110	103	88	102
111	52	35	136
112	82	55	153
116	105	75	182
117	84	66	154
118	115	92	140
166	107	64	127
164	97	80	185
158	98	97	120
153	128	100	108
151	57	41	153
145	92	71	102
143	129	123	122
141	122	85	145
140	71	42	96
134	115	97	138
133	77	87	84
132	121	104	84
130	114	67	182
128	116	60	162
189	118	116	116
192	44	39	147
194	43	39	197
183	133	97	141
185	89	68	170
195	135	105	123
203	137	67	133
219	95	76	119

<sup>\*</sup>For compound 80 the test was not fully validated.

### Composition Examples

"Active ingredient" (a.i.) as used throughout these examples relates to a compound of Formula (I), including any 50 tautomer or stereoisomeric form thereof, or a pharmaceutically acceptable addition salt or a solvate thereof; in particular to any one of the exemplified compounds.

Typical examples of recipes for the formulation of the invention are as follows:

### 1. Tablets

Active ingredient	5 to 50 mg	_
Di-calcium phosphate	20 mg	
Lactose	30 mg	60
Talcum	10 mg	
Magnesium stearate	5 mg	
Potato starch	ad 200 mg	

### 2. Suspension

An aqueous suspension is prepared for oral administration so that each milliliter contains 1 to 5 mg of active ingredient, 50 mg of sodium carboxymethyl cellulose, 1 mg of sodium benzoate, 500 mg of sorbitol and water ad 1 ml.

### 3. Injectable

A parenteral composition is prepared by stirring 1.5% (weight/volume) of active ingredient in 0.9% NaCl solution or in 10% by volume propylene glycol in water.

### 4. Ointment

Active ingredient	5 to 1000	mg
Stearyl alcohol	3	g
Lanoline	5	g
White petroleum	15	g
Water	ad 100	g

In this Example, active ingredient can be replaced with the same amount of any of the compounds according to the present invention, in particular by the same amount of any of the exemplified compounds.

The invention claimed is:

1. A compound of Formula (I)

35 wherein

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 $R^1$  is hydrogen,  $C_{1-4}$ alkyl or Ar; provided however that  $R^1$  is  $C_{1-4}$ alkyl

provided however that  $R^1$  is  $C_{1-4}$ alkyl or Ar when L is a covalent bond;

- $\rm R^2$  is hydrogen, phenyl, cycloC $_{3-7}$ alkyl, tetrahydro-2H-pyranyl, tetrahydro-2H -thiopyranyl, piperidinyl, or  $\rm C_{1-4}$ alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, hydroxy,  $\rm C_{1-4}$ alkyloxy and NR $^7\rm R^8$ ;
- Z is methylene or 1,2-ethanediyl, wherein methylene or 1,2-ethanediyl is optionally substituted with one or two  $C_{1-4}$ alkyl substituents;
- L is a covalent bond, — $(CH_2)_{m-n}$ —Y— $(CH_2)_n$ —, 1,2-cyclopropanediyl,  $C_{1-6}$ alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  $C_{1-4}$ alkyloxy $C_{1-4}$ alkyl, or  $C_{1-6}$ alkanediyl wherein two geminal hydrogen atoms are replaced by  $C_{2-6}$ alkanediyl;

m is 3, 4, 5, 6 or 7;

n is 1, 2 or 3;

Y is O or NH;

Ar is a ring system selected from the group consisting of phenyl, indolyl, oxazolyl, benzo[b]thienyl, 1,3-benzodioxolyl, 1,4-benzodioxanyl, 3,4-dihydro-2H-1,5-benzodioxepinyl, pyridyl, indanyl, 1,2,3,4-tetrahydro-1-naphthalenyl, and naphthalenyl;

wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>, C<sub>1.4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1.4</sub>alkyloxy and cycloC<sub>3.7</sub>alkyl, and C<sub>1.4</sub>alkyloxy

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optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl;

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and C<sub>1-4</sub>alkyl optionally substituted with one or more halo substituents;

 $R^3$  is hydrogen, halo or  $C_{1-4}$ alkyl;

R<sup>4</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

R<sup>5</sup> is hydrogen or C<sub>1-4</sub>alkyl;

X is CR<sup>6</sup> or N;

 $R_{-}^{6}$  is hydrogen or  $C_{1-4}$ alkyl;

R<sup>7</sup> is hydrogen or C<sub>1-4</sub>alkyl;

R<sup>8</sup> is hydrogen or C<sub>1-4</sub>alkyl; and,

 $R^{\rm o}$  is a ring system selected from the group consisting of 15 piperidinyl, morpholinyl, cycloC<sub>3-7</sub>alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or C<sub>1-4</sub>alkyl optionally substituted with one or more halo 20 atoms;

or a tautomer, stereoisomer or pharmaceutically acceptable addition salt thereof.

2. The compound according to claim 1, wherein

R<sup>2</sup> is hydrogen, phenyl, cycloC<sub>3-7</sub>alkyl, tetrahydro-2H- pyranyl, or C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of hydroxy, C<sub>1-4</sub>alkyloxy and NR<sup>7</sup>R<sup>8</sup>;

m is 3;

n is 2;

Y is 0;

R<sup>3</sup> is hydrogen or halo;

R<sup>4</sup> is hydrogen, halo or C<sub>1-4</sub>alkyl;

 $R^8$  is  $C_{1-4}$ alkyl; and,

 $R^0$  is a ring system selected from the group consisting of 35 piperidinyl, morpholinyl, cyclo $C_{3-7}$ alkyl, pyrazolyl and pyrrolidinyl; wherein said ring system is optionally substituted with one or more  $C_{1-4}$ alkyl groups optionally substituted with one or more halo atoms.

- 3. The compound according to claim 1, wherein  $R^1$  is Ar. 40
- **4**. The compound according to claim **1**, wherein X is CH.
- 5. The compound according to claim 1, wherein n is 1 or 2.
- 6. The compound according to claim 1, wherein

R<sup>1</sup> is Ar;

R<sup>2</sup> is hydrogen or C<sub>1-4</sub>alkyl;

Z is methylene;

L is a  $C_{1-6}$ alkanediyl;

Ar is a ring system selected from the group consisting of phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, Ar<sup>2</sup>, R<sup>0</sup>, C<sub>1-4</sub>alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo, C<sub>1-4</sub>alkyloxy and cycloC<sub>3-7</sub>alkyl, and C<sub>1-4</sub>alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and cycloC<sub>3-7</sub>alkyl:

Ar<sup>2</sup> is phenyl optionally substituted with one or more substituents each independently selected from the group consisting of halo and  $C_{j-4}$ alkyl optionally substituted 60 with one or more halo substituents;

R<sup>3</sup> is hydrogen;

 $R^4$  is  $C_{1-4}$ alkyl;

R<sup>5</sup> is hydrogen;

X is CH; and,

R<sup>0</sup> is a ring system selected from the group consisting of piperidinyl, morpholinyl, cycloC<sub>3.7</sub>alkyl, pyrazolyl and

pyrrolidinyl; wherein said ring system is optionally substituted with one or more substituents each independently selected from the group consisting of halo or  $C_{1-4}$ alkyl optionally substituted with one or more halo atoms.

7. The compound according to claim 1, wherein

L is a covalent bond, 1,2-cyclopropanediyl,  $C_{1-6}$ alkanediyl optionally substituted with one or more substituents selected from the group consisting of tetrahydro-2H-pyranyl, phenyl and  $C_{1-4}$ alkyloxy $C_{1-4}$ alkyl, or  $C_{1-6}$ alkanediyl wherein two geminal hydrogen atoms are replaced by  $C_{2-6}$ alkanediyl.

8. The compound according to claim 1, wherein

L is a covalent bond or  $C_{1-6}$ alkanediyl;

R<sup>1</sup> is Ar; and,

Ar is indolyl optionally substituted with one or more substituents each independently selected from the group consisting of halo,  $\operatorname{Ar}^2$ ,  $\operatorname{R}^0$ ,  $\operatorname{C}_{1-4}$ alkyl optionally substituted with one or more substituents each independently selected from the group consisting of halo,  $\operatorname{C}_{1-4}$ alkyloxy and  $\operatorname{cycloC}_{3-7}$ alkyl, and  $\operatorname{C}_{1-4}$ alkyloxy optionally substituted with one or more substituents each independently selected from the group consisting of halo and  $\operatorname{cycloC}_{3-7}$ alkyl.

**9**. The compound according to claim **1**, wherein  $R^2$  is  $C_{1-4}$ alkyl and L is  $C_{1-6}$ alkanediyl.

10. The compound according to claim 1, wherein the position of  $\mathbb{R}^3$  is fixed as shown in formula (I-x)

$$R^{1}$$
 $R^{2}$ 
 $Z$ 
 $N$ 
 $N$ 
 $X$ 
 $R^{4}$ .

11. The compound according to claim 1, wherein the compound is selected from the group consisting of

2-[1-[1-ethyl-5-(trifluoromethyl)-1H-indol-3-yl]ethyl]-3, 4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione, and,

2-[1-(5-chloro-1-ethyl-7-fluoro-1H-indol-3-yl)ethyl]-3,4-dihydro-3-methyl-7-(4-methyl-1H-imidazol-1-yl)-2H-pyrido[1,2-a]pyrazine-1,6-dione, or a tautomer, stereoisomer or pharmaceutically acceptable addition salt thereof.

12. A pharmaceutical composition comprising a therapeutically effective amount of a compound according to claim 1 and a pharmaceutically acceptable carrier.

13. A method for treating a disease or condition in a subject selected from the group consisting of Alzheimer's disease, traumatic brain injury, mild cognitive impairment, senility, dementia, dementia with Lewy bodies, cerebral amyloid angiopathy, multi-infarct dementia, dementia pugilistica, Down's syndrome, dementia associated with Parkinson's disease and dementia associated with beta-amyloid, comprising administering to said subject a therapeutically effective amount of a compound according to claim 1.

14. The method according to claim 13, wherein the disease is Alzheimer's disease.

\* \* \* \* \*